

Generation and Reactions of Phosphorin 1-Sulfides

Hideyuki TANAKA and Shinichi MOTOKI*

Department of Chemistry, Faculty of Science, Science University of Tokyo, Kagurazaka, Shinjuku-ku, Tokyo 162

(Received September 26, 1986)

Synopsis. Phosphorin 1-sulfides generated by the thermolysis of 1-phospha-2-thiabicyclo[4.4.0]deca-3,7,9-triene 1-sulfides reacted with triphenylphosphine, dienophiles, or nucleophiles to give the corresponding phosphorins, 1-phosphabicyclo[2.2.0]octa-2,5-diene 1-sulfide, or 1,2-dihydrophosphorin 1-sulfide derivatives, respectively.

Previously, we reported that 2,9-dithia-1-phosphabicyclo-[4.3.0]nona-3,7-diene 1-sulfides (**2**)¹⁾ derived from α,β -unsaturated ketones **1** reacted with phenylacetylene to afford 1-phospha-2-thiabicyclo[4.4.0]deca-3,7,9-triene 1-sulfides (**3**).²⁾ From the mass spectral fragments, it is expected that thermolysis of **3** generates phosphorin 1-sulfides (**4**) along with α,β -unsaturated thiones (**5**)³⁾ in the manner of the retro-Diels-Alder reaction.

Recently, Mathey et al. reported that 4,5-dimethyl-2-phenylphosphorin 1-sulfide formed by the treatment of the corresponding phosphorin with sulfur in boil-

ing xylene was trapped by 2,3-dimethylbutadiene or dimethyl acetylenedicarboxylate.⁴⁾ However, no other papers on the synthesis or reactions of phosphorin 1-sulfides have been presented.

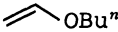
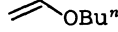


In this paper, we wish to describe some reactions of phosphorin 1-sulfides **4** generated by the thermolysis of the precursors **3**.

Results and Discussion

Treatment of **3** with triphenylphosphine in refluxing toluene produced phosphorins **6** and triphenylphosphine sulfide (Table 1). Another component, thiones **5** were decomposed by triphenylphosphine. The phosphorins **6** were easily identified by comparison of the ¹H NMR spectral data with those of **6a** in the literature.⁵⁾ The ¹³C NMR spectra of **6** were assigned as shown in Table 2.⁶⁾

Phosphorins were synthesized by Märkl et al. by the

Table 1. Reaction of **3** with Various Reagents

Material	Reagent	Reaction time/h	Product from 4		Product from 5	
			Mp $\theta_m/^\circ\text{C}$	Yield/% ^{a)}	Yield/% ^{a)}	
3a	Ph ₃ P	3	6a	170—172 ^{b)}	60	—
3b	Ph ₃ P	3	6b	173—175	59	—
3a	 OBu ⁿ	5	7a	134—136	13	8a 56
3b	 OBu ⁿ	5	7b	113—116	31	8b 77
3a		2	9a	187—189	47	11a 94
			10a	186—188	48	
3b		2	9b	128—130	45	11b 95
			10b	171—173	46	
3a	PhCH ₂ OH	1	12a ^{c)}	113—115	96	14a 98
3b	PhCH ₂ OH	1	12b ^{c)}	152—154	95	14b 95
3a	PhSH	2.5	13a ^{c)}	210—212	48	14a 60
3b	PhSH	2.5	13b ^{c)}	196—198	44	14b 56

a) The yield of the products was based on **3**. b) Lit.⁵⁾ 164—166°C. c) Yellow crystals.

