

Efficient Desulfurization of 2-Thiopyrimidine Nucleosides to the Corresponding 4-Pyrimidinones

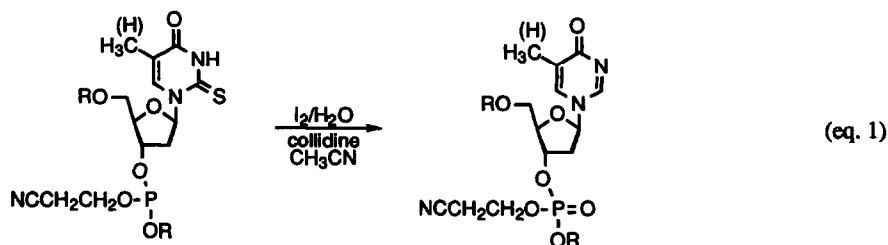
Robert G. Kuimelis and Krishnan P. Nambiar*

Department of Chemistry, University of California, Davis
Davis, California 95616

Abstract: A procedure is described for the preparation of 2'-deoxy-4-pyrimidinone (dH²U) and 2'-deoxy-5-methyl-4-pyrimidinone (dH²T) nucleosides. The key transformation is a nearly quantitative desulfurization of the corresponding 2-thio analogue by a brief treatment with a *m*-chloroperbenzoic acid/pyridine solution. The phosphoramidites of these nucleosides have also been synthesized.

Ligand and protein/DNA interactions can be studied by incorporating nucleoside analogues into DNA sequences at predetermined positions.¹ Hence there is growing interest in the synthesis of such molecules and their incorporation into oligodeoxynucleotides. The 4-pyrimidinones are a class of nucleoside analogues lacking both the N³-amide hydrogen that participates in Watson-Crick base pairing and the O²-oxygen situated in the minor groove of B-type DNA.² Although the O²-oxygen does not participate in base pairing, it would normally serve as a hydrogen bond acceptor while interacting with ligands and proteins.³ In addition, pyrimidinone nucleotides are susceptible to mild acid hydrolysis, thus generating abasic sites at unique positions.^{4,5} A recent report⁶ describes the desulfurization of 2-thiothymidine (dS²T) with Raney-nickel to give dH²T in moderate yield. In this communication, we describe a more efficient and general procedure to effect such a transformation.

While attempting to prepare synthetic oligodeoxynucleotides containing 2-thiopyrimidines, we found the thio-carbonyl moiety to be sensitive to the aqueous iodine oxidation reagent used in the phosphoramidite⁷ DNA synthesis method.⁸ Further investigation revealed that the major product was the 4-pyrimidinone nucleotide formed via desulfurization (eq. 1).^{9,10} Oxidative desulfurization of heterocycles has previously been observed.¹¹



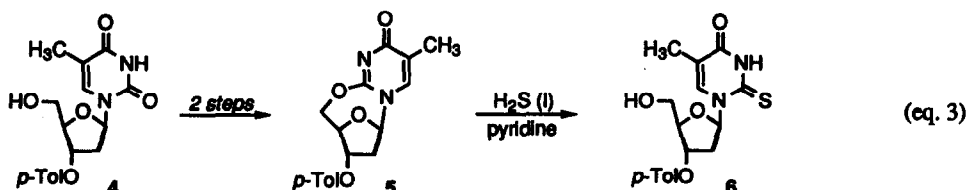
During the evaluation of alternative oxidizing agents, we discovered that treatment of 2-thiopyrimidine nucleosides with *m*-chloroperbenzoic acid/pyridine solutions¹² rapidly produced the 4-pyrimidinone nucleosides in high yield. This desulfurization method represents a significant improvement over current methods. Raney-nickel reductions^{6,13} proceed in only moderate yields (30-61%) and are not amenable to post-synthetic modifications of oligodeoxynucleotides. Ogihara and Mitsunobu's dipotassium diazenedicarboxylate treat-

ment and related procedures¹⁴ are quite sluggish (one week) and are not suitable for large-scale preparations⁶. Direct glycosylation methods generally afford complex mixtures of regio- and stereo-isomers.¹⁵

