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Topochemistry. Part XVII.¹ The Crystal Structures of Two Modifications of trans-β-2-Thienylacrylic Acid

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The partial three-dimensional analyses of two crystal modifications of trans-β-2-thienylacrylic acid are reported. The C-C bond lengths have been determined with an e.s.d. of about 0.02 Å; the differences in bond lengths and angles measured independently in the two packing arrangements are not significantly greater than the standard deviations of their measurements.

According to their cell dimensions as well as the geometry of contact of exocyclic C=C groups the two modifications belong, respectively, to the β - and γ -type of the cinnamic acid series. The photochemistry of the two forms is discussed in terms of their packing arrangements.

According to current work in this laboratory² the two modifications of trans- β -2-thienylacrylic acid (I) undergo different photo-reactions: whereas the metastable



 β -2-Thienylacrylic acid (I). The numbering of atoms used in the present analysis.

form gives the cyclobutane (II) in high yield together with small quantities of polymer the stable form polymerises exclusively and affords no dimeric material whatever. This variation in photo-behaviour with packing arrangement has led us to analyse the two crystal structures of the monomer. Since we were primarily interested in the crystal structures rather than exact molecular dimensions only partial three-dimensional data were collected.

EXPERIMENTAL

The acid (I) was prepared from thiophen-2-aldehyde by condensation with malonic acid in pyridine solution in the presence of piperidine. The crystals obtained from slowly cooled solutions in ethanol or acetic acid had m. p. 147° $(lit., ^{3} 147^{\circ});$ they are colourless monoclinic laths elongated along [010]. This form, the cell dimensions of which are listed in Table 1, is designated γ in accordance with the

TABLE 1

Cell constants of the two forms of thienylacrylic acid, $C_7H_6O_2S$, M, 154·2

- γ -Modification: $a = 11.41_2, b = 5.04_0, c = 13.00_5 \text{ Å}; \beta = 98.2^\circ.$ Absences: hol for l odd, 0k0 for k odd; space group: $P2_1/c;$ $V = 740 \text{ Å}^3; d$ (flotation): 1.37; d (calc.) (n = 4): 1.384 $g_{\rm cm.^3}$. F(000) = 320.
- β-Modification: $a = 9.58_5$, $b = 3.91_1$, $c = 20.19_2$ Å; $\beta = 109.5_1^\circ$. Absences: hol for l odd; 0k0 for k odd; space group: P2/c; V = 714 Å³; d (flotation): 1.45; d (calc.) (n = 4): 1.436 g./cm.³. F(000) = 320.

nomenclature of Part III.⁴ A second, also monoclinic, modification has been grown from solutions in dimethyl-

¹ Part XVI, S. E. Filippakis and G. M. J. Schmidt, preceding

Paper. ² M. Lahav and G. M. J. Schmidt, Part XVIII, following

formamide by evaporation; slowly heated on a microscope hot-stage the crystals of this modification transform into the γ -modification between 130 and 135° without actually melting, and liquefy at 147°, the m. p. of the stable form. According to its cell dimensions (Table 1) this (metastable) modification belongs to the β -type. Cell dimensions were measured from high-angle reflections on zero-level Weissenberg photographs calibrated for film shrinkage; densities were measured by flotation in liquids of calibrated density (carbon tetrachloride-toluene mixtures).

The hol and hll reflections of each modification were recorded on Weissenberg photographs with Ni-filtered Cu K_{α} radiation. Absorption was not allowed for; spotshape corrections according to Phillips 5 were applied to the h1lintensities. The number of measured and "unobserved " reflections were F(h0l): (168 + 28) and (225 + 16); F(h1l): (304 + 57) and (383 + 29), for the γ and β modifications, respectively. Both h0l zones were solved by means of sharpened Patterson projections which showed the $S \cdots S$ space-group vectors as well as the vectors between sulphur and all other atoms. After the short-axis projections had been refined by a full-matrix programme on the CDC 1604 installation to acceptable agreement factors (γ -form: $r[=\Sigma w (k^2 F_0^2 - F_c^2)^2 / \Sigma w k^4 F_0^4] = 0.019; R[= \Sigma | kF_0 - |F_c| | / \Sigma kF_0] = 0.075; \beta$ -form: r = 0.040; R =0.093) the y co-ordinates were derived from approximate molecular models projected onto the (xz) co-ordinates; the refinement of the hll reflections was carried out first separately and later combined with their h0l reflections. In the absence of experimental scale factors connecting F(h0l)and F(h1l) the scale factors of the levels k = 1 were included as parameters in the least-squares procedure. Hydrogen co-ordinates were derived by means of the usual assumptions and introduced into the anisotropic refinement of all the atoms, the thermal parameters of the hydrogens being treated as isotropic throughout. Iterations were stopped when the shifts in all parameters had become random and reached about a third of their standard deviations. At this stage the agreement factors for the combined h0land hll reflections stood at r = 0.011, R = 0.062 (y-form) and r = 0.032 and R = 0.085 (β -form). The following scattering factor curves were used: $f_{C,O}$, Berghuis et al.,⁶ $f_{\rm S}$, Dawson.⁷

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TABLE 2(a)

