

# A CONVENIENT SYNTHESIS OF BENZO[C]PHENANTHRIDINE ALKALOID, CHELERYTHRINE, BY THE PALLADIUM-ASSISTED INTERNAL BIARYL COUPLING REACTION

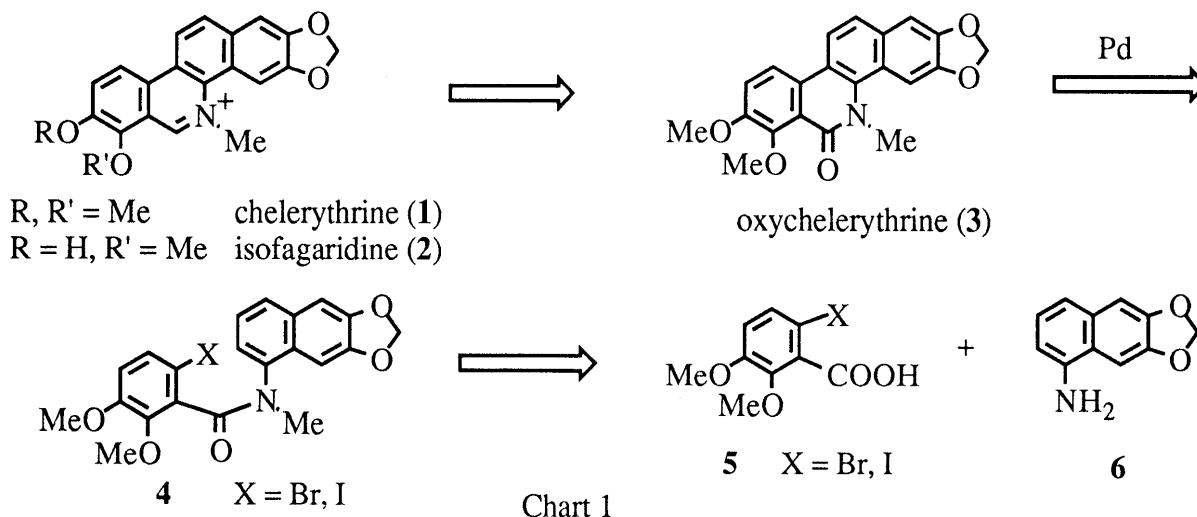
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Total synthesis of chelerythrine, a benzo[c]phenanthridine alkaloid, was accomplished *via* the internal aryl-aryl coupling reaction of halo-amide (4) by the palladium-assisted cyclization reaction.

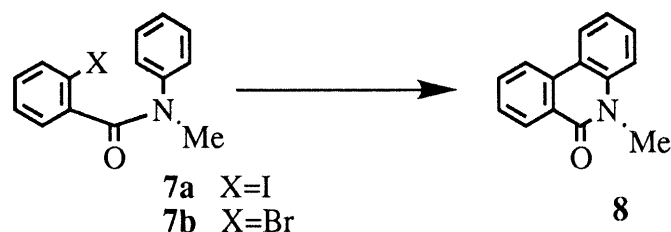
**KEY WORDS** internal biaryl coupling; Heck type reaction; halo-amide; benz[c]phenanthridine alkaloid; chelerythrine

Benzo[c]phenanthridine alkaloids have attracted much attention because of their potent pharmacological activities.<sup>1)</sup> Recently, it was reported that chelerythrine (1)<sup>2)</sup> and isofagaridine (2)<sup>3)</sup> inhibited protein kinase C and DNA topoisomerase I, respectively. Although several reports on the total synthesis of chelerythrine have been published,<sup>4)</sup> the methods had some disadvantages such as many steps, low yield, and/or no generality. Therefore, we planned to develop a convenient synthetic method for 1.



The cross-coupling reaction with palladium catalyst has been an extremely useful tool in organic synthesis.<sup>5)</sup> We designed the synthetic plan for 1 involving an internal biaryl cyclization by palladium as a key reaction,<sup>6)</sup> as shown in Chart 1. Since the coupling product (3), oxychelerythrine, had already been converted into chelerythrine (1),<sup>4e)</sup> the synthesis of 3 indicates a formal synthesis of 1. It was reported that an internal coupling reaction of bromo-amide (7b) with a Pd reagent proceeded in 50% yield.<sup>6a)</sup> In order to improve the yield, the reaction was re-examined using purified  $\text{Pd}(\text{OAc})_2$ ,<sup>7)</sup> phosphine ligand and base. The results are summarized in Table 1. On

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**Table 1.** Results of Cyclization Reaction of 2-Halo-*N*-methyl-*N*-phenylbenzamide (**7**)<sup>a)</sup>