Table 2. ¹³C NMR Spectral Data of the Products [$\delta(J_{\text{CP}}/\text{Hz})$]

Product	C(2)	C(3)	C(4)	C(5)	C(6)	$J_{\text{C(2)P}}$	$^2J_{\text{C(3)P}}$	$^3J_{\text{C(4)P}}$	$^2J_{\text{C(5)P}}$	$J_{\text{C(6)P}}$
6a	169.5	136.6	142.7	141.1	156.4	48.8	11.0	3.7	17.1	52.5
6b	169.4	136.3	142.7	141.1	156.4	48.8	12.2	3.7	17.1	53.7
7a	141.4	147.0	59.6	164.7	127.0	61.0	2.4	31.7	2.4	69.6
7b	141.5	147.0	59.8	165.0	127.0	61.0	2.4	31.7	2.5	69.6
9a	140.5	151.7	57.4	164.2	128.3	61.0	0	36.6	2.4	67.1
9b	140.3	150.9	57.3	164.2	128.3	61.0	0	36.6	2.4	67.1
10a	139.4	148.4	57.6	166.9	128.0	58.6	0	36.6	3.7	69.6
10b	139.2	147.5	57.5	166.9	128.1	58.6	0	35.4	2.4	67.1
12a	135.5	141.1	a)	a)	42.2	61.0	6.1	a)	a)	73.2
12b	134.3	140.5	a)	a)	42.1	61.0	6.1	a)	a)	72.0
13a	139.6	141.4	a)	a)	44.2	56.2	3.7	a)	a)	53.7
13b	139.6	141.1	a)	a)	44.0	56.2	3.7	a)	a)	52.5

a) These signals were not confirmed.

reaction of the corresponding pyrylium salts with $P(CH_2OH)_3$ or $P(SiMe_3)_3$.⁷⁾ By our method, **6** can be readily obtained from α,β -unsaturated ketones **1** by three step reactions.

Phosphorin sulfides **4** reacted as heterodienes at their 1, 4 positions with some dienophiles to give the corresponding cycloadducts.^{1,4)} When the reaction of **3** with butyl vinyl ether was carried out in refluxing toluene, the cycloadducts **7** were obtained along with **8** (Table 1). The cycloadducts **8** were easily identified by comparison with the authentic compounds reported previously.³⁾ In the ^{13}C NMR spectra of the cycloadducts **7**, the signals for the methine carbons [$\delta=77.6$ ($J_{CP}=8.5$ Hz)] and the methylene carbons [$\delta=37.5$ ($J_{CP}=47.6$ Hz)] indicated that butoxyl groups were linked to C(β) carbons,¹⁾ but the configuration of the C(β) carbons could not be determined. The signals for the C(2)-(6) carbons in **7** were assigned as shown in Table 2.^{2,4)}

The reaction of **3** with norbornene gave the cycloadducts **9**, **10**, and **11** in high yields (Table 1). The thione(5)-norbornene cycloadducts (**11**) were identified by comparing the spectral data with those for the authentic compounds.³⁾ The elemental analyses and mass spectra indicated that both **9** and **10** were 1:1 adducts of **4** with norbornene. In the ^{13}C NMR spectra of these adducts, seven signals due to the norbornane ring system were observed and the C(α), C(β) methine carbons showed C-P coupling [C(α), $\delta=52.2-53.2$ ($J_{CP}=46.4$ Hz); C(β), $\delta=51.1-53.7$ ($J_{CP}=8.5-9.8$ Hz)]. The signals for the C(2)-C(6) carbons in 1,4-dihydrophosphorin 1-sulfide ring system were assigned by comparison with those for phosphabarrelene derivatives^{2,4)} (Table 2). The chemical shifts for the C(2), C(3), and C(5) carbons gave the information on the configuration of **9** and **10**. The cycloadducts **10** showed the resonance for C(2) and C(3) carbons shifted upfield by 1.1 and 3.4 ppm, respectively, compared with those for **9**, whereas **9** showed that the resonance for C(5) carbon shifted upfield by 2.7 ppm compared with that for **10**. On the basis of these shifts, the norbornyl skeleton is considered to be situated on the side of the C(5) and C(6) carbons in **9** and on the side of the C(2) and C(3) carbons in **10**.^{1,8)} The signals for the bridged methylene carbons (C*) in **9** and **10** shifted upfield by 4.8-4.6 ppm compared with that in unsubstituted norbornane and appeared at $\delta=33.9$ and 34.1, respectively. This suggested that the configuration of these isomers would be *exo* with respect to the norbornyl ring.^{1,8)}

The reactions of **3** with styrene or acrylonitrile were unsuccessful.

