

1,4-Addition Reaction of 5H-Oxazol-4-ones to Allenic Esters and Ketones Catalyzed by Chiral Guanidines

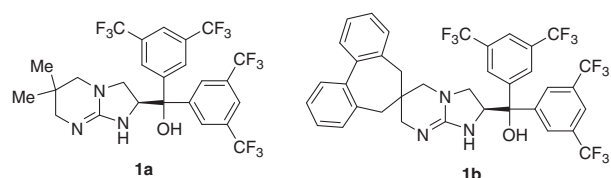
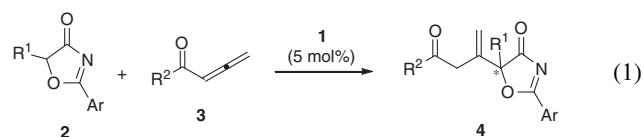
Nari Jin, Tomonori Misaki,* and Takashi Sugimura*

Graduate School of Material Science, University of Hyogo, 3-2-1 Kohto, Kamigori, Ako-gun, Hyogo 678-1297

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In this paper, a chiral guanidine-catalyzed 1,4-addition reaction of 5H-oxazol-4-ones to allenic esters and ketones is described. 5H-Oxazol-4-ones substituted with a 2-chlorophenyl group were suitable pronucleophiles that gave high enantioselectivities. Subsequent hydrolysis of the obtained adduct gave the corresponding γ -butenolide ester without loss of enantiopurity and was transformed into (+)-crobarbic acid in a few steps.

Recently, electron-deficient allene compounds have been used as reactive electrophilic reactants in several organic reactions¹ including catalytic asymmetric carbon–carbon bond-forming reactions.² Some conjugate 1,4-addition reactions of carbon nucleophiles to allenyl carbonyl compounds have also been developed.³ However, Jørgensen et al. reported the only successful example of catalytic asymmetric 1,4-addition of carbon nucleophiles to allenyl carbonyl compounds in 2008.^{2c} We have successfully developed addition reactions of 5H-oxazol-4-ones as pronucleophiles catalyzed by bicyclic chiral guanidines⁴ bearing a hydroxy group,⁵ and disclosed that the combination of 5H-oxazol-4-one and the guanidine is quite effective for the highly enantioselective carbon–carbon bond formation producing a less-accessible chiral building block, α -oxygen atom-substituted carboxylates bound to a chiral quaternary α -carbon atom.⁶ Because the catalytic enantioselective carbon–carbon bond formation producing synthetically useful α -hydroxy carboxylates is limited because of the difficulty in the effective enolate generation of glycolate derivatives,⁷ we here developed a method for the asymmetric 1,4-addition of 5H-oxazol-4-ones to allenic esters and ketones catalyzed by chiral guanidines as part of our continuous study (eq 1).



We selected octyl allenic ester **3a** as an electrophile for the development of the new 1,4-addition method, and the initial attempt at reacting α -methyl group-substituted 5H-oxazol-4-one (**2a**: Ar = Ph) to **3a** using guanidine **1a** revealed that the 1,4-addition proceeded with appreciable enantioselectivity, as shown in Table 1, Entry 1. Note that, the migration of the β,γ

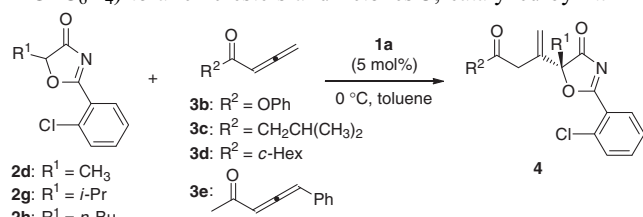
Table 1. Optimization of the 1,4-addition of 5H-oxazol-4-ones **2** to allenic esters **3** using catalyst **1a** or **1b**^a

