INTRAMOLECULAR CYCLISATION-N-DEALKYLATION OF AZETIDINE-3-ACETIC ACIDS David Bartholomew* and Michael J.Stocks.

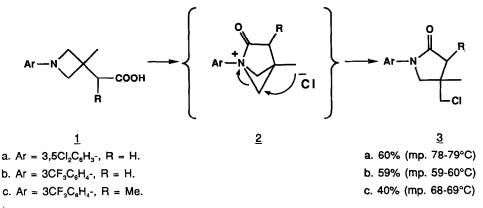
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<u>Summary</u>. N-Aryl-azetidine-3-acetic acids are converted to pyrrolidinones when treated with oxalyl chloride. The intramolecular cyclisation-N-dealkylation reaction occurs via an azabicyclic intermediate. N-Aryl-azetidine-3-acetic acids are prepared from N-aryl-3-(chloromethyl)azetidin-2-ones.

Intramolecular cyclisation-N-dealkylation reactions occur if an acyl chloride moeity is formed within a molecule which contains a tertiary amine.¹ In this way, novel N-aryl-azetidine-3-acetic acids <u>1</u> on attempted conversion into acid chlorides with oxalyl chloride give 4-(chloromethyl)pyrrolidine-2-ones <u>3</u> via the azabicyclic intermediate <u>2</u>. (Scheme 1.)

Intramolecular cyclisation-N-dealkylation reactions, via an azabicylic intermediate, have not, to our knowledge, been previously described with azetidines. However, it is known that pyrrolidinones are obtained, via azabicyclic intermediates, when piperidine-4-carboxylic acids² and also pyrrolidine-3-acetic acids³ react with thionyl chloride.

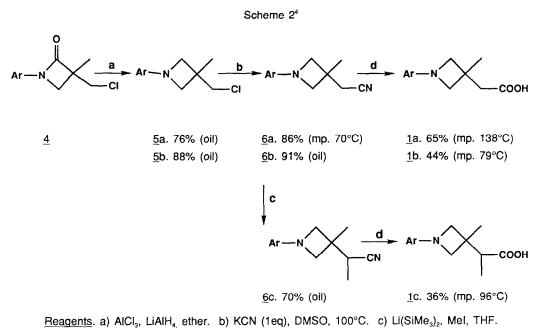
SCHEME 1⁴



The N-aryl-azetidine-3-acetic acids $\underline{1}$ are prepared (Scheme 2) from N-aryl-3-(chloromethyl)-3methylazetidin-2-ones $\underline{4}^5$ by reduction with hydrolane⁶ to give $\underline{5}$. Nucleophilic substitution with cyanide anion gives $\underline{6}$ followed by alkaline hydrolysis to azetidines $\underline{1}a$ and $\underline{1}b$. Azetidine-3-acetic acid $\underline{1}c$ is obtained by methylation of 6b, followed by alkaline hydrolysis.

Typical reaction conditions for conversion of $\underline{1}$ to $\underline{3}$ are as follows:- To a stirred solution of $\underline{1}$ (1 eq.) in dichloromethane (20ml) and dimethylformamide (4 drops) is added oxalyl chloride (1 eq.), dropwise at room temperature. The reaction mixture is stirred at room temperature for 3 hours and then left to stand overnight. The solvent is removed by evaporation and the residue purified by flash column

chromatography (ether : hexane; 1:1). 1a and 1b are isolated as R/S mixtures and 6c is an 8 : 1 mixture in which the 3S,4R; 3R,4S pair is the major product and the 3S,4S; 3R,4R pair the minor product.



d) NaOH (4eq.) aq. EtOH.

The synthesis of the N-aryl-azetidine-3-acetic acids, described here, is limited. Further studies are in hand to develop a new route to these acids to enable further investigation of cyclisation-N-dealkylation reactions of azetidines which will be reported shortly.

References and footnotes.

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- 2. R.L. Clarke, A. Mooradian, P. Lucas and T.J. Slauson. J. Amer. Chem. Soc., 71, 2821 (1949).
- C.D. Lumsford, A.D. Cale Jr., J.W. Ward, B.V. Franco and H. Jenkins. <u>J. Med. Chem.</u>, 7, 302 (1964).
- All quoted yields are pure and isolated. All products give satisfactory microanalytical (M.F. Jamieson, ICI Agrochemicals) and NMR data (P. Stanley, ICI Agrochemicals)
- 5. D. Bartholomew and M.J. Stocks. *Tet. Lett.*, (1991).
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 N-Alkyl-3-(chloromethyl)-3-methylazetidin-2-ones are not reduced to the corresponding azetidines. Ring opening and decomposition are thought to occur.

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