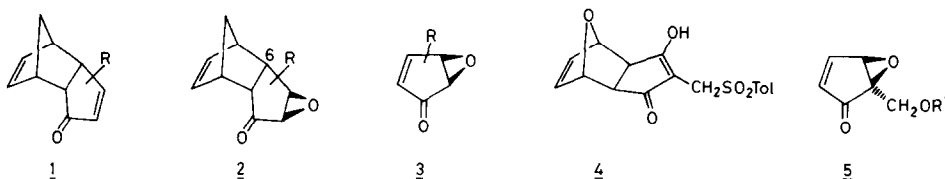


SULFONE MEDIATED SYNTHESIS OF CYCLOPENTADIENONE EPOXIDES FROM  
 10-OXATRICYCLO[5.2.1.0<sup>2,6</sup>]DECADIENONES.  
 A CONVENIENT ROUTE TO *epi*-PENTENOMYCINS

A.J.H. Klunder, A.A.M. Houwen-Claassen, M.G. Kooy and B. Zwanenburg<sup>\*</sup>  
 Department of Organic Chemistry, University of Nijmegen,  
 Toernooiveld, 6525 ED NIJMEGEN, The Netherlands

**Abstract:** Sulfonylmethylation of 10-oxatricyclodecadienone **6** leads to sulfone **4** which forms the key intermediate in the synthesis of 5-alkoxymethylcyclopentadienone epoxides **5**. Acid catalyzed hydrolysis of **5** followed by acylation affords *epi*-pentenomycin derivatives **15**

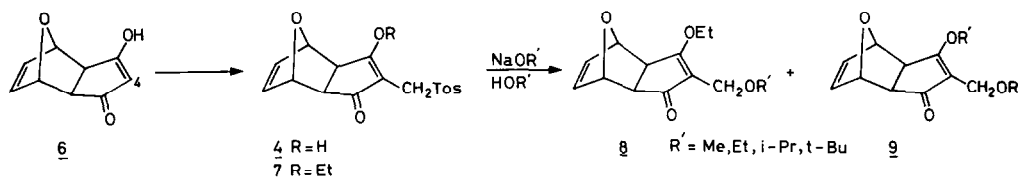
Tricyclo[5.2.1.0<sup>2,6</sup>]decadienones **1** have great potential as synthetic equivalents of cyclopentadienone and serve as building blocks in cyclopentanoid natural product synthesis. *E.g.*, stereoselective chemical transformation of the enone moiety in **1**, followed by thermal cycloreversion, produces functionalized cyclopentenones. Based on this strategy we recently reported the total synthesis of ( $\pm$ )terrein<sup>1</sup> and ( $\pm$ )pentenomycin<sup>2</sup>. Key intermediates in these syntheses are tricyclodecadienone epoxides **2** which are readily obtained from **1** by nucleophilic epoxidation.



Thermal cycloreversion of these tricyclic epoxides applying flash vacuum thermolysis, generates cyclopentadienone epoxides **3**, which, however, can only be isolated as such, if the temperature, required for efficient cycloreversion, is sufficiently low to prevent their thermal rearrangement to the corresponding  $\alpha$ -pyrones<sup>3</sup>. The scope of this approach turned out to be limited to tricyclodecadienone epoxides **2** which possess a  $\pi$ -containing functionality R at C<sub>6</sub> (*e.g.* R: CH=O, CO<sub>2</sub>Et or C=CCH<sub>3</sub>)<sup>3</sup>. This severe limitation prompted us to explore synthetic routes to the 10-oxa-analogues of **1** and **2**. These furan derived Diels-Alder adducts will presumably allow [4+2] cycloreversions at lower temperatures<sup>4</sup> and hence offer a more general route to functionalized cyclopentadienone epoxides. In this communication we wish to report on the synthesis of oxatricyclodecadienone sulfone **4** and its ultimate transformation into 5-alkoxymethyl cyclopentadienone epoxides **5**. These latter structures serve as precursors for *epi*-pentenomycin analogues **15**.

The 10-oxatricycloclodecadienone system can readily be prepared by a Diels-Alder reaction of furan and cyclopentene-3,5-dione<sup>4</sup>. A single adduct **6** is formed which is completely enolic and possesses the *exo*- configuration. This structure is ideally suited for functionalization of the active methylene position at C<sub>4</sub>. With the object to eventually synthesize *epi*- pentenomycins, the introduction of a hydroxymethyl function at this position was investigated. Direct hydroxymethylation of **6** using formaldehyde failed and led to minor amounts of polar products, which according to <sup>1</sup>H-NMR, did not contain any of the desired hydroxymethyl alcohol. A more successful result was achieved by utilizing a modification of Hellmann's method for the  $\beta$ -sulfonylmethylation of 1,3-diketones<sup>5</sup>. Treatment of **6** with sodium *p*-toluene sulfinate and paraformaldehyde in glacial acetic acid/DMF at room temperature afforded sulfone **4** which precipitated from the reaction mixture (yield 85%; Scheme I).