Entry	Substrates		Cat.	Time /h	Product 4	Yield /%	ee ^b /%
	2 : Ar	3					
1	2a : Ph	3a	1a	12	4a	66	75
2	2a : Ph	3a	1b	19	4a	65	62
3	2a : Ph	3b	1a	3.5	4b	76	86
4	2b : 4-Cl-C ₆ H ₄	3b	1a	0.5	4c	68	3
5	2c : 3-Cl-C ₆ H ₄	3b	1a	4	4d	68	53
6	2d : 2-Cl-C ₆ H ₄	3b	1a	1	4e	85	94
7	2e : 2-F-C ₆ H ₄	3b	1a	1	4f	81	93
8	2f : 2-Br-C ₆ H ₄	3b	1a	1	4g	83	93
9 ^c	2d : 2-Cl-C ₆ H ₄	3b	1a	1	4e	88	97
10 ^c	2d : 2-Cl-C ₆ H ₄	3b	1b	2	4e	84	93

^aReactions were performed on a 0.3 mmol scale in 1.0 mL of toluene using 1.2 equiv of allenyl carbonyl compound **3** and 5 mol % of catalyst **1a** or **1b**. ^bDetermined by chiral HPLC analysis. ^cReaction was performed at –20 °C.

carbon–carbon double bond on adduct **4** was not observed during the 1,4-addition. The use of guanidine **1b** did not enhance the enantioselectivity in the 1,4-addition to **3a** (Entry 2), whereas guanidine **1b** was a suitable catalyst for the 1,4-addition to alkynones, as shown in our previous report.^{5c} The use of phenyl allenic ester **3b** instead of **3a** enhanced the reaction rate and enantioselectivity (Entry 3). Thus, by using **3b** as an electrophile and **1a** as a catalyst, we investigated the effect of the aromatic substituent (Ar) on 5H-oxazol-4-one **2** to further improve enantioselectivity (Entries 4–8). The 1,4-additions of 5H-oxazol-4-ones **2b** and **2c** showed low enantioselectivity (Entries 4 and 5). When the 2-chlorophenyl group-substituted 5H-oxazol-4-one **2d** was used as a pronucleophile, the enantioselectivity of the 1,4-addition was improved to 94% ee (Entry 6). The 1,4-additions of the 2-fluoro and 2-bromo analogues **2e** and **2f**, respectively, also showed high chemical yields and enantioselectivities (Entries 7 and 8). The enantioselectivity was improved to 97% ee by lowering the reaction temperature (Entry 9).

Next, we applied the optimized conditions to the 1,4-addition of some 5H-oxazol-4-ones **2d**, **2g**, and **2h** (Ar = 2-Cl-C₆H₄) to phenyl allenic ester **3b** using catalyst **1a** (Table 2, Entries 1–3). As a result, adducts **4**^{8,9} were obtained in high chemical yields with high enantioselectivities, even in the case of bulky pronucleophile **2g**. We also performed the 1,4-addition with allenic ketones **3c** and **3d** as electrophiles, and the

Table 2. Catalytic 1,4-addition of 5*H*-oxazol-4-ones **2** (Ar = 2-Cl-C₆H₄) to allenic esters and ketones **3**, catalyzed by **1a**^a


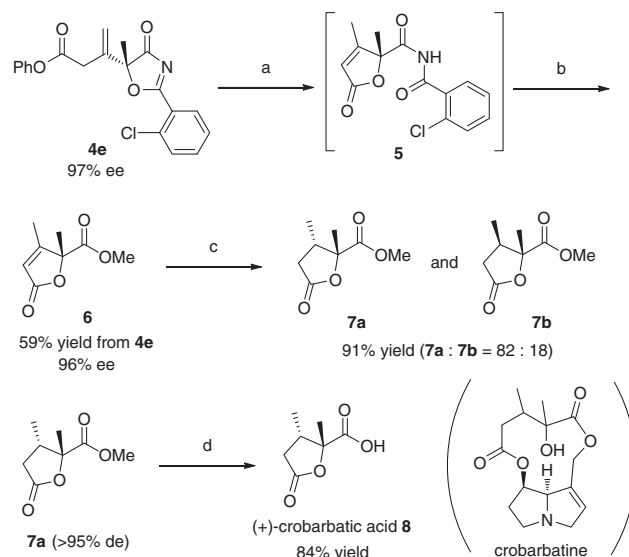
Entry	Substrates 2	3	Time /h	Product 4	Yield /%	ee ^b /%
1 ^c	2d	3b	1	4e	88	97
2 ^c	2g	3b	2	4h	81	96
3	2h	3b	0.5	4i	77	93
4	2d	3c	1	4j	83	98
5	2d	3d	1	4k	81	96
6	2d	3e	24 ^d	—	<40 ^e	—

