

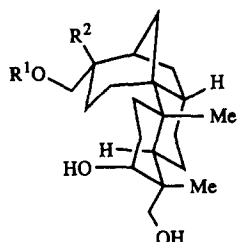
Enantioselective Total Synthetic Route to (+)-Aphidicolin

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Abstract: Enantioselective total synthetic route to (+)-aphidicolin (1) has been described. Cycloisomerization reaction (9→10) and intramolecular Diels-Alder reaction (12→13) were employed for the key steps of the sequence.

In 1972, Hesp and co-workers at ICI reported the isolation of a tetracyclic diterpenoid antibiotic, aphidicolin (1), from the fungus *Cephalosporium aphidicola* Petch.¹ Aphidicolin (1) displays marked activity against Herpes simplex type I virus in cultures of human embryonic cells.² Apart from its antifeedant property,³ 1 exhibits considerable antitumor activity in the C6 mouse colon and B16 mouse melanoma screens⁴ and has been shown to inhibit the growth of leukemic T- and B-lymphocytes.⁵ The antitumor activity of aphidicolin (1) observed against rodent tumors has stimulated interest in clinical investigation to the effectiveness of 1 against human tumors. Although the development of 1 as an antitumor agent has been hampered by its poor water solubility, a recent report of enhanced antitumor activity associated with the more water-soluble compounds such as aphidicolin-17-glycinate HCl salt (2)⁶ and 16-fluoroaphidicolin (3),⁷ synthesized as pro-drugs, might revive interest in aphidicolin (1) and its analogues as a specific reversible inhibitor of DNA polymerase α .



$R^1 = H, R^2 = OH$; Aphidicolin (1)

$R^1 = COCH_2NH_2 \cdot HCl, R^2 = OH$; Aphidicolin-17-glycinate HCl salt (2)

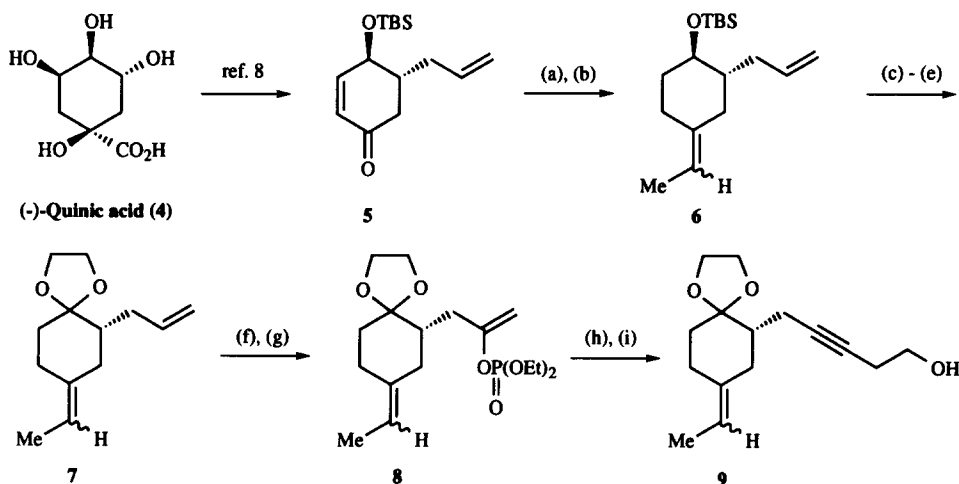
$R^1 = H, R^2 = F$; 16-Fluoroaphidicolin (3)

The physiological properties coupled with its structural complexity have made aphidicolin (1) an attractive synthetic target. 1 incorporates three structural features which have posed long-standing synthetic problems: (i) a spiro fused bicyclo[3.2.1]octane moiety which comprises the C and D rings, (ii) adjacent quaternary stereogenic centers (C9 and C10), and A, B-*trans* ring juncture.

Herein, we describe an enantioselective total synthetic route to (+)-aphidicolin (**1**) featuring methodologies designed to address the outstanding issues confronted during the enantioselective total synthesis of (+)-**1**, viz (a) cycloisomerization reaction for the construction of CD ring system, and (b) intramolecular Diels-Alder reaction producing A, B-*trans* ring juncture.