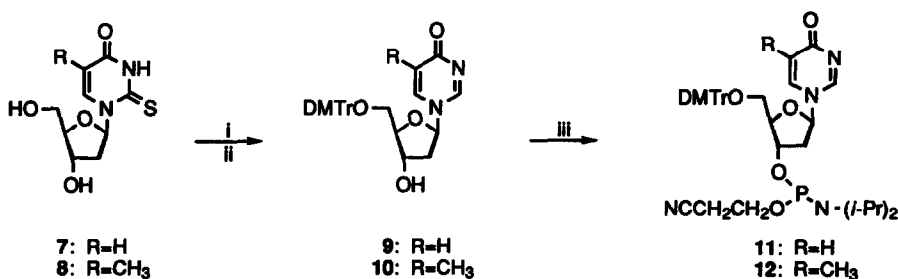
We have used a modification^{9,16} of the silyl Hilbert-Johnson reaction¹⁷ to synthesize β -2'-deoxy-3',5'-*p*-toluoyl-2-thiouridine (3)¹⁸ in 94% isolated yield from the α -chlorosugar (eq. 2). 3'-*p*-Toluoyl-2-thiothymidine



(6)¹⁹ was synthesized from thymidine by a variation of Faerber and Scheit's procedure²⁰ utilizing sulphydrolysis of the cyclic 2,5'-anhydro intermediate (5) with liquid H_2S /pyridine under pressure (eq. 3). The 2-thiopyrimidines were deprotected by transesterification with 0.1M NaOMe to give (7)²¹ and (8),²² which were



next treated with 4,4'-dimethoxytrityl chloride to give the 5'-*O*-4,4'-dimethoxytrityl ethers (scheme 1). After purification, treatment with 0.1M *m*-chloroperbenzoic acid in CH_2Cl_2 /pyridine for ten minutes afforded the 4-pyrimidinones **9**²³ and **10**²⁴ in 88% and 98% isolated yield, respectively. The diester **3** also undergoes a similar desulfurization in quantitative yield.²⁵ The work-up involves quenching the excess reagent with 10% aqueous NaHSO_3 , washing the organic layer with 5% aqueous NaHCO_3 , drying over Na_2SO_4 , and purifying by silica gel chromatography. Reaction of **9** and **10** with β -cyanoethyl-bis-(diisopropylamino)-phosphoramidite²⁶ and diisopropylammonium tetrazolid²⁷ gave the dH²U and dH²T phosphoramidites, **11**²⁸ and **12**²⁹, in good yield. The phosphoramidite **12** was recently used to synthesize an oligonucleotide.⁶



Scheme 1. Synthetic route for the preparation of the dH²U and dH²T phosphoramidites. Reaction conditions were as follows: i) 4,4'-dimethoxytrityl chloride, DMAP, pyridine; ii) 0.1M MCPBA in CH_2Cl_2 /pyridine (9:1), 10 min; iii) $(i\text{-Pr}_2\text{N})_2\text{POCH}_2\text{CH}_2\text{CN}$, diisopropylammonium tetrazolid, CH_2Cl_2 .

