A Long-range Intramolecular Functionalization by Alkoxyl Radicals: a Long-range Intramolecular Hydroxylation of C(25) of Cholestane Side Chain¹

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Alkoxyl radicals generated by the irradiation of hypoiodites of 5α -chlolestan- 7α -yl-4-(hydroxyphenylmethyl)-phenylacetates and 5α -cholestan- 7α -yl-3-[4-(hydroxyphenylmethyl)phenyl] propionates, respectively, abstracted a hydrogen from the remote C(25) of their cholestane side chain to give novel macrocyclic ether lactones which gave 5α -cholestane- 7α , 25-diol by reduction with Na and liquid ammonia in good yields.

Many studies have been carried out on functionalization of unactivated C-H bonds *via* an intramolecular abstraction of hydrogen attached to a carbon atom by an alkoxyl radical²

since the importance of the process in organic synthesis has been shown by Barton and his colleagues.³ The intramolecular hydrogen abstraction by an alkoxyl radical that demands a

Scheme 1. Reagents and conditions: i, PhCOC₆H₄COCl-pyridine, 50-60 °C; ii, PhCOC₆H₄CH₂COCl-pyridine, 50-60 °C; iii, PhCOC₆H₄(CH₂)₂COCl-pyridine, 50-60 °C; iv, NaBH₄-tetrahydrofuran (THF), 25 °C; v, HgO-I₂-benzene; vi, hv; vii, Pb(OAc)₄-hv-benzene; viii, NH₃-Na.

6-membered cyclic transition state has been repeatedly demonstrated ever since by numerous examples with a variety of substrates.² There have been, however, few examples of successful intramolecular hydrogen abstraction *via* a manymembered cyclic transition state by alkoxyl radicals which are generated by the photolysis of nitrites^{2,3} or hypohalites.^{2,4}

On the other hand, Breslow and his colleagues have devised an ingenious extension of intramolecular abstraction of a hydrogen through a 6-membered cyclic transition state by an excited carbonyl group to a functionalization of remote unactivated methylene groups, as part of their model study of biomimetic control of chemical reactivity.^{5,6} They used a series of esters derived from benzophenone-3- or -4-carboxylic acid and steroidal alcohols.

In this communication, we wish to report on a two-step long-range hydroxylation of a steroidal skeleton, based on a long-range intramolecular hydrogen abstraction by alkoxyl radicals generated by the irradiation of hypoiodites of esters carrying a benzhydryl group, derived simply by reducing Breslow-type esters with NaBH₄. Steroidal-25-ols are of importance since there are several biologically-active steroids belonging to this group.⁷

Thus three esters, 5α -cholestan- 7α -yl-4-(hydroxyphenylmethyl)benzoates (5), 5α -cholestan- 7α -yl-4-(hydroxyphenylmethyl)phenylacetates (6), and 5α -cholestan- 7α -yl-3-[4-(hydroxyphenylmethyl)phenyl]propionates (7) were prepared by the reduction of the corresponding esters (2), (3), and (4), derived from appropriate benzophenone-4-carboxylic acids and 5α -cholestan- 7α -ol (1), with NaBH₄.† Each ester was a mixture of epimers with regard to the carbon atom carrying the hydroxy group. The epimeric esters (6) in CCl₄ were first transformed into the corresponding hypoiodites with 3 equiv. of both mercury(II) oxide and iodine. The solution was then irradiated with a 450W high pressure Hg arc for 7 h in a

nitrogen atmosphere to give a mixture of products from which (8a) (4.3%), (8b) (3.8%), the benzophenone esters (3) (29%), as well as unreacted starting alcohol (6) (31%) were isolated by means of preparative TLC.

The molecular formulae of crystalline products (8a) and (8b) were determined to be $C_{42}H_{58}O_3$ by means of high resolution mass spectrometry and by elemental analysis. IR spectra of (8a) and (8b) showed the absorption bands at 1722 and 1720 cm⁻¹ assignable to the lactone carbonyl groups, respectively. The ¹H NMR spectrum (400 MHz) of product (8a) exhibited a 1H singlet at δ 5.43 and two 3H singlets at δ 1.25 and 1.30 assignable to -C₆H₄-CH(OR)-C₆H₅ and the gem dimethyl group, besides the signals due to the H(18), H(19), and $H(7\beta)$. These spectral results indicated that the structure of the product is a macrocyclic ether lactone (8a). The ¹H NMR of the product (**8b**) similarly exhibited a singlet at δ 5.56 (1H) and two singlets (each 3H) at δ 1.09 and 1.20 assignable to the - C_6H_4 -CH(OR)- C_6H_5 and the gem dimethyl group. These spectral data suggested that it is a macrocyclic ether lactone (8b) epimeric with (8a).

A similar irradiation of the epimeric esters (6) in CCl₄, each containing 3 equiv. of lead tetra-acetate and iodine, for 4 h gave lactones (8a) (5.2%), (8b) (3.9%), benzophenone ester (3) (32%), and the recovered alcohol (6) (13%). The lactones (8a) and (8b) were not formed, however, when esters (6) in CCl₄, each containing 3 equiv. of iodosylbenzene diacetate, 8 were irradiated.

Reduction of the macrocyclic lactone (8a) or (8b) with Na–liquid ammonia cleanly removed the non-steroidal portion of the lactones to give 5α -cholestane- 7α , 25-diol (10)† in 84 and 75% yields, respectively.

