

**A CONVENIENT STEREOSELECTIVE SYNTHESIS OF CONJUGATED  
(2Z)-EN-4-YNOIC AND (2Z,4Z)- AND (2Z,4E)-DIENOIC ACID  
DERIVATIVES FROM PROPIOLIC ACID DERIVATIVES**

Xiyan Lu<sup>\*</sup>, Xiaoling Huang and Shengming Ma  
Shanghai Institute of Organic Chemistry, Academia Sinica,  
345 Lingling Lu, Shanghai 200032, China

**Abstract:** The palladium-catalysed one-pot sequential reaction of propiolic acid derivatives with lithium halides first, and then with terminal alkynes or alkenes affords conjugated (2Z)-en-4-ynoic and (2Z,4Z)- and (2Z,4E)-dienoic acid derivatives with high stereoselectivity.

Stereodefined conjugated 2,4-dienoic acid and 2-en-4-ynoic acid derivatives<sup>1</sup> represent a class of important synthetic intermediates<sup>2</sup> and a wide variety of natural products of biological interest, such as sorbic acid, dehydromatricaria ester and arachidonic acid metabolites<sup>3</sup>.

In recent years, many stereoselective synthetic methods for conjugated enynes and dienes have been developed<sup>4</sup>, especially the cross-coupling reaction of vinyl halides with organometallic reagents<sup>5</sup>. All these methods suffer from the preparation of stereodefined vinyl halides, and organometallic reagents, which require lengthy procedures. Recently, a very convenient stereoselective method for the synthesis of 3-halo-(2Z)-propenoic acid derivatives was developed in our laboratory<sup>6</sup>. Based on this work, we developed a new, simple, and one-pot method for introducing a cis carbon-carbon double bond into organic molecule, i.e., the synthesis of (2Z)-en-4-ynoic and (2Z,4Z)- and (2Z,4E)-dienoic acid derivatives via palladium(0) catalysed cross-coupling reaction of propiolic acid derivatives under very mild condition with alkynes or alkenes.

For the synthesis of (2Z)-en-4-ynoic acid derivatives, a propiolic acid derivative was treated with lithium halide and acetic acid in acetonitrile first, without isolation, the mixture was allowed to undergo cross-coupling reaction with a terminal alkyne<sup>7</sup> 3 in the presence of  $\text{PdCl}_2(\text{PPh}_3)_2$  and CuI at rt in triethylamine to afford conjugated (2Z)-en-4-ynoic acid derivatives<sup>2</sup> 4. It is noted that both lithium iodide and lithium bromide reacted smoothly, but lithium bromide was better. The coupling products showed high stereospecificity with high yields. The results are summarized in Table 1. The yields were slightly lower when  $\text{Pd}(\text{OAc})_2$ ,  $\text{PPh}_3$  and  $\text{Et}_3\text{N}$ <sup>8</sup> were used as the catalytic system.

A typical procedure is as follows Lithium bromide(104 mg, 1.2 mmol) was added to an acetonitrile(1 ml) solution of methyl propiolate(84 mg, 1 mmol) and acetic acid(66 mg, 1.1 mmol). The mixture was then stirred and refluxed

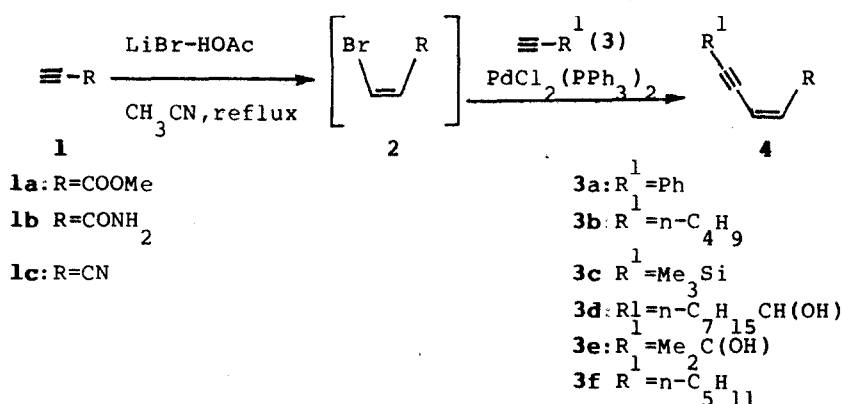


Table 1. Synthesis of (2Z)-En-4-yoic Acid Derivatives  
from Propiolic Acid Derivatives

1	3	Time(h)	Product	Isolated yield(%) <sup>9</sup>
1a	3a	6	4aa	85
1a	3b	6	4ab	84
1a	3c	6	4ac	77
1a	3d	5	4ad	80 <sup>a</sup>
1a	3e	4	4ae	60 <sup>a</sup>
1a	3f	6	4af	80
1b	3a	6	4ba	85
1b	3b	6	4bb	80
1b	3c	6	4bc	79
1b	3d	8	4bd	92
1b	3e	6	4be	82
1c	3a	8	4ca	90 <sup>b</sup>

a. 2 eq Et<sub>3</sub>N was added.

b. The reaction was carried out in a sealed tube.

