

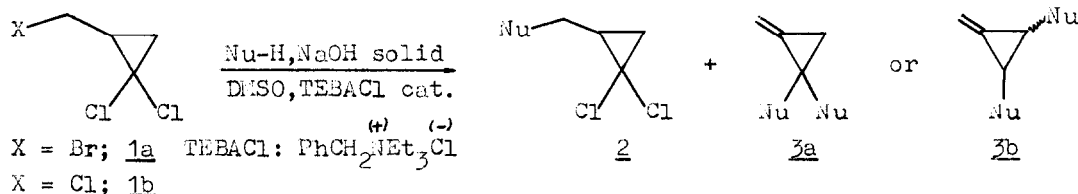
A NEW REACTION OF 1,1-DICHLORO-2-HALOMETHYLCYCLOPROPANES IN BASIC MEDIUM¹

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Summary: Utilization of the reaction of nucleophilic reagents with **1b** allows for the simple preparation of methylenecyclopropane derivatives **3** via elimination-addition routes.

It has been reported² that **1a** reacts with nucleophilic reagents (e.g. potassium phthalimide, sodium diethylmalonate) to give chain substituted products **2**. We wish to report that the reactivity pattern of **1**³ is strongly influenced by the nature of nucleophilic reagent and reaction conditions leading either to the formation of product **2**, or a mixture of products **2** and **3**.



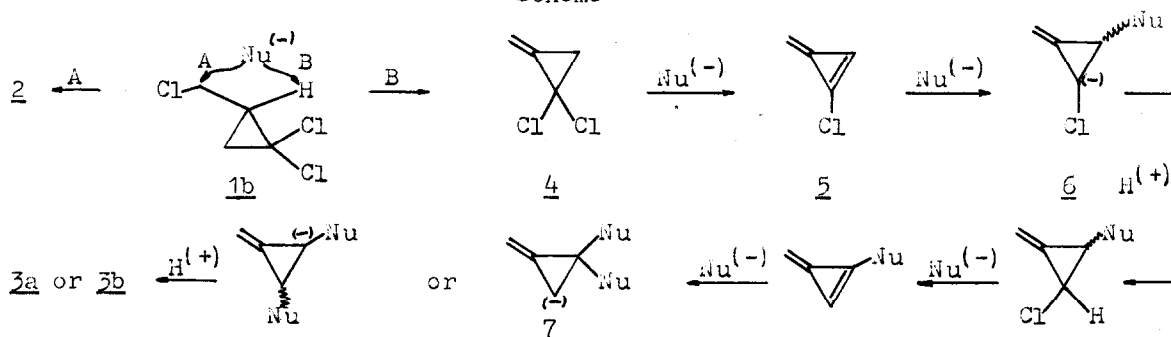
Nu-	Product(s) ^a		B.p. (°C/mmHg) m.p. (°C)	Nu-	Product(s) ^a		B.p. (°C/mmHg) m.p. (°C)
	No.	yield ^b (%)			No.	yield ^b (%)	
PhO-	2a	10	105-106/0.1	Me PhC CN	2d	9	107-110/0.05
	3a	73.5	94-96		3b1	88 ^c	163-170/0.05; 95-93
PhS-	2b	91	96-99/0.1	Ph PhC CN	2e	34	150-152/0.01
PhCH ₂ O-	2c	74	135-133/0.01		3b2	46	145-143 (trans)

^a all compounds were fully characterized; ^b the yields were not optimized;

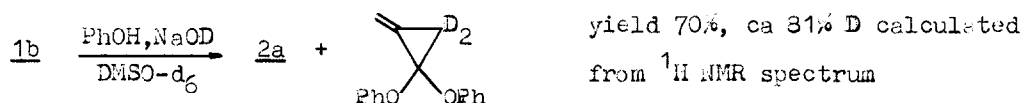
^c mixture of diastereoisomers, cis-trans stereochemistry not determined.

Clearly, the products **2** are formed via a S_N process (Scheme, path A). Analysis of the literature⁴ as well as the results of our experiments (see below) enable us to suggest that the formation of **3** proceeds via elimination-addition reactions (Scheme, path B).

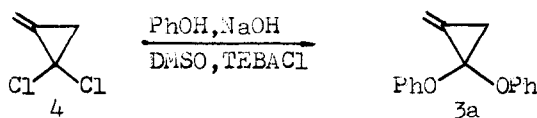
Scheme



The interaction of 1b which takes place with phenol in NaOD/DMSO- d_6 confirms that carbanions 6 and 7 are generated during the reaction course.



Methylenedichlorocyclopropane (4)⁵, a reactive intermediate reacts with phenolate to yield the anticipated product 3a⁶ according to Scheme, path B.



Our paper presents evidence for the intermediacy of substituted methylenecyclopropanes as reactive intermediates⁴. We have also described a new, simple procedure for the preparation of methylenecyclopropane derivatives.

REFERENCES AND NOTES

1. Reactions of Organic Anions, Part CIII: A.Jończyk, T.Pytlewski, J.Org.Chem., in press.
2. K.Steinbeck, Liebigs Ann.Chem., 920 (1979).
3. Qualitatively similar results were obtained both with 1a and 1b, only the results with 1b are described in this paper.
4. For recent examples see: W.E.Billups, A.J.Blakeney and W.T.Chamberlain, J.Org. Chem., 41, 3771 (1976), and literature cited therein.
5. T.Greibrokk, Acta Chem.Scand., 27, 3207 (1973).
6. 3a1: ¹H NMR (CDCl₃) δ: 1.59-1.76 (m, 2H, Δ), 5.14-5.61 (m, 2H, CH₂=), 6.55-7.11 (m, 10H, Ar-H); ¹³C NMR (CDCl₃): 17.85 (t), 81.42 (s), 105.29 (s), 115.11 (d), 120.43 (d), 127.37 (d), 130.10 (s), 154.27 (s).

Acknowledgements: We are grateful to the Polish Academy of Sciences for financial support (Grant MR-I-12).

(Received in UK 4 January 1983)