Scheme I



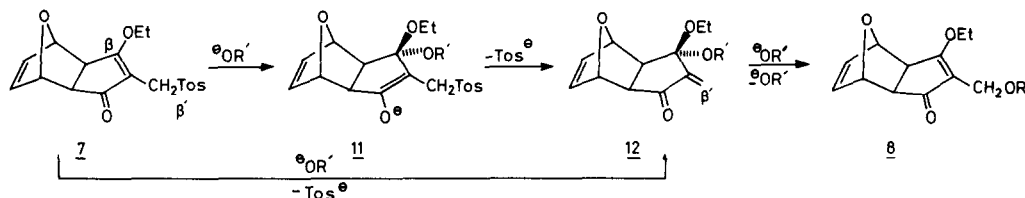
The poorly soluble sulfone was converted into a more convenient material by reacting it with Meerwein's reagent to give the crystalline enolether **7** (m.p. 170- 172 °C, yield 90%). This multiply functionalized compound turned out to be a synthetically promising species as the tolylsulfonyl function can be readily replaced by other heteroatom containing groups.

Heating **7** in EtOH with 1.2 equivalent of NaOEt for 10 min. under reflux gave diethoxy ether **8** (R' = Et) in almost quantitative yield. When NaOMe in MeOH was used under identical conditions ethoxymethoxy ether **8** (R' = Me) was obtained also in high yield. Longer reaction times produced mixtures of **8** (R' = Me) and the bismethoxy ether **9** (R' = Me). The latter ether was isolated as the only product after 15 hrs of heating under reflux, but in considerably lower yield (60%)<sup>6</sup>. Similar behavior was observed for other alcohols *i.e.* iso-propyl and *t*-butyl alcohol. Thiols *e.g.* benzylthiol and thiophenol also reacted smoothly with sulfone **7** to give the corresponding thioethers in excellent yields.

For the sulfone displacement in **7** two pathways can be envisaged, *viz.* conjugate addition/elimination and vinylogous substitution (S<sub>N</sub>2')<sup>7</sup>. Initial  $\beta$ -attack of the nucleophile on sulfone **7** will occur exclusively from the sterically less hindered *exo*-side<sup>9</sup> leading either to enolate anion **11** ( $\beta$ -addition product) or directly to methylene ketone **12** (S<sub>N</sub>2' process) (Scheme II). In case of initial formation of **11**, subsequent  $\beta'$ -elimination also leads to **12**. As under the applied reaction conditions, the enone **12** could not be detected, it will have a very short lifetime and will rapidly react with a second nucleophile at the  $\beta'$ -position to form eventually enolether **8** under the expulsion of the *exo*- $\beta$ -substituent introduced at the outset. This second transformation can again be envisaged either as a conjugate addition followed by elimination or as a S<sub>N</sub>2'-type substitution. The high stereoselectivity observed in the expulsion of the *exo*- $\beta$ -alkoxy

group when applying relatively short reaction times, indicates the  $\text{SN}_2'$  reaction to be the most important one.

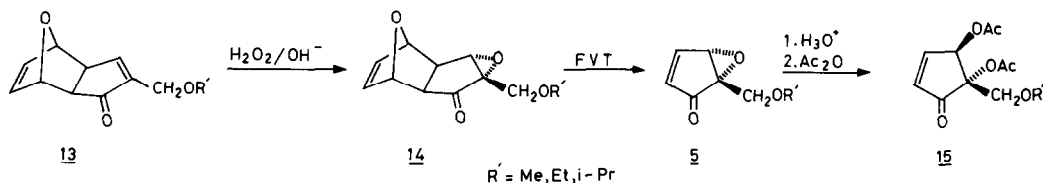
Scheme II



The nature of the  $\beta'$ -function strongly effects the course of this nucleophilic process. Whereas replacement of the  $\beta'$ -tosyl group is almost instantaneous, treatment of bis-methoxy ether 9 ( $\text{R}'=\text{Me}$ ) under identical conditions with  $\text{NaOEt}$  in  $\text{EtOH}$ , only showed a relatively slow exchange of the  $\beta'$ -methoxy group. Even by applying extended reaction times (heating for 2 days) substitution of the  $\beta'$ -methoxy group could only partly be accomplished. Studies to further unravel the mechanism of the nucleophilic transformation in the uniquely functionalized tricyclic system 7 are currently under way<sup>10</sup>.