Irradiation of epimeric esters (7) having a longer spacer in CCl_4 containing 3 equiv. of mercury(II) oxide and iodine similarly gave a mixture of homologous macrocyclic ether lactones (9a) and (9b),† albeit in low yield (2%), together with homologous benzophenone ester (4) (46%), and the recovered starting alcohol (7) (14%). Reduction of ether lactones (9a) and (9b) with Na-liquid ammonia gave 5α -cholestane- 7α , 25-diol (10) in a high yield.

A long-range intramolecular functionalization by the alkoxyl radical generated from epimeric esters (5), carrying a shorter spacer, failed to give any macrocyclic ether lactone corresponding to lactones (8) and (9), resulting only in the formation of the benzophenone ester (2).

The macrocyclic ethers (8a), (8b), (9a), and (9b) should be formed through a cyclization of a carbocation which is formed from one-electron oxidation of the C(25) tertiary radical generated by intramolecular hydrogen abstractions as outlined in Scheme 2.

The present long-range intramolecular functionalization involving 1,20 or 1,21 hydrogen transfer is the first example in which an oxygen atom is directly introduced into the remote

[†] Selected spectroscopic data for (5); a glass; IR v_{max} (Nujol) 3400 (OH), 1710 (C=O), and 1275 cm⁻¹ (C-O); ¹H NMR (270 MHz): δ $0.66\,[3H,\,s,\,H(18)],\,0.83\,[3H,\,s,\,H(19)],\,5.12\,[1H,\,br.\,s,\,7H(7\beta)],$ and 5.90 (1H, s, -CHOH). For (6); m. p. $107 \,^{\circ}$ C; IR v_{max} (Nujol) 3450 (OH), 1700 (C=O), and 1270 cm⁻¹ (C-O); ¹H NMR: δ 0.59 [3H, s, H(18)], 0.74 [3H, s, H(19)], 3.59 (2H, s, -COCH₂-), 4.82 [1H, br. s, $H(7\beta)$], and 5.82 (1H, s, -CHOH-). For (7); a glass; IR v_{max} (Nujol) 3450 (OH), 1700 (C=O), and 1270 cm⁻¹; 1 H NMR: δ 0.59 [3H, s, H(18)], 0.74 [3H, H(19)], 4.82 [1H, br. s, $H(7\beta)$], and 5.83 (1H, s, –CHOH–). For (2); a glass; IR ν_{max} (Nujol) 1710 (C=O), 1640 (Ph C=O), and 1270 cm⁻¹ (C-O); ¹H NMR: δ 0.68 [3H, s, H(18)], 0.85 [3H, s, H(19)], and 5.18 [1H, br. s, H(7 β)]. For (3); m. p. 129—130 °C; IR v_{max} (Nujol) 1660 (PhC=O) and 1270 cm⁻¹ (C-O); ¹H NMR: δ 0.58 [3H, s, H(18)], 0.76 [3H, s, H(19)], 3.71 (2H, s, -OCH₂-), 4.86 [1H, br. s, H(7 β)]. For (4); m.p. 97—101 °C; IR v_{max} (Nujol) 1725 (C=O), and 1650 (PhCO); ¹H NMR: δ 0.62 [3H, s, H(18)], 0.77 [3H, s, H(19)], 2.73 and 2.71 (each 2H, each t, J 7.3 Hz, $-OCOCH_2CH_2-$), and 4.90 [1H, br. s, $H(7\beta)$]. For (8a); m.p. 292—298 °C (light petroleum); IR v_{max} (Nujol) 1722 (C=O) and 1265 cm⁻¹ (C-O); ¹H NMR (400 MHz): $\delta 0.50 [3\text{H}, \text{s}, \text{H}(18)], 0.76 [3\text{H}, \text{s}, \text{H}(19)], 3.43 \text{ and}$ 3.49 (each 1H, each d, J 11.72 Hz, $-COCH_{2-}$), 4.73 [1H, br. s, $H(7\beta)$], and 5.43 (1H, s, –CHO–); m/z 610 (M^+ , 100) and 592 [(M – 18)⁺, 30%]. For (8b); m.p. 262—270 °C (light petroleum); IR v_{max} (Nujol) 1720 (C=O) and 1275 cm⁻¹ (C-O); ¹H NMR (400 MHz): δ 0.52 [3H, s, H(18)], 0.77 [3H, s, H(19)], 0.93 [3H, d, J 6.8 Hz, H(21)], 1.09 and 1.20 [each 3H, each s, H(26) and/or H(27)], 3.43 and 3.57 (each 1H, each d, J 12.2 Hz, $-COCH_{2}$ -), 4.81 [1H, br. d, J 2.73 Hz, $H(7\beta)$], and 5.56 (1H, s, -CHO-); m/z 610 (M^+ , 100) and 592 [(M-18)+, 6%]. For (10); m.p. 146—148 °C (light petroleum); IR v_{max} 3380 cm⁻¹ (OH); ¹H NMR (270 MHz): δ 0.66 [3H, s, H(18)], 0.78 [3H, s, H(19)], 0.92 [3H, d, J 6.6 Hz, H(21)], 1.21 [6H, s, H(26) and H(27)], and 3.81 [1H, br. s, H(7 β)]; m/z [field desorption (FD)MS], 404 (M^+ , 44), 386 $[M - H_2O, 34]$, 369 $[(M - H_2O - OH)^+, 71]$, and 59 $[(Me_2 C=O + H)^+, 100\%]$. For (**9a** and **b**); IR v_{max} (neat) 1720 cm⁻¹ (C=O); ¹H NMR (90 MHz): δ 0.53 and 0.66 [each 3H, each s, H(18) and H(19)].

position as a result of a long-range intramolecular hydrogen abstraction.

The results of our further investigation of this long-range oxygenation of steroid skeletons will be reported in our future publications.‡

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