Structure factors, γ -modification

h	k l	F_{o}	F_{e}	h k l	F_{o}	F_{c}	h k l	F_{o}	F_{c}	h k l	F_{o}	F_{e}	h k l	F_{o}	F_{c}
0	0 2	14.74	15.85	-12	6.93	-).38	-9	3.71 6.82	-4.50	8	24.69 3.31	25.63 4.13	2 3	5.47 5.94	4.85 9 .17
	6	40.57	-20.24 -39.64	7 0 0 2	6.22	5.63	-11	7.47	7.40	10	3.36	3.50	14	3.29u	1.48
	10	9.27 3.71	-11.06 -2.81	4	9.51 4.72	-3.89	-12	10.42	10.99	12	3.08u	- 28	20	4.75	4.80
	12 14	8.82	8.41 -7.88	2 10	3.95u 4.44	-3.00 -4.56	-14	10.18	-4.25	5 1 - 1	8.89	9.05	8	3.19	-2.79
1	16 0 -2	2.76 24.29	-1. 49 24.96	7012 80-2	8.37 12.06	7.26 -11.91	. 2 3	18.59 10.03	18.31 9.60	-3	8.15	7.69	9 10	5.02	-0.95
	-4 -6	5.33 18.86	-6.07 -17.28	_4 -6	3.57u 3.75	77 -3.07	4 5	1.94 11.09	68 -9.87	-5	11.57	11.78	91-1	2.25 11.95	2.23 11.97
	-3 -10	9.21 4.52	6.84 4.02	-8 -10	13.00	13.15	6 7	5.89 19.22	-6.90 18.81	-6 -7	4.74	-4.78 14.31	-2 -3	21.95 3.28u	-22.34 -1.48
1	-12 0 0	17.21	15.46 -7.88	800	25.22	23.95	8 119	13.93 8.46	-13.35 8.36	-8 -9	3.08 3.94	-2.85	-4 -5	5.20 3.32	-5.93 -3.32
-	2	92.75z	123.74	4	3.81	-3.93	10	7.65	7.13	-10 -11	3.32 14.63	-2.41 -14.68	-6 -7	9.76	11.03 -9.08
	6	14.27	14.57	8	10.12	-9.99	12	11.74	11.77	5 1 0 1	12.75 20.98	13.56 -19.87	-8 -9	3.34u 6.10	90 6.91
	10	20.80	22.09	12	5.98	5.91	14	5.24	.5.03 82 h7	2 3	8.36 6.93	-10.57 7.08	9 1 Ó 1	4.01 3.29u	-3.72
	14	4.18	-4.67	9 0 -2	22.55	-22.26	-2	81.59z	102.46	514	6.57	7.06	23	9.68	9.71
2	0 -5 10	65.38:	74.86	-8	3.95u	-2.21	-9	18.62	-17.89	5.6	7.51	7.44 7.01	4	10.36	10.07
	-4 -6	14.01	-13.55	-10 -12	3.75u 3.93	4.72	-> -6	22.78	-20.20	7 3	3.16u 3.29u	-2.22 29	6	3.28u	2.34
	-3 -10	7.40	-24.66	900 2	5.96 9.83	5.44 8.65	-7 -8	4.89	-4.55	9 10	11.62 3.80	-12.38 -8.82	8	4.14	-4.51
2	0 0 2	25.06 4.74	-26.22 -4.68	4 6	9.65 3.89u	-9.56 1.52	-9 10	3.03u 3.23u	-3.33	11 6 1 -1	5.00 23.41	-5.88 -22.02	9 1 9 10	2.24u	-2.12
	4 6	53.10 4.85	-53.69 -4.86	8 10	3.55u 6.85	.90 5.81	210	5.27 36.75	-4.11 40.45	-2 -3	2.54u 3.67	-1.07 2.70	10 1 -1	4.51 3.36u	-2.04
	3 10	13.46 27.21	13.28 26.95	10 0-2 -4	21.49 7.38	21.23	1 2	27.50 20.87	-26.01 16.79	-4 -5	9.84 4.82	11.25	-2 -3	3.36u 3.36	2.90
	12 14	7.79	-8.09	-6 -8	9.65 3.830	-8.94	3 4	52.91 2.54	-48.90	-6 -7	5.84	-6.31	-4	5.82 3.36	-5.73 -3.96
3	0 -2	52.13z	57.18 26.78	-10	6.97	6.72	5	22.33	-20.37	-8	10.08	-9.94	-6 -7	5.77 4.02	5.10 -3.15
	-6	23.66	22.55	2	5.58	5.84	7	2.74	-3.93	-10	7.66	-6.19	-8 -9	5.05 3.06	5.06 2.59
	-10	19.44	-19.06	6	5.15	-12.54	9	6.70	7.01	-11	4.72	-5.98 5.68	10 1 Ó	14.83 3.36	14.66
	-12	5.95u 9.35	S.28	10	5.10u	20 5.22	10 11	0.15 3.36u	.11	-13 -14	5.12 3.66	-3.88	23	3.36u 7.90	.86
3	0 0 2	1.834 10.97	52 -10.55	11 0 -2 -4	8.76 15.81	8.42 15.39	12 13	3.28u 3.04u	-1.84	610 1	20 .73 3.67	-18.57 -3.98	4	3.28u	-1.06
	4 6	2.49u 33.66	88 32.26	-6 -8	3.79u 7.54	-2.35 -7.16	14 3 1 -1	4.99 14.37	-4.40 -14.62	23	26.34 21.65	-26.94 22.41	161	6.40	-6.41
	8. 10	11.35 17.64	11.32. -17.23	-10 11 0 0	4.85 3.9 1 u	4.57 97	-2 -3	4.22	-5.59 45.92	ů, 5	5.04 9.88	-5.53 11.03	8	5.02	-5.04
4	12 0 -2	17.76 3.51	-17.49 -2.89	2	9. 37 9.57	-9.18 -9.05	-4 -5	32.46 8.60	-29 .15 8.07	67	11.69 6.17	13.10	<u>и</u> 1-1	4.04	3.98
	-4 -6	3.51	-2.42	6 8	3.20u 7.28	-2.24	-6 -7	14.51 4.97	-16.32 -3.79	8	9.19	9.94 - 3.42	-2	4.67 8.06	-4.69 8.94
	-8 -10	3.35u	2.95	12 0 -2	8.21 3.61	-8.57	-8 -9	6.42 14.75	7.73	10	3.94	4.06	-4 -5	8.94 3.23	-9.29 -3.22
4	ີ້ວ່	36.47	-37.36	-6	3.45	-3.06	•10 •11	13.95	12.94	12	2.61	-2.85	-6 -7	5.49 4.34	-5.72 -4.31
	ů,	6.12	5.80	12 0 0	4.44	4.75	-12	3.54	3.90	-2	21.12 2.81u	-23.() 2.44	-8 -9	4. 1 4 4.74	4.77 -5.13
	3	8.58	8.73	4	4.09 3.12u	-0.50	-14	4.11.	4.55	-3 -4	12.55	13.25	11 1 Ö	3.28u 6.50	19 6.56
_	13	5.25	4.28	13 0 -2	5.52	4.91 5.51	1	24 46	-24.62	-5 -6	15.80	16.37	23	3.19u 7.63	1.33
5	0 -2	6.20 3 1. 53	5.81 32.22	-4	7.32 5.01	6.67 4.62	2 3	19.89	-19.10	-7 -8	9.61 5.70	10.63 5.72	4 5	4.24	4.91
	-6 -8	40.49 6.55	40.66 6.7೦	13 0 0	3.10u 9.67	.12 -8.89	314	3.56 25.21	-4.77 24.27	-9 -10	6.27 7.10	6.84 -6.66	67	3.18 3.64	-3.30
_	-10 -12	17.50 6.75	-19.39 -5.43	14 0-2 _4	2.45u 2.37u	98 .47	6 7	23.16 7 .3 2	-32.59 7.14	-11 -12	6 .1 0 9 . 48	4.53 -9.74	12 1 -1	3.03u	1.05
5	002	51.70 17.38	-48.97 -17.37	-6 14 0 0	3.20 6.69	2.58 -5.88	8 9	7.83. 9.39	8 .12 9 . 75	7 1 O 1	19.49 8.08	-18.64 7.49	-3	4.79	4.23
	4 6	39.98 11.86	41.95 12.32	011	29.93 27.50	33 .84 28 .73	10	9.46 5.79	9.90 -5.92	2	6.03	6.73 10.80	-5	4.14	3.77
	8 10	5.39 6.81	-5.46	3 4	6.83 4.59	-7.78 -6.09	4 1 -1	29.56	-29.60	γ Ι 5 Ι	8.28	8.13	-7	3.29	-2.04
6	12	6.55	-6.90	5 6	21.65 7.58	-19.58 -6.07	-3	17.03	-15.90	5	8.11 8.11	-1.40 8.99	12 1 0	2.10	-2.46 6.91
Ū	-4	31.63	-30.59	7	8.94	-9.76	-5	6.37	6.58	7 8	4.75	-3.36 -5.11	1 2	4.64 2.34u	2.96
	0- 3-	16.58	16.84	9	4.31u	39	-0	9.79	10.46	9 10	3.23u 6.72	.60 -6.56	12 1 3	5.62 4.01	-4.50
	-10	5.95u 6.06	-4.77	11	14.76	14.61	-0 -9	10.87	-11.83	11 12	2.68u 3.52	.01 -3.41	5 6	2.29u 3.96	-4. 3 8
6	ນ 0 2	8.76 36.90	-6.90 35.51	13	3.19u	-2.29	-10 -11	3.28u 3.36u	1.80	6 1 -1 -2	13.45 5.70	14.51 -4.63	13 1 -1	3.09 2.58	3.23 -1.26
	ц б	24.6) 6.33	27.62 -7.14	15	2.00u 3.42	-3.27	-12 -13	3.32u 5.47	1.13 5.24	-3 -4	10.44 3.131	-10.81 2.77	-2 -3	2.59 2.58a	.85 66
	8 10	8.76 8.15	-9.34 -8.85	1 1 -1 -2	50.05	57.24 -48.04	4 1 0 1	23.33 25.70	20 .73 -24.85	-5	6.77	-6.87 1.28	-4 -5	2.53u 6.70	.46 5.07
7	12 0 -2	3.93 23.90	-3.35 -24.71	-3 -4	57.91 69.23	-59.29 -72.32	2 3	27.47 2.29u	-27.76 -1.70	-7	3.31u	-2.44	1	4.81:	-6.35 -1.30
•	-4 -6	6.59 6.63	6.42 -6.91	-5 -6	8.53 13.07	-7.87 12.46	14 5	33.99 20.33	-34.21 19.84	-0 -9	5.80 6.13	5.73	3 1	2.09u 3.64	71
	-8 -10	14.76 3.95u	14.25 .32	-7 -8	26.20 25.95	-25.42 26.45	6 7	3.46 12.97	-3.16 13.72	810	12.82 19.27	-12.50 18.69	14 1 -1 -2	1.86u 4.66	-1.21 4.69

