## Two Amphoteric Galactocerebrosides Possessing a Tri-Unsaturated Long-Chain Base from the Leech (*Hirudo nipponica*)

Naoki Noda,\*,<sup>a</sup> Ryuichiro Tanaka,<sup>a</sup> Kazushige Tsujino,<sup>a</sup> Machiko Miura,<sup>a</sup> Kazumoto Miyahara,<sup>a</sup> and Junko Hayakawa<sup>b</sup>

Faculty of Pharmaceutical Sciences, Setsunan University,<sup>a</sup> 45–1, Nagaotoge-cho, Hirakata, Osaka 573–01, Japan and Aichi Prefectural Institute of Public Health,<sup>b</sup> Nagare 7–6, Tsuji-machi, Kita-ku, Nagoya 462, Japan. Received October 17, 1994; accepted December 12, 1994

Six amphoteric galactocerebrosides were isolated from the land annelid (*Hirudo nipponica*). Two of them have a tri-unsaturated long-chain base, D-erythro-(4E,8Z,11Z)-docosasphingatrienine. The position and geometry of the double bonds in the long-chain base unit were determined on the bases of chemical and spectral data.

**Key words** amphoteric galactosylceramide; tri-unsaturated long-chain base; D-erythro-(4E,8Z,11Z)-docosasphingatriene; annelid; leech; *Hirudo nipponica* 

As described in our previous papers, members of the phylum Annelida, such as *Pheretima asiatica*, and *Neanthes diversicolor*, contain amphoteric galactosylceramides (AGSLs). In a continuation of our systematic survey of the constituents in this phylum, we have examined the lipid composition, especially AGSLs, of the land annelid, the leech (*Hirudo nipponica*), and have isolated and characterized two new AGSLs together with four known ones. This paper deals with the isolation and structure determination of these compounds.

The CHCl<sub>3</sub>-MeOH extract (72.2 g) of the dried bodies (1 kg) of the leech (sold as a crude drug "Suitetsu")<sup>3)</sup> was treated with CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O (1:1:1), and the lower phase was concentrated to give a total lipid fraction. It was subjected to silica gel and Cosmosil 75C<sub>18</sub>-OPN column chromatographies with various solvents to yield a phospholipid fraction. This was further separated by preparative HPLC in a recycling mode to furnish six AGSLs 1—6. Among them, 3, 4 and 5 were respectively identified as *N*-hexadecanoyl-, *N*-heptadecanoyl-<sup>2a)</sup> and

*N*-octadecanoyl-1-O-[6-O-(2-trimethylammonioethoxy)-phosphinate- $\beta$ -D-galactopyranosyl]-(4E)-sphingenine, <sup>2b</sup>) and **6** as *N*-tetracosanoyl-1-O-[6-O-(2-trimethylammonioethoxy)phosphinate- $\beta$ -D-galactopyranosyl]-(4E)-17-methylsphingenine<sup>1)</sup> by comparison of the physical and spectroscopic data with those of corresponding authentic samples.

The <sup>13</sup>C-NMR spectrum of **1** (m/z: 1030 [M+H]<sup>+</sup>, fast atom bombardment mass spectrum (FAB-MS)) gave six olefinic carbon signals together with typical signals due to a phosphocholine group and a monogalactosylceramide moiety (Fig. 1). <sup>1,2)</sup> The <sup>1</sup>H-NMR spectrum showed signals ascribable to olefinic protons (4H,  $\delta$  5.3—5.4), two methylenes next to a double bond (4H,  $\delta$  2.0—2.1), a methylene (2H,  $\delta$  2.78) lying between two double bonds in addition to the signals arising from the common AGSL. These findings indicated that **1** has a 2,5-heptadiene-1,7-diyl (-CH=CH-CH<sub>2</sub>-CH=CH-) group.

