

dried. Evaporation of the solvent yielded a thick liquid (2.2 g, 86%). ( $C_{14}H_{20}O_3$  requires: C, 71.15; H, 8.53; Found C, 71.17; H, 8.51%). IR  $\nu_{\max}^{CHCl_3}$ ,  $cm^{-1}$ : 1680, 1460, 1380, 1360, 1310, 1290, 1060.  $^1H$  NMR:  $\delta$  1.15 (d,  $J=6.5$  Hz, 6H), 1.36 (s, 12H), 3.52 (m, 1H), 7.28 (d,  $J=1$  Hz, 1H).

**Conversion of alkylidene 5 to dienol 6.** The alkylidene (2 g) was dissolved in benzene (50 ml) and refluxed under  $N_2$  for 1 hr. Evaporation of the  $C_6H_6$  under red. pres. on a rotavapor under  $N_2$  yielded a white solid (2 g), mp  $71^\circ$ . ( $C_{14}H_{20}O_3$  requires: C, 71.15; H, 8.53. Found C, 71.10; H, 8.55%). IR  $\nu_{\max}^{CHCl_3}$ ,  $cm^{-1}$ : 3450, 1680, 1640, 1590, 1460, 1380, 1360, 1320, 1230, 1210, 1080, 1010.  $^1H$  NMR:  $\delta$  1.48 (s, 12H), 1.68 (br s, 3H), 1.96 (br s, 3H), 5.72 (m, 1H).

**Reaction of dienol (6) with diazomethane.** The dienol (1 g) was treated with an excess of an ethereal soln of  $CH_2N_2$  and, after 30 min at  $0^\circ$ , the evaporation of solvent yielded a thick yellow liquid (1 g). ( $C_{15}H_{22}O_3$  requires: C, 71.97; H, 8.86. Found C, 71.95; H, 8.88%). IR  $\nu_{\max}^{Nujol}$ ,  $cm^{-1}$ : 1680, 1646, 1585, 1455.  $^1H$  NMR:  $\delta$  1.44 (s, 3H), 1.93 (s, 3H), 1.56 (br s, 3H), 1.92 (br s, 3H), 3.8 (s, 3H), 5.84 (m, 1H).

**Reaction of dienol (6) with oxygen.** A soln of dienol (2 g) in  $C_6H_6$  (50 ml) was stirred under an atmosphere of  $O_2$ . After 30 min, the evapn of solvent gave a white crystalline solid which showed IR, NMR and mass spectra superimposable with those of an authentic sample of  $G_3$  ( $R_1=R_2=Me$ ). By this method were also obtained  $G_1$  ( $R_1=Me$ ,  $R_2=Et$ ) and  $G_2$  ( $R_1=Et$ ,  $R_2=Me$ ).

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## ACYLRESORCINOLS FROM SEED KERNELS OF *MYRISTICA DACTYLOIDES*

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**Key Word Index**—*Myristica dactyloides*; Myristicaceae; seeds; polyketides;  $^{13}C$  NMR.

**Abstract**—A new polyketide 1-(2,6-dihydroxyphenyl)-9-(4-hydroxy-3-methoxyphenyl)nonan-1-one and five other polyketides 1-(2,6-dihydroxyphenyl)tetradecan-1-one and malabaricones A–D have been isolated from *Myristica dactyloides* seeds.  $^{13}C$  NMR of the first and the second mentioned compounds are also reported for the first time.

## INTRODUCTION

*Myristica dactyloides* is used in native medicine [1] in Sri Lanka and its seeds and aril are used as adulterants to *M. fragrans* (nutmeg). Previously only myoinositol [2], malabaricone A (2) and 1-(2,6-dihydroxyphenyl)tetradecan-1-one (1) [3] have been isolated from the bark of *M. dactyloides*. We now report the isolation of six acylresorcinols (1–6) from seeds of *M. dactyloides*.

## RESULTS AND DISCUSSION

Column chromatography of an acetone seed extract of *M. dactyloides* yielded six phenols. All six compounds had similar UV spectra, showed bathochromic shifts with alkali and gave a violet colouration with fast blue B salt, thus confirming their phenolic nature.

The peak around  $3360\text{ cm}^{-1}$  in the IR spectrum confirmed the presence of a hydrogen-bonded hydroxyl.