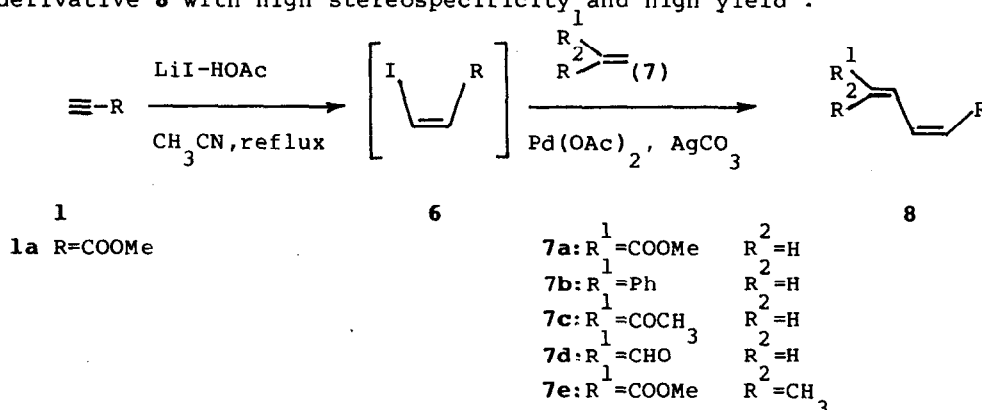
for 22 h. After cooling, the mixture was purged with nitrogen, then phenylacetylene(153 mg, 1.5 mmol), PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> (14 mg, 0.02 mmol), CuI(7.6 mg, 0.04 mmol) and triethylamine(4 ml) were added. The reaction was carried out at rt with stirring and monitored by TLC. After the reaction was complete, water(5 ml) was added. The mixture was extracted with ethyl acetate(3x10 ml), dried (MgSO<sub>4</sub>) and concentrated. The crude product was purified by column chromatography on silica gel(eluent:petroleum ether:ethyl acetate=50:1) to afford the desired product 4aa(160 mg, 85%).

It is worth noting that this method could also be used to synthesize (2Z)-en-4-ynamides (4ba-4be) which were rarely reported.

4 could be transformed into (2Z,4Z)-dienoic acid derivatives under

suitable condition. As an example, methyl (2Z,4Z)-deca<sup>10</sup>dienoate 5, which was reported to show interesting biological activity<sup>10</sup>, could be prepared by catalytic hydrogenation of methyl deca-(2Z)-en-4-ynoate 3af over Lindlar catalyst in 87% yield<sup>11</sup>.

From the above coupling reaction, it occurred to us that if an alkene is used instead of an alkyne, a (2Z,4E)-dienoic acid derivative might be prepared directly. Thus the intermediate 3-iodo-(2Z)-propenoic acid derivative 6 was treated with an alkene 7 in the presence of Pd(OAc)<sub>2</sub> and Ag<sub>2</sub>CO<sub>3</sub><sup>12</sup>, the reaction did give the desired product, (2Z,4E)-dienoic acid derivative 8 with high stereospecificity and high yield.



The title compounds, (2Z,4E)-dienoic acid derivatives are rarely reported. In our case, Heck's<sup>13</sup> and Jeffery's<sup>4,14</sup> catalytic systems were unsuccessful. The use of alkenyl iodides and Ag<sub>2</sub>CO<sub>3</sub> as the base, which makes the reaction easier to occur at rt, might be responsible for the retaining of cis configuration from 6 to 8. The results are shown in Table 2. Even with acrolein, which is easy to polymerize in Heck reaction<sup>13</sup>, as an olefinic component, the reaction occurred in high yield and high stereospecificity.

Table 2. Synthesis of (2Z, 4E)-Dienoic Acid Derivatives from Propiolic Acid Derivatives<sup>a</sup>

1	7	Product	Isolated yield(%) <sup>g</sup>
1a	7a	8aa	81
1a	7b	8ab	94
1a	7c	8ac	75
1a	7d	8ad	89
1a	7e	8ae	50

a. All reactions were carried out for 24h.