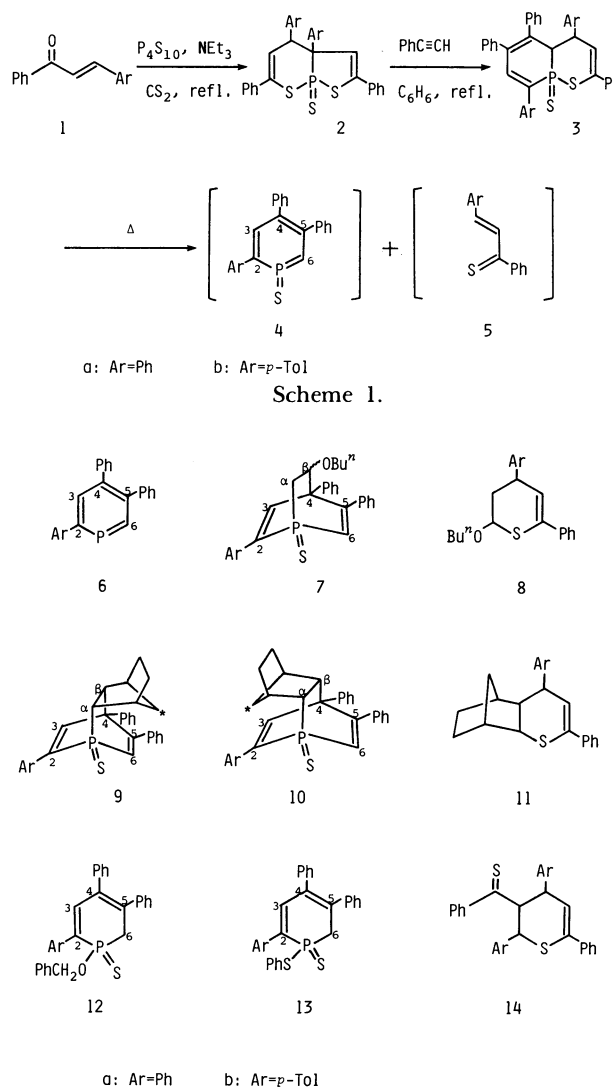
Finally, we examined the reaction of **3** with nucleophiles.⁹⁾ Treatment of **3** with benzyl alcohol in boiling toluene gave **12** in high yields. The products showed strong IR absorptions due to the P-O bonding at about 1000 cm^{-1} . In the 1H NMR spectra, the multiplets of two protons at $\delta=3.33-3.67$ were assigned to the H(6) AB protons. The signals for the C(6) methylene carbon showed large C-P coupling constant (Table 2).

Similarly, the reaction of **3** with benzenethiol gave the expected adducts **13**. The structure of **13** was con-

firmed by comparison of the spectral data with those for **12**.

In these reactions, addition of the hydrogen of the nucleophiles Nu-H to the C(6) carbons took place preferentially. These adducts **12** and **13** seem to be more stable than their isomeric 1,2- or 1,4-dihydrophosphorin 1-sulfides due to the effective conjugation of the double bond in dihydrophosphorin ring with three aryl groups at the 2-, 4-, and 5-positions. In both cases thiones **5** did not react with the nucleophiles, and were recovered as their dimers **14**.³⁾

Reaction of **3** with amines was complicated and no identified product was isolated.



