

Specific Phase-Transfer Catalyzed *N*-Monoalkylation of 2-Aminobenzophenones

Gilbert MOUZIN, Henri COUSSE, Jean-Marie AUTIN

Centre de Recherches Pierre Fabre, 17 Avenue Jean Moulin, F-81 106 Castres, France

The 2-alkylaminobenzophenones, especially 5-chloro-2-methylaminobenzophenone, are synthetic intermediates used in the manufacture of anxiolytics such as diazepam* and in the synthesis of a new generation of non-diazepam-like anxiolytics^{1,2,3}, the glycanilides^{4,5,6}.

The presently used methods of alkylating 2-aminobenzophenone are as follows:

- tosylation of 2-aminobenzophenones, alkylation of the sulfonamide intermediate, and hydrolysis with concentrated sulfuric acid⁷;
- amidation of 2-aminobenzophenones with benzoyl chloride, alkylation of the secondary amide thus formed, and hydrolysis⁷;
- direct alkylation of 2-aminobenzophenones, in solution in acetic acid, with an alkyl sulfate⁸;
- formylation of 2-aminobenzophenones⁹;
- alkylation of 2-aminobenzophenones with polyphosphate esters¹⁰.

Methods (a) and (b) are selective, but relatively slow (three steps), whereas with the direct methods (c), (d), and (e) the alkylation is not selective, and obtaining pure monoalkyl derivatives proves difficult. To reduce these disadvantages, we have developed a new method of alkylation using a phase-transfer catalytic technique.

2-Aminobenzophenones (**1**) are alkylated in solution in tetrahydrofuran in the presence of powdered sodium hydroxide and tetrabutylammonium bromide.

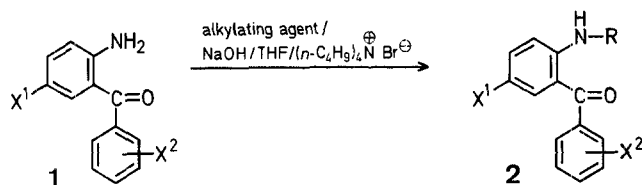
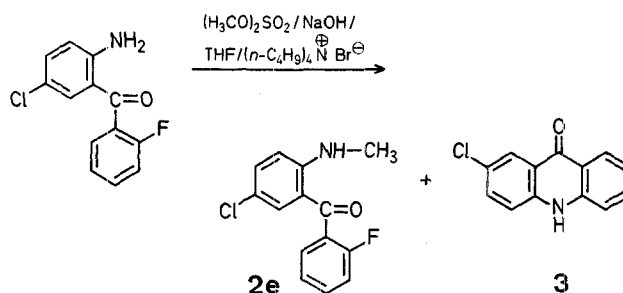


Table. 2-Alkylaminobenzophenones (**2**)

Product 2	X ¹	X ²	R	Reaction time at 60 °C	Yield [%]	m.p. [°C]	
						found	reported ⁷
a	Cl	H	CH ₃	1 h	100	95°	94–95°
b	Br	H	CH ₃	2 h	94	97°	97–98°
c	Cl	2-Cl	CH ₃	1 h	98	89°	88–90°
d	Cl	4-Cl	CH ₃	2 h	90	123°	C ₁₄ H ₁₁ Cl ₂ NO (280.2)
e	Cl	2-F	CH ₃	2 h	68	121°	118–119°
f	Cl	H	—C ₂ H ₅	2 h	95	58°	56–57°
g	Cl	H	—CH ₂ —CH=CH ₂	30 min	100	76°	76–77°

By this method¹¹, the 2-alkylaminobenzophenones (**2**) listed in the Table may be obtained in a single step and in high yields. Product **2d** is a new compound.

The relatively low yield (68%) of product **2e** is caused by the formation of 7-chloroacridanone (**3**; 20%) as a side product by intramolecular cyclization of compound **1**.



Compound **3** was identified by comparison with an authentic sample (m.p. 394–396 °C) prepared by the method of Ref.¹².