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TABLE 2(b)

Structure factors, β-modification

h k l	F_{o}	F_{c}	h k l	F_{o}	F_{c}	h k l	F_{o}	F_{c}	h k l	F_{o}	$F_{,i}$	h k l	F_{o}	F_{c}
002	24.53	-24.89	16 16	2.46	2.73 -12.55	-10	2.52	-2.11	-14	12.70	-12.23	-24	5.51	-5.12
4 6 8	6.89 72.89	-6.13	. 20 5 0 - 2	7.30 47.65	6.78 -44.30	-14 -16	8.90 2.16	8.07 -2.12	-16 -17	29.34 13.24	29.74 -14.21	4 1 Ó 1	47.21	52.71 29
10. 12	22.07 29.07	19.93 28.00	_4 _6	26.48 17.02	24.61 -16.79	-18 -20	3.30 5.27	-3.16 4.80	-18 -19	15.51 6.83	-15.58 -6.14	2 3	1.47 3.43	1.24 7.13
14 16	13.08 16.10	-11.23 -14.79	-8 -10	19.46 10.17	-17.81 -9.99	1000.	6.07 8.38	-6.37	-20 -21	2.32 1.26u	-2.50 .68	5	4.75	.83 -5.45
13 20	5.91 3.10	-4.61	-12 -14	6.55	-8.30 6.77	<u>п ó -</u> 2	5.21	-4.27 4.80	-22	6.92 8.54	5.89 6.86 8.70	6 7 8	25.85 12.80	24.88 -13.34 -13.34
24 1 0 -2	3.72	4.15	-18 -20	3.03 1.32u	3.70	-6 -8	1.071	56 -6.37	210	7.18	7.12	9 10	10.37	10.40
-6	47.28 30.67	43.98 -28.97	-22	1.17u 8.73	1.19 -8.66	-10 -12	2.12	-2.67 5.59	23	13.56 20.04	19.61 -22.09	11 12	5.63 9.82	4.88 9 .1 3
-8 -10	31.40 2.80	34.20 14	500	31.88 29.57	30.92 30.38	11.00 12.0-4	8.94 9.81	8.75 9.58	4	23.01 1.311	-25.16 -1.77	13 14	7.02	-6.71 -16.52
-12 -14	19.65 28.72 33.09	-18.85	4 6 8	26.76 23.65	-25.06 28.28	-6 -8	8.00 1.81 6.86	-7.55 -2.19	6 7 8	23.13	-18.54	15 16 17	7.16 5.59	7.26 4.65
-18 -20	3.59	-3.81	10	16.56 17.29	-16.14 16.60	012	32.64	36.10 -10.09	9 10	3.08	-3.07	18 19	.97u 3.78	88
-22 -2 ¹	8.61 8.24	-8.97 6.42	14 16	3.32 7.41	-3.87 -6.16	14 5	32.17 1.51	-32.37 -1.13	-11 12	19.29 31.30	16.60 26.81	51-1 -2	1.10u 28.91	-1.43 30.22
100	11.27 59.81	-11.37 54.70	18 6 0 - 2	2.98 45.38	· 3.26 42.99	6 7	1.64 39.15	-1.03 -34.96	13 14	1.38 1.41u	-1.49	-3 -4	9.65	-6.57 11.40
4 . S	19.41 53.10 15 52	-17.55 -58.12	-4 -6 -8	16.68	-15.79 8.05	9 10	24.57 1.77 37.19	-24.15 -35.	15 16 17	7.27	-6.98	-5 -6 -7	21.28	-19.23
1)	33.26 1.16u	-29.63 51	-10 -12	20.63 10.89	-21.50 11.77	11 12	28.97 30.62	23.31	18 19	1.28u 7.46	.63 -6.77	-8 -9	16.57 10.48	17.11
14 16	22.06 1.91u	18.49 .06	-14 -16	5.40 19.54	5.47 -23.22	13 14	2.87 1.34u	-2.15 .16	·20 21	10.11 6.12	-10.54 5.79	-10 -11	4.70 4.50	-4.58 4.41
18 20 20	5.16. 1.19u	4.60 1.32	18 -20	8.54 1.28u	-1.16	15 16	12.21 9.98	-11.32 9.25	3 1 -1 -2	34.08 17.86	-37.52	-12 -13	17.66	-18.12
2 0 -2 -4	64.99 z 22.10	-9.20 119.12 -21.50	-24 6 0 0	9.90 3.63	10.58	18 19	11.34 7.64	11.40 -8.06	-4 -5	19.00 14.79 .86u	-15.09 .14	-15 -16	6.37 11.63	6.87 -12.33
-6 -8	21.71 42.61	21.43 -40.82	24	8.99 28.23	7.47 27.88	20 21	8.75 1.15u	-9.74 1.43	-6 -7	21.17 13.29	-20.48 -13.96	-17 -18	1.42u 5.80	1.51 -5.90.
-10 -12	14.67 51.07	13.62 52.17	6 8	33.63 1.35u	-39.32 1.74	22 23	1.01u 8.25	20 7.68	-8 -9	18.21 3.72	18.07 -2.73	-19 -20	7.02 8.70	-8.08 9.59
-14. -16 -18	2.90	-12.29 2.73 -9.82	10 12 14	9.73 9.78	-10.75 8.42	-2	60.02z 31.71	78.44	-10 -11 3 1-12	3.77	-20.33 3.63	5 1-21 -22	5.17	5.20 -4.14
-20 -22	22.39 5.69	-22.60 5.64	70-2 -4	12.28 3.26	-12.57 75	_4 11-5	6.41 27.73	-6.29 -27.62	-13 -14	8.64 18.40	8.64 18.54	-22 -24 5 1 0	1.03u 3.41 17.86	25 2.95 -18.49
-24 2 0 0	5.86 5.88	6.69 -5.40	-6 7 0 -5	12.32 15.20	11.64 -15.04	-3 -7	15.21 29.86	14.67 -30.21	-15 -16	11.29 9.91	11.74 -9.71	1 2	21.20 23.92	22.79