First of all, the readily available optically pure enone (**5**), prepared from (-)-quinic acid (**4**) by using Overman's protocol,⁸ was converted to the alcohol (**9**), as shown in Scheme I. Namely, regioselective reduction of **5** with sodium hydrosulfite⁹ in the presence of Adogen 464[®] afforded the corresponding ketone (71%), which was subjected to Wittig reaction, furnishing the ethylidene derivative (**6**)¹⁰ in 89% yield. Deprotection (83%) of the TBS group, followed by oxidation¹¹ with tetrapropylammonium perruthenate (TPAP) in the presence of N-methylmorpholin N-oxide (NMO) yielded, in 91% yield, the ketone, which was transformed into **7** after ketalization (100%). The conversion of **7** to the phosphonate (**8**) was achieved in two steps, including Wacker oxidation¹² (88%) and enolization of the resulting methyl ketone under kinetic conditions. Upon treatment of **8** with LDA, the corresponding acetylene was produced in 86% yield. Alkylation of lithium salt of the above acetylene was conducted with ethylene oxide by means of Kotsuki's procedure¹³ to give rise to the alcohol (**9**) in 87% yield.

Scheme I



Reaction Conditions: (a) $\text{Na}_2\text{S}_2\text{O}_4$, NaHCO_3 , Adogen 464[®], C_6H_6 - H_2O (1 : 1 v/v), reflux. (b) $\text{Ph}_3\text{P}^+\text{EtBr}^-$, $^n\text{BuLi}$, THF, reflux. (c) $^n\text{Bu}_4\text{N}^+\text{F}^-$, THF. (d) TPAP, NMO, 4 Å MS, CH_2Cl_2 . (e) ethylene glycol, PPTS, C_6H_6 , reflux. (f) O_2 , PdCl_2 , CuCl , H_2O -DMF (1 : 4 v/v), 45 °C. (g) LDA, THF, -78 °C; ClP(O)(OEt)_2 . (h) LDA, THF, -78 °C $\rightarrow$ 0 °C. (i) $^n\text{BuLi}$, THF, -78 °C; ethylene oxide, $\text{BF}_3 \cdot \text{Et}_2\text{O}$.

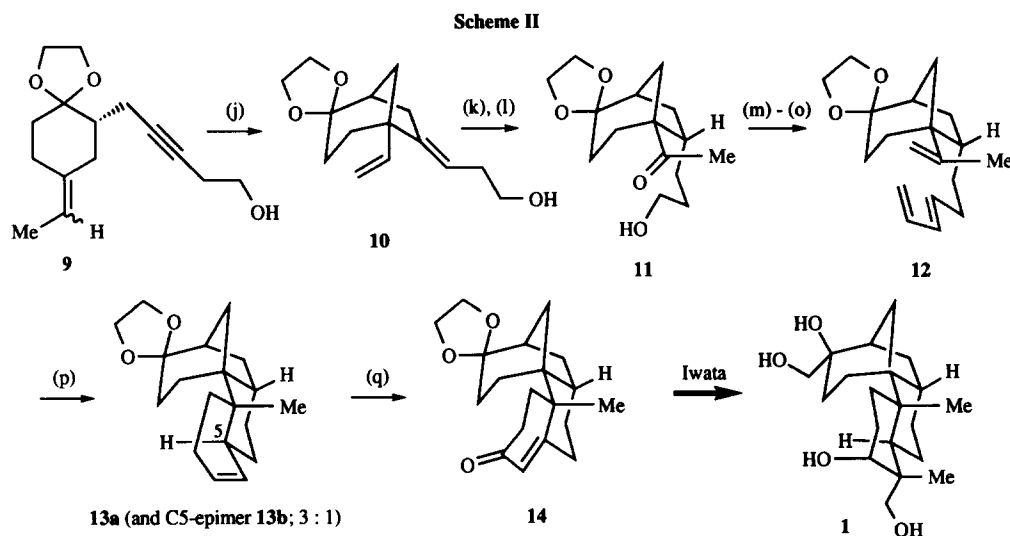
With **9** in hand, the crucial cycloisomerization reaction¹⁴ for the construction of the CD ring system of aphidicolin (**1**) was examined. As a result of testing, the cycloisomerization of **9** in C_6H_6 in the presence of $(\text{dba})_3\text{Pd}_2 \cdot \text{CHCl}_3$ (2.5% mol), $\text{P}(o\text{-tolyl})_3$ (5.0% mol) and AcOH (5.0% mol) proceeded quite nicely to furnish the olefinic alcohol (**10**), $[\alpha]_{\text{D}}^{23} -81.88^\circ$ ($c = 1.02$, CHCl_3), in 76% yield.

Our synthetic efforts were next focused on the introduction of diene and dienophile portions for the second key step. The requisite triene (**12**) was prepared as describe below. Wacker oxidation at 60 °C (69%)

followed by catalytic hydrogenation of the resulting olefinic ketone gave the keto alcohol (**11**), $[\alpha]_{\text{D}}^{25} -19.25^\circ$ ($c = 0.267$, CHCl_3), in 52% yield. Wittig olefination (62%) of **11** followed by oxidation with PCC in the presence of NaOAc of the resulting alcohol led to the corresponding aldehyde (85%), which was subjected to Yamamoto reaction¹⁵ (73%) to give **12** $\{[\alpha]_{\text{D}}^{25} -7.96^\circ$ ($c = 0.470$, CHCl_3) $\}$.

With the efficient synthesis of the triene (**12**) realized, the stage was now step for the construction of aphidicolane-type ring system. An intramolecular Diels-Alder reaction of **12** was performed in toluene at 220°C for 11 days in a sealed tube to provide the tetracyclic compounds (**13**), in 70% yield as a 3 : 1 diastereomeric mixture at C5,¹⁶ which was used directly in next step without separation.

Finally the pivotal transformation of **13** into the enone (**14**), $[\alpha]_{\text{D}}^{25} +91.57^\circ$ ($c = 0.375$, CHCl_3), was conducted, 55% overall yield for 3 steps, under standard conditions¹⁷ (Scheme II).



Reaction Conditions: (j) $(\text{dba})_3\text{Pd}_2\text{-CHCl}_3$, $\text{P}(o\text{-tolyl})_3$, AcOH , C_6H_6 , 60°C , reflux, tube. (k) O_2 , PdCl_2 , CuCl , $\text{H}_2\text{O-DMF}$ (1 : 4 v/v), 60°C . (l) H_2 , PtO_2 , EtOH . (m) $\text{Ph}_3\text{P}^+\text{MeBr}^-$, $^n\text{BuLi}$, DME, reflux. (n) PCC, NaOAc, Florisil®, CH_2Cl_2 . (o) $\text{Ph}_2\text{P}(\text{O})\text{CH}_2\text{CH=CH}_2$, $^n\text{BuLi}$, HMPA, THF, $-78^\circ\text{C} \rightarrow \text{room temperature}$. (p) 220°C , toluene, sealed tube. (q) O_2 , hv, pyridine, hematoporphyrin; NaI, AcOH , EtOH , Et_2O ; TPAP, NMO, $4 \text{ \AA MS}$, CH_2Cl_2 .

An important advantage of the present synthetic strategy is that each diastereoisomer (**13a** and **13b**), isomeric only at C5 position, affords the same enone (**14**),¹⁸ which displays the same spectra with those provided by Iwata and co-workers in a total synthesis of ($\pm$)-aphidicolin (**1**). Thus, the present enantioselective synthesis of **14** gives rise to the formal total synthetic route to (+)-aphidicolin (**1**).

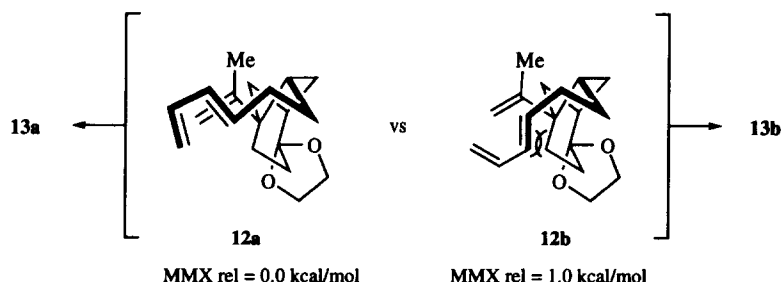
In conclusion, we have demonstrated enantioselective formal total synthetic route to (+)-aphidicolin (**1**) based on cycloisomerization reaction and intramolecular Diels-Alder reaction.

