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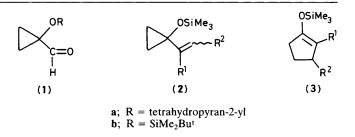
Cyclopentenones from the Acid-induced Ring Expansion of 1-Alkenylcyclopropanol Derivatives

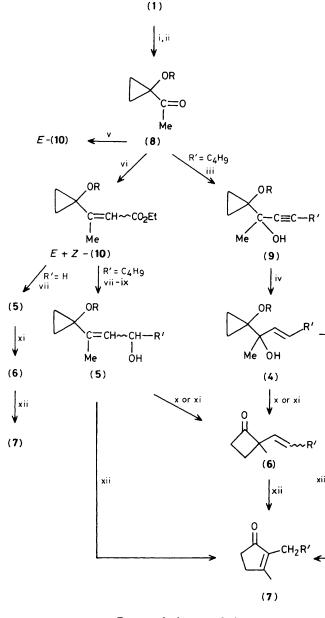
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1-Alkenylcyclopropanols (**4a,b**) and (**5a,b**) underwent acid-induced ring expansion into cyclopentenones (**7**), *via* 2-alkenylcyclobutanones (**6**); the 2-methylcyclopropanols (**15a,b**) are also synthesised using the same method.

Cyclopropanecarbaldehyde derivatives (1) constitute useful building blocks for the construction of five-membered ring moieties as illustrated by the syntheses of jasmonoid,¹ spirovetivane,² and dicranenone³ compounds. These syntheses are based on thermal vinylcyclopropanecyclopentene ring expansion of the cyclopropanes (2), readily available from (1),¹⁻⁴ into the regiospecific cyclopentanone enol ethers (3) which then undergo either acidic and basic hydrolysis or dehydrosilation to provide the corresponding



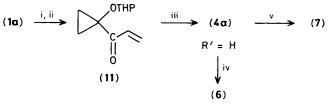


a; $\mathbf{R} = \text{tetrahydropyran-2-yl}$ **b**; $\mathbf{R} = \text{SiMe}_2\text{Bu}^t$

Scheme 1. Reagents and conditions: i, MeMgI, Et₂O reflux, 2 h; ii, dimethylsulphoxide (DMSO)–(COCl)₂, CH₂Cl₂, -60 °C; iii, LiC=C-[CH₂]₃Me, tetrahydrofuran (THF), 0 °C; iv, LiAlH₄, THF reflux, 3 h; v, (EtO)₂P(O)CHCO₂Et, THF reflux, 35% yield; vi, LiC-(SiMe₃)HCO₂Et, THF, -78 °C, 75%; vii, Buⁱ₂AlH, toluene, -70 °C, 98%; viii, DMSO–(COCl)₂, CH₂Cl₂, -60 °C; ix, BuⁿMgBr, Et₂O reflux, 2 h; x, BF₃–Et₂O (0.1 mol equiv.), CHCl₃, room temp., 5 min; xi, MeSO₃H–P₂O₅ (17 equiv.), 6 h, room temp.

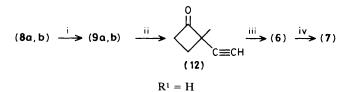
 α,β -disubstituted cyclopentanones or cyclopentenones. This communication reports that (1) can also provide 1-alkenylcyclopropanol derivatives such as (4) and (5) which undergo acid induced ring expansion, *via* the intermediacy of the cyclobutanones (6), into cyclopentenone derivatives (7).

Addition of MeMgI to (1a,b) followed by oxidation⁵ gave (8a,b) in 96% yield. Addition of hex-1-ynyl-lithium led to the octynols (9a,b) ($\mathbf{R}' = C_4\mathbf{H}_9$) which on reduction provided the *trans* vinyl alcohols (4a,b) ($\mathbf{R}' = C_4\mathbf{H}_9$) (90%). On the other

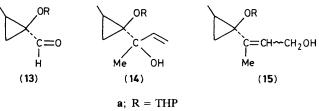


THP = tetrahydropyran-2-yl

Scheme 2. Reagents and conditions: i, $CH_2=CHMgBr$, THF, 20 °C; ii, $DMSO-(COCl)_2$, CH_2Cl_2 , -60 °C; iii, MeMgBr, Et_2O , 0 °C, 50%; iv, $MeSO_3H-P_2O_5$ (0.1 equiv.), Et_2O , 5 min, 95%; v, $MeSO_3H-P_2O_5$ (15 equiv.), 55–68%.



