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REACTION OF ETHYL 3-OXOBUTANOATE AND ETHYL 4-BROMO-3-OXOBUTANOATE WITH BENZENE IN THE PRESENCE OF ALUMINUM CHLORIDE

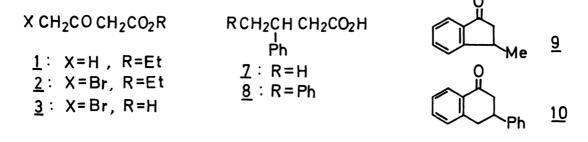
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Refluxing of ethyl 3-oxobutanoate $(\underline{1})$ in benzene in the presence of aluminum chloride gave ethylbenzene $(\underline{4})$, 9,10-dimethylanthracene $(\underline{5})$, 3-phenylbutanoic acid $(\underline{7})$, and 3-methyl-1-indanone $(\underline{9})$. Similar reaction of ethyl 4-bromo-3-oxobutanoate $(\underline{2})$ yielded $\underline{4}$, $\underline{5}$, 4-bromo-3-oxobutanoic acid $(\underline{3})$, 3,4-diphenylbutanoic acid $(\underline{8})$, and 3-phenyl-1-tetralone $(\underline{10})$.

While investigating some potential uses of diketene, we have studied reactions of 4-halo-3-oxobutanoate which is most easily prepared from diketene.¹⁾ In the present paper, we wish to report its Friedel-Crafts reaction. Concerning this reaction Labunskii reported the reaction between ethyl 3-oxobutanoate (<u>1</u>) and benzene in the presence of aluminum chloride to give ethylbenzene (<u>4</u>), 9,10-di-methylanthracene (5), and phenylacetic acid (6).²

First, we reinvestigated this reaction and obtained some variant results. According to the procedure reported by Labunskii², two equivalents of aluminum chloride were added gradually in portions to a solution of the ester (<u>1</u>) in benzene with stirring. After heating at 70-80° for 3 hr, the reaction mixture was poured into cold HCl, and the benzene layer was fractionally distilled to give ethylbenzene (<u>4</u>) (35%), bp 130-133°, and the starting ester (<u>1</u>) (1%), bp 75-80° (20 mmHg). The residue was dissolved in ether, and the ether solution was washed with 5% NaHCO₃. The NaHCO₃ washing was acidified with 10% HCl to give 3-phenylbutanoic acid (<u>7</u>) (1%) bp 107-109° (1 mmHg) (11t³) bp 113-115° (2 mmHg)). The ether layer was purified by silica gel column chromatography to give 9,10-dimethylbanthracene (trace) (<u>5</u>), mp 177-178° (11t⁴) mp 180°). When the reaction was carried out in the presence of five equivalents of aluminum chloride, products obtained were <u>4</u> (9.4%), <u>5</u> (29%), <u>7</u> (41.5%), and 3-methyl-1-indanone (<u>9</u>) (50%), bp 80° (1 mmHg) (11t⁵) bp 118-119° (11 mmHg)). Phenylacetic acid (<u>6</u>) was not detected.



Next, Friedel-Crafts reaction of ethyl 4-bromo-3-oxobutanoate (2) was carried out. To a boiling suspension of aluminum chloride in dry benzene, was added dropwise a solution of the bromoester (2) in benzene. After additional refluxing, the reaction mixture was poured into a mixture of conc. HCl and ice with stirring. The benzene layer separated was washed with 10% Na_2CO_3 . The aqueous layer was acidified to give 3,4-diphenylbutanoic acid (8), mp 93-94° (lit⁶⁾ mp 96-97°), and 4-bromo-3oxobutanoic acid (3), mp 66-67° (lit⁷⁾ mp 69-69.5°). The benzene layer was distilled to give ethylbenzene (4), and the residue was purified by silica gel column chromatography to give 9,10-dimethylanthracene (5), and 3-phenyl-1-tetralone (10), mp 64-65° (lit⁸⁾ mp 65°). The results are summarized in Table I.

| | | | Reaction | Reaction | Yield(%) | | | | | |
|---------|--------------------|---------------------|----------|-----------------|----------|-----------------|------------------|------------------|----------|-----------------|
| Benzene | 2 | A1C13 | Time(hr) | Temperature(°C) | 2 | <u>3</u> | | <u>5</u> | <u>8</u> | <u>10</u> |
| 20 m1 | | 1.34 g (0.01mol) | 3 | 20 | 71 | ₊ a) | + ^b) | _ | _ | _ |
| 20 mi | 2.1 g (0.01mo1) | 1.34 g (0.01mol) | 3 | 80 | 52 | + ^{a)} | 9 | + ^a) | _ | |
| 20 m1 | 2.1 g (0.01mo1) | 2.68 g (0.02mol) | 1 | 80 | 46 | 15 | 12 | ₊ a) | _ | _ |
| 20 m1 | 2.1 g (0.01mo1) | 2.68 g (0.02mol) | 3 | 80 | - | - | + ^b) | 2 | 11 | ₊ a) |
| 20 m1 | 2.1 g (0.01mo1) | | 3 | 80 | - | _ | + ^{b)} | 28 | 47 | 5 |
| 20 m1 | 2.1 g (0.01mo1) | 6.7 g (0.05mol) | 3 | 80 | | _ | + ^p) | 29 | 38 | 41 |

| Table | I | Reaction | of | ethv] | 4-bromo-3-oxobutanoate | (2) | with | benzene |
|-------|---|----------|----|-------|------------------------|-----|------|---------|
|-------|---|----------|----|-------|------------------------|-----|------|---------|

a) These compounds were identified by silica gel thin layer chromatography.

b) This compound was identified by gas chromatography on a 2 m×2.5 mm silicon OV-17 (5% on Chromosorb AW-HMDS) column at 100°.

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