Preliminary communication

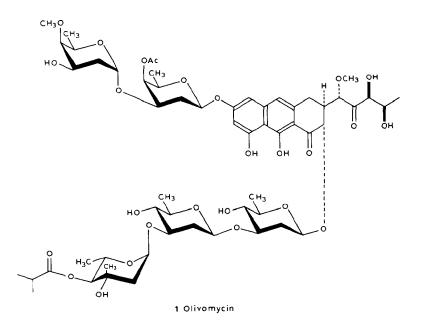
Preparation of 2-deoxy-β-D-lyxo-hexosides (2-deoxy-β-D-galactosides)

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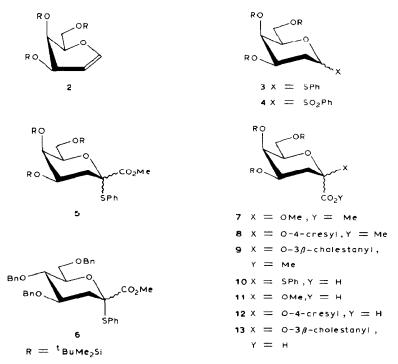
With a view to the eventual synthesis of the trisaccharide chain of the antitumor antibiotic olivomycin A¹(1), we have studied the preparation² of heptulosonic acid O-glycosides and their stereoselective radical decarboxylation³ to 2-deoxy- β -Darabino-hexosides (2-deoxy- β -D-glucosides). We now report the preparation of 2deoxy- β -D-lyxo-hexosides (2-deoxy- β -D-galactosides), relevant to the disaccharide chain of olivomycin A, and improvements in the synthesis of heptulosonic acids.



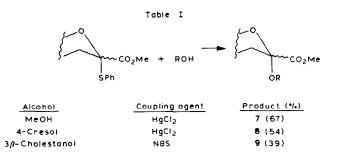
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 Treatment of D-galactal⁴ with 3.5 equiv. of *tert*-butyldimethylsilyl chloride and 7 equiv. of imidazole in N, N-dimethylformamide at 60° for 2 days gave 86% of the D-galactal derivative 2⁺ as a colourless oil, $[\alpha]_D^{25} = -37^\circ$ (c 1, chloroform). ¹H-N.m.r. data (CDCl₃): δ 6.22 (d, 1 H, $J_{1,2}$ 7 Hz, H-1), 4.65 (m, 1 H, H-2). Reaction of **2** with 1 equiv. of hydrogen chloride in toluene at 0°, and subsequently with 1.3 equiv. of thiophenol and 1.3 equiv. of ethyldi-isopropylamine in dichloromethane at 25°, afforded 76% of the thioglycoside **3**, after chromatography on silica gel, as a 1:14 $\alpha\beta$ -mixture. ¹H-N.m.r. data (CDCl₃): β anomer, δ 4.74 (dd, 1 H, $J_{1,2e}$ 2.2, $J_{1,2a}$ 11.9 Hz, H-1); α anomer, δ 5.62 (d, 1 H, J 4.8 Hz, H-1). Alternatively, the camphor-10-sulphonic acid-catalysed reaction of **2** with 4 equiv. of thiophenol in dichloromethane in the presence of powdered molecular sieves gave 60% of **3** isolated as a 15:1 $\alpha\beta$ -mixture.

Oxidation of crude 3 (obtained by the former procedure) with 2.4 equiv. of 3-chloroperoxybenzoic acid in dichloromethane at 0° in the presence of sodium hydrogencarbonate gave the corresponding sulphone 4 (67% from 2). ¹H-N.m.r. data (CDCl₃): β anomer, δ 4.45 (dd, 1 H, $J_{1,2e}$ 2, $J_{1,2a}$ 12 Hz, H-1); α anomer, δ 4.78 (dd, 1 H, J 2.2 and 6.3 Hz, H-1). The sulphone 4 was obtained (96%) by oxidation of pure 3 with 1.05 equiv. of commercial magnesium monoperoxyphthalate⁵ in ethanol at 5°. Sequential treatment of 4 in tetrahydrofuran at -70° with a solution of butyl-lithium (1.2 equiv.), dimethyl carbonate (1.3 equiv.), a solution of lithium naphthalenide (2.5 equiv.) in tetrahydrofuran, and diphenyl disulphide (2.0 equiv.),

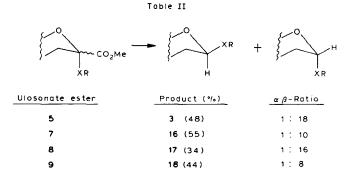


[†]All new products gave satisfactory spectroscopic and microanalytical data (C,H $\pm 0.4\%$). With the exception of 5, all of the new compounds were isolated as syrups.

