

## Note

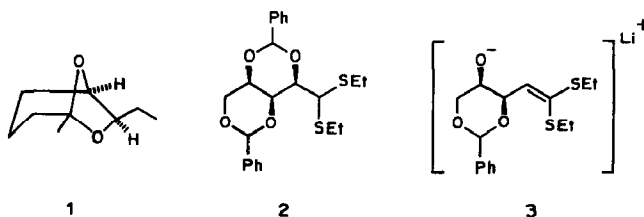
### Enantiospecific synthesis of (+)-*exo*-brevicomine from D-xylose\*

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*exo*-Brevicomine (**1**) is a principal component of the sex attractant of the western pine beetle, *Dendroctonus brevicomis* (Coleoptera: Scolytidae), the pest of the ponderosa pine tree<sup>1</sup>. Due to its importance in the pest control of pine forests coupled with its novel morphological features, several syntheses of **1**, both racemic<sup>2</sup> as well as optically active<sup>3</sup>, have appeared. In spite of the fact that **1** has a striking similarity to 1,6-anhydro- $\alpha$ -D-hexopyranoses, the utility of sugars in the chiral synthesis of **1** has not been fully exploited, as only two syntheses<sup>4</sup> of **1** from D-glucose have been reported so far. We now describe a synthesis of (+)-*exo*-brevicomine starting from D-xylose.



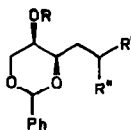
2,4:3,5-Di-*O*-benzylidene-D-xylose diethyl dithioacetal<sup>5</sup> (**2**), when treated with butyl-lithium at  $-10^\circ$ , gave an unstable ketene dithioacetal derivative **3**, which was reduced *in situ* with lithium aluminium hydride to afford the 2-deoxy derivative **4** (overall yield of 29%). The structure of **4** was demonstrated by the <sup>1</sup>H-n.m.r. data (loss of the signal for the benzylidene protons, m at  $\delta$  2.00 for H-2,2').

Compound **4** was desulfurised with W-2 Raney nickel in refluxing ethanol to give 80% of the 1,2-dideoxy product **5**, the <sup>1</sup>H-n.m.r. spectrum of which contained signals characteristic of a CHEt group. Hydrogenolysis of **5** with aluminium chloride-lithium aluminium hydride (1:1) in refluxing dichloromethane gave a

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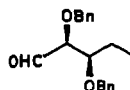
product which appeared to be chromatographically homogeneous but was a 1:1 mixture of isomeric diols ( $^1\text{H-n.m.r.}$  data).



- 4  $R = \text{H}, R', R'' = \text{SEt}$   
 5  $R, R', R'' = \text{H}$   
 6  $R = \text{Bn}, R', R'' = \text{H}$



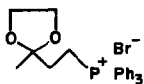
- 7  $R, R' = \text{Bn}, R'' = \text{H}$   
 8  $R = \text{H}, R', R'' = \text{Bn}$   
 9  $R'' = \text{Ac}, R', R = \text{Bn}$



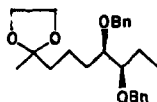
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Compound **5** was conventionally *O*-benzylated to give 90% of **6** which, with aluminium chloride–lithium aluminium hydride, gave 50% of **7** and 6.25% of **8**, after chromatography. The structure **7** was assigned by comparing its  $^1\text{H-n.m.r.}$  spectrum with that of the derived acetate **9**. The resonances for H-1,1' appeared in each spectrum as a doublet, but in that of **9** they were shifted downfield by 0.5 p.p.m., whereas the chemical shifts of all the other resonances were comparable, thereby indicating the presence of a primary hydroxyl group in **7**.