204	24.86 50.87	-23.25	-10) -12 -14	27.75 16.51	-15.60	-0 -9 -10	18.29 28.91	17.20	-17 -18	8,23 15.05	-7.58	3	1.21 8.20	68 8.25
8 10	4.82	-4.58 43.37	-16 -18	9.20 11.21	10.01	-11 -12	2.97 22.77	-2.99 19.29	-20 -21	5.58 .1.28u	5.41	567	7.40 27.81	-29.25
12 14	8.10 23.51	8.17 -22.95	-20 -22	1.73 7.05	2.57 7.20	-13 -14	9.61 10.80	9 .33 9 .5 9	-22 -23	9.38 7.07	-8.01 -5.99	8	10.13	-12.51
16 18	24.71 8.09	-8,57 -00	700	4.08 23.02	-3.65	-15 -16 -17	3.70	-3.51 -20.43	-24 310	5.38 16.67	5.28 -17.57	10	13.74 3.15	-14.26 3.81
20 22 3 0 - 2	12.01 .69u	12.52 -1.23	4	1.34 6.07	1.20 -5.91	-18 -19	2.82	1.87	2 3	11.00	11.46 -2.21	12 13	2.38 8.89	-2.41 9.61
-4 -6	23.20 42.37	22.93 -49.62	్ 1ం	13.96 1.23u	-17.29 -2.81	-20 -21	10.19 1.22u	-9.91 52	<u>ц</u> 5	29.71 29.88	31.50 -31.37	14 15 16	4.61 5.26	-4.10 -4.08
-8 -10	13.32 55.65	-2.12 53.38	12 14	14.55	13.50 -2.75	-22 -23 -24	1.91	-0.90 2.51 4 10	6 7	11.15 26.79	·11.11 26.85	17 18	2.32 3.13	1.62 3.38
-12 -14 -16	18.75 24.65	18.81 24.66	-4 -6	2.66	2.36 -17.99	1 1 0 1	69.18z 20.00	107.55 -20.35	9 10	7.87 6.43	-7.40	6 1 -1 -2	17.72	16.08 7.08
-18 -20	36.86 1.33u	-36.92 2.34	-8 -10	14.37 4.85	14.55 4.24	·2 3	47.23 9.45	-52.56 8.62	11 12	1.36u 23.49	60 -21.47		21.45 13.67	-21.00
-22 3 C D	3.17 53.46	-3.05 -51.45	-12 -14	3.32 16.98	-2.53 17.02	4 5 6	19.30 37.73	-37.58	13 14	8.25 23.26	6.73 17.68	-6 -7	20.20 6.06	20.71 5.01
4 6	35.06 15.42	-34.64 -14.64	-10 -18 -20	9.15 8.41 1.12	-9.04 -9.15 2.03	7 8	22.85 33.12	-19.78 27.09	15 16 17	1.094 6.43 1.264	-6.75	-8 -9	2.50 9.36	-3.00 9.10
8 10	20.91 18.27	19.53 -16.63	-22 -24	6.98 6.39	-6.32 5.69	9 10	15.23 27.32	1 0.97 21.54	18 19	4.63 1.02u	-4.33 89	-11	10.17 15.78	-16.10 7.90
12 14	28.80 17.38	25.66 16.47	.800	12.48 6.37	-12.74 7.43	11 12 13	26.66 25.88	-22.18 -22.18	20	5.74 1.76	5.58 -1.99	-13 -14	11.43 19.21	11.28 -21.47
18 18 20	3.73 6.18	-19.41 3.78 -6.80	4 6 3	0.29 12.22 12.71	-13.93 -13.93 9.86	15 14 15	3.37 15.16	2.21	4 1 -1 -2 -3	10.48 50.15 12.73	-29.61 13.04	-15 -16	1.41: 3.47	-1.32 -3.34
4 0 -2 -4	.84u 52.22	-1.51 -54.24	90-2 -4	7.50 17.29	-6.63 -15.62	16 17	18.96 4.86	18.64 -4.54	-4 -5	11.45 21.73	-10.87 -21.39	-17 -18 -19	7.04 3.40 6.17	-7.78 -3.76
-6 -8	10.54 .93u	-9.36 -1.40	-6 -3	6.21 8.68	5.62 -8.09	18 19	6.22 5.58	-6.29 5.50	-6 -7	15.60 4.08	15.62 -3.31	-20 -21	2.87	-1.85
-10 -12	29.00 37.73	-25.90 36.59	-12 -14	29.22 10.50	2.55 27.30	20 21 22	3.31 3.04	-3.35 -3.64	-8 -9	16.31 27.29	-16.72 -28.48	-22 -23	6.60 2.93	6.98 2.39
-16 -13	3.95 11.75	-4.28 11.83	-16 -18	1.24u 1.60	40 -1.41	2 1 -1 -2	52.56 z 48.30 z	75.18 -65.76	-10 -11 -12	15.15 3.89	16.31 3.06	610 1	:81: 17.60	1.38
-20	10.70 1.20u	-9.81 01	-20 -22	6.88 8.02	-7.22	-3 -4	32.42 5.85	-35.33	-13 -14	17.87 18.35	19.38 -19.22	2	15.43	-15.22
400	5.93 24.00 30.45	0.16 .21.21 .28.25	900 2 1	15.97 1.29u 10 41	18.62 52 -12.49	-> -6 -7	19.00 3.61 2.02	-20.51 -3.46 -2.84	-15 4 1-16 -17	9.55 1.40 10.06	-10.12 • 1.90 11 80	د ۲ o 4 5	0.12 9.72 3.67	2.47 9.82 3.52
4 6	19.87 19.05	10.47 18.6)	6 8	8.18 12.52	7.53	2 1 -8 -9	27.58 13.18	-29.45 -13.64	-18 -19	14.08 9.35	15.02	6 7	11.91 12.74	-13.40 -14.43
S. 10	12.64 24.10	-12.90 24.02	10 0 -2 4	1.73	1.22	-10 -11	45.61 22.74	45.65	-20 -21	2.33 7.93	-3.03	8	13.87	-15.22 4.61
4 '0 12 14	16.73 5.99	-15.84 6.86	-6 -8	13.90 1.28u	-13.37 1.71	-12 -13	4.40 13.05	5.80 13.57	-22 -23	8.44 5.40	8.36 4.99	10	4.18	-4.16