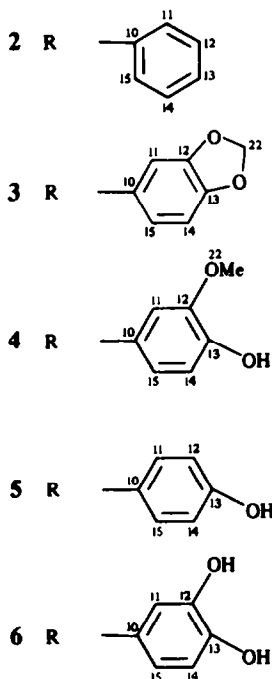
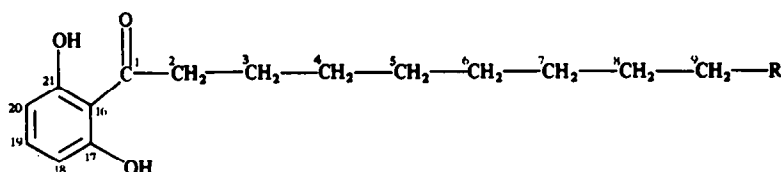
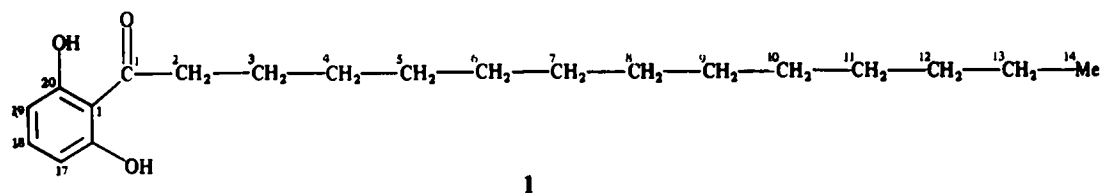
A base peak at ( $m/z$ ) 137 and a less intense peak at 109 in the mass spectrum and a sharp band around  $1640\text{ cm}^{-1}$  in the IR spectrum confirmed the presence of a 2,6-dihydroxyacetophenone moiety in all the compounds isolated.

The molecular formula of **1** was confirmed to be  $\text{C}_{20}\text{H}_{32}\text{O}_3$  [ $M^+$  ( $m/z$ ) 320.2356; found, C, 74.89%; H, 9.96%].  $^1\text{H}$  NMR showed the presence of three aromatic protons [ $\delta$  7.22 (1H,  $t$ ,  $J = 8\text{ Hz}$ ) and 6.38 (2H,  $d$ ,  $J = 8\text{ Hz}$ )] and hence the presence of one benzene ring. Further  $J = 8\text{ Hz}$  showed *ortho* coupling of the hydrogens. The triplet at  $\delta$  3.12 indicated that a  $-\text{CH}_2-\text{CH}_2-$  group is attached to a carbonyl group. The distorted triplet at  $\delta$  0.88 indicated the presence of a  $-\text{Me}$  group connected to an  $n$ -alkyl group and other signals  $\delta$  1.69 (2H,  $m$ ,  $-\text{CH}_2-$ ) and 1.27 (2OH,  $m$ ,  $-\text{CH}_2-$ ) account for 22H. The  $^{13}\text{C}$  NMR spectrum of this compound was in agreement with the structure 1-(2,6-dihydroxyphenyl)tetradecan-1-one (**1**).

Compounds **2**, **3**, **5** and **6** were identified as malabaricones A, D, B and C, respectively from their UV, IR,  $^1\text{H}$  NMR, MS data, mmps and co-TLC.

Compound **4** which is a new acylresorcinol had a molecular formula  $\text{C}_{22}\text{H}_{28}\text{O}_5$  [ $M^+$  ( $m/z$ ) 372.1947; found: C, 71.03%; H, 7.48%]. IR(KBr) supporting the presence of one free and one hydrogen bonded hydroxyl ( $3450$  and  $3340\text{ cm}^{-1}$ ) and hydrogen bonded carbonyl group.  $^1\text{H}$  NMR revealed two benzene rings one 1,2,3-trisubstituted [ $\delta$  7.20 (1H,  $t$ ,  $J = 8\text{ Hz}$ ), 6.37 (2H,  $d$ ,  $J = 8\text{ Hz}$ )] and the other 1,3,4 trisubstituted  $\delta$  6.98 (1H,  $d$ ,  $J = 8\text{ Hz}$ ), 6.73 (1H,  $d$ ,  $J = 2\text{ Hz}$ ), 6.65 (1H,  $dd$ ,  $J = 8$  and  $2\text{ Hz}$ ), singlets at  $\delta$  3.80, 5.60 and 9.50 correspond to a methoxy group, a free hydroxyl and two hydrogen bonded hydroxyl groups.