Experimental

All melting points are uncorrected. IR spectra were recorded on a Hitachi Model 260-10 spectrometer. 1H and ^{13}C NMR spectra were measured in $CDCl_3$ solution using Me_4Si as an internal standard [spectrometer: 1H ; JEOL JNM-PMX60SI (60 MHz), ^{13}C ; JEOL JNM-FX 100 (25.05 MHz)]. Mass spectra were recorded on a Hitachi M-80 spectrometer or a JEOL JMS-D300 spectrometer, operating at an ionizing potential of 70 eV. Elemental analyses were performed using a Yanaco Model MT-3 CHN corder. 1-Phospha-2-thiabicyclo[4.4.0]deca-3,7,9-triene 1-sulfides

(3) were prepared according to the method previously described.¹⁾

General Procedure for the Reaction of 3 with Triphenylphosphine: A solution of 3 (1.0 mmol) and triphenylphosphine (1.2 mmol) in dry toluene (5.0 cm³) was refluxed under a nitrogen atmosphere for 3 h. The toluene was removed, and the residue was chromatographed on Wakogel C-200 with benzene-hexane (1:3) as an eluent. The solvent was evaporated, and the residue was recrystallized from hexane to give 6.

2,4,5-Triphenylphosphorin (6a): ¹H NMR δ =6.93–7.87 (m, 15H, Ar-H), 8.08 [d, 1H, J_{HP} =6.0 Hz, H(3)], and 8.73 [d, 1H, J_{HP} =38.0 Hz, H(6)]; MS m/z 324 (M^+ , 100). Found: C, 84.99; H, 5.23%. Calcd for C₂₃H₁₇P: C, 85.17; H, 5.28%. The ¹H NMR spectrum is identical with that reported by Märkl et al.⁵⁾

4,5-Diphenyl-2-*p*-tolylphosphorin (6b): ¹H NMR δ =2.43 (s, 3H, CH₃), 6.97–7.90 (m, 14H, Ar-H), 8.08 [d, 1H, J_{HP} =6.0 Hz, H(3)], and 8.73 [d, 1H, J_{HP} =38.0 Hz, H(6)]; MS m/z 338 (M^+ , 100). Found: C, 85.14; H, 5.62%. Calcd for C₂₄H₁₉P: C, 85.19; H, 5.66%.

General Procedure for the Reaction of 3 with Dienophiles or Nucleophiles: A solution of 3 (1.0 mmol) and large excess of dienophile or nucleophile (10.0 mmol) in dry toluene (2.5 cm³) was refluxed under a nitrogen atmosphere until 3 was consumed, as indicated by TLC. After evaporation of the solvent, the residue was chromatographed on Wakogel C-200 with benzene-hexane (2:1–3:1) as an eluent. The products were recrystallized from ethanol or ligroine.

8-Butoxy-2,4,5-triphenyl-1-phosphabicyclo[2.2.2]octa-2,5-diene 1-Sulfide (7a): IR(KBr) 1085 cm⁻¹ (C–O); ¹H NMR δ =0.67–2.67 (m, 9H), 3.33–4.83 (m, 3H), 6.53 [d, 1H, J_{HP} =20.0 Hz, H(3)], and 6.83–7.67 (m, 16H); MS m/z 456 (M^+ , 1), 356 (4a, 72), 324 (6a, 50), 254 (4a-PhC=CH, 100), and 100 (CH₂=CHOBuⁿ, 75). Found: C, 76.37; H, 6.45%. Calcd for C₂₉H₂₉OPS: C, 76.29; H, 6.40%.

8-Butoxy-4,5-diphenyl-2-*p*-tolyl-1-phosphabicyclo[2.2.2]octa-2,5-diene 1-Sulfide (7b): IR(KBr) 1085 cm⁻¹ (C–O); ¹H NMR δ =0.67–2.27 (m, 9H), 2.33 (s, 3H, *p*-Tol-CH₃), 3.33–4.00 (m, 3H), 6.52 [d, 1H, J_{HP} =20.0 Hz, H(3)], and 6.93–7.67 (m, 15H); MS m/z 470 (M^+ , 1), 370 (4b, 94), 338 (6b, 100), 268 (4b-PhC=CH, 82), and 100 (CH₂=CHOBuⁿ, 76). Found: C, 76.50; H, 6.66%. Calcd for C₃₀H₃₁OPS: C, 76.57; H, 6.64%.