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						Table	2 (b)	(Contin	nued)					
h k	l F _o	$F_{\rm e}$	h k l	F_{o}	F_{c}	h k l	F_{o}	$F_{\mathbf{c}}$	h k l	$F_{ m o}$	${F}_{ m c}$	$h \ k \ l$	F_{o}	F_{c}
l	2 1. 25u	1.30	2	11.71	12.42	-17	6.39	-6.95	-16	6.75	6.86	-1 6	5.34	-5.03
1	5 4.02	3.82	3	4.46	-5.13	-18	6.12	6.74	-17	4.55	4.25	-17	5.87	-5.54
1.	4 1.05u	-2.05	4	1/.04	-19.52	-20	8.91	-7.99	-18	2.93	-2.84	-18	3.81	3.17
1	2. 4.4/ 6 7.10	4.19	2	3.14 z hh	-0.55	210	0.50	6.01	+19	6.16	-5.61	-19	3.07	-2.54
7 1	1 7 15	7 38	7	1 38	4.00	1	1.410	1.00	9 1-20	4.15	3.84	10 1 0	3.44	-2.42
1	2 11 68.	-11.40	á	10.34	9.55	4	13.86	-(+()	-21	1.14	1.15	1	5.71	5.24
_	3 11 46	-11 38	ă	8.18	-7.67	í.	1 37.9	10.92	,	1.91	1.00	2	6.26	-5.46
Ū	4 22.51	20.99	าด์	10.31	-9.41	Ę	15 48	-16 11	1	2.12	2.11	3	2.39	-2.75
- 1	5 4.81	4.45	ĩĩ	8.94	7.87	é	3 66	-10.11	2 *	1.30	2.99	4	11.77	10.96
-0	6 12.53	-11.66	12	6.26	5.61	7	1.23	85	ñ	2.08	-2.85	5	1.93	-2.15
-	7 6.29	-6.02	13	5.95	5.01	à	4.19	-3.28	5	2.53	3.06	6	7.72	-8.02
-6	6 5.23	4.79	1 1 14	5.25	4.80	9	4.30	3.57	6	9.69	8,38	11 1 -1	6.36	6.53
-9	9 1.3 6u	43	81-1	2.0 1 u	94	15	11.23	10.86	7	5.47	-4.90	-2	8.46	-7.57
-1	ა 16.3 9	1 6.84	-2	8.49	7.77	9 1 - 1	6.26	6.26	10 1-i	1.68	96	-2	2.01	-2.59
-1	1 11.03	'1 0.88	-3	6.32	5.31	-2	12.41	-11.74	-2	10.93	11.69		2.04	2.24
-12	2 11.31	-11.88	-4	17.89	-19.58	-3	5.72	-5.05	-3	8.03	7.81	-5	2.19	-2.00
-1	3 1.410	-1.23	-5	15.62	-13.62	_4	5.75	4.70	_4	3.08	-3.34	-0	2.09	1.90
-1	4 10.31	10.30	-6	1.9 9±	.05	-5	1.98	-2.59	-5	6.95	-7.41	-8	6.23	-5 49
-1	5 1.41	19	-7	2.00	-1.38	-6	1.98	-2.52	-6	3.37	-4.01	-9	3.58	-3.40
-10	6 4.43	4.64	-8	3.99	1.91	-7	2.42	-3.61	-7	7.98	-7.69	·10	8.10	7.62
-1	7 4.13	-4.12	-9	2.83	2.79	-8	7.26	-5.99	-8	2.55	2.67	-11	1.41	62
-10	0. 70.05	-11.58	-10	12.85	-11.95	-9	5.57	-4.86	-9	5.52	5.98	-12	7.47	-5.86
-1)	9 3.04		-11-	10.15	9.54	-10	5.91	5.70	-10	8.14	-8.00	-13	7.43	7.21
-2	1 2.20	5.10	-12	21.40	21.11	-11	14.95	13.74	-11	5.11	-4.50	-14	3.08	-3.52
-2	2.24	1.02	-19	1 400	5.95	-12	2.04	-2.19	-12	2.11	-2.07			
7 1 6	0 13.66	-13.16	-15	3 64		-1)	1.86	-2.49	-15	(-92	6.60	11 1-15	4.06	-4.24
	1 14.21	15.59	-16	7.37	-8.10	-14	1.81	-1.72	-14	h 03	(.22	-10	4.50	4.02
-	- 14.124	-/.//	-10		5.10	-+/		-4.10	-13	95	2.94	H I 0	2.00	2.44