A typical procedure is as follows: After preparing methyl 3-iodo-(2Z)-

propenoate (from methyl propiolate (84 mg, 1 mmol)) as described in ref. 6, the reaction mixture was purged with nitrogen, then acrolein (224 mg, 4 mmol),  $\text{Ag}_2\text{CO}_3$  (414 mg, 1.5 mmol), and  $\text{Pd}(\text{OAc})_2$  (11 mg, 0.05 mmol) were added. The reaction was carried out at rt with stirring and monitored by TLC. After the reaction was complete, the mixture was filtered and concentrated. The residue was purified by column chromatography on silica gel (petroleum ether:ethyl acetate=15:1) to afford the desired product **8ad** (105 mg, 75%).

In conclusion, this procedure provides a method for introducing a cis carbon-carbon double bond into organic molecules with easy availability of the starting materials, simplicity of operation and high stereoselectivity, and will show its synthetic utility in organic chemistry.

**Acknowledgement:** We thank the National Natural Science Foundation of China and Academia Sinica for financial support.

#### References and Notes:

1. F. Bohlmann, T. Burkhardt, C. Zdero, "Naturally Occurring Acetylenes", Academic Press, Inc. London, 1973; E.R.H. Jones, V. Thaller, "The Chemistry of Carbon-Carbon Triple Bond", Part 2, Patai S., Ed., Interscience, New York, 1978, 261. H. Suemune, N. Hayashi, K. Funakoshi, H. Akita, T. Oishi, K. Sakai, Chem. Pharm. Bull., 1985, **33**, 2186; A. Kobayashi, S. Kouya, K. Yamashita, Agri. Biol. Chem., 1976, **40**, 2257, and references cited therein.
2. P. Beckstrom, U. Jacobsson, T. Norin, Tetrahedron, 1988, **44**, 2541; Y. Ikeda, J. Ukai, N. Ikeda, H. Yamamoto, Tetrahedron, 1987, **43**, 743; F. D. Lewis, D. K. Howard, S. V. Baranczyk, J. D. Oxman, J. Am. Chem. Soc., 1986, **108**, 3016; B. M. Trost, T. N. Nanninga, D. M. T. Chan, Organometallics, 1982, **1**, 1543.
3. I. E. Ronning, H. A. Frank, J. Food Prot., 1988, **51**, 651; B. Samuelsson, Angew. Chem. Int. Ed. Engl., 1983, **22**, 805.
4. T. Jeffery, Synthesis, 1987, 70; D. Ma, X. Lu, Tetrahedron, 1990, **46**, 3189. R. M. Moriarty, J. Am. Chem. Soc., 1991, **113**, 6315; K. Takagi, N. Hayama, Chem. Lett., 1983, 637.
5. A. Suzuki, Acc. Chem. Res., 1982, **15**, 178; N. Jabri, A. Alexakis, J. F. Normant, Tetrahedron Lett., 1981, **22**, 959; E.-I. Negishi, T. Yoshida, A. Abramovitch, G. Lew, R. M. William, Tetrahedron, 1991, **47**, 343; E.-I. Negishi, Acc. Chem. Res., 1982, **15**, 340.
6. S. Ma, X. Lu, J. Chem. Soc. Chem. Commun., 1990, 1643; S. Ma, X. Lu, Tetrahedron Lett., 1990, **31**, 7653; S. Ma, X. Lu, Z. Li, J. Org. Chem., 1992, **57**, 709.
7. J. Cassar, J. Organomet. Chem., 1975, **93**, 253; K. Sonogashira, Y. Tohda, N. Hagihara, Tetrahedron Lett., 1975, 4467.
8. J. R. Weir, B. A. Patel, R. F. Heck, J. Org. Chem., 1980, **45**, 4926.
9. All the products of the reactions gave satisfactory  $^1\text{H}$ NMR, IR, and MS data, and the new compounds were identified by elemental analysis.
10. J. A. Byers, H.-E. Hogberg, R. Unelius, G. Bergstrom, J. Lofqvist, J. Chem. Ecol., 1989, **15**, 685.
11. H. Lindlar, R. Dubuis, Org. Synth., 1966, **46**, 89.
12. M. M. Abelman, L. E. Overman, J. Am. Chem. Soc., 1988, **110**, 2328; M. M. Abelman, T. Oh, L. E. Overman, J. Org. Chem., 1987, **52**, 4130.
13. R. F. Heck, Org. React., 1982, **27**, 345.
14. T. Jeffery, Tetrahedron Lett., 1985, **26**, 2667.