Methanolysis of 1 gave a fatty acid methyl ester, which was identified as methyl *n*-tetracosanoate by gas chro-

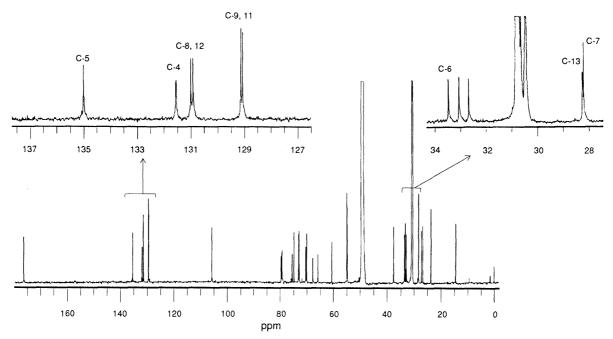


Fig. 1.  ${}^{13}\text{C-NMR}$  Spectrum of 1 (150 MHz, CD<sub>3</sub>OD: CDCl<sub>3</sub>=4:1)

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<sup>\*</sup> To whom correspondence should be addressed.

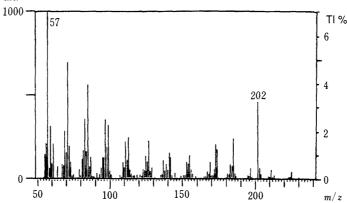


Fig. 2. EI-MS of the Dimethyl Disulfide Derivative of 1

matography (GC), electron impact mass spectrometry (EI-MS) and  $^1\text{H-NMR}$  spectroscopy. In view of the molecular weight of 1 (M.W. 1029) and its components (galactose 6-phosphocholine and n-tetracosanoyl units), the long-chain base (LCB) was considered to have a  $\text{C}_{22:3}$  tri-unsaturated carbon chain.

The dimethyl disulfide derivative of 1 gave a fragment ion peak at m/z 202 [ $C_{12}H_{25}S+H$ ]<sup>+</sup>, which was regarded as being due to a fragment ion produced by cleavage between the C-11 and C-12 sulfided carbons, <sup>2b,4</sup>) suggesting that the 2,5-heptadiene-1,7-diyl group is located at C-7 (Fig. 2).

To confirm the above suggestion as well as the geometry of the double bonds, the proton and carbon signals of 1 were assigned by two-dimensional shift correlation spectroscopy (1H-1H and 13C-1H COSY) and twodimensional nuclear Overhauser effect spectroscopy (NOESY) experiments. The <sup>1</sup>H-<sup>1</sup>H COSY spectrum exhibited a series of correlation peaks from H-3 to H<sub>2</sub>-13, showing the locations of three double bonds (C4, C8 and C11). Furthermore, in the NOESY spectrum, the signals of the methylene (H<sub>2</sub>-10) lying between two double bonds (C8 and C11) were correlated with those of H<sub>2</sub>-7 and H<sub>2</sub>-13, and an nuclear Overhauser effect (NOE) correlation between H-4 and H<sub>2</sub>-6 appeared (Fig. 3). These observations revealed that the geometry of the double bond at C4 is trans and that of the double bonds at C8 and C11 is cis. The chemical shifts of the carbon signals next to the double bonds, C-6 ( $\delta$  33.47), C-7 ( $\delta$  28.20), C-10 ( $\delta$  26.60) and C-13 ( $\delta$  28.24), supported<sup>5)</sup> the above

With regard to the configuration of the LCB unit, all AGSLs so far isolated from members of Annelida contain the D-erythro-sphingenine moiety, so it was presumed that the LCB part of 1 also has the erythro form. Hydrogenation

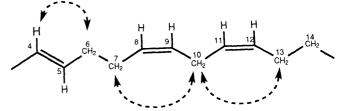


Fig. 3. NOE Correlations of 1

over palladium carbon of 1 followed by methanolysis gave a saturated LCB, which was acetylated to give an LCB (docosasphinganine) triacetate (1a). When the <sup>1</sup>H-NMR spectral data of 1a were compared with those of two pairs of D,L-erythro- and D,L-threo-sphinganine triacetates reported by Shibuya et al., <sup>6)</sup> the chemical shifts and splitting patterns of H<sub>2</sub>-1, H-2 and H-3 of 1a were in good accord with those of the former. Moreover, the sign of the specific rotation was the same as that of D-erythro-sphinganine triacetate. From the above findings, the configuration was concluded to be D-erythro, and hence the LCB of 1 was assigned the structure D-erythro-(4E,8Z,11Z)-docosasphingatrienine.

