

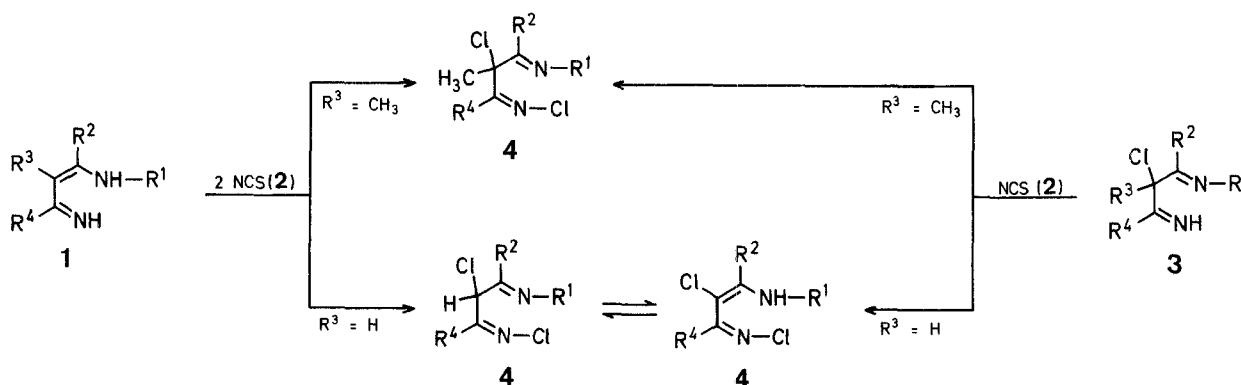
Synthesis of 1,3-Dichloro-1,5-diazapenta-1,4-dienes

José BARLUENGA, Miguel TOMÁS, J. Francisco LÓPEZ-ORTIZ, Vicente GOTOR

Departamento de Química Orgánica, Facultad de Química, Universidad de Oviedo, Oviedo, Spain.

In an earlier paper we have reported the reaction of 1-azabutadienes (**1**, 3-amino-2-alkenimines) with *N*-chlorosuccinimide (**2**) in stoichiometric ratio leading exclusively to the 2-halogenated derivatives **3**. These compounds were found to be useful precursors of chloro-substituted heterocycles¹. On the other hand, *N*-chloroamidines² and *N*-chloro-4- and -5-alkenamines³ show special reactivity due to the N—Cl bond and can therefore be used for the preparation of a variety of azaheterocycles, some of them with an alkaloid partial structure. Accordingly, we focused our attention on the *N*-chlorination of azabutadienes **1**. *N*-Chlorosuccinimide cannot only be used for the preparation of halogenated enamines⁴ but also for the *N*-chlorination of imine groups such as in amidines⁵. Here we report the preparation of 1,3-dichloro-1,5-diazapenta-1,4-dienes (**4**) starting from either **1** or **3** as well as their rearrangement-cleavage reaction upon acid hydrolysis.

When 4-amino-1-azabutadienes (**1**) or 2-chloroalkane-1,3-diimines (**3**) are allowed to react with *N*-chlorosuccinimide (**2**) in a molar ratio of 1 : 2 or 1 : 1, respectively, in toluene at 60°C, dichloro derivatives (**4**) are obtained in high yields.

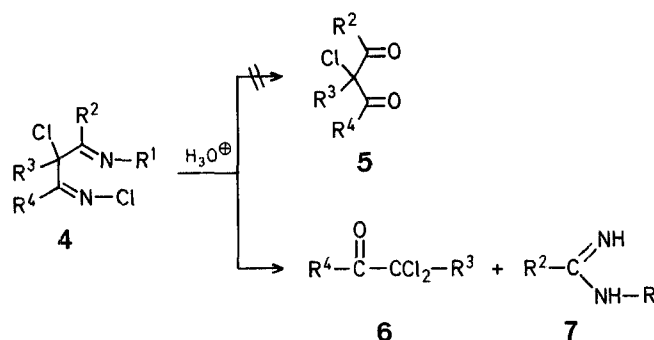


Compounds **4** were characterized by microanalyses and spectroscopic data. Thus, the ¹³C-N.M.R. spectra of **4** show characteristic signals at $\delta \approx 167$ (s), 157 (s), and 86 (s) ppm which are assignable to the imine C-atoms and to the C-atom bearing the Cl-atom, respectively.

In the case of $R^3 = \text{H}$ there is spectroscopic evidence of an equilibrium between the tautomeric forms **4** and **4'**. Whereas the form **4'** is present according to the I. R. (Nujol) and ¹H-N.M.R. (CDCl₃) spec-

tra (see Table 2, entries **g**, **h**, **i**), the form **4** is found almost exclusively in the ¹³C-N.M.R. (CDCl₃) spectrum since the C-3 atom appears as a doublet at ~ 68.5 ppm. Low- and high-temperature N.M.R. studies on the tautomerism **4** \rightleftharpoons **4'** as well as studies on the mechanism of the reaction are in progress.

