Novel Skeletal Rearrangement Reaction of Tetrasilanes with Aryl Isocyanides

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A novel skeletal rearrangement of tetrasilanes in the palladium catalysed reaction with aryl isocyanides gives 3,3-disilyl-2,4-disila-1-azacyclobutane derivatives, the structure of one example of which has been determined by X-ray crystallography.

Organopolysilanes are known to undergo skeletal rearrangements catalysed by Lewis acids as well as transition metal complexes. We now report a novel skeletal rearrangement of tetrasilanes (1) in the palladium catalysed reaction with aryl isocyanides (2) to give 3,3-disilyl-2,4-disila-1-azacyclobutane derivatives (3). The rearrangement is unique and interesting in that the product (3) is completely reconstituted from the

four fragments of tetrasilane (1) and two fragments of isocyanide (2).

This novel rearrangement reaction was found in the course of our recent study² on the palladium catalysed insertion of isocyanide into the silicon-silicon linkage of polysilanes; *i.e.*, the reaction of polysilanes with an excess of aryl isocyanides was catalysed by a palladium complex to afford the corres-

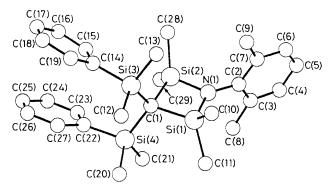


Figure 1. Molecular structure of (**3c**). Selected bond lengths (Å) and angles (°): C(1)–Si(1) 1.925(3), C(1)–Si(2) 1.913(3), N(1)–Si(1) 1.733(3), N(1)–Si(2) 1.746(3), N(1)–C(2) 1.436(4), Si(1)–C(1)–Si(2) 84.9(1), Si(1)–N(1)–Si(2) 96.2(1), C(1)–Si(1)–N(1) 89.4(1), C(1)–Si(2)–N(1) 89.5(1), Si(1)–N(1)–C(2) 135.5(2), Si(2)–N(1)–C(2) 128.2(2).

ponding poly[sila(*N*-aryl)imine]. However, the palladium catalysed reaction of tetrasilane (1) with 1.5 equivalents of aryl isocyanide (2) in the presence of 1,1,3,3-tetramethylbutyl isocyanide† gave unexpectedly (3) in moderate yields.‡ The rearrangement of tetrasilanes occurred in the reaction with aryl isocyanides, but not with alkyl isocyanides. The use of sterically bulky 2,6-di-isopropylphenyl isocyanide gave better yields of the corresponding rearranged products (3) than 2,6-xylyl isocyanide.

The reactions were carried out by heating a mixture of tetrasilane (1) (0.26 mmol), aryl isocyanide (2) (0.39 mmol), 1,1,3,3-tetramethylbutyl isocyanide (0.26 mmol), and Pd(OAc)₂ (0.026 mmol) in toluene (1 ml) at reflux for 6 h under a nitrogen atmosphere and (3) was isolated by

preparative t.l.c. (Scheme 1). The 3,3-disilyl-2,4-disila-1-azacyclobutane (3) thus formed was stable and the Si-N bond of (3) was not cleaved even on treatment with aqueous acid. The structure of (3) was in accord with the n.m.r., \S i.r., and mass spectra, and established by X-ray crystal structure analysis. The crystal structure of (3c) is shown in Figure 1, together with selected bond lengths and angles. Puckering of the four-membered ring is slight: the two half-rings are folded along the Si(1)-Si(2) diagonal, forming a dihedral angle of 2.2(2)°. The 2,6-xylyl ring is perpendicular [89.7(1)°] to the mean plane of the four-membered ring.

Intermediacy of a palladium-silylene complex might be involved in the present skeleton rearrangement of tetrasilanes, but this still remains to be clarified.

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References

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§ ¹H N.m.r. for (**3c**) (CDCl₃, 200 MHz): δ 0.21 (12 H, s), 0.59 (12 H, s), 2.37 (6 H, s), 6.86—7.08 (3 H, m), 7.23—7.41 (10 H, m).

¶ Crystal data for (3c): $C_{29}H_{43}NSi_4$, M=518.0, monoclinic, space group $P2_1/c$, a=9.344(1), b=12.233(1), c=26.197(3) Å, $\beta=90.34(1)^\circ$, Z=4, U=2994.4(5) ų, $D_c=1.149$ g cm⁻³, radiation, Cu- K_α ($\lambda=1.54178$ Å), $\mu=18.9$ cm⁻¹, crystal size $0.4\times0.3\times0.2$ mm, 4455 unique reflections measured ($2\theta_{\rm max}=120^\circ$). The structure was solved using the program MULTAN87³ and refined by the block-diagonal least-squares method to R=0.040, $R_{\rm w}=0.061$ and S=1.020 for 3973 reflections. All H atoms were located on a difference electron density map. The thermal parameter of each H atom was assumed to be isotropic and equal to $B_{\rm eq}$ of the bonded atom. Weights were taken as $w=[\sigma^2(F_o)+0.00290]F_0]^2]^{-1}$ for the reflections with $w^{1/2}|F_o|>3$ and $w^{1/2}|\Delta F|<4$, and w=0 otherwise. Atomic coordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.

^{† 1,1,3,3-}Tetramethylbutyl isocyanide was not incorporated in the product. In the absence of 1,1,3,3-tetramethylbutyl isocyanide, (3) was also obtained but in lower yield.

[‡] Under similar conditions [Ar–NC (1.8 mmol), Pd(OAc)₂ (0.15 mmol) in toluene (5 ml) at reflux for 6 h], trisilanes [RMe₂Si–(SiMe₂)–SiMe₂R] (1.5 mmol) gave the corresponding mono-insertion products [RMe₂Si–(C=N–Ar)–(SiMe₂)–SiMe₂R].