X	Run	Pd(OAc) <sub>2</sub> (eq.)	Ligand	Base	Solvent	Temp.	Time	Yield (%)	
								<b>8</b>	S.M.
I	1	0.05	PPh <sub>3</sub>	Ag <sub>2</sub> CO <sub>3</sub>	DMF	Refl.	40 min	79	—
	2	0.2	PPh <sub>3</sub>	Ag <sub>2</sub> CO <sub>3</sub>	DMF	Refl.	15 min	93	—
	3	0.2	POT	Ag <sub>2</sub> CO <sub>3</sub>	DMF	Refl.	15 min	93	—
	4	0.2	PPh <sub>3</sub>	Ag <sub>2</sub> CO <sub>3</sub>	DMF	30-35°C	35 h	85	—
	5	0.2	PPh <sub>3</sub>	Ag <sub>2</sub> CO <sub>3</sub>	Xylene	30-35°C	23 h	93	—
	6	0.2	PPh <sub>3</sub>	Ag <sub>2</sub> CO <sub>3</sub>	Benzene	Refl.	10 min	98	—
	7	0.2	PPh <sub>3</sub>	Ag <sub>2</sub> CO <sub>3</sub>	CH <sub>3</sub> CN	Refl.	15 min	95	—
	8	0.2	PPh <sub>3</sub>	<i>i</i> Pr <sub>2</sub> NEt	DMF	Refl.	4.5 h	21	7
	9	0.2	PPh <sub>3</sub>	<i>i</i> Pr <sub>2</sub> NEt	Benzene	Refl.	6 h	45	14
Br	10	1.0	PPh <sub>3</sub>	Ag <sub>2</sub> CO <sub>3</sub>	DMF	Refl.	60 h	75	7
	11	0.2	POT	Ag <sub>2</sub> CO <sub>3</sub>	DMF	Refl.	1.5 h	99	—

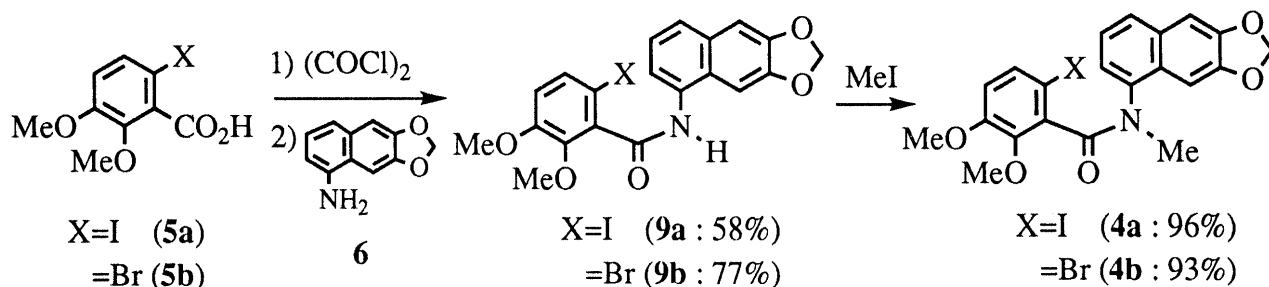
a) All reactions were carried out using Pd(OAc)<sub>2</sub> and ligand in a ratio of 1 : 2 and 2 mol equivalent of base.

Chart 2

**Table 2.** Results of Cyclization Reaction of 6-Halo-2,3-dimethoxy-*N*-methyl-*N*-(6,7-methylenedioxy-1-naphthyl)benzamide (**4**) to Oxachelerythrine (**3**)<sup>a)</sup>

X	Run	Pd(OAc) <sub>2</sub> (eq.)	Ligand	Base	Solvent	Temp.	Time	Yield (%)
								<b>3</b>
I	1	0.2	PPh <sub>3</sub>	Ag <sub>2</sub> CO <sub>3</sub>	DMF	Refl.	20 min	85
	2	0.2	POT	Ag <sub>2</sub> CO <sub>3</sub>	DMF	Refl.	20 min	94
Br	3	0.2	PPh <sub>3</sub>	Ag <sub>2</sub> CO <sub>3</sub>	DMF	Refl.	2 h	79
	4	0.2	POT	Ag <sub>2</sub> CO <sub>3</sub>	DMF	Refl.	3 h	96

a) All reactions were carried out in the presence of ligand (0.4 eq) and base (2 eq).

using 0.2 eq of  $\text{Pd}(\text{OAc})_2$ ,  $\text{PPh}_3$ , and  $\text{Ag}_2\text{CO}_3$ , the solvent had no crucial effect on the internal coupling reaction of iodo-amide (**7a**), although  $\text{Ag}_2\text{CO}_3$  was superior to Hünig base. On the other hand, the coupling reaction of bromo-amide (**7b**) proceeded slowly even when using a stoichiometric amount of  $\text{Pd}(\text{OAc})_2$  in the presence of  $\text{PPh}_3$  as a ligand in DMF (see run 10 in Table 1). However, on using tris(2-methylphenyl)phosphine (POT) as ligand, the reaction proceeded smoothly using 0.2 eq of  $\text{Pd}(\text{OAc})_2$  and gave phenanthridone (**8**) in an excellent yield (see run 11 in Table 1).

Next, we applied these methods to the synthesis of oxychelerythrine (**3**) from halo-amides (**4**), which were prepared from iodo-acid (**5a**)<sup>8</sup> or bromo-acid (**5b**)<sup>9</sup> and naphthylamine (**6**)<sup>10</sup> as shown in Chart 2. The results are summarized in Table 2. The coupling reaction of both halo-amides (**4**) with  $\text{Pd}(\text{OAc})_2$ ,  $\text{PPh}_3$  or POT and  $\text{Ag}_2\text{CO}_3$  in DMF under reflux afforded oxychelerythrine (**3**) in an excellent yield, although iodo-amide (**4a**) was more reactive than bromo-amide (**4b**).

In conclusion, the present method using the Pd reagent is very convenient and effective for preparing benzo[c]phenanthridine alkaloids. We are now investigating the generality of this method.

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