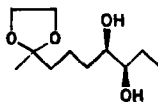
Swern oxidation<sup>6</sup> of **7** at  $-50^\circ$  afforded 90% of the aldehyde **10**, which was immediately treated with the phosphonium salt **11** in the presence of butyl-lithium to yield an *EZ*-mixture of the Wittig product **12** ( $^1\text{H-n.m.r.}$  data). Hydrogenation of **12** in the presence of Raney nickel at normal pressure and temperature reduced the double bond and removed the benzyl groups to give 90% of the diol **13**. Brief treatment of **13** with toluene-*p*-sulfonic acid at  $-10^\circ$  effected cyclisation to give 74% of (+)-**1**, the analytical and spectral data of which were comparable with those reported for *exo*-brevicomin.



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13

#### EXPERIMENTAL

All evaporations were performed under diminished pressure.  $^1\text{H-N.m.r.}$  spectra were recorded for solutions in  $\text{CDCl}_3$  (internal  $\text{Me}_4\text{Si}$ ) with a Varian FT-80A or Bruker WH-90 spectrometer. Optical rotations were measured with a JASCO DIP-181 polarimeter. All solvents were purified and dried. E.i.-mass spectra were obtained using an AEI MS-30 spectrometer. Column chromatography was performed on silica gel (Acme) (60–120 mesh).

*3,5-O-Benzylidene-2-deoxy-D-threo-pentose diethyl dithioacetal* (**4**). — Butyl-

lithium (100 mL, 2M solution in hexane) was added to a solution of **2** (43.2 g, 100 mmol) in tetrahydrofuran (300 mL) at  $-10^{\circ}$ . The mixture was stirred for 2 h, lithium aluminium hydride (1.9 g, 50 mmol) was added, and, after further reaction for 4 h at the same temperature, the mixture was quenched with saturated aqueous sodium hydrogencarbonate (20 mL). The precipitate was collected and the solvent was distilled off. Column chromatography (1:4 ethyl acetate–benzene) of the gummy product gave **4** (9.51 g, 29%),  $[\alpha]_{\text{D}}^{28} +16^{\circ}$  (*c* 2.9, chloroform);  $\nu_{\text{max}}^{\text{Nujol}}$  3450  $\text{cm}^{-1}$  (OH). Mass spectrum:  $m/z$  328 ( $\text{M}^+$ ).  $^1\text{H-N.m.r.}$  data:  $\delta$  1.25 (t, 6 H, 2  $\text{CH}_3$ ), 2.15 (m, 2 H, H-2,2'), 2.65 (q, 4 H, 2  $\text{CH}_2$ ), 3.45 (bs, OH), 3.60 (bs, 1 H, H-1), 3.90–4.50 (m, 4 H, H-3,4,5,5'), 5.65 (s, 1 H, PhCH), 7.43 (m, 5 H, Ph).

*Anal.* Calc. for  $\text{C}_{16}\text{H}_{24}\text{O}_3\text{S}_2$ : C, 58.52; H, 7.37; S, 19.49. Found: C, 58.60; H, 7.25; S, 19.70.

(4*R*,5*R*)-4-Ethyl-2-phenyl-1,3-dioxan-5-ol (**5**). — A mixture of **4** (6.56 g, 20 mmol), ethanol (50 mL), and W-2 Raney nickel (30 g) was boiled under reflux for 6 h, then filtered, and concentrated under reduced pressure to yield **5** as a colourless oil (3.33 g, 80%),  $[\alpha]_{\text{D}}^{28} -2.5^{\circ}$  (*c* 1, chloroform);  $\nu_{\text{max}}^{\text{liq}}$  3450  $\text{cm}^{-1}$  (OH). Mass spectrum:  $m/z$  208 ( $\text{M}^+$ ).  $^1\text{H-N.m.r.}$  data:  $\delta$  1.02 (t, 3 H,  $\text{CH}_3$ ), 1.70 (m, 2 H,  $\text{CH}_2$ ), 2.64 (bs, OH), 3.45 (bs, 1 H, H-4), 3.75–4.10 (m, 3 H, H-3,5,5'), 5.50 (s, 1 H, H-2), 7.35–7.65 (m, 5 H, Ph).