On the basis of all the above results, the structure of 1 is defined as N-tetracosanoyl-1-O-[6-O-(2-trimethyl-ammonioethoxy)phosphinate- $\beta$ -D-galactopyranosyl]-(4E,8Z,11Z)-docosasphingatrienine (Fig. 4).

Compound 2 showed a quite similar <sup>1</sup>H-NMR spectrum to that of 1, while in the FAB-MS, the [M+H]<sup>+</sup> ion peak was 28 mass units less than that of 1. Methanolysis of 2 provided methyl *n*-docosanoate. High-resolution NMR spectroscopic analyses (COSY, NOESY) in the same manner as for 1 revealed that 2 differs from 1 only in the fatty acid residue; the *n*-tetracosanoyl group in 1 is replaced by a *n*-docosanoyl group in 2. Therefore, 2 was

Fig. 4. Structures of Compounds 1—6

characterized as N-docosanoyl-1-O-[6-O-(2-trimethyl-ammonioethoxy)phosphinate- $\beta$ -D-galactopyranosyl]-(4E,8Z,11Z)-docosasphingatrienine (Fig. 4).

Glycosphingolipids obtained so far from natural sources have mainly a mono- or di-unsaturated LCB group, and only two tri-unsaturated LCBs ( $C_{18:3}$ ,  $C_{19:3}$ ) were detected in a methanolyzed mixture of the glycolipids fraction of the starfish, *Asterias amurensis*. In regard to intact glycosphingolipids with a tri-unsaturated LCB unit, only four glycocerebrosides have been isolated recently from marine animals. All of them consist of sphingatrienine or sphingatrienine possessing a branched methyl group, and the geometry and position of the three double bonds in LCB part are 4E,8E,10E. The two compounds (1 and 2) isolated here are the first examples of AGSLs with a  $C_{22:3}$  LCB, (4E,8Z,11Z)-docosasphingatrienine unit.

## Experimental

The NMR spectra were recorded on a GE NMR OMEGA 600 instrument at 600 MHz (1H) and 150 MHz (13C) at a probe temperature of 35 °C using tetramethylsilane (TMS) as an internal reference. MS were acquired on a JEOL JMS DX-300 spectrometer (EI-MS: ionization voltage, 30 eV; accelerating voltage, 3-10 kV. Positive ion FAB-MS: accelerating voltage, 3 kV; matrix, glycerol; collision gas, Xe). Optical rotations were measured with a JASCO DIP-140 polarimeter. TLC was carried out on silica gel HPTLC with Al sheets (Merck Art. 5556). Spots were visualized with 5% H<sub>2</sub>SO<sub>4</sub> in MeOH (by heating). Column chromatography was carried out on Merck silica gel (230-400 mesh, Art. 9385), and Cosmosil 75C<sub>18</sub>-OPN (Nacalai Tesque). Preparative HPLC was conducted over an L-column octadecyl silica (ODS) (10 mm × 250 mm, Chemicals Inspection & Testing Ins.) on a JASCO PU-980 equipped with a model 504R unit (GL Sciences). Recycling HPLC was carried out on a JASCO PU-980 equipped with a JASCO preparative recycle valve.