^{a,b}See corresponding footnote in Table 1. ^cReaction was performed at –20 °C. ^dReaction was performed at 0 °C for 10 h then at rt for 14 h. ^eConversion yield.

corresponding adducts **4j** and **4k** were obtained in high chemical yields with excellent enantioselectivities (Entries 4 and 5). Unfortunately, the 1,4-addition to γ -phenyl allenic ketone **3e** slowly proceeded under the optimized conditions, and a low conversion yield (<40%) was observed after prolonged stirring at room temperature (Entry 6).

Adduct **4e** formed by the 1,4-addition of **2d** to **3b** can be easily converted into the corresponding γ -butenolide **6** without loss of enantiopurity.⁹ As illustrated in Scheme 1, the γ -butenolide ring formation, after the ring-opening hydrolysis of 5*H*-oxazol-4-one through the treatment of **4e** with aqueous NaOH in THF, readily afforded the corresponding imide **5**, which was converted into γ -butenolide **6** by CH₃OLi-mediated methanolysis. The enantiopurity of **6** was confirmed by HPLC analysis. To assign the absolute configuration of adduct **4e** and also to demonstrate the utility of the present 1,4-addition, we converted γ -butenolide **6** into (+)-crobarbatic acid **8**,^{10,11} which is a hydrolysis product of a pyrrolizidine alkaloid crobarbatine.¹¹ The Pd–C-catalyzed hydrogenation of γ -butenolide **6** afforded a mixture of diastereoisomers **7a** and **7b** (**7a**:**7b** = 82:18), and the major isomer **7a** was isolable through purification by simple column chromatography. Finally, the hydrolysis of isomer **7a** (>95% de) gave (+)-crobarbatic acid (**8**) in 84% yield. The absolute configuration of obtained **8** was assigned as (2*R*, 3*S*) by comparing its optical rotation value with the reported value;¹⁰ hence, the absolute configuration of adduct **4e** was *R*. All spectral data for obtained **8** were also consistent with the reported data.¹⁰

In conclusion, we developed a highly enantioselective 1,4-addition of 5*H*-oxazol-4-ones **2** (Ar = 2-Cl-C₆H₄) to allenic esters and ketones **3** using chiral guanidine **1a**. By examining the effect of Ar on **2**, we found that the 2-chlorophenyl group was suitable for the 1,4-addition and gave high enantioselectivities. An obtained adduct **4e** of the 1,4-addition was derived into (+)-crobarbatic acid in a few steps.



Scheme 1. Derivatization of adduct **4e** to **6** and (+)-crobarbatic acid (**8**). (a) 1.0 M NaOH (aq), THF, 0 °C, 1 h. (b) 1.0 M CH₃OLi, CH₃OH, 0 °C, 2.5 h. (c) 10% Pd–C, H₂, MeOH, rt, 24 h. (d) 4.0 M NaOH (aq), THF, 100 °C, 4 h, then 2 M HCl (aq), 100 °C, 1 h. Optical rotation value of **8**: [α]_D²⁰ = +3.76 (*c* 0.66, H₂O) [lit.¹⁰ [α]_D²⁵ = +3.56 (*c* 1.4, H₂O) (2*R*, 3*S*)].

