# DILATATIN, THE FIRST EXAMPLE OF A C-ALLOSYLATED NATURAL PRODUCT

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**Abstract**—Dilatatin, the xanthone-C-glycoside previously isolated from the filmy fern Hymenophyllum dilatatum, is identified as  $2-C-\beta$ -D-allopyranosyl-1,3,6,7-tetrahydroxyxanthone. The glycosyl moiety is demonstrated to be allopyranose by <sup>13</sup>C NMR spectroscopy using synthetic  $\beta$ -D-allopyranosyl-2,4,6-trimethoxybenzene as a model.

## INTRODUCTION

In 1980 Markham and Wallace [1] reported the isolation of a new C-glycosylxanthone from the filmy fern Hymenophyllum dilatatum. The aglycone moiety of this glycoside was shown unequivocally to be 1,3,6,7-tetrahydroxyxanthone and elemental analysis and acid treatment defined dilatatin as a mono-C-hexoside of this xanthone, containing no other substituents. On the basis of absorption data and chromatographic evidence dilatatin was initially thought to be mangiferin  $[2-C-\beta-D-gluco$ pyranosyl-1,3,6,7-tetrahydroxyxanthone (1)], a known constituent of many other leptosporangiate ferns [1, 2]. Closer study, however, revealed differences in the optical rotation and <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra. The present communication extends this work and identifies dilatatin as 2-C- $\beta$ -D-allopyranosyl-1,3,6,7-tetrahydroxyxanthone (3).

#### **RESULTS AND DISCUSSION**

The site of the sugar linkage to the aglycone in dilatatin was considered to be at C-2 of 1,3,6,7-tetrahydroxyxanthone, since in the <sup>13</sup>C NMR spectrum, the one protonated carbon (C-4) in the phloroglucinol ring appeared at  $\delta$ 94 (cf. mangiferin,  $\delta$ 93.7) and the glycosylated carbon (C-2) at  $\delta$ 107.8 (cf. mangiferin,  $\delta$ 107.7)[1]. Furthermore, dilatatin (3) bore the same chromatographic relationship to its Wessely–Moser rearrangement isomer, isodilatatin, as mangiferin did to isomangiferin (2).

The hexose substituent at C-2 gave a pattern of  ${}^{13}C$  NMR signals markedly different from those exhibited by the C-linked glucose in mangiferin (Table 1) and from other C-linked hexoses for which  ${}^{13}C$  NMR data were available, e.g. galactose and rhamnose [3]. It was possible, however, to calculate approximately the spectra of unknown C-hexosides from those of O-hexosides by using differences observed between the spectra of known C- and O-linked hexosides. Such

calculations suggested allose to be the most likely sugar residue in dilatatin. This was supported also by <sup>13</sup>C NMR data on a comprehensive series of 1,5-anhydrohexitols.\* Only the spectrum of 1,5-anhydroallitol approximated closely to that of the sugar in dilatatin when allowance was made for the effects on the chemical shifts brought about by bonding to a phloroglucinol ring system (cf. ref. [3]).

It was therefore decided to synthesize an authentic C-alloside for <sup>13</sup>C NMR comparison, and for convenience  $\beta$ -D-allopyranosyl-2,4,6-trimethoxybenzene (4) was chosen.  $\beta$ -Linkage of the sugar was indicated by the <sup>1</sup>H NMR spectrum of dilatatin in which the sugar H-1' signal at  $\delta$  5.05 exhibited *trans*-diaxial coupling (J = 9.4 Hz) with the adjacent C-2 proton. The tetraacetate (5) of the chosen model was synthesized in 44 % yield by reacting 2,3,4,6-tetraacetyl- $\alpha$ -D-allopyranosyl bromide



1  $R^1 = \beta$ -D-glucopyranosyl;  $R^2 = H$  (mangiferin)

2  $R^1 = H; R^2 = \beta$ -D-glucopyranosyl (isomangiferin)

3  $R^1 = \beta$ -D-allopyranosyl;  $R^2 = H$  (dilatatin)



<sup>\*</sup>Some available from ref. [3], others (1,5-anhydro-mannitol, -allitol, -talitol and altritol) kindly made available by Dr. V. M. Chari, formerly of the University of Munich.

Compound	Sugar carbon resonances $(\delta)$					
	C-1	C-2	C-3	C-4	C-5	C-6
Mangiferin (1)	73.4	70.6*	79.2	70.8*	81.7	61.7
Dilatatin (3)	71.9	68.9†	68.2†	68.1†	76.7	61.8
$\beta$ -D-Allopyranosyl-2,4,6-trimethoxybenzene (4)	71.7	68.5‡	68.5‡	67.8‡	76.2	62.2
$\beta$ -D-Glucopyranosyl-2,4,6-trimethoxybenzene	73.2	70.5§	79.2	70.9§	81.3	61.8

Table 1. Chemical shifts of the sugar carbons in the  ${}^{13}C$  NMR spectra of various C-glycosylated phenols

\*, †, ‡, \$Assignments bearing the same superscript may be reversed.

C-H coupling: triplet J = ca 140 Hz, all other signals doublets J = ca 140 Hz.

with tri-O-methylphloroglucinol in dry carbon tetrachloride in the presence of zinc oxide. The constitution of the product was confirmed by elemental analysis and <sup>1</sup>H NMR spectroscopy. Deacetylation of this product with sodium methoxide in dry methanol gave  $\beta$ -Dallopyranosyl-2,4,6-trimethoxybenzene (4) in 75% yield. <sup>13</sup>C NMR and <sup>1</sup>H NMR spectroscopy confirmed the identity of the product, the latter also confirming that the  $\beta$ -linkage of the allose had been produced (H-1', d, J = 9.9 Hz at  $\delta$  4.93).

The <sup>13</sup>C NMR spectrum of 4 (Table 1) resembled very closely that of dilatatin in the sugar carbon region, and this evidence was taken to provide firm confirmation that the *C*-linked sugar in dilatatin was indeed  $\beta$ -D-allopyranose. That the nature of the aglycone in 4 had little effect on the chemical shifts of the sugar carbon was evident from a comparison (see Table 1) of the sugar carbon resonances in mangiferin and  $\beta$ -D-glucopyranosyl-2,4,6-trimethoxybenzene (also synthesized by the same route as 5).

Allose is a sugar which is rare in nature. Only very recently has it been found in glycosidic combination with naturally occurring phenols [4, 5]. Dilatatin appears to be the first example of a natural product containing a C-linked  $\beta$ -D-allopyranosyl moiety.

#### EXPERIMENTAL

NMR measurements were made on a Varian FT-80A instrument at 20 MHz (<sup>13</sup>C) and 80 MHz (<sup>1</sup>H).

2,3,4,6-Tetraacetyl- $\alpha$ -D-allopyranosyl bromide. 2,3,4