Isolation of AGSLs 1—6 The CHCl<sub>3</sub>-MeOH extractives (72.2 g) of the crushed powder (1 kg) of the leech, *Hirudo nipponica* (sold as a crude drug "Suitetsu," purchased from Tochimoto Tenkaido, 1993), were treated with CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O (1:1:1,600 ml), and the lower soluble portion was collected and concentrated to give a total lipid fraction

Table 1. <sup>1</sup>H-NMR Chemical Shifts ( $\delta$ ) of 1 and 2 (CD<sub>3</sub>OD: CDCl<sub>3</sub> = 4:1)

No.	1	2
1	3.53 (1H, dd, $J=8.0$ , 10.0 Hz)	3.52  (1H, dd,  J=8.0, 10.0  Hz)
	4.18 (1H, dd, J=4.0, 10.0 Hz)	4.19 (1H, dd, $J=4.0$ , 10.0 Hz)
2	3.96 (1H, ddd, $J=4.0$ , 8.0, 8.0 Hz)	3.96 (1H, ddd, $J = 4.0$ , 8.0, 8.0 Hz)
3	4.07 (1H, t, J = 8.0 Hz)	4.07  (1H, t,  J=8.0  Hz)
4	5.45 (1H, dd, $J = 8.0$ , 15.0 Hz)	5.45 (1H, dd, $J=8.0$ , 15.2 Hz)
5	5.69 (1H, td, $J=7.0$ , 15.0 Hz)	5.69 (1H, td, $J=7.0$ , 15.2 Hz)
6	2.03 (2H, m)	2.03 (2H, m)
7	2.07 (2H, m)	2.07 (2H, m)
8	5.35 (1H, m) <sup>a)</sup>	5.35 (1H, m) <sup>a)</sup>
9	$5.34 (1H, m)^{b}$	5.34 (1H, m) <sup>b)</sup>
10	2.78 (2H, t, J = 8.0 Hz)	2.78 (2H, t, J = 8.0 Hz)
11	5.34 (1H, m) <sup>b)</sup>	5.34 (1H, m) <sup>b)</sup>
12	5.35 (1H, m) <sup>a)</sup>	5.35 (1H, m) <sup>a)</sup>
13	2.04 (2H, m)	2.05 (2H, m)
14	1.37 (2H, m)	1.37 (2H, m)
22	0.90 (3H, t, J = 7.0 Hz)	0.90 (3H, t, J = 7.0 Hz)
2'	2.17 (2H, t, J = 7.5 Hz)	2.17 (2H, t, J=7.5 Hz)
3'	1.58 (2H, m)	1.59 (2H, m)
22'		0.90 (3H, t, J = 7.0 Hz)
24'	0.90 (3H, t, $J = 7.0 \text{ Hz}$ )	,
Gal-1	4.24 (1H, d, J=7.5 Hz)	4.24 (1H, d, J = 7.0 Hz)
Gal-2	3.55 (1H, dd, J=7.5, 10.0 Hz)	3.55 (1H, dd, J=7.0, 10.0 Hz)
Gal-3	3.51 (1H, dd, $J=3.0$ , 10.0 Hz)	3.50  (1H, dd,  J=3.0, 10.0  Hz)
Gal-4	3.86 (1H, dd, $J=0.8$ , 3.0 Hz)	3.87  (1H, dd,  J=0.7, 3.0  Hz)
Gal-5	3.71 (1H, td, $J = 0.8$ , 6.5 Hz)	3.71 (1H, td, $J=0.7$ , 6.5 Hz)
Gal-6	4.03 (2H, t, J = 6.5 Hz)	4.03 (2H, t, J = 6.5 Hz)
	4.30 (2H, m)	4.30 (2H, m)
	3.65 (2H, t, $J = 4.5 \text{ Hz}$ )	3.64 (2H, t, J=4.5 Hz)
$CH_3$	3.22 (9H, s)	3.23 (9H, s)

a, b) Signals with the same superscripts are overlapping.