"u": "Unobserved" reflections; "z" reflections given zero weight

The structure factors corresponding to the set of coordinates listed in Table 3(a) and (b), are given in Table 2(a)and (b). Little significance is attached to the values of

TABLE 3								
Atomic co-o	ordinates and	standard devia	tions (Å)					
(a) γ -Modification								
Atom	х с	y σ	<i>z</i> σ					
C(1) *	4·472 (6)	1.318 (13)	0.449 (7)					
C(2)	3.515(6)	$2 \cdot 361$ (14)	0.746(7)					
C(3)	3.677 (6)	3.087(14)	1.842 (7)					
C(4)	2.766 (6)	4.105(13)	$2 \cdot 222$ (7)					
C(5)	2.926(7)	4·855 (14)	3.340(6)					
C(6)	1.836 (7)	5.764 (16)	3.387 (8)					
C(7)	0.878 (8)	5.611(16)	2.351 (7)					
S	1.237 (2)	4.425 (5)	1.232 (2)					
O(1)	4·148 (5)	0.642 (10)	-0.643 (5)					
O(2)	$5 \cdot 563$ (4)	1·119 (9)	1.210(4)					
	(b) β-Mo	odification						
C(1)	1.124(5)	0.439(12)	1.901(5)					
C(2)	2.054(6)	0.793(12)	3.350(5)					
C(3)	1.711 (5)	0.384(13)	4·473 (5)					
C(4)	2.553(5)	0.599(12)	5·934 (5)					
Č(5)	2.171(6)	0.232(13)	7.104 (5)					
C(6)	3·270 (7)	0.577(13)	8.399 (6)					
C(7)	4.404 (7)	1.132(13)	8.195 (6)					
S`	4·216 (1)	1.305(4)	6.448(1)					
O(1)	1.635(5)	0.818 (9)	0.915(4)					
O(2)	-0.007 (4)	-0.131 (9)	1.661 (4)					
			,					

* Refer to diagram for numbering of atoms.

the thermal parameters, in particular of u_{22} , and these are not listed.