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REFERENCES AND NOTES

1. (a) Mascareñas, J.L.; Hayashibara, K.C.; Verdine, G.L. *J. Am. Chem. Soc.* **1993**, *115*, 373-374. (b) Loontjens, F.G.; McLaughlin, L.W.; Diekman, S.; Clegg, R.M. *Biochemistry* **1991**, *30*, 182-189. (c) Mazzarelli, J.M.; Rajur, S.B.; Iadarola, P.; McLaughlin, L.W. *Biochemistry* **1992**, *31*, 5925-5936.
2. Saenger, W. *Principles of Nucleic Acid Structure*; Springer-Verlag: New York, 1984.
3. (a) Pabo, C.O.; Sauer, R.T. *Ann. Rev. Biochem.* **1992**, *61*, 1053-1095. (b) Steitz, T.A. *Q. Rev. Biophys.* **1990**, *23*, 205-280.
4. Iacono, J.A.; Gildea, B.; McLaughlin, L.M. *Tetrahedron Lett.* **1990**, *31*, 175-178.
5. (a) Stuart, G.R.; Chambers, R.W. *Nucleic Acids Res.* **1987**, *15*, 7451-7462. (b) Manoharan, M.; Ransom, S.C.; Mazumder, A.; Gerlt, J.A.; Wilde, J.A.; Withka, J.A.; Bolton, P.H. *J. Am. Chem. Soc.* **1988**, *110*, 1620-1622.
6. Sharanabasava, B.R.; McLaughlin, L.W. *Tetrahedron Lett.* **1992**, *41*, 6081-6084.
7. Caruthers, M.H., in *Synthesis and Applications of DNA and RNA*; Narang, S.A., Ed; Academic: Orlando, 1987; pp 47-94.
8. This contradicts an earlier report stating that dS²T is unreactive toward the aqueous iodine oxidation reagent. See Connolly, B.A.; Newman, P.C. *Nucl. Acids Res.* **1989**, *17*, 4957-4974.
9. Kuimelis, R.G. Presented in part at the 203rd National Meeting of the American Chemical Society, San Francisco, CA, April 1992; paper ORGN 489.
10. McLaughlin recently reported the synthesis of a short oligonucleotide containing dS²T using iodine oxidation.⁶ Base composition analysis showed the presence of unknown nucleosides.
11. (a) Asinger, F.; Saus, A.; Offermanns, H.; Krings, P.; Andree, H. *Liebigs. Ann. Chem.* **1971**, *744*, 51-64. (b) Ikehara, M.; Ogiso, Y. *Tetrahedron* **1972**, *28*, 3695-3704.
12. (a) Ogilvie, K.K.; Nemer, M.J. *Tetrahedron Lett.* **1981**, *22*, 2531-2532. (b) Tanaka, T.; Lestinger, R.L. *Nucleic Acids Res.* **1982**, *10*, 3249-3260.
13. (a) Lee, H.J.; Wigler, P.W. *Biochemistry* **1968**, *7*, 1427-1431. (b) Ueda, T.; Shibuya, S. *Chem. Pharm. Bull.* **1974**, *22*, 930-937.
14. (a) Ogihara, T.; Mitsunobu, O. *Chem. Lett.* **1982**, 1621-1624. (b) Mitsunobu, O.; Ito, N.; Satta, S.; Ogihara, J.; Tamaoki, H.; Nagasawa, H.; Suzuki, H.; Kimura, J. *Tetrahedron Lett.* **1982**, *23*, 517-520.
15. (a) Niedballa, U.; Vorbrüggen, H. *J. Org. Chem.* **1974**, *39*, 3668-3671. (b) Seela, F.; Bindig, U. *Liebigs Ann. Chem.* **1989**, *9*, 895-901.
16. Kuimelis, R.G.; Hope, H.; Nambiar, K.P. *Nucleosides and Nucleotides*, in press.
17. Lukevics, E.; Zablocka, A. *Nucleoside Synthesis, Organosilicon Methods*; Ellis-Horwood: Chichester, 1991.
18. ¹H NMR (CDCl₃, 300.7 MHz) δ= 9.8(s,1H,NH) 7.9(dd,4H,Ar) 7.7(d,1H,H₆) 7.3(d,4H,Ar) 7.0(dd,1H,H_{1'}) 5.8(dd,1H,H₅) 5.6(m,1H,H_{3'}) 4.8(ddd,2H,H_{5'}) 4.6(d,1H,H_{4'}) 3.1(ddd,1H,H_{2'}) 2.4(d,6H,CH₃) 2.2(m,1H,H_{2''}); ¹³C NMR (CDCl₃, 75.5 MHz) δ= 174.3, 160.3, 144.6, 144.5, 139.4, 130.0, 129.9, 129.3, 129.1, 126.3, 126.2, 106.8, 89.8, 83.2, 74.0, 63.6, 38.1.