(55.6 g). It was placed on a silica gel column and eluted successively with CHCl<sub>3</sub>–MeOH (8:2 $\rightarrow$ 7:3) $\rightarrow$ CHCl<sub>3</sub>–MeOH–H<sub>2</sub>O (7:3:0.5 $\rightarrow$ 6:4:1 $\rightarrow$ 5:5:1). The eluates were monitored by TLC (mobile phase: CHCl<sub>3</sub>–MeOH–H<sub>2</sub>O, 6:4:1) and those showing a positive spot with Dittmer–Lester's reagent<sup>9)</sup> were combined and evaporated to give six fractions, fr. 1 (38.3 g), fr. 2 (3.3 g), fr. 3 (1.9 g), fr. 4 (4.6 g), fr. 5 (1.8 g) and fr. 6 (4.5 g). Fraction 6 was subjected to chromatography on a 75C<sub>18</sub>-OPN column using MeOH $\rightarrow$ CHCl<sub>3</sub>–MeOH (1:1) as the eluent to yield four fractions, fr. 7 (570 mg), fr. 8 (2.46 g), fr. 9 (1.28 g) and fr.

Table 2.  $^{13}$ C-NMR Chemical Shifts ( $\delta$ ) of 1 and 2 (CD<sub>3</sub>OD: CDCl<sub>3</sub> = 4:1)

No.	1	2
1	70.18	70.20
	54.68	54.74
2 3	72.89	72.94
4	131.52	131.54
5	134.97	135.00
6	33.47	33.49
7	28.20	28.24
8	130.98 <sup>a)</sup>	131.01 <sup>a)</sup>
9	$129.12^{b)}$	$129.14^{b}$
10	26.60	26.62
11	$129.05^{b)}$	$129.08^{b}$
12	130.90 <sup>a)</sup>	$130.92^{a)}$
13	28.24	28.25
22	14.45	14.45
1′	175.80	175.87
2′	37.41	37.45
3′	27.14	27.17
22'		14.45
24'	14.45	
Gal-1	105.45	105.49
Gal-2	72.70	72.74
Gal-3	74.61	74.67
Gal-4	69.89	69.95
Gal-5	75.25	75.31
Gal-6	65.67	65.70
Cho-1"	60.45	60.48
Cho-2"	67.58	67.64
$CH_3$	54.82	54.85

a, b) Assignments with the same superscripts in each column may be interchangeable.

10 (190 mg). Fraction 9 was separated by HPLC (mobile phase: CHCl<sub>3</sub>–MeOH, 1.5:10) in a recycling mode to give compounds 1 (20.5 mg), 2 (54.7 mg), 3 (27.8 mg), 4 (15.5 mg), 5 (19.8 mg) and 6 (23.6 mg). 1: mp 170—183 °C,  $[\alpha]_D^{26} + 8.76^\circ$  (c = 0.39, MeOH: CHCl<sub>3</sub> = 1:1). Positive ion FAB-MS m/z: 1030 (M+H)+. <sup>1</sup>H-NMR δ: Table 1. <sup>13</sup>C-NMR δ: Table 2. 2: mp 159—172 °C,  $[\alpha]_D^{26} + 8.62^\circ$  (c = 0.58, MeOH: CHCl<sub>3</sub> = 1:1). Positive ion FAB-MS m/z: 1002 (M+H)+. <sup>1</sup>H-NMR δ: Table 1. <sup>13</sup>C-NMR δ: Table 2. Compounds 3, 4 and 5 were respectively identified as N-hexadecanoyl, N-heptadecanoyl- and N-octadecanoyl-1-O-[6-O-(2-trimethylammonioethoxy)phosphinate-β-D-galactopyranosyl]-(4E)-sphingenine, and 6 as N-tetracosanoyl-1-O-[6-O-(2-trimethylammonioethoxy)phosphinate-β-D-galactopyranosyl]-(4E)-17-methylsphingenine by comparison with authentic samples obtained previously. <sup>1,2)</sup>