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References and Notes

- a) S. Ma, *Chem. Rev.* **2005**, *105*, 2829. b) S. Yu, S. Ma, *Angew. Chem., Int. Ed.* **2012**, *51*, 3074.
- a) D. Zhao, K. Oisaki, M. Kanai, M. Shibasaki, *J. Am. Chem. Soc.* **2006**, *128*, 14440. b) K. Oisaki, D. Zhao, M. Kanai, M. Shibasaki, *J. Am. Chem. Soc.* **2007**, *129*, 7439. c) P. Elsner, L. Bernardi, G. D. Salla, J. Overgaard, K. A. Jørgensen, *J. Am. Chem. Soc.* **2008**, *130*, 4897. d) Y.-Q. Fang, E. N. Jacobsen, *J. Am. Chem. Soc.* **2008**, *130*, 5660. e) H. Xiao, Z. Chai, C.-W. Zheng, Y.-Q. Yang, W. Liu, J.-K. Zhang, G. Zhao, *Angew. Chem., Int. Ed.* **2010**, *49*, 4467. f) X. Han, Y. Wang, F. Zhong, Y. Lu, *J. Am. Chem. Soc.* **2011**, *133*, 1726. g) J.-B. Denis, G. Masson, P. Retailleau, J. Zhu, *Angew. Chem., Int. Ed.* **2011**, *50*, 5356. h) X. Wang, T. Fang, X. Tong, *Angew. Chem., Int. Ed.* **2011**, *50*, 5361. i) Q.-Y. Zhao, C.-K. Pei, X.-Y. Guan, M. Shi, *Adv. Synth. Catal.* **2011**, *353*, 1973. j) A. T. Brusoe, E. J. Alexanian, *Angew. Chem., Int. Ed.* **2011**, *50*, 6596. k) X. Han, F. Zhong, Y. Wang, Y. Lu, *Angew. Chem., Int. Ed.* **2012**, *51*, 767. l) Q. Zhao, X. Han, Y. Wei, M. Shi, Y. Lu, *Chem. Commun.* **2012**, *48*, 970. m) C.-K. Pei, Y. Jiang, Y. Wei, M. Shi, *Angew. Chem., Int. Ed.* **2012**, *51*, 11328.
- a) Y. H. Paik, P. Dowd, *J. Org. Chem.* **1986**, *51*, 2910. b) R. K. Dieter, K. Lu, S. E. Velu, *J. Org. Chem.* **2000**, *65*, 8715. c) S. Ma, S. Yu, S. Yin, *J. Org. Chem.* **2003**, *68*, 8996. d) S. Lucas, U. Kazmaier, *Synlett* **2006**, 255. e) Z. Lu, G. Chai, S. Ma, *J. Am. Chem. Soc.* **2007**, *129*, 14546. f) G. Chai, Z. Lu, C. Fu, S. Ma, *Adv. Synth. Catal.* **2009**, *351*, 1946.

