

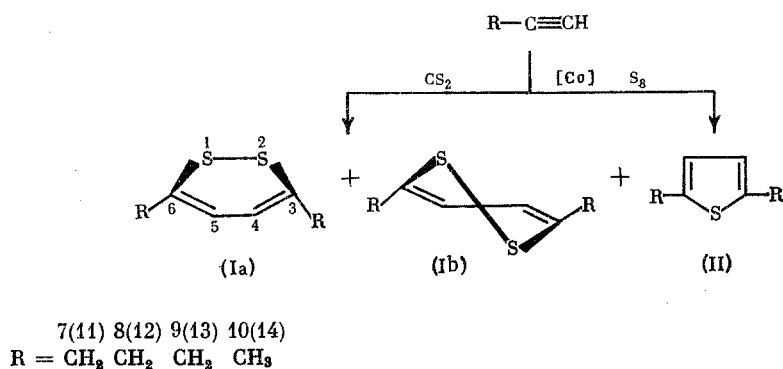
A NEW CATALYTIC REACTION OF ELEMENTAL SULFUR WITH ACETYLENES  
BY THE ACTION OF COBALT COMPLEXES

U. M. Dzhemilev, F. A. Selimov, V. R. Khafizov,  
L. M. Khalilov, and G. A. Tolstikov

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We are the first to report that the reaction of 1-hexyne with  $\text{CS}_2$  taken in 3:1 mole ratio by the action of a catalyst prepared by the reduction of  $\text{Co(2-ethylhexanoate)}_2$  by triethylaluminum in the presence of absolute DMSO ( $\text{Co:Al:DMSO} = 1:3:10\text{--}20$ ) in absolute toluene solution at  $150^\circ\text{C}$  for 6 h gives a 1:1 mixture of stereoisomeric 1,2-dithia-3,6-dibutyl-3,5-cyclohexadienes (Ia) and (Ib) in about 40% total yield. By analogy, 1-hexyne and  $\text{S}_8$  ( $\text{S}_\alpha$ -cyclooctasulfane) give 40% disulfides (Ia) and (Ib), 15% 2,5-dibutylthiophene (II), and about 45% of a mixture of 1,3,4- and 1,3,5-tributylbenzenes (III) in 45% total yield identified by comparison with authentic samples [1].

The  $^{13}\text{C}$  NMR spectrum of (I) has a double set of signals of virtually identical intensity. The upfield signals (13.7–36.4 ppm) are related to the butyl groups. The singlets at 139.7 and 140.7 ppm correspond to substituted atoms at  $\text{C}^3$  and  $\text{C}^6$  of the six-membered ring, while the doublets at 115.02 and 115.54 ppm are related to  $\text{C}^4$  and  $\text{C}^5$ . The doubling of the signals in the  $^{13}\text{C}$  NMR spectrum is apparently a result of the formation of two conformers or isomers differing in the symmetry elements (Ia) and (Ib), which are frozen at  $\sim 20^\circ\text{C}$ . A symmetry plane  $\sigma$  is found for (Ia) which is perpendicular to the S–S and  $\text{C}^2\text{--C}^3$  bonds and this compound apparently has boat form, while (Ib) has a twofold axis  $\text{C}_2$  and half-chair or twist form:



Derivatives (I) and (II) were separated and purified by vacuum distillation. 1,2-Dithia-3,6-dibutyl-3,5-cyclohexadiene (Ia,b), bp  $95\text{--}97^\circ\text{C}$  (2 mm),  $n_D^{20}$  1.5475. IR spectrum ( $\nu$ ,  $\text{cm}^{-1}$ ): 620, 760, 920, 1070, 2940, 3020.  $^{13}\text{C}$  NMR spectrum ( $\delta$ , ppm): 13.77 q ( $\text{C}^{10}$ ,  $\text{C}^{14}$ ), 21.93 t ( $\text{C}^9$ ,  $\text{C}^{13}$ ), 30.58 t, 30.65 t ( $\text{C}^8$ ,  $\text{C}^{12}$ ), 36.07 t, 36.42 t ( $\text{C}^7$ ,  $\text{C}^{11}$ ), 115.02 d, 115.54 d ( $\text{C}^4$ ,  $\text{C}^5$ ), 139.73 s, 140.41 s ( $\text{C}^3$ ,  $\text{C}^6$ ). Found, %: C 63.20; H 8.58; S 28.16. Calculated, %: C 63.16; H 8.77; S 28.07.  $M^+$  228.

2,5-Dibutylthiophene (II), bp  $82\text{--}83$  (2 mm),  $n_D^{20}$  1.4940 [2].

LITERATURE CITED

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Institute of Chemistry, Bashkir Branch of the Academy of Sciences of the USSR, Ufa.  
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