*Anal.* Calc. for  $\text{C}_{12}\text{H}_{16}\text{O}_3$ : C, 69.21; H, 7.74. Found: C, 69.05; H, 7.80.

(4*R*,5*R*)-5-Benzoyloxy-4-ethyl-2-phenyl-1,3-dioxane (**6**). — To a suspension of sodium hydride (0.264 g, 11 mmol) in tetrahydrofuran (10 mL) were added, sequentially, solutions of **5** (2.0 g, 9.6 mmol) in tetrahydrofuran (10 mL) and benzyl bromide (1.9 g, 11 mmol) in tetrahydrofuran (10 mL), and the reaction was continued for 2 h. Water (10 mL) was added, the mixture was extracted with ether ( $4 \times 25$  mL), the combined extracts were concentrated, and the residue was crystallised from light petroleum to give **6** (2.6 g, 90%), m.p.  $90^{\circ}$ ,  $[\alpha]_{\text{D}}^{28} -20^{\circ}$  (*c* 1.4, chloroform). Mass spectrum:  $m/z$  298 ( $\text{M}^+$ ).  $^1\text{H-N.m.r.}$  data:  $\delta$  0.88 (t, 3 H,  $\text{CH}_3$ ), 1.70 (m, 2 H,  $\text{CH}_2$ ), 3.50–3.90 (m, 2 H, H-4,5), 4.50 (m, 2 H, H-6,6'), 4.60 (s, 2 H, Ph $\text{CH}_2$ ), 5.50 (s, 1 H, H-2), 7.25 (m, 10 H, 2 Ph).

*Anal.* Calc. for  $\text{C}_{19}\text{H}_{22}\text{O}_3$ : C, 76.48; H, 7.43. Found: C, 76.00; H, 7.20.

(2*R*,3*R*)-2,3-Dibenzoyloxy-pentan-1-ol (**7**). — To an ice-cooled mixture of aluminium chloride (1.50 g), lithium aluminium hydride (1.50 g), and ether (20 mL) was added a solution of **6** (2.98 g, 10 mmol) in dichloromethane (30 mL). The mixture was stirred for 6 h at ambient temperature, saturated aqueous sodium carbonate (3 mL) was added, and insoluble material was collected and washed with dichloromethane. The combined filtrate and washings were concentrated, and column chromatography (1:20 acetone–light petroleum) of the residue (2.50 g) afforded (2*R*,3*R*)-1,2-dibenzoyloxy-pentan-3-ol (**8**; 0.19 g, 6.25%), isolated as a syrup,  $[\alpha]_{\text{D}}^{28} -12^{\circ}$  (*c* 1, chloroform).  $^1\text{H-N.m.r.}$  data:  $\delta$  0.90 (t, 3 H,  $\text{CH}_3$ ), 1.45 (m, 2 H,  $\text{CH}_2$ ), 2.20 (bs, 1 H, OH), 3.40–3.70 (m, 3 H, H-1,1',2), 4.50 (s, 2 H, Ph $\text{CH}_2$ ), 4.55 (m, 1 H, H-3), 4.65 (ABq, 2 H, Ph $\text{CH}_2$ ), 7.30 (bs, 10 H, 2 Ph).

*Anal.* Calc. for  $\text{C}_{19}\text{H}_{24}\text{O}_3$ : C, 76.00; H, 8.00. Found: C, 76.18; H, 8.00.

Eluted second was **7**, isolated as a syrup (1.50 g, 50%),  $[\alpha]_D^{28} -3.5^\circ$  (c 1, chloroform).  $^1\text{H-N.m.r.}$  data:  $\delta$  0.90 (t, 3 H,  $\text{CH}_3$ ), 1.55 (m, 2 H,  $\text{CH}_2$ ), 2.10 (bs, 1 H, OH), 3.35–3.80 (m, 4 H, H-1,1',2,3), 4.58, 4.65 (2 s, 4 H, 2  $\text{PhCH}_2$ ), 7.28 (bs, 10 H, 2 Ph).