Scheme 3. Reagents and conditions: i, $LiC=CH-NH_2CH_2CH_2NH_2$ (2 equiv.), THF, 20–40 °C, 70%; ii, MeSO₃H–P₂O₅ (0.1 equiv.), Et₂O, 83%; iii, Pd–CaCO₃, PbO, pentane; iv, MeSO₃H–P₂O₅ (17 equiv.), 6 h, room temp., 65%.



b; $\mathbf{R} = \mathrm{SiMe}_2\mathrm{Bu}^{\mathrm{t}}$

hand, addition of triethylphosphonoacetate carbanion to (8a,b) gave the α,β -unsaturated *E*-ethyl carboxylates (10a,b),† while addition of ethyl lithiotrimethylsilylacetate⁶ gave a mixture of *E*- and *Z*-(10a,b) (ratio 1:2).† Reduction of (10a,b) led to the allylic alcohol (5a,b) (R' = H). Then, DMSO-(COCl)₂ oxidation⁵ and addition of BuⁿMgBr provided the alcohols (5a,b) (R' = C₄H₉) (92.5%), Scheme 1.

As recently reported, upon treatment in mild acidic conditions⁴ octenols (**4a,b**) and (**5a,b**) ($\mathbf{R}' = \mathbf{C}_4\mathbf{H}_9$) were converted quantitatively into the cyclobutanone (**6**) ($\mathbf{R}' = \mathbf{C}_4\mathbf{H}_9$). Furthermore, treatment of neat (**4a,b**) or (**5a,b**) with methanesulphonic acid-phosphorus pentoxide⁷ led directly to dihydrojasmone (7) ($\mathbf{R}' = \mathbf{C}_4\mathbf{H}_9$) in 65–90% yields, as did (**6**) upon treatment under the same conditions (Scheme 1).^{8,9}

Addition of vinylmagnesium bromide to (1a) followed by oxidation⁵ gave (11) which, on treatment with MeMgBr gave (4a) ($\mathbf{R}' = \mathbf{H}$), Scheme 2. The butenol (4a) or (5a,b) ($\mathbf{R}' = \mathbf{H}$) could then undergo either $C_3 \rightarrow C_4$ ring expansion into the cyclobutanone (6) ($\mathbf{R}' = \mathbf{H}$) or $C_3 \rightarrow C_5$ ring expansion to cyclopentenone (7) ($\mathbf{R}' = \mathbf{H}$) (55–68%) a precursor of methylenomycin B,¹⁰ Scheme 1.

Addition of lithiumacetylide–ethylenediamine complex¹¹ to (8a,b) provided the propynols (9a,b) ($\mathbf{R}' = \mathbf{H}$), which underwent $C_3 \rightarrow C_4$ ring expansion to (12). Then, partial hydrogenation of (12) led to (6) ($\mathbf{R}' = \mathbf{H}$), quantitatively, which was also prone to acid induced rearrangement into (7) ($\mathbf{R}' = \mathbf{H}$), Scheme 3.

 $[\]dagger$ As shown by the chemical shifts of the olefinic protons of *E*- and *Z*-(10b) at δ 5.80 and 5.22 respectively.

a; $R^1 = Me$, $R^2 = H$ b; $R^1 = H$, $R^2 = Me$

The aldehydes (13a,b), prepared from the readily available 1-hydroxy-2-methylcyclopropanecarboxylic acid,¹² allowed the synthesis of the alcohols (14a,b) and (15a,b), using the same route as for the formation of (4a,b) and (5a,b) from (1a,b). They also underwent acid-induced ring expansion, *via* the isomeric cyclobutanones (16a,b) into the cyclopent-2-en-1-ones (17a)¹³ and (17b) (ratio 9:1) (50-70%).

This mild acid induced $C_3 \rightarrow \hat{C}_4$ ring expansion of 1-alkenylcyclopropanol derivatives, into cyclobututanones¹⁴ prone to $C_4 \rightarrow C_5$ ring enlargement^{8,9} provides a convenient alternative pathway to five-membered ring compounds from 1-hydroxycyclopropanecarbaldehyde derivatives.

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