

CONCURRENT AMINATION OF CARBONYL GROUPS IN UNSYMMETRICAL β -DIKETONES BEARING POLYFLUOROALKYLARYL GROUPS

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The condensation of unsymmetrical β -diketones which have alkyl and fluoroalkyl substituents with amines takes place at the carbonyl linked to the unfluorinated substituent with the formation of only one of the isomers of the β -aminovinyl ketones (β -AVK) [1, 2]. It has turned out that unsymmetrical fluorine-bearing β -diketones with phenyl substituents (Ia-c) form a mixture of isomers of β -aminovinyl ketones (II), (III) and (IV), (V) on condensation with ammonia and methylamine, respectively.

*Deceased.

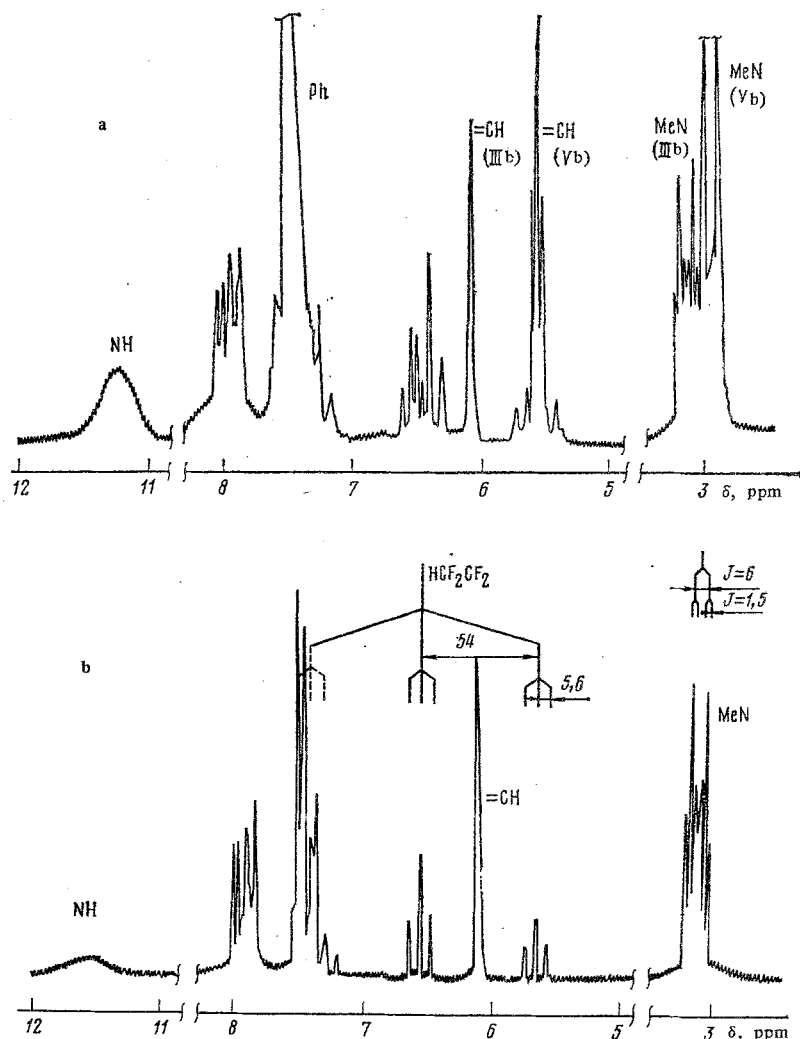


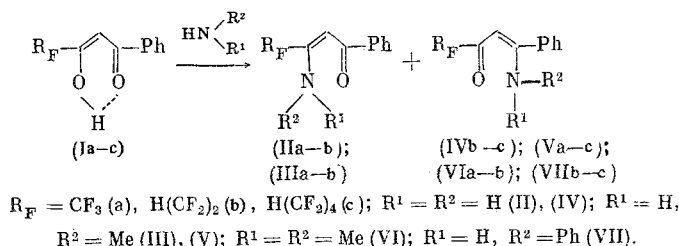
Fig. 1. PMR spectrum of the mixture of products (Vb) and (IIIb) (a) and 1-phenyl-3-methylamino-4,4,5,5-tetrafluoro-2-penten-1-one (comparison standard) (IIIb) (b) (J, Hz).

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TABLE 1. Isomeric β -Aminovinyl ketones