*Anal.* Calc. for  $\text{C}_{19}\text{H}_{24}\text{O}_3$ : C, 76.00; H, 8.00. Found: C, 76.10; H, 8.00.

(2R,3R)-2,3-Dibenzylxypentanal (**10**). — To a mixture of dichloromethane (10 mL), oxalyl chloride (0.504 g, 4 mmol), and methyl sulfoxide (0.312 g, 4 mmol) at  $-60^\circ$  was added a solution of **7** (1.00 g, 3.3 mmol) in dichloromethane (5 mL). After 15 min, a solution of triethylamine (0.404 g, 4 mmol) in dichloromethane (2 mL) was added, the mixture was allowed to attain room temperature, and water (10 mL) was added. The mixture was extracted with dichloromethane ( $4 \times 15$  mL), and the combined extracts were washed with aqueous 10% HCl (10 mL), dried, and concentrated to give **10** as a syrup (0.90 g, 90%),  $[\alpha]_D^{28} -3^\circ$  (c 1.5, chloroform);  $\nu_{\text{max}}^{\text{liq}}$  1725  $\text{cm}^{-1}$  (C=O). Mass spectrum:  $m/z$  298 ( $\text{M}^+$ ).  $^1\text{H-N.m.r.}$  data:  $\delta$  0.85 (t, 3 H,  $\text{CH}_3$ ), 1.60 (m, 2 H,  $\text{CH}_2$ ), 3.60 (m, 1 H, H-3), 3.75 (dd, 1 H, *J* 6.5 and 2.0 Hz, H-2), 4.50 (s, 2 H,  $\text{PhCH}_2$ ), 4.60 (ABq,  $\text{PhCH}_2$ ), 7.25 (bs, 10 H, 2 Ph), 9.60 (d, 1 H, *J* 2.0 Hz, H-1).

*Anal.* Calc. for  $\text{C}_{19}\text{H}_{22}\text{O}_3$ : C, 76.48; H, 7.43. Found: C, 76.00; H, 7.25.

2-[(4R,5R)-4,5-Dibenzylxyhept-2-enyl]-2-methyl-1,3-dioxolane (**12**). — A mixture of 2-(2-bromoethyl)-2-methyl-1,3-dioxolane<sup>7</sup> (0.225 g, 1.1 mmol), triphenylphosphine (0.4 g, 1.5 mmol), and benzene (10 mL) was boiled under reflux and treated with tetrahydrofuran (10 mL) followed by butyl-lithium (0.5 mL, 1.1 mmol) at  $-10^\circ$ . The resulting orange-coloured phosphorane was treated with a solution of **10** (0.298 g, 1 mmol) in tetrahydrofuran (5 mL). After 3 h, the mixture was poured into water (10 mL) and extracted with ethyl acetate ( $3 \times 20$  mL), and the combined extracts were concentrated. Column chromatography (light petroleum–acetone, 20:1) gave *EZ*-**10** (0.3 g, 75%), isolated as a syrup. Mass spectrum:  $m/z$  396 ( $\text{M}^+$ ).  $^1\text{H-N.m.r.}$  data:  $\delta$  0.90 (t, 3 H,  $\text{CH}_3$ ), 1.30 (s, 3 H,  $\text{CH}_3$ ), 1.60 (m, 2 H,  $\text{CH}_2$ ), 2.35 (dd, 2 H, *J* 7.5 and 3 Hz, allylic protons), 3.35 (m, 1 H, H-5'), 3.90 (s, 4 H), 4.20 (dd, 1 H, *J* 9 and 7 Hz, 1 H, H-4'), 4.50 (ABq,  $\text{PhCH}_2$ ), 5.35–5.90 (m, 2 H, olefinic protons), 7.30 (bs, 10 H, 2 Ph).