- 4 For the chiral guanidine catalysis, see: a) T. Ishikawa, *Guanidines in Organic Synthesis in Superbases for Organic Synthesis: Guanidines, Amidines, Phosphazenes and Related Organocatalysts*, ed. by T. Ishikawa, John Wiley & Sons, Chippenhams, **2009**, pp. 93–143. doi:10.1002/9780470740859. For a review, see: b) D. Leow, C.-H. Tan, *Chem.—Asian J.* **2009**, *4*, 488, and references therein.
- 5 a) T. Misaki, G. Takimoto, T. Sugimura, *J. Am. Chem. Soc.* **2010**, *132*, 6286. b) T. Misaki, K. Kawano, T. Sugimura, *J. Am. Chem. Soc.* **2011**, *133*, 5695. c) T. Misaki, N. Jin, K. Kawano, T. Sugimura, *Chem. Lett.* **2012**, *41*, 1675. For the theoretical study of an 1,4-addition of 5*H*-oxazol-4-one, see: d) N. Lu, H. Wang, *Int. J. Quantum Chem.* **2013**, in press. doi:10.1002/qua.24447.
- 6 Examples of catalytic chiral quaternary α -carbon atom constructions, see: a) B. M. Trost, K. Dogra, M. Franzini, *J. Am. Chem. Soc.* **2004**, *126*, 1944. b) T. Ooi, K. Fukumoto, K. Maruoka, *Angew. Chem., Int. Ed.* **2006**, *45*, 3839. c) P. S. Hynes, D. Stranges, P. A. Stupple, A. Guarna, D. J. Dixon, *Org. Lett.* **2007**, *9*, 2107. d) K. Zhang, Q. Peng, X.-L. Hou, Y.-D. Wu, *Angew. Chem., Int. Ed.* **2008**, *47*, 1741. e) T. Hashimoto, K. Fukumoto, N. Abe, K. Sakata, K. Maruoka, *Chem. Commun.* **2010**, *46*, 7593. f) H. Huang, K. Zhu, W. Wu, Z. Jin, J. Ye, *Chem. Commun.* **2012**, *48*, 461. g) B. M. Trost, K. Hirano, *Angew. Chem., Int. Ed.* **2012**, *51*, 6480.
- 7 Addition reactions to α -ketoesters are also important methods for the preparation of α -hydroxy carboxylates bound to a chiral quaternary α -carbon atom. Examples of transition-metal catalysis: a) K. Aikawa, Y. Miyazaki, K. Mikami, *Bull. Chem. Soc. Jpn.* **2012**, *85*, 201. b) T.-S. Zhu, S.-S. Jin, M.-H. Xu, *Angew. Chem., Int. Ed.* **2012**, *51*, 780. Examples of organocatalysis: c) C. Zheng, Y. Wu, X. Wang, G. Zhao, *Adv. Synth. Catal.* **2008**, *350*, 2690. d) M. Rueping, T. Theissmann, A. Kuenkel, R. M. Koenigs, *Angew. Chem., Int. Ed.* **2008**, *47*, 6798. e) J. Nie, G.-W. Zhang, L. Wang, D.-H. Zheng, Y. Zheng, J.-A. Ma, *Eur. J. Org. Chem.* **2009**, 3145. f) J. Luo, H. Wang, X. Han, L.-W. Xu, J. Kwiatkowski, K.-W. Huang, Y. Lu, *Angew. Chem., Int. Ed.* **2011**, *50*, 1861.
- 8 General procedure: To a stirred solution of 5*H*-oxazol-4-one **2** (0.30 mmol) and allenyl carbonyl compound **3** (0.36 mmol) in toluene (1.0 mL) was added chiral guanidine **1** (15 μ mol) at -20 or 0°C . After stirring at same temperature for indicated time in Table 1 or 2 under a N_2 atmosphere. The resulting reaction mixture was concentrated to give a crude mixture, which was purified by SiO_2 -column chromatography to give the adduct **4**.
- 9 Supporting Information is available electronically on the CSJ-Journal Web site, <http://www.csj.jp/journals/chem-lett/index.html>.
- 10 D.-P. Jang, J.-W. Chang, B.-J. Uang, *Org. Lett.* **2001**, *3*, 983.
- 11 a) A.-F. M. Rizk, *Naturally Occurring Pyrrolizidine Alkaloids*, CRC Press, Boca Raton, FL, **1991**. b) S. C. Puri, R. S. Sawhney, C. K. Atal, *Experientia* **1973**, *29*, 390. c) J. Huang, J. Meinwald, *J. Am. Chem. Soc.* **1981**, *103*, 861. d) M.-Y. Chen, J.-M. Fang, *J. Org. Chem.* **1992**, *57*, 2937. e) T. Honda, F. Ishikawa, S.-i. Yamane, *J. Chem. Soc., Perkin Trans. 1* **1996**, 1125. f) T. J. Donohoe, C. A. Stevenson, M. Helliwell, R. Irshad, T. Ladduwahetty, *Tetrahedron: Asymmetry* **1999**, *10*, 1315.