Compound	mp or bp, °C (p, mm Hg)	Found/ % calculated				Empirical formula ^b	PMR spectrum, δ , ppm ^a						Isomer content in mixture, %
		C	H	F	N		H(CF ₃) _n ^b	CH	NH...O	NH	H-N-Me ₂ ^c	Me ₂ N	
(IIa)	84-85	55.49 55.81	4.03 3.74	26.62 26.48	6.80 6.51	C ₁₀ H ₉ F ₃ ON	—	6.26 6.30 ^d	7-8 (2H) 7-8 (2H) ^d	—	—	—	90
(IIb)	51-51.5	53.34 53.44	3.45 3.67	30.76 30.74	5.48 5.65	C ₁₁ H ₉ F ₄ ON	5.94 t. t	6.18 6.20 ^d	7-8 (2H) 7-8 (2H) ^d	—	—	—	80
(IVb)	57-57.5	53.42 53.44	3.40 3.67	30.79 30.74	5.40 5.65	—	6.1 t. t	5.9	10.5 (1H)	6.6 (1H)	—	—	20
(IVc)	93-94 (1)	44.65 44.97	2.80 2.61	44.08 43.77	—	C ₁₃ H ₉ F ₈ ON	6.28 t. t	5.88	10.65 (1H)	6.5 (1H)	—	—	90
(IIIa)	41-41.5	57.80 57.64	4.33 4.39	24.55 24.86	6.10 6.11	C ₁₁ H ₁₀ F ₃ ON	—	6.25	10.9	3.14 d. q	—	—	70
(Va)	67-68	57.80 57.64	4.30 4.39	24.60 24.86	6.20 6.11	—	—	5.47	11.2	3.05 d	—	—	30
(IIIb) ^e	127-129 (1)	55.09 55.17	4.06 4.24	28.60 29.09	—	C ₁₂ H ₁₁ F ₄ ON	6.5 t. t	6.15	11.2	3.1 d. t	—	—	40
(Vb) ^e	131-132 (3)	46.79 46.54	3.16 3.06	—	4.05 3.87	C ₁₁ H ₁₁ F ₆ ON	6.35 t. t	5.55	11.3	2.95 d	—	—	60
(Vc)	91-92	59.52 59.25	4.78 4.97	—	5.88 5.75	C ₁₂ H ₁₂ F ₃ ON	6.4 t. t	5.52	11.4	3.05 d	—	—	90
(VIa)	61-62	44.88 45.07	5.27 5.20	—	6.45 6.37	C ₁₃ H ₁₃ F ₄ ON	6.04 t. t	5.8	—	—	—	3.1 s	—
(VIb)	182 (2)	63.20 63.15	4.16 4.05	23.30 23.50	4.33 3.28	C ₁₇ H ₁₃ F ₄ ON	6.06 t. t	5.88	12.60	—	—	2.95 s	—
(VII b)	70-71	53.49 53.91	2.99 3.09	—	3.31	C ₁₉ H ₁₃ F ₉ ON	6.25 t. t	5.66	12.50	—	—	—	—

^aIntegral intensities of all signals in the PMR spectrum correspond to the structures given.^b $J_{\text{HCF}_2} = 54$, $J_{\text{HCF}_2\text{CF}_2} = 5.6$ Hz.^cFor (Va), (Vb), and (Vc), $J_{12} = 6$ Hz; for (IIIa) and (IIIb), $J_{12} = 6$, $J_{\text{Me...RF}} = 1.3-1.5$ Hz.^dSolvent was octafluorotoluene.^eThe mixture of isomers (IIIb) and (Vb) could not be separated; the ratio was determined from the PMR spectrum (see Fig. 1). The analytical data and boiling point are given for the mixture.



The PMR spectra of the lower-boiling compounds (IIa-b) contain a very broad two-proton signal for the NH_2 group at 7-8 ppm and the methine proton signal at 6.0-6.3 ppm characteristic of β -AVK which contain fluoroalkyl substituents in the α position to NH_2 [3]. The PMR spectra of the high-boiling polymers (IVb-c) show two broad one-proton signals: an NH_2 proton that participates in the intramolecular hydrogen bond (10.5-10.65 ppm) and an exocyclic NH proton at 6.5-6.6 ppm. The CH= signal position with 5.5-5.9 ppm was characteristic of β -AVK with an amino group at the C linked to the unfluorinated substituent [1]. Furthermore, the PMR spectra of compounds (IIIa-b) show a spin-spin interaction of the MeN protons with the F atoms of the fluoroalkyl substituents (Fig. 1) which is absent in the isomeric β -AVK (Va-c), also confirming the isomer classification. In the reaction between the phenyl-substituted β -diketones (Ia-c) and dimethylamine and aniline, we succeeded in separating only β -AVK (VI) and (VII), due apparently to steric hindrance.

The direction of the reaction with respect to both electrophilic centers of the unsymmetrical polyfluoroalkylaryl-containing β -diketones as a whole confirms the scheme given in [1].

EXPERIMENTAL

The PMR spectra were obtained on a Perkin-Elmer R-12B spectrometer (60 MHz, CCl_4 , internal standard TMS).

The reaction of β -diketones (Ia-c) with the amines was performed as in [1]. To increase the β -aminovinyl ketone yield, we carried out a multiple "amine bubbling-azeotropic distillation" operation. The isomers were separated by redistillation [the boiling point of (II) and (III) is lower than that of (IV) and (V)] with subsequent recrystallization from *n*-hexane.

1-Phenyl-3-methylamino-4,4,5,5-tetrafluoro-2-penten-1-one (IIIb) (Comparison Standard). A solution of 5 g (20 mmoles) of 1-phenyl-3-amino-4,4,5,5-tetrafluoro-2-penten-1-one in 10 ml absolute ethanol was placed in an ampul, and then a 2.5- to 3-fold molar excess of methylamine was condensed by cooling with liquid nitrogen. The cooled ampul was evacuated, sealed, and allowed to stand 2-3 days at $\sim 20^\circ\text{C}$. After the alcohol had been driven off, the residue was distilled under vacuum. We obtained 2.8 g (51%) of (IIIb). Composition found: C 55.35; H 4.21; N 5.38%. Calculated composition for $\text{C}_{12}\text{H}_{11}\text{F}_4\text{ON}$: C 55.17; H 4.24; N 5.36%. The properties of the compounds synthesized are given in Table 1.

CONCLUSIONS

Unlike the unsymmetrical fluorine-containing β -diketones with alkyl substituents, the polyfluorinated β -diketones with phenyl substituents form a mixture of regioisomers of the β -aminovinyl ketones on condensation with ammonia and methylamine.

LITERATURE CITED

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