*Anal.* Calc. for  $\text{C}_{25}\text{H}_{32}\text{O}_4$ : C, 75.72; H, 8.16. Found: C, 75.10; H, 8.00.

2-[(4R,5R)-4,5-Dihydroxyheptyl]-2-methyl-1,3-dioxolane (**13**). — A solution of **12** (0.12 g, 0.5 mmol) in ethanol was hydrogenated in the presence of W-2 Raney nickel (1 g) until 3 mol of hydrogen had been absorbed (2 h). The insoluble material was collected and washed with ethanol, and the combined filtrate and washings were concentrated under reduced pressure to give **13** (0.1 g, 90%) as an oil,  $[\alpha]_D^{28} +15^\circ$  (c 3, chloroform);  $\nu_{\text{max}}^{\text{liq}}$  3500  $\text{cm}^{-1}$  (OH). Mass spectrum:  $m/z$  200 ( $\text{M}^+ - 18$ ).  $^1\text{H-N.m.r.}$  data:  $\delta$  1.00 (t, 3 H,  $\text{CH}_3$ ), 1.30 (s, 3 H,  $\text{CH}_3$ ), 1.50 (m, 8 H, 4  $\text{CH}_2$ ), 2.6 (bs, 2 OH), 3.40 (m, 2 H), 3.95 (s, 4 H, 2  $\text{CH}_2$ ).

*Anal.* Calc. for  $\text{C}_{11}\text{H}_{22}\text{O}_4$ : C, 60.52; H, 10.16. Found: C, 60.10; H, 9.70.

(+)-*exo*-Brevicomine (**1**). — To a moist solution of **13** (0.10 g, 0.46 mmol) in ether was added toluene-*p*-sulfonic acid (0.01 g) at  $-10^\circ$ . After 3 h, the reaction

was quenched with solid sodium hydrogencarbonate, filtered, and concentrated. Distillation of the residue at 180° (bath) gave **1** (53 mg, 74%) as a colourless liquid,  $[\alpha]_D^{28} +75^\circ$  (*c* 1.5, ether). Mass spectrum: *m/z* 156 (*M*<sup>+</sup>). <sup>1</sup>H-N.m.r. data:  $\delta$  0.87 (t, 3 H, *J* 7.0 Hz, CH<sub>3</sub>), 1.30 (s, 3 H, CH<sub>3</sub>), 1.45 (m, 8 H, 4 CH<sub>2</sub>), 3.78 (t, 1 H, 7 Hz, H-7), 3.98 (bs, 1 H, H-1).

*Anal.* Calc. for C<sub>9</sub>H<sub>16</sub>O<sub>2</sub>: C, 69.19; H, 10.32. Found: C, 69.10; H, 10.20.

#### ACKNOWLEDGMENT

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#### REFERENCES

- 1 R. M. SILVERSTEIN, T. E. BELLAS, D. L. WOOD, AND L. E. BROWNE, *Science*, 159 (1968) 889–891.
- 2 P. S. REDDY, V. VIDYASAGAR, AND J. S. YADAV, *Synth. Commun.*, 14 (1984) 197–201, and references therein.
- 3 M. LARCHEVEQUE AND S. S. BIGELOW, *J. Chem. Soc., Chem. Commun.*, (1985) 83–84, and references therein.
- 4 (a) A. E. SHERK AND B. FRASER-REID, *J. Org. Chem.*, 47 (1982) 932–935; (b) R. J. FERRIER AND P. PRASIT, *J. Chem. Soc., Perkin Trans 1*, (1983) 1645–1647.
- 5 E. J. C. CURTIS AND J. K. N. JONES, *Can. J. Chem.*, 38 (1960) 1305–1315.
- 6 A. J. MANUSCO, S. L. HUANG, AND D. SWERN, *J. Org. Chem.*, 43 (1978) 2480–2482.
- 7 A. A. PONARAS, *Tetrahedron Lett.*, (1976) 3105–3108.