The base peak in the HRMS at  $m/z$  137.0605 ( $\text{C}_8\text{H}_9\text{O}_2$ ) and acetylvanillic acid (mmp and co-TLC) in the oxidative products of acetylated **4** indicated the presence of a 4-hydroxy-3-methoxybenzylic moiety in [3] the second ring. The  $^{13}\text{C}$  NMR of **4** was similar to that



of **6** except for a (–) signal at  $\delta$ 5.60 due to a methoxy group.

## EXPERIMENTAL

Mps: uncorr.  $^1\text{H}$ NMR was run at ( $\text{CD}_2\text{Cl}_2$ ) 100 MHz and 400 MHz and  $^{13}\text{C}$  NMR spectra ( $\text{CD}_2\text{Cl}_2$ ) at 74.2 MHz. HRMS,  $^{13}\text{C}$  NMR and micro-analysis were performed at the Department of Chemistry, University of British Columbia, Canada.

*Isolation and identification of phenolics.* The dried powdered seeds (500 g) of *M. dactyloides* Gaertn., from Hanguranketa, Sri Lanka, were extracted with  $\text{Me}_2\text{CO}$ . The concd  $\text{Me}_2\text{CO}$  extract was defatted with petrol giving a viscous brown solid (165 g), which after CC on silica gel using petrol:EtOAc mixtures of increasing proportions yielded compounds 1–6, which were purified by prep. TLC.

1-(2,6-Dihydroxyphenyl)tetradecan-1-one (**1**). Needles mp, 91–91.5° (petrol, lit [3] 91–92°); Found: C, 74.89%; H, 9.96%;  $\text{C}_{20}\text{H}_{32}\text{O}_3$  requires: C, 74.96%; H, 10.06%; UV  $\lambda_{\text{max}}^{\text{EtOH}}$  nm, 223, 268, 339,  $\lambda_{\text{max}}^{\text{EtOH/OH}^-}$  nm 239, 283, 390; IR  $\nu_{\text{max}}^{\text{KBr}}$  3360 (br), 1640 (s)  $^1\text{H}$  NMR (400 MHz,  $\text{CD}_2\text{Cl}_2$ )  $\delta$ 9.37 (2H, br s, OH-16, 20  $\text{D}_2\text{O}$  exchangeable), 7.22 (1H, t,  $J$  = 8 Hz, H-18), 6.38 (2H, d,  $J$  = 8 Hz, H-17, 19), 3.12 (2H, t,  $J$  = 7 Hz, H-2), 1.69 (2H, m, H-3), 1.27 (20H, m, H-4–13), 0.88 (3H, t,  $J$  = 7 Hz, H-14); HRMS,  $m/z$  (rel. int. %) 320.2356 ( $\text{C}_{20}\text{H}_{32}\text{O}_3$ , 10), 302.2246 ( $\text{C}_{20}\text{H}_{30}\text{O}_3$ , 17), 189.0920 ( $\text{C}_{12}\text{H}_{13}\text{O}_2$ , 17), 165.0553 ( $\text{C}_9\text{H}_9\text{O}_3$ , 26), 152.0473 ( $\text{C}_8\text{H}_8\text{O}_3$ , 31), 137.0247 ( $\text{C}_7\text{H}_5\text{O}_3$ , 100)  $^{13}\text{C}$  NMR assignments are given in Table 1.

Malabaricone A (**2**). Crystals (petrol) mp, 80–82° (lit. [4] 81–82°).

Malabaricone D (**3**). Pale yellow crystals (toluene) mp, 89–91° (lit. [4] 90–91°).

1-(2,6-Dihydroxyphenyl)-9-(4-hydroxy-3-methoxyphenyl)nonan-1-one (**4**). White needles (toluene) mp 109–111°; found: C, 71.03%; H, 7.48%;  $\text{C}_{22}\text{H}_{28}\text{O}_5$  requires: C, 70.95%; H, 7.58%; UV  $\lambda_{\text{max}}^{\text{EtOH}}$  nm 340, 272, 269, 220  $\lambda_{\text{max}}^{\text{EtOH/OH}^-}$  nm 386, 283, 238; IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$  3450 (s), 3340 (br), 1635 (s),  $^1\text{H}$  NMR ( $\text{CD}_2\text{Cl}_2$  run at 400 MHz)  $\delta$ 9.50 (2H, s, OH-17, 21,  $\text{D}_2\text{O}$  exchangeable), 7.20 (1H, t,  $J$  = 8 Hz, H-19), 6.98 (1H, d,  $J$  = 8 Hz, H-15), 6.73 (1H, d,  $J$  = 2 Hz, H-11), 6.65 (1H, dd,  $J$  = 2 and 8 Hz, H-14), 6.37 (2H, d,  $J$  = 8 Hz, H-20, 18), 5.60 (1H, s,  $\text{D}_2\text{O}$  exchangeable, OH-13), 3.80 (3H, s, H-22), 3.00 (2H, t,  $J$  = 7 Hz, H-2), 2.44 (2H, t,  $J$  = 7 Hz, H-9), 1.64 (4H, m, H-3, 8), 1.32 (8H, m, H-4–7); HRMS  $m/z$  (rel. int. %) 372.1947 ( $\text{C}_{22}\text{H}_{28}\text{O}_5$ , 52), 262.1568 ( $\text{C}_{16}\text{H}_{22}\text{O}_3$ , 21), 234.1618 ( $\text{C}_{15}\text{H}_{22}\text{O}_4$ , 15), 165.0548 ( $\text{C}_9\text{H}_9\text{O}_3$ , 22), 137.0605 ( $\text{C}_8\text{H}_9\text{O}_2$ , 100), 137.0244 ( $\text{C}_7\text{H}_5\text{O}_3$ , 79);  $^{13}\text{C}$  NMR assignments are given in Table 2.