5-Chloro-2-methylaminobenzophenone (**2a**); Typical Procedure:

Finely powdered sodium hydroxide (160 g, 4 mol) is added to a stirred solution of 5-chloro-2-aminobenzophenone (231 g, 1 mol) and tetrabutylammonium bromide (3.22 g, 0.01 mol) in tetrahydrofuran (2000 ml) and stirring is continued for 5 min at room temperature. Then, dimethyl sulfate (378 g, 3 mol) is added and the stirred mixture is heated at 60 °C for 1 h. The tetrahydrofuran is evaporated under reduced pressure and the residue taken up in ethyl acetate. This organic solution is washed with water until it is neutral, dried with sodium sulfate, filtered, and evaporated to dryness; yield of **2a**: 245 g (100%); m.p. 95 °C (Ref.⁷, m.p. 94–95 °C); purity: >99% [G.L.C. analysis (10% SE 30)].

The 2-alkylaminobenzophenones **2b–g** are prepared following the same procedure except that in the preparation of **2f** diethyl sulfate is used as alkylating agent and in the preparation of **2g** allyl bromide is used.

The I.R. and ¹H-N.M.R. spectra of all products **2** (except **2d**) thus obtained were identical with those of authentic samples prepared by method (a)⁷.

5,4'-Dichloro-2-methylaminobenzophenone (**2d**); yield: 90%; m.p. 123 °C, golden-yellow crystals.

C₁₄H₁₁Cl₂NO calc. C 60.02 H 3.95 Cl 25.31 N 4.99
(280.2) found 59.97 3.87 25.33 5.02

I.R. (KBr): ν = 3340 (NH); 1625 (C=O); 1560 (C=C_{arom}) cm⁻¹.

¹H-N.M.R. (CDCl₃): δ = 7.5–7.1 (m, 7 H_{arom}); 6.6 (d, 1H); 2.9 ppm (s, 3H).

Received: November 24, 1980

* Trade names of commercial (diazepam preparation): Valium, Ansolin, Apaurin, Apozepam, Atensine, Atilen, Bialzepam, Calmpose, Cereglart, Dipum, Eridan, Faustan, Lembrol, Levium, Morosan, Noan, Puritram, Tranimul, Vival, and Vivol.

¹ R. Roques, J. P. Declercq, G. Germain, G. Mouzin, H. Cousse, Congrès de Chimie Thérapeutique, Bordeaux, France, 1979.

² G. Mouzin, H. Cousse, A. Stenger, J. P. Rieu, XVIèmes Rencontres Internationales de Chimie Thérapeutique, Marseille, France, 1980.

³ G. Mouzin, H. Cousse, J. P. Rieu, A. Stenger, M. Charveron, M. Morre, XVIèmes Rencontres Internationales de Chimie Thérapeutique, Marseille, France, July 1980.

⁴ G. Mouzin, H. Cousse, A. Stenger, S. Casadio, *European Patent* 299 (1979), Centre de Recherches Pierre Fabre; C. A. **91**, 20092 (1979).

⁵ G. Mouzin, H. Cousse, A. Stenger, *European Patent* 10030 (1980), Centre de Recherches Pierre Fabre; C. A. **93**, 221099 (1980).

- ⁶ G. Mouzin, H. Cousse, *French Patent* 7904837 (1979), Centre de Recherches Pierre Fabre.
- ⁷ L. H. Sternbach et al., *J. Org. Chem.* **27**, 3781 (1962).
- ⁸ E. B. Åkerblom, *French Patent* 2081393 (1972), Pharmacia Aktiebolag; *C. A.* **77**, 48072 (1972).
- ⁹ C. Podesva, C. Solomon, K. Vagi, *Can. J. Chem.* **46**, 435 (1968).
- ¹⁰ M. Oklobdžija et al., *Synthesis* **1975**, 596.
- ¹¹ G. Mouzin, H. Cousse, J. M. Autin, *French Patent* 2436774 (1978) Centre de Recherches Pierre Fabre; *C. A.* **93**, 94996 (1980).
- ¹² R. I. Fryer, J. Earley, L. H. Sternbach, *J. Chem. Soc.* **1963**, 4979.