DISCUSSION

Molecular Dimensions.—Bond lengths and angles, which must be regarded as approximate in view of the necessarily incomplete refinement of the y parameters, are shown in Figures 1 and 2. The e.s.d. of the C-C and C-S bonds are 0.02 and 0.01 Å in both structures; the e.s.d. of the bond angles are 1° . Since these are of the order of the differences in length of equivalent bonds in the two structures we do not wish to attribute significance to these differences. The C-C and C-S bond lengths averaged between the two structures are given in Table 4, as are also the averaged bond angles within the thiophen ring. Comparison with the comparable bond



FIGURE 1 Bond lengths (Å)

lengths and angles of thiophen-2-carboxylic acid derived from a three-dimensional analysis ⁸ shows fair agreement. On the other hand a recent three-dimensional analysis of thiophen-3-carboxylic acid ⁹ reports unexpectedly large variations in the hetero-ring compared with the thiophen derivatives substituted in the 2-position. Evidently, the bond lengths of substituted thiophens merit further work.

The best planes of the four-carbon chain C(4-7),

- ⁸ M. Nardelli, G. Fava, and G. Giraldi, Acta Cryst., 1962, 15, 727; D. Hudson and J. H. Robertson, *ibid.* p. 913
- 737; P. Hudson and J. H. Robertson, *ibid.*, p. 913.
 P. Hudson and J. H. Robertson, *Acta Cryst.*, 1964, 17, 497.



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the thiophen ring [C(4-7), S], the ethylenic C(1-4), and carboxylic [C(2), C(1), O(1), O(2)] groups and the deviations of the atoms from these planes are listed in Table 5 (a) and (b). The hetero-ring is planar in both structures. The angles of twist between the thiophen rings and the ethylenic groups are $3^{\circ} 10' (\gamma)$ and $0^{\circ} (\beta)$,



FIGURE 2 Bond angles (degrees)

and between the ethylene and carboxylic groups $1^{\circ}\,50'$ (γ) and $4^{\circ}\,25'$ $(\beta).$

As in furylacrylic acid¹ the exocyclic and endocyclic [C(4)--C(5)] double bonds are in the *trans*-position; the

TABLE 4 Comparison of bond lengths and angles

Thienylacry (averag	ylic acid ged)	Thiophen-2-carb- oxylic acid ⁸	Thiophen-3-carb- oxylic acid ⁹
C(1) - C(2)	1.47- Å	5	
C(2) = C(3)	1.32_{5}		
C(3) - C(4)	1.44		
C(4) = C(5)	1.367	1.362	1.406
C(5) - C(6)	1.42	1.414	1.437
C(6) = C(7)	1.32_{7}	1.363	1.508
C(7)-S	1.70_{3}	1.701	1.699
C(4)-S	1.72_{4}	1.693	1.708
SC(4)C(5)	111·4°	111·8°	110·4°
C(4)C(5)C(6)	111.6	112.4	112.5
C(5)C(6)C(7)	112.6	111.9	113.3
C(6)C(7)S	113.3	111.8	106.0
C(7)SC(4)	91.0	92.0	98.0

cis-conformation of the exocyclic >C=C< and the carboxylic >C=O groups follows the rule established in Part XI.¹⁰

Packing Arrangement.—No abnormally short intermolecular contacts have been found; distances less than 3.6 Å are given in Tables 6 (a) and (b). OH \cdots O distances of 2.57 Å (γ) and 2.64 Å (β) are normal.

The packing arrangements of the two forms follow the pattern established in the cinnamic acid series. The stable form consists of two stacks of centrosymmetric, nearly planar, hydrogen-bonded molecule pairs arranged in parallel along the b axis, the two sets being related by the twofold screw axis parallel to b. The mean molecular plane is inclined at 62° to the stack axis; as a result of this steep inclination intrastack contacts between carbon atoms C(2) and C(3) of the exocyclic double bond are too long for interaction; consequently,

TABLE 5

		(a) γ-mo	onncati	on		
	(i) E	quations	of best	planes *		
		- m ₁	m	2 11	1 ₃	d
Plane A:		-		-	0	
C(4), C(5), C(6)	6),C(7)	. 5.990	6 3.4	961 - 7	3230	3.0382
Plane B:						
C(4),C(5),C(6	6),C(7),S	-5.971	5 - 3.4	874 7.	3735 -	-3.0144
Plane C:						
C(1),C(2),C(3	3),C(4)	. 6.432	2 3.4	570 6·	9706	3.1929
Plane D:						
C(1),C(2),O(1),O(2)	. 6.145	1 3.3	733 - 7.	5890	3.0357
	(ii) Dev	viations (.	Å) from	best plan	.es	
Plane		Plane	,	Plane		Plane
Atom A	Atom	в	Atom	С	Atom	D
C(4) 0.010	C(4)	-0.013	C(1)	-0.009	C(1)	-0.008
C(5) - 0.016	C(5)	0.018	C(2)	0.008	C(2)	0.002
C(6) 0.016	C(6)	-0.014	C(3)	0.010	O(1)	0.003
C(7) = -0.010	C(7)	0.006	C(4)	-0.009	O(2)	0.003
	S	0.004	.,			
S -0.014	C(3)	-0.005	O(1)	-0.070	C(3)	-0.064
C(3) - 0.004	()		O(2)	0.062		

(b) β -Modification

(i) Equations of best planes *

Diana Ar		m_1	m_2	11	n ₃	d	
Flane A:		4 3 6 5 6		0.00			
C(4), C(5), C(6)),C(7)	. 4.1075	-3.5	333 - 2	7053 -	-0.2377	
Plane B:							
C(4), C(5), C(6)),C(7),S	-4.1039	$3 \cdot 5$	340 2·	6982	0.2363	
Plane C:							
C(1),C(2),C(3)	,C(4)	-5.0566	3.3	209 3·	0865	0.0871	
Plane D:							
C(1),C(2),O(1)),O(2)	4.4883	-3.4	$443 - 1 \cdot$	8058 -	-0.0346	
(ii) Deviations (Å) from best planes							
Plane		Plane		Plane		Plane	
Atom A	Atom	В	Atom	C	Atom	D	
C(4) = -0.004	C(4)	0.005	C(1)	-0.017	C(1)	0.004	
C(5) 0.007	C(5)	-0.007	$\tilde{C}(2)$	0.015	$\tilde{C}(2)$	-0.001	
C(6) - 0.007	C(6)	0.007	$\tilde{C}(3)$	0.021	O(1)	-0.005	
C(7) 0.004	C(7)	0.004	C(4)	-0.018	O(2)	-0.005	
	S	-0.000	• /		()		
S 0.001	C(3)	-0.024	O(1)	-0.116			
C(3) 0·024	()		O(2)	0.059	C(3)	0.097	
* Compute Marsh, and E	d accor 2. Bergn	ding to V nan, Acta	Cryst.,	maker, J 1959, 12 ,	. Wase 600.	r, R. E.	