Unlike the monochloro derivatives **3**¹, compounds **4** are not converted into the 2-chloro-1,3-diketones **5** upon treatment with 2 normal sulfuric acid in tetrahydrofuran; instead, a rearrangement-cleavage process takes place to afford a $\sim 1/1$ mixture of α,α -dichloroketones **6** and amidines **7**, the yields ranging from 80–90 %.



By the above-described method, *N*, 2-dichloroalkane-1,3-diimines (**4**) can be prepared in a regioselective manner. The high yields reached in all instances combined with the easy preparation of the starting materials make this synthesis a convenient route to compounds **4**.

The 1-azabutadienes **1**⁶ and the monohalogenated diimines **3**¹ were prepared as previously reported. Dichloroketones **6**⁷ and amidines **7**⁸ are known compounds.

1,3-Dichloro-3-methyl-5-phenyl-4-(4-chlorophenyl)-2-(4-methylphenyl)-1,5-diazapenta-1,4-diene (**4e**); Typical Procedure:

N-Chlorosuccinimide (1.67 g, 20 mmol) is added to a stirred solution of 3-methyl-5-phenyl-4-(4-chlorophenyl)-2-(4-methylphenyl)-1,5-diazapenta-1,3-diene (**1e**; 3.26 g, 10 mmol) in toluene (60 ml). The resultant mixture is heated at 50°C for 3 h, then cooled, and treated with aqueous 3 normal potassium hydroxide (150 ml). The organic layer is washed with water (50 ml) and dried with sodium sulfate. Removal of the solvent leaves a solid which is purified by recrystallization from hexane to give **4e**; yield: 3.44 g (87 %); m. p. 115–117°C.

C₂₃H₁₉Cl₃N₂ calc. C 64.28 H 4.55 Cl 24.65 N 6.52
(429.8) found 64.41 4.46 24.75 6.46

I. R. and ¹H-N.M.R., see Table 2.

¹³C-N.M.R. (CDCl₃/TMS_{int}): $\delta = 167.71$ (s), 157.11 (s), 147.83 (s), 139.84 (s), 137.09 (s), 134.06 (s), 130.55 (s), 86.12 (s), 35.46 (q), 21.12 (q) ppm.

The preparation of compounds **4** from **3** is carried out in an analogous manner, except for the use of equimolecular amounts of **3** and *N*-chlorosuccinimide.

Table 1. 1,3-Dichloro-1,5-diazapenta-1,4-dienes(**4**) from 4-Amino-1-azabutadienes (**1**)

4	R ¹	R ²	R ³	R ⁴	Yield [%]	m.p. [°C] (hexane)	Molecular Formula ^a	M.S. m/e (M ⁺)
a	C ₆ H ₅	C ₆ H ₅	CH ₃	C ₆ H ₅	89	128–130°	C ₂₂ H ₁₈ Cl ₂ N ₂ (381.3)	380
b	C ₆ H ₅	C ₆ H ₅	CH ₃	4-H ₃ C–C ₆ H ₄	91	120–122°	C ₂₃ H ₂₀ Cl ₂ N ₂ (395.3)	394
c	4-H ₃ C–C ₆ H ₄	C ₆ H ₅	CH ₃	C ₆ H ₅	83	83–85°	C ₂₃ H ₂₀ Cl ₂ N ₂ (395.3)	
d	4-H ₃ C–C ₆ H ₄	C ₆ H ₅	CH ₃	4-H ₃ C–C ₆ H ₄	86	104–106°	C ₂₄ H ₂₂ Cl ₂ N ₂ (409.4)	
e	C ₆ H ₅	4-Cl–C ₆ H ₄	CH ₃	4-H ₃ C–C ₆ H ₄	87	115–117°	C ₂₃ H ₁₉ Cl ₃ N ₂ (429.8)	
f	C ₆ H ₅	C ₆ H ₅	CH ₃	4-Cl–C ₆ H ₄	90	127–129°	C ₂₂ H ₁₉ Cl ₃ N ₂ (415.7)	
g	C ₆ H ₅	C ₆ H ₅	H	C ₆ H ₅	74	86–88°	C ₂₁ H ₁₆ Cl ₂ N ₂ (366.3)	
h	C ₆ H ₅	C ₆ H ₅	H	4-H ₃ C–C ₆ H ₄	79	120–122°	C ₂₂ H ₁₈ Cl ₂ N ₂ (381.3)	380
i	4-H ₃ C–C ₆ H ₄	C ₆ H ₅	H	C ₆ H ₅	80	92–94°	C ₂₂ H ₁₈ Cl ₂ N ₂ (381.3)	

^a Satisfactory microanalyses obtained: C ± 0.30; H ± 0.23; N ± 0.17; Cl ± 0.25.