Table 2.  $^{13}\text{C}$  NMR assignment (attached proton test) of compound **4**

| Carbon no. | $\delta$ (ppm) | Intensity |
|------------|----------------|-----------|
| 1          | 207.95         | 6         |
| 2          | 44.71          | 41        |
| 3          | 24.37          | 46        |
| 4          | 29.12          | 47        |
| 5, 6       | 29.02          | 69        |
| 7          | 28.75          | 45        |
| 8          | 31.25          | 45        |
| 9          | 35.08          | 61        |
| 10         | 136.35         | 12        |
| 11         | 110.58         | –48       |
| 12         | 145.05         | 8         |
| 13         | 144.52         | 5         |
| 14         | 114.57         | –41       |
| 15         | 119.93         | –39       |
| 16         | 110.05         | 6         |
| 17, 21     | 161.18         | 15        |
| 18, 20     | 108.42         | –101      |
| 19         | 135.62         | –50       |
| 22         | 56.01          | –19       |

372.1947 ( $\text{C}_{22}\text{H}_{28}\text{O}_5$ , 52), 262.1568 ( $\text{C}_{16}\text{H}_{22}\text{O}_3$ , 21), 234.1618 ( $\text{C}_{15}\text{H}_{22}\text{O}_4$ , 15), 165.0548 ( $\text{C}_9\text{H}_9\text{O}_3$ , 22), 137.0605 ( $\text{C}_8\text{H}_9\text{O}_2$ , 100), 137.0244 ( $\text{C}_7\text{H}_5\text{O}_3$ , 79);  $^{13}\text{C}$  NMR assignments are given in Table 2.

*Oxidation of 4.* Compound **4** (75 mg) was heated in HOAc (0.3 ml) with one drop of concd  $\text{H}_2\text{SO}_4$  at 60° for 15 min. Acetylated **4** (30 mg) in  $\text{Me}_2\text{CO}$  was stirred with a soln of  $\text{KMnO}_4$  (0.2 g) in  $\text{H}_2\text{O}$  (0.5 ml) and  $\text{Me}_2\text{CO}$  (2 ml), decolourized with  $\text{NaHSO}_3$  in dil.  $\text{H}_2\text{SO}_4$  and extracted with  $\text{Et}_2\text{O}$ . Prep. TLC of the extract on silica gel with the upper layer of toluene–HOAc– $\text{H}_2\text{O}$  (2:3:1) [5] gave acetylvannillic acid which was characterized by mmp and co-TLC with an authentic sample.

Malabaricone B (**5**). Pale yellow crystals (toluene) mp 100–102° (lit. [4] 102°).

Malabaricone C (**6**). Yellow crystals (toluene) mp 122–124° (lit. [4] 123–124°).

Table 1.  $^{13}\text{C}$  NMR assignment (attached proton test) of compound **1**

| Carbon no. | $\delta$ (ppm) | Intensity |
|------------|----------------|-----------|
| 1          | 207.79         | 6         |
| 2          | 44.82          | 32        |
| 3          | 24.41          | 48        |
| 4–11       | 29.66          | 90        |
|            | 29.55          | 75        |
|            | 29.40          | 52        |
|            | 29.37          | 48        |
|            | 31.93          | 17        |
| 12         | 22.70          | 23        |
| 13         | 14.13          | –14       |
| 14         | 110.03         | v. weak   |
| 16, 20     | 161.03         | 15        |
| 17, 19     | 108.52         | –100      |
| 18         | 135.57         | –52       |

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