8,10,12-Triphenyl-1-phosphatetracyclo[6.2.2.1^{3,6}.0^{2,7}]trideca-9,11-diene 1-Sulfide (9a): ¹H NMR δ =0.75–3.17 (m, 9H), 3.67–3.83 (m, 1H), 6.45 [d, 1H, J_{HP} =20.0 Hz, H(3)], and 6.67–7.75 (m, 16H); MS m/z 450 (M^+ , 3), 356 (4a, 100), 324 (6a, 65), 254 (4a-PhC=CH, 68), 121 (39), 94 (13), and 66 (81). Found: C, 79.99; H, 5.92%. Calcd for C₃₀H₂₇PS: C, 79.97; H, 6.04%.

10a: ¹H NMR δ =0.67–3.17 (m, 10H), 6.37 [d, 1H, J_{HP} =19.0 Hz, H(3)], and 6.50–7.67 (m, 16H); MS m/z 450 (M^+ , 2), 356 (4a, 100), 324 (6a, 42), 254 (4a-PhC=CH, 69), 121 (22), 94 (4), and 66 (31). Found: C, 79.74; H, 6.04%. Calcd for C₃₀H₂₇PS: C, 79.97; H, 6.04%.

8,12-Diphenyl-10-*p*-tolyl-1-phosphatetracyclo[6.2.2.1^{3,6}.0^{2,7}]trideca-9,11-diene 1-Sulfide (9b): ¹H NMR δ =0.75–3.17 (m, 9H), 2.35 (s, 3H, CH₃), 3.67–3.83 (m, 1H), 6.45 [d, 1H, J_{HP} =18.0 Hz, H(3)], 6.67–7.50 (m, 13H), and 7.50 (d, 2H, J_{HH} =8.0 Hz, *p*-Tol-H); MS m/z 464 (M^+ , 1), 370 (4b, 100), 338 (6b, 36), 268 (4b-PhC=CH, 80), 135 (38), 94 (10), and 66 (69).

Found: C, 79.81; H, 5.99%. Calcd for C₃₁H₂₉PS: C, 80.14; H, 6.29%.

10b: ¹H NMR δ =0.67–3.17 (m, 10H), 2.40 (s, 3H, CH₃), 6.35 [d, 1H, J_{HP} =20.0 Hz, H(3)], 6.50–7.50 (m, 13H), and 7.62 (d, 2H, J_{HH} =8.0 Hz, *p*-Tol-H); MS m/z 464 (M^+ , 2), 370 (4b, 100), 338 (6b, 63), 268 (4b-PhC=CH, 80), 135 (28), 94 (7), and 66 (47). Found: C, 80.28; H, 6.17; S, 6.96%. Calcd for C₃₁H₂₉PS: C, 80.14; H, 6.29; S, 6.90%.

1-Benzylloxy-3,4,6-triphenyl-1,2-dihydrophosphorin 1-Sulfide (12a): IR(KBr) 1000, 970 cm⁻¹ (P–O); ¹H NMR δ =3.33–3.67 [m, 2H, H(6)], 4.75–5.50 (m, 2H, PhCH₂O), 6.82 [d, 1H, J_{HP} =22.0 Hz, H(3)], and 6.67–7.83 (m, 20H, Ar-H); MS m/z 464 (M^+ , 72), 356 (4a, 22), 340 (75), 324 (6a, 10), 293 (66), 254 (4a-PhC=CH, 23), 215 (44), and 91 (100). Found: C, 77.58; H, 5.44%. Calcd for C₃₀H₂₅PS: 77.56; H, 5.42%.