Enol ethers 8 were subjected to hydride reduction, in order to remove the enolic ether function. With DIBAL in hexane, a regioselective reduction led to the expected enones 13 ( $\text{R}'=\text{Me}$ ;  $\text{Et}$ ;  $i\text{-Pr}$ ) in 90% overall yields<sup>11</sup> (Scheme III).

Scheme III



Alkaline epoxidation of these enones with  $\text{H}_2\text{O}_2$  led to the corresponding epoxides 14 in almost quantitative yield.  $^1\text{H-NMR}$  analysis of these tricyclic epoxides established the predicted *exo*-configuration of the epoxide ring<sup>3</sup>. Thermal cycloreversion of tricyclic epoxides 14, using flash vacuum thermolysis, proceeds at considerably lower temperatures than for 2. At temperatures as low as  $330^\circ\text{C}$  complete cycloreversion of 14 ( $\text{R}'=\text{Me}$ ;  $\text{Et}$ ;  $i\text{-Pr}$ ) was observed and the cyclopentadiene epoxides 5 could be isolated in almost quantitative yield. At this temperature no  $\alpha$ -pyrones were detected at all. Cyclopentadienone epoxides 5 were unequivocally characterized by their IR and  $^1\text{H-NMR}$  spectra. Typically, the  $^1\text{H-NMR}$  ( $\text{CCl}_4$ ) spectrum of 5 ( $\text{R}'=\text{CH}_3$ ) displays a doublet of doublets ( $J_{\text{H}_3\text{H}_2}=6\text{ Hz}$ ;  $J_{\text{H}_3\text{H}_4}=1.5\text{ Hz}$ ) at  $\delta 7.52$  for  $\beta$ -proton  $\text{H}_3$ , a doublet of doublets ( $J_{\text{H}_2\text{H}_3}=6\text{ Hz}$ ;  $J_{\text{H}_2\text{H}_4}=2.3\text{ Hz}$ ) at  $\delta 5.96$  for  $\alpha$ -proton  $\text{H}_2$ , a narrow multiplet at  $\delta 3.97$  for  $\text{H}_4$ , AB-doublets at  $\delta 4.04$  and  $3.61$  ( $J=11.4\text{ Hz}$ ) for the diastereotopic methylene protons and a singlet at  $\delta 3.33\text{ ppm}$  for the methoxy group.

This efficient synthesis of alkoxymethyl cyclopentadienone epoxides 5 (overall yields from 6 are > 65%) shows the potential of the 10-oxatricyclodecadienone system as a synthetic equivalent for cyclopentadienones.

The applicability of 4-alkoxymethyl cyclopentadienone epoxides 5 in natural product synthesis is demonstrated by their conversion into acylated alkyl *epi*-pentenomycin derivatives 15 by acid catalyzed hydrolysis followed by acylation<sup>12</sup>. The moderate yields (25-30%) observed for this epoxide ring opening reaction are attributable to the reactive nature of these cyclopentenones.

#### NOTES AND REFERENCES

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5. H. Hellmann and K. Mueller, *Chem. Ber.* 88, 638 (1965).
6. It is very likely that under the basic conditions applied,  $\beta$ -elimination of the 10-oxabridge becomes competitive, *cf.* F. Brion, *Tetrahedron Lett.*, 1982, 5299.
7. Direct  $SN_2$ -type replacement of the sulfone function is sterically and electronically excluded<sup>8</sup>.
8. See J. Strating in 'Organic Sulfur Compounds I', p. 146-153; Ed. by N. Kharash, NY (1961); F.G. Bordwell and G.D. Cooper, *J. Amer. Chem. Soc.*, 73, 5184 (1951).
9. We found that the alkaline epoxidation of *exo*-tricyclodecadienones 1<sup>3</sup> and *exo*-oxatricyclodecadienone 13 (*vide infra*) both proceed with complete *exo*-stereoselectivity.
10. Related processes are the 1,4-conjugate addition to  $\beta$ -functionalized enones and the bis- $\beta,\beta'$ -1,4-conjugate addition to  $\beta'$ -functionalized enones; see for leading references on both processes: C.J. Kowalski and K.W. Fields, *J. Org. Chem.*, 46, 197 (1981) and A.B. Smith, III, B.A. Wexler, and J.S. Slade, *Tetrahedron Letters* 1980, 3237.
11. The hydride reduction of enolether sulfone 7 using  $LiAlH_4$  in THF deviated from the expected reaction pattern and did not lead to the anticipated enone. Details on this interesting reaction will be reported in a forthcoming paper.
12. See for other approaches to *epi*-pentenomycins: T. Shono, Y. Matsumura, S. Yamnae and M. Suzuki, *Chem. Lett.* 1980, 1619; A.B. Smith III, S.J. Branca, N.N. Pilla and M.A. Guaciaro, *J. Org. Chem.* 47, 1855 (1982).

(Received in UK 2 January 1987)