Table 2. Spectral Data of Compounds **4**

4	I.R. (Nujol) ν [cm ⁻¹]	¹ H-N.M.R. (CDCl ₃ /TMS _{int}) δ [ppm]
a	1650, 1620	2.4 (s, 3H, CH ₃); 6.3–8.0 (m, 15H _{arom})
b	1655, 1625	2.2 (s, 3H, CH ₃); 2.4 (s, 3H, CH ₃); 6.7–8.0 (m, 14H _{arom})
c	1660, 1620	2.3 (s, 3H, CH ₃); 2.4 (s, 3H, CH ₃); 6.6–8.0 (m, 14H _{arom})
d	1650, 1610	2.2 (s, 3H, CH ₃); 2.3 (s, 3H, CH ₃); 2.45 (s, 3H, CH ₃); 6.5–8.0 (m, 13H _{arom})
e	1650, 1615	2.3 (s, 3H, CH ₃); 2.4 (s, 3H, CH ₃); 6.5–7.9 (m, 13H _{arom})
f	1650, 1600	2.4 (s, 3H, CH ₃); 6.6–7.9 (m, 14H _{arom})
g	3350, 1620, 1590	6.3–6.5 (br., 1H, NH); 6.8–7.6 (m, 15H _{arom})
h	3450, 1630, 1600	2.2 (s, 3H, CH ₃); 6.4 (br., 1H, NH); 6.8–7.7 (m, 14H _{arom})
i	3450, 1645, 1620	2.2 (s, 3H, CH ₃); 6.9 (br., 1H, NH); 6.9–7.5 (m, 14H _{arom})

Hydrolysis of Compounds 4e; Typical Procedure:

A solution of compound **4e** (1.97 g, 5 mmol) in tetrahydrofuran (30 ml) is stirred with 2 normal sulfuric acid (25 ml) for 4 h at room temperature. The resultant mixture is extracted with ether (3 × 40 ml). The organic extract is washed with saturated sodium chloride solution (40 ml), and dried with sodium sulfate. The solvent is removed and the residue distilled in vacuo to afford 2,2-dichloro-1-(4-methylphenyl)-propanone (**6**, R³ = CH₃, R⁴ = 4-H₃C–C₆H₄); yield: 0.93 g (86%); b.p. 76°C/0.05 torr.

C₁₀H₁₀Cl₂O calc. C 55.33 H 4.64 Cl 32.66
(217.1) found 55.06 4.41 32.82

M.S.: m/e = 216 (M⁺).

I.R. (neat): ν = 1680 cm⁻¹.

¹H-N.M.R. (CDCl₃/TMS_{int}): δ = 2.35 (s, 3H, CH₃); 2.4 (s, 3H, CH₃); 7.2 (d, 2H_{arom}); 8.1 ppm (d, 2H_{arom}).

¹³C-N.M.R. (CDCl₃/TMS_{int}): δ = 187.96 (s), 145.37 (s), 132.39 (d), 129.89 (d), 129.65 (s), 84.08 (s), 35.27 (q), 22.53 (q) ppm.

The aqueous layer is made alkaline with aqueous 6 normal potassium hydroxide and extracted with ether (3 × 40 ml). The organic extract is dried with sodium sulfate, the solvent is removed, and the

residue is recrystallized from hexane to give N-phenyl-4-chlorobenzamidine (**7**, R¹ = C₆H₅, R² = 4-Cl–C₆H₄); yield: 1.01 g (88%); m.p. 138–140°C (Ref.⁹, m.p. 140–141°C).

I.R. (Nujol): ν = 3450, 3300, 1620 cm⁻¹.

¹H-N.M.R. (CDCl₃/TMS_{int}): δ = 4.7 (br., 2H, NH); 6.8–7.8 ppm (m, 9H_{arom}).

Received: December 28, 1983

¹ J. Barluenga, M. Tomás, F. López-Ortiz, V. Gotor, *J. Chem. Soc. Perkin Trans. 1* **1983**, 2273.

² L. Citero, D. Pocar, R. Stradi, B. Gioia, *Tetrahedron* **35**, 69 (1979); and references cited therein.

³ L. Stella, *Angew. Chem.* **95**, 368 (1983); *Angew. Chem. Int. Ed. Engl.* **22**, 337 (1983).

⁴ N. De Kimpe, N. Schamp, *Org. Prep. Proced. Int.* **13**, 241 (1981).

⁵ L. Citero, D. Pocar, R. Stradi, *J. Chem. Soc. Perkin Trans. 1* **1978**, 309.

⁶ H. Hoberg, J. Barluenga, *Synthesis* **1970**, 142.

⁷ N. De Kimpe et al., *Synth. Commun.* **8**, 75 (1978); **9**, 575 (1979). *Beilsteins Handbuch der Organischen Chemie*, Vol. 7, III, 972.

⁸ P. Oxley et al., *J. Chem. Soc.* **1946**, 147; **1947**, 1112.

⁹ F.C. Cooper, M.W. Partridge, *J. Chem. Soc.* **1953**, 255.