this form of the acid belongs to the γ -type and should be light-stable. We report elsewhere ² that, in accordance with these crystallographic results, this form does not yield dimer. However, the material forms a photopolymer: if we admit, here as in the instance of furylacrylic acid,¹ that the excited exocyclic >C=C< group can react with the double bonds of the hetero-ring, and

¹⁰ L. Leiserowitz and G. M. J. Schmidt, Acta Cryst., 1965, 18, 1058.

postulate further that polymerisation is initiated by an attack of the exocyclic on an endocyclic double bond, the crystal structure must show sufficiently close contacts ($<4\cdot3$ Å) between C(2) or C(3) and the carbon

TABLE 6

Short intermolecular contacts < 3.6 Å

	(a) γ-Form			(b) β-Form					
A ₀₀₀ C(1) C(3)	A _{0Ĩ0} C(4) C(6)	3∙46 3∙49	A ₀₀₀ O(1) O(1)	A ₀₁₀ O(2) O(2)	$3.58 \\ 3.51$				
C(1) C(2)	A ₁₁₀ O(2) O(2)	3∙48 3∙60	C(4) C(6)	A _{0ī0} S C(7)	3∙57 3∙57				
s	А ₀₂₀ S	3.46	C(6) C(5)	B ₀₀₀ O(2) O(2)	$3.51 \\ 3.30$				
C(5)	В ₁₀₀ О(2)	3.49	C(6)	Б ₀₀₀ О(1)	3.56				
C(5) C(6)	Б ₀₁₀ О(1) О(1)	$3.36 \\ 3.34$	C(2)	В _{1ї0} S	3.54				
	Co-ordinates of equivalent positions Symbol								
	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$								
Appr 6	A _{yer} denotes fractional co-ordinates $p + x$, $q + y$, $r + z$.								

atoms of the hetero-ring. That such contacts indeed exist is shown by Table 7(a).

The metastable form can be described as consisting of two stacks of nearly planar hydrogen-bonded molecules arranged in parallel along the short b axis, the two stacks being related by the twofold screw axis parallel to b. The mean molecular plane is inclined at 27° to the stack axis. As a result of this arrangement, and by analogy with the cinnamic acids crystallising in the 4 Å packing-type, the exocyclic >C=C< groups should, on irradiation, give the dimer of symmetry m(II), which has indeed been isolated in high yield.

However, the β -modification also yields polymer, though in small amounts. It has so far not been possible to establish unambiguously whether the polymer arises indirectly from the thermodynamically stable modification via a (thermal) phase transformation or whether it is formed in the β -lattice. Analysis of the distances from C(2) and C(3) to the carbon atoms of the hetero-ring shows in the β -, as in the γ -form, sufficiently numerous and short distances (Table 7b) to permit initiation of polymerisation by the mechanism postulated above, so that direct polymerisation in the β -structure is geometrically at least possible. The modification is thus seen to resemble in its packing arrangement and photo-behaviour the furylacrylic acid analysed in Part XVI.

In conclusion we draw attention to the somewhat unusual situation whereby the metastable modification is significantly denser than the stable form (Table 1). This problem will be discussed in detail elsewhere; here we note that in the β -form each sulphur atom is

TABLE 7
Intermolecular contacts (<4.3 Å) between carbon atoms
of exocyclic and endocyclic double bonds

			5		
	(a) γ-Form			(b) β -Form	ı
$A_{000} *$	Aoio		A_{000}	A ₀₁₀	
C(2)	C(4)	3.73	C(2)	C(3)	3.73
C(2)	C(5)	3.74	C(4)	C(5)	3.79
C(2)	C(6)	3.71			
C(2)	C(7)	3.73		Aoio	
C(3)	C(4)	4.12	C(3)	C(4)	3.96
C(3)	C(5)	3.72	C(4)	C(7)	4.14
C(3)	C(6)	3.49	C(5)	C(6)	3.83
C(3)	C(7)	3.85	C(5)	C(7)	3.69
C(4)	C(6)	3.74	C(6)	C(7)	3.57
C(4)	C(7)	4.02		Ā	
C(5)	C(6)	4.27	C(0)	A 101	
	ង		C(6)	C(6)	3.90
C(B)	C(7)	9.76		C(7)	3.80
C(0)	C(7)	3.60	C(I)	C(I)	4.71
C(I)	C(I)	3.09		Ā.,,	
	Bain		C(6)	C(7)	4.11
C(6)	C(7)	3.96	$\tilde{C}(7)$	C(7)	3.98
Č(7)	$\tilde{C}(7)$	3.69	0(1)	0(1)	0.00
-(.)	- (.,	0.00		B_{000}	
	$\mathbf{B}_{01\overline{1}}$		C(3)	C(5)	4.08
C(2)	C(5)	3.88	. ,	• •	
C(2)	C(6)	4.03		Boio	
			C(3)	C(5)	4.23
			(-)		
				$B_{1\overline{1}0}$	
			C(2)	C(7)	4.17

* For definition of symbols see Table 6.

close-packed to four other sulphurs: of these two make contact of 3.91 Å along the *b* translation while two other contacts of 3.89 Å are generated by the twofold screw axis along *b*. The stable but less dense form shows only one rather short contact of 3.46 Å between sulphur atoms related by a centre of symmetry.

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