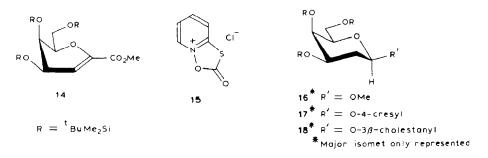


followed by aqueous work-up and chromatography on silica gel (ether-light petroleum 1:19), gave 63% of methyl [phenyl 4,5,7-tri-*O*-(*tert*-butyldimethylsilyl)-3-deoxy-2-thio-D-*lyxo*-2-heptulopyranosid]onate (**5**) as an 8:1 $\alpha\beta$ -mixture, m.p. 99– 100° (from ether-light petroleum). ¹H-N.m.r. data (CDCl₃): α anomer, δ 2.39 (dd, 1 H, H-2*a*) and 2.14 (dd, 1 H, H-2*e*). The above procedure gave significantly better yields of **5** than when the intermediates were isolated and purified.

Compound 5 was coupled with alcohols using N-bromosuccinimide, mercuric acetate, or mercuric chloride as the activating agent, as described for the glycosyl donor 6. Thus, storage of a mixture of 5, alcohol (2 equiv.), powdered molecular sieves, and the activating agent (1.1 equiv.) at room temperature gave, after chromatography, the corresponing 2-deoxy- α , β -galactosides 7-9 (see Table I). The unsaturated ester 14 was a minor product of these reactions. ¹H-N.m.r. data (CDCl₃): δ 5.83 (d, 1 H, J 3.2 Hz, H-2). Saponification of the esters 5 and 7-9 afforded the corresponding acids (10-13) which were not purified but decarboxylated by reaction with 3-oxa-2-oxo-1-thiaindolizinium chloride⁶ (15) and triethylamine in dichloromethane, followed by photolysis of the resulting *O*-acyl thiohydroxamates⁷, in the presence of *tert*-dodecanethiol. For example, the ester 8 was saponified with aqueous potassium hydroxide (1.5 equiv.) in methanol-tetra-hydrofuran and the crude acid 12 was obtained by extraction with ether after acidification to pH 2. Treatment of 12 in dichloromethane with the salt 15 (1 equiv.) in the presence of powdered molecular sieves at 20° under nitrogen for 5 min,



followed by the addition of triethylamine (1 equiv.), gave the yellow thiohydroxamate which was treated with *tert*-dodecanethiol (5 equiv.) and irradiated for 1.5 h with a 500-W tungsten lamp under nitrogen at 5°. Chromatography of the product gave 34% of the 4-cresyl galactoside **17** with a 1:16 $\alpha\beta$ -ratio. ¹H-N.m.r. data (CDCl₃): β anomer, δ 4.99 (dd, 1 H, $J_{1,2e}$ 2.3, $J_{1,2a}$ 9.8 Hz, H-1). The results of the radical decarboxylation reactions are given in Table II. Each of the 2-deoxy-Dgalactosides (**3**, and **16–18**) had good β -selectivities, demonstrating that the new method for their preparation tolerates both axial and equatorial 4-substituents.



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REFERFNCES

- 1 W. A. REMERS, in *The Chemistry of Antitumor Antibiotics*, Vol. 1, Wiley Interscience, New York, 1979, pp. 133-175.
- 2 D. CRICH AND T. J. RITCHIE, J. Chem. Soc., Chem. Commun., (1988) 985-986.
- 3 D. CRICH AND T. J. RITCHIE, J. Chem. Soc., Chem. Commun., (1988) 1461-1463.
- 4 F. SHAFIZADEH, Methods Carbohydr. Chem., 2 (1963) 409-410.
- 5 P. BROUGHMAN, M. S. COOPER, D. A. CUMMERSON, H. HEANEY, AND N. THOMPSON, Synthesis, (1987) 1015–1017.
- 6 D. H. R. BARTON, D. CRICH, AND W. B. MOTHERWELL, Tetrahedron, 41 (1985) 3901-3924.
- 7 D. CRICH, Aldrichimica Acta, 20 (1987) 35-42.