19. ^1H NMR ($\text{CDCl}_3 + \text{CD}_3\text{OD}$, 300.7 MHz) δ = 8.1(s,1H,H₆) 7.8(d,2H,Ar) 7.3(d,2H,Ar) 7.1(dd,1H,H_{1'}) 5.5(m,1H,H_{3'}) 4.3(dd,1H,H_{4'}) 3.9(d,2H,H_{5'}) 2.8(m,1H,H_{2'}) 2.4(s,3H,CH₃) 2.3(m,1H,H_{2''}) 1.9(s,3H,CH₃). ^{13}C NMR ($\text{CDCl}_3 + \text{CD}_3\text{OD}$, 75.5 MHz) δ = 174.4, 166.5, 160.9, 144.5, 136.9, 129.7, 129.2, 126.3, 116.3, 89.8, 85.9, 74.6, 61.7, 38.0, 21.6, 12.6.
20. Faerber, P.; Scheit, K.H. *Chem. Ber.* **1970**, *103*, 1307-1311.
21. ^1H NMR (CDCl_3 , 300.7 MHz) δ = 9.8(s,1H,NH) 7.9(dd,4H,Ar) 7.7(d,1H,H₆) 7.3(d,4H,Ar) 7.0(dd,1H,H_{1'}) 5.8(dd,1H,H₅) 5.6(m,1H,H_{3'}) 4.8(ddd,2H,H_{5'}) 4.6(d,1H,H_{4'}) 3.1(ddd,1H,H_{2'}) 2.4(d,6H,CH₃) 2.2(m,1H,H_{2''}). ^{13}C NMR (CDCl_3 , 75.5 MHz) δ = 174.3, 160.3, 144.6, 144.5, 139.4, 130.0, 129.9, 129.3, 129.1, 126.3, 126.2, 106.8, 89.8, 83.2, 74.0, 63.6, 38.1.
22. ^1H NMR ($\text{CDCl}_3 + \text{CD}_3\text{OD}$, 300.7 MHz) δ = 8.0(s,1H,H₆) 6.8(t,1H,H_{1'}) 4.3(dd,1H,H_{3'}) 3.8(ddd,2H,H_{5'}) 3.3(m,1H,H_{4'}) 2.4(m,1H,H_{2'}) 2.1(m,1H,H_{2''}) 1.8(s,3H,CH₃). ^{13}C NMR ($\text{CDCl}_3 + \text{CD}_3\text{OD}$, 75.5 MHz) δ = 173.9, 161.4, 137.2, 115.6, 89.4, 87.3, 69.3, 60.6, 40.5, 12.2.
23. ^1H NMR (CDCl_3 , 300.7 MHz) δ = 8.2(s,1H,H₂) 7.8(dd,1H,H₆) 6.8-7.4(m,13H,Ar) 5.9(m,2H,H₁+H₅) 4.8(s,1H,H_{3'-OH}) 4.7(m,1H,H_{3'}) 3.8(d,6H,OCH₃) 3.4(ddd,2H,H_{5'}) 2.6(m,1H,H_{2'}) 2.4(m,1H,H_{2''}). ^{13}C NMR (CDCl_3 , 75.5 MHz) δ = 171.4, 158.6, 150.6, 144.2, 138.3, 135.2, 135.0, 130.0, 128.0, 128.0, 127.1, 113.3, 112.7, 91.6, 87.5, 87.0, 71.9, 63.5, 55.2, 42.1.
24. ^1H NMR (CDCl_3 , 300.7 MHz) δ = 8.2(s,1H,H₂) 7.8(s,1H,H₆) 6.8-7.4(m,13H,Ar) 6.0(m,1H,H_{1'}) 4.7(d,1H,H_{3'}) 4.2(s,1H,H_{4'}) 3.8(d,6H,OCH₃) 3.4(d,2H,H_{5'}) 2.5(m,2H,H₂+H_{2''}) 1.6(s,3H,CH₃). ^{13}C NMR (CDCl_3 , 75.5 MHz) δ = 172.2, 158.7, 149.7, 149.6, 144.2, 135.3, 135.1, 134.2, 130.1, 128.1, 128.0, 127.1, 113.3, 91.8, 87.6, 87.0, 72.4, 63.9, 55.2, 42.1, 13.7.
25. ^1H NMR (CDCl_3 , 300.7 MHz) δ = 8.3(d,1H,H₂) 7.9(dd,4H,Ar) 7.5(dd,1H,H₆) 7.3(d,4H,Ar) 6.1(d,1H,H₅) 5.9(dd,1H,H_{1'}) 5.6(m,1H,H_{2'}) 4.7(d,2H,H_{5'}) 4.6(m,1H,H_{4'}) 2.8(dd,1H,H_{2'}) 2.5(m,1H,H_{2'}) 2.4(d,6H,CH₃); ^{13}C NMR (CDCl_3 , 75.5 MHz) δ = 170.1, 165.9, 165.7, 150.0, 144.7, 144.6, 136.6, 129.4(m,Ar), 126.3, 126.0, 113.1, 91.0, 83.4, 74.4, 63.7, 39.0, 21.7, 21.6.
26. Nielsen, J.; Dahl, O. *Nuc. Acids Res.* **1987**, *15*, 3626.
27. Barone, A.D.; Tang, J.Y.; Caruthers, M.H. *Nucleic Acids Res.* **1984**, *12*, 4051-4061.
28. ^{31}P NMR (CDCl_3 , 121.7 MHz) δ = 150.5, 150.2.
29. ^{31}P NMR (CDCl_3 , 121.7 MHz) δ = 150.3, 150.2.

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