Analysis of the Fatty Acids Each (ca. 1 mg) of 1—6 was treated with 5% methanolic HCl at 90°C for 1 h. The reaction mixture was extracted with n-hexane and the fatty acid liberated was analyzed by EI-MS,  $^1$ H-NMR and GC (fused silica capillary column Bonded MPS-50, Quadrex, 0.25 mm × 50 m; column temperature, 230 °C (hold, 12 min)  $\rightarrow$  240 °C at 1 °C/min; carrier gas, He at 33.4 ml/min):  $t_R$  (min): 6.54 (methyl hexadecanoate) from 3, 7.63 (methyl heptadecanoate from 4), 9.11 (methyl octadecanoate from 5), 21.02 (methyl docosanoate from 2), 31.75 (methyl tetracosanoate from 1 and 6). Methyl n-docosanoate: EI-MS

m/z: 354 (M<sup>+</sup>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 0.89 (3H, t, J=7.0 Hz, H<sub>3</sub>-22), 1.21—1.72 (-C $\underline{\text{H}}_2$ -), 2.30 (2H, t, J=7.0 Hz, H<sub>2</sub>-2), 3.66 (3H, s, OC $\underline{\text{H}}_3$ ). Methyl n-tetracosanoate: EI-MS m/z: 382 (M<sup>+</sup>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 0.90 (3H, t, J=7.0 Hz, H<sub>3</sub>-24), 1.23—1.72 (-C $\underline{\text{H}}_2$ -), 2.29 (2H, t, J=7.0 Hz, H<sub>2</sub>-2), 3.65 (3H, s, OC $\underline{\text{H}}_3$ ).

Preparation of the Dimethyl Disulfide Derivative of 1 Carbon disulfide  $(0.2\,\mathrm{ml})$  and iodine  $(1\,\mathrm{mg})$  were added to 1  $(2\,\mathrm{mg})$  in dimethyl disulfide  $(0.2\,\mathrm{ml})$ , and the mixture was kept at  $60\,^{\circ}\mathrm{C}$  for  $40\,\mathrm{h}$ . The reaction was quenched with 5% aqueous solution of sodium thiosulfate, then the mixture was shaken with  $\mathrm{CHCl_3}$ -MeOH  $(1:1,3\,\mathrm{ml})$ . The lower soluble portion was separated and concentrated under a nitrogen stream. The residue was subjected to analysis by EI-MS (Fig. 2).

Determination of the LCB Part (1a) of 1 and 2 Each (15 mg) of compounds 1 and 2 was hydrogenated over 10% palladium carbon (mg) in CHCl<sub>3</sub>-MeOH (1:4, 50 ml). The catalyst was filtered off and the filtrate was evaporated to dryness to give a product. It was methanolyzed with 7.5% methanolic HCl at 90 °C for 2 h. The fatty acid liberated was extracted with n-hexane and the methanolic layer was neutralized by adding a small excess of AgCO<sub>3</sub>. After centrifugation, the supernatant was evaporated to dryness to give a residue. The residue was passed through a Sephadex LH-20 column with MeOH, yielding a product. This was acetylated with acetic anhydride-pyridine (1:1, 1 ml) at room temperature for 1 d to give an LCB triacetate (1a, 4 mg). 1a:  $\lceil \alpha \rceil_D^{25} + 8.4^{\circ}$  $(c = 0.13, \text{ CHCl}_3)$ . Positive ion FAB-MS m/z: 484 [M+H]<sup>+</sup>. <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.88 (3H, t,  $J = 7.0 \,\text{Hz}$ , H<sub>3</sub>-22), 1.26—1.75 (-CH<sub>2</sub>-), 1.99, 2.06, 2.07 (each, s, OCOC $\underline{H}_3 \times 2$  and NHCOC $\underline{H}_3$ ), 4.06 (1H, dd, J = 3.9, 12.0 Hz, H-1), 4.24 (1H, dd, J = 6.1, 12.0 Hz, H-1), 4.39 (1H, m, H-2), 4.90 (1H, ddd, J = 5.4, 5.4, 7.8 Hz, H-3), 5.81 (1H, m, NH).

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