1-Benzylloxy-3,4-diphenyl-6-*p*-tolyl-1,2-dihydrophosphorin 1-Sulfide (12b): IR(KBr) 980, 960 cm⁻¹ (P–O); ¹H NMR δ =2.38 (s, 3H, CH₃), 3.33–3.67 [m, 2H, H(6)], 4.75–5.50 (m, 2H, PhCH₂O), 6.77 [d, 1H, J_{HP} =20.0 Hz, H(3)], 6.60–7.57 (m, 17H, Ar-H), and 7.57 (d, 2H, J_{HH} =8.0 Hz, *p*-Tol-H); MS m/z 478 (M^+ , 67), 370 (4b, 11), 354 (59), 338 (6b, 22), 307 (100), 268 (4b-PhC=CH, 15), 215 (33), and 91 (74). Found: C, 77.54; H, 5.57; S, 6.66%. Calcd for C₃₁H₂₇OPS: C, 77.80; H, 5.69; S, 6.70%.

1-(Phenylthio)-3,4,6-triphenyl-1,2-dihydrophosphorin 1-Sulfide (13a): ¹H NMR δ =3.50–4.00 [m, 2H, H(6)], 6.68 [d, 1H, J_{HP} =18.0 Hz, H(3)], and 6.68–7.83 (m, 20H, Ar-H); MS m/z 466 (M^+ , 10), 356 (4a, 100), 324 (6a, 69), 254 (4a-PhC=CH, 87), 121 (36), and 110 (46). Found: C, 74.40; H, 4.88%. Calcd for C₂₉H₂₃S₂P: C, 74.65; H, 4.97%.

3,4-Diphenyl-1-(phenylthio)-6-*p*-tolyl-1,2-dihydrophosphorin 1-Sulfide (13b): ¹H NMR δ =2.37 (s, 3H, CH₃), 3.50–4.00 [m, 2H, H(6)], 6.68 [d, 1H, J_{HP} =19.0 Hz, H(3)], 6.67–7.50 (m, 17H, Ar-H), and 7.65 (d, 2H, J_{HH} =8.0 Hz, *p*-Tol-H); MS m/z 480 (M^+ , 3), 370 (4b, 87), 338 (6b, 100), 268 (4b-PhC=CH, 80), 135 (40), 110 (78), and 66 (17). Found: C, 75.03; H, 5.21%. Calcd for C₃₀H₂₅S₂P: C, 74.97; H, 5.24%.

References

- 1) H. Tanaka, S. Kametani, T. Saito, and S. Motoki, *Bull. Chem. Soc. Jpn.*, **58**, 667 (1985).
- 2) H. Tanaka and S. Motoki, *Bull. Chem. Soc. Jpn.*, **59**, 2047 (1986).
- 3) T. Karakasa and S. Motoki, *J. Org. Chem.*, **43**, 4147 (1978); **44**, 4151 (1979).
- 4) J. -M. Alcaraz and F. Mathey, *J. Chem. Soc., Chem. Commun.*, **1984**, 508.
- 5) G. Märkl, G. Y. Jin, and E. Silbereisen, *Angew. Chem., Int. Ed. Engl.*, **21**, 370 (1982).
- 6) G. Märkl, H. Baier, and R. Liebl, *Justus Liebigs Ann. Chem.*, **1981**, 919.
- 7) G. Märkl, *Angew. Chem., Int. Ed. Engl.*, **5**, 846 (1966) and later papers.
- 8) J. B. Stothers, "Carbon-13 NMR Spectroscopy," Academic Press, New York (1972), pp. 68, 414; G. C. Levy, R. L. Lichter, and G. L. Nelson, "Carbon-13 Nuclear Magnetic Resonance for Organic Chemists," Wiley, New York (1972), p. 61.
- 9) H. Tanaka, T. Saito, and S. Motoki, *Bull. Chem. Soc. Jpn.*, **59**, 59 (1986).