Novel Synthesis of the 7-oxo-1,3-diazabicyclo[3.2.0]heptane and 8-oxo-1,3-diazabicyclo[4.2.0]octane Ring Systems

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Summary 4-Vinyl- and 4-allyl-azetidin-2-ones (1) and (2) have been converted into the imines (5) and (19) respectively, while use of azetidin-2-ones in which the double bond was substituted with a methoxycarbonyl group allowed the synthesis of the enamines (13) and (21).

As a continuation of our studies¹ concerned with the preparation of fused β -lactams by intramolecular cycloaddition reactions, we now report the synthesis of the 7-oxo-1,3-diazabicyclo[3.2.0]hept-3-ene and -heptane ring systems. The homologous 4,6-ring systems have also been obtained. All compounds in the following report are (\pm) mixtures, but in some instances only one enantiomer is depicted for convenience.

Condensation of 4-vinylazetidin-2-one (1)² with benzyl glyoxylate afforded the α -hydroxy-ester (3)† which was converted into the azide (4)† by established procedures.³ When (4) was heated in refluxing toluene (1 mg ml⁻¹, under argon), the two imines (5)† (35%), δ 5.95 (2-H) and (6)† (23%), m.p. 76—77 °C, δ 5.23 (2-H) were isolated.‡ The intermediate 1,2,3-triazoline (7) is assumed to undergo ready loss of nitrogen to form the isolated products. Heating⁴ (8) has been shown to give (10), the reaction proceeding via the isolable triazoline (9).

† Satisfactory elemental analysis and/or accurate mass data were obtained.

‡ In all cycloadditions described, both C(2) epimers are formed. However, epimerisation to the epimer possessing the natural penicillin stereochemistry at C(2) was achieved by treatment of the mixture with 1,5-diazabicyclo [5.4.0] undec-5-ene (DBU) in methyland diablarida at 100 °C

Ozonolysis of (3), followed by addition of methoxycarbonylmethylenetriphenylphosphorane provided the $\alpha\beta$ unsaturated ester (11) \dagger as an inseparable mixture of E- and Z-isomers. The azide (12)† was then prepared and heated in toluene at 110 °C for 23 h to give a mixture of enamines (13) † (24%), m.p. 154.5—156 °C, δ 5.54 (2-H) and (14) † (15%), m.p. 137—139.5 °C, $\delta 5.02$ (2-H). The olefinic proton of each epimer appeared at δ ca. 4.7, a chemical shift consistent with an enamine structure. Although the azidoolefin (12) was a mixture of geometric isomers, each enamine epimer was a single olefinic isomer. An X-ray study of the epimer with the natural penicillin stereochemistry at C(2) showed that the double bond had the Z-configuration. Presumably the Z-olefin is favoured owing to stabilisation via hydrogen bonding between the enamine N-H and the carbonyl of the methyl ester. Ozonolysis of (13) and (14) in ethyl acetate at -76 °C provided (15),† m.p. 147.5— 148.5 °C, δ 5.42 (2-H) and (16), † m.p. 154.5—155.5 °C, δ 5.07 (2-H), respectively.

CO₂Me

(11)
$$X = OH$$

(12) $X = N_3$

(13) $R^1 = CO_2CH_2Ph$, $R^2 = H$

(14) $R^1 = H$, $R^2 = CO_2CH_2Ph$

(15) $R^1 = CO_2CH_2Ph$, $R^2 = H$

(16) $R^1 = H$, $R^2 = CO_2CH_2Ph$

We have also demonstrated the cyclisation process in the homologous series. Thus 4-allylazetidin-2-one (2)5 was converted into the azide (17), which was heated in toluene at reflux for 7 h. Treatment of the crude product¶ with

DBU afforded the imine (19)† (30%), m.p. 94.5—95 °C $\delta(\text{CDCl}_3)$ 2.03 (s, CH₃) and 5.72 (br. s, 2-H), possessing the natural penicillin stereochemistry at C(2).

The azide (18)§ was also prepared and cyclised (refluxing toluene, $5\frac{1}{2}$ h) to give a mixture of enamines (21) and (22), from which (21)† (20%), m.p. 149—150 °C, δ (CDCl₃) 4.66 (br. s, =CH) and 5.53 (d, J 1.7 Hz, 2-H), $\lambda_{max}(EtOH)$ 283 nm (ϵ 18,700), could be crystallised. Purification of the mother liquors on 'Florisil' gave a further quantity of material (30%) which was an inseparable mixture of epimers (21) and (22) in a ratio of 1:4. The latter showed $\delta(\text{CDCl}_3)$ 4.68 (br. s, =CH) and 5.15 (dd, J 1.7 and 4 Hz, 2-H). The Z-geometry of the double bond in each epimer is assigned by analogy with the corresponding 5-membered ring derivatives. Ozonolysis of (21) in methylene dichloride containing methanol (ca. 2%) gave the amide (23)† (87%), m.p. 141—142 °C, δ 5.61 (2-H). All compounds showed the expected spectroscopic properties; none of the bicyclic esters or corresponding free acids showed any antibacterial activity.

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§ Azides in this series were unstable oils, and were generally used immediately after preparation

¶ In this case the epimer (20) with the proton α at C(2) was unstable to chromatography and could not be selectively crystallised from the mixture.

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