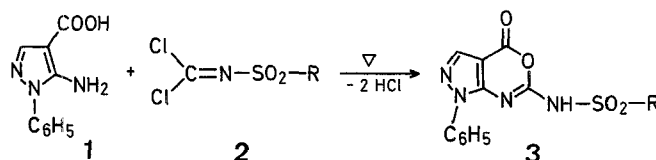


tional compounds to give sulfonylamino- and sulfonylimino-substituted five- and six-membered heterocyclic compounds. In an extension of this reaction to other bifunctional compounds, we report the preparation of 4-oxo-1-phenyl-6-sulfonylamino-1*H*,4*H*-pyrazolo[3,4-*d*] [1,3]oxazines (**3a-f**) by condensation of *N*-(dichloromethylene)-sulfonamides (**2a-f**) and 5-amino-4-carboxy-1-phenylpyrazole⁴ (**1**).



The reactions are carried out by addition of compound **2** in toluene at room temperature to a suspension of compound **1** in toluene. After reflux, the crude product **3** is obtained and may be purified by recrystallization (Table). The new heterocyclic compounds **3** with a condensed pyrazole ring^{4,5} may be used as intermediates in the synthesis of 5-alkyl- and 5-aryl-1-phenyl-4-oxo-3,4-dihydropyrazolo[3,4-*d*]pyrimidines bearing a sulfonylamino group⁴.

Melting points were determined on a Gallenkamp capillary apparatus and are uncorrected. The I.R. spectra were recorded with a Perkin-Elmer Model 257 instrument. Microanalyses were performed at the Centro Nacional de Química Orgánica, C.S.I.C., Madrid.

4-Oxo-1-phenyl-6-sulfonylamino-1*H*,4*H*-pyrazolo[3,4-*d*] [1,3]oxazines **3**; General Procedure:

To a suspension of 5-amino-4-carboxy-1-phenylpyrazole (**1**; 10.2 g, 0.05 mol) in toluene (125 ml) at room temperature is added dropwise with stirring a solution of the *N*-(dichloromethylene)-sulfonamide **2** (0.05 m) in toluene (125 ml). When the addition is complete, the mixture is refluxed during 15 h and the precipitate obtained is isolated by filtration, dried, and purified by recrystallization from acetonitrile.

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Preparation of 4-Oxo-1-phenyl-6-sulfonylamino-1*H*,4*H*-pyrazolo[3,4-*d*] [1,3]oxazines

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Recently, it has been reported¹ that *N*-(dichloromethylene)-sulfonamides^{2,3} undergo condensation with various bifunc-

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Table. 4-Oxo-1-phenyl-6-sulfonylamino-1*H*,4*H*-pyrazolo[3,4-*d*] [1,3]oxazines **3a-f**

Product No.	R	Yield [%]	m.p. [°C]	Molecular Formula ^a	IR (Nujol) [cm ⁻¹]			
					ν_{NH}	$\nu_{\text{C=O}}$	$\nu_{\text{C=N}}$	ν_{SO_2}
3a	CH ₃	70	147–149°	C ₁₂ H ₁₀ N ₄ O ₄ S (306.3)	3205	1790	1640	1260, 1150, 1095
3b	C ₆ H ₅	85	181–183°	C ₁₇ H ₁₂ N ₄ O ₄ S (368.3)	3100	1760	1620	1260, 1160, 1095
3c	<i>p</i> -H ₃ C–C ₆ H ₄	73	174–176°	C ₁₈ H ₁₄ N ₄ O ₄ S (382.3)	3105	1770	1620	1275, 1145, 1095
3d	<i>p</i> -Cl–C ₆ H ₄	88	204–206°	C ₁₇ H ₁₁ ClN ₄ O ₄ S (402.8)	3100	1765	1630	1255, 1175, 1095
3e	<i>p</i> -H ₃ CO–C ₆ H ₄	65	191–193°	C ₁₈ H ₁₄ N ₄ O ₅ S (398.3)	3125	1780	1635	1295, 1145, 1090
3f	<i>p</i> -O ₂ N–C ₆ H ₄	57	210–212°	C ₁₇ H ₁₁ N ₅ O ₆ S (413.3)	3185	1785	1640	1280, 1140, 1075

^a Satisfactory microanalyses obtained: C ± 0.23, H ± 0.27, N ± 0.28, S ± 0.33.