Acknowledgment

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References and Notes

1. Bundret, K. M.; Dalziel, W.; Hesp, B.; Jarvis, J. A. J.; Neidle, S. *J. Chem. Soc., Chem. Commun.*, 1972, 1027-1028. For total and formal syntheses of aphidicolin (1): Trost, B. M.; Nishimura, Y.; Yamamoto, K.; McElvain, S. S. *J. Am. Chem. Soc.*, 1979, **101**, 1328-1330; McMurry, J. E.; Andrus, A.; Ksander, G. M.; Musser, J. H.; Johnson, M. A. *J. Am. Chem. Soc.*, 1979, **101**, 1330-1332; *idem*, *Tetrahedron*, 1981, Suppl. 1, 319-327; Corey, E. J.; Tius, M. A.; Das J. *J. Am. Chem. Soc.*, 1980, **102**, 1742-1744; Ireland, R. E.; Godfrey, J. D.; Thaisrivongs, S. *ibid.*, 1981, **103**, 2446-2448; Ireland, R. E.; Dow, W. C.; Godfrey, J. D.; Thaisrivongs, S. *J. Org. Chem.*, 1984, **49**, 1001-1013; van Tamelen, E. E.; Zawacky, S. R.; Russell, R. K.; Carlson, J. G. *J. Am. Chem. Soc.*, 1983, **105**, 142-143; van Tamelen, E. E.; Zawacky, S. R. *Tetrahedron Lett.*, 1985, **26**, 2833-2836; Bettolo, R. M.; Tagliatesta, P.; Lupi, A.; Bravetti, D. *Helv. Chim. Acta*, 1983, **66**, 1922-1928; Lupi, A.; Patamia, M.; Bettolo, R. M. *ibid.*, 1988, **71**, 872-875; Tanis, S. P.; Chuang, Y.-H.; Head, D. B. *Tetrahedron Lett.*, 1985, **26**, 6147-6150; *idem*, *J. Org. Chem.*, 1988, **53**, 4929-4938; Holton, R. A.; Kennedy, R. M.; Kim, H.-B.; Krafft, M. E. *J. Am. Chem. Soc.*, 1987, **109**, 1597-1600; Toyota, M.; Nishikawa, Y.; Fukumoto, K. *Tetrahedron Lett.*, 1994, **35**, 6495-6498; Toyota, M.; Nishikawa, Y.; Seishi, T.; Fukumoto, K. *Tetrahedron*, 1994, **50**, 10183-10192; Toyota, M.; Nishikawa, Y.; Fukumoto, K. *ibid.*, 1994, **50**, 11153-11166; Tanaka, T.; Murakami, K.; Okuda, O.; Inoue, T.; Kuroda, T.; Kamei, K.; Murata, T.; Yoshino, H.; Imanishi, T.; Kim, S.-W.; Iwata, C. *Chem. Pharm. Bull.*, 1995, **43**, 193-197.
2. Dalziel, W.; Hesp, B.; Stevenson, K. M.; Jarvis, J. A. *J. Chem. Soc., Perkin Trans. 1*, 1973, 2841-2851.
3. Koskinen, A. In *Asymmetric Synthesis of Natural Products*, John Wiley & Sons: Chichester, 1993; p 6.
4. Douros, J.; Suffness, M. In *New Anticancer Drugs*, Carter, S. K.; Sakurai, Y., Eds.; Springer-Verlag: Berlin, 1980; p 29.
5. Pedrali-Noy, G.; Belvedere, M.; Crepaldi, T.; Focher, F.; Spadari, S. *Cancer Res.*, 1982, **42**, 3810-3813.
6. Personal information from Dr. Anthony B. Mauger (National Cancer Institute).
7. Hiramitsu, T.; Mouri, A.; Suzuki, H. (Nippon Mektron Ltd.). Japan Patent Kokai Tokyo Koho JP 5-310621.
8. Kamenecka, T. M.; Overman, L. E. *Tetrahedron Lett.*, 1994, **35**, 4279-4282.
9. Camps, F.; Coll, J.; Guitart, J. *Tetrahedron*, 1986, **42**, 4603-4609.
10. E/Z ratio (ca. 1 : 1) was determined after the initial Wacker oxidation reaction.
11. Griffith, W. P.; Ley, S. V. *Aldrichimica Acta*, 1990, **23**, 13-19.
12. Tsuji, J.; Nagashima, H.; Nemoto, H. *Org. Synth.*, 1990, Coll. Vol. 7, 137-139.
13. Kotsuki, H.; Ushio, Y.; Kadota, I.; Ochi, M. *J. Org. Chem.*, 1989, **54**, 5153-5161.
14. A recent review: Trost, B. M. *Acc. Chem. Res.*, 1990, **23**, 34-42.
15. Ukai, J.; Ikeda, Y.; Ikeda, N.; Yamamoto, H. *Tetrahedron Lett.*, 1983, **24**, 4029-4032.
16. The stereochemistry of **13a** was deduced from the preference of the conformer (**12a**) in the transition state, in which the nonbonding interactions are minimized. MMX steric energy calculations¹⁹ were carried out on the two transition states available to the triene (**12**).



These calculations suggested that the transition state (**12a**) is considerably lower in energy than the alternative one (**12b**). Thus, this analysis supports the tentative assignment of **13a** as the major product of the intramolecular Diels-Alder reaction of **12**.

17. Nickon, A.; Swartz, N.; DiGiorgio, J. B.; Widdowson, D. A. *J. Org. Chem.*, 1965, **30**, 1711-1717.
18. The consumption of both cycloadducts (**13a** and **13b**) was deduced from the TLC analysis.
19. Molecular mechanics calculations were performed by means of the MMX PC modification of MM2. Gilbert, K. E.; Gajewski, J. J. Serena Software, P. O. Box 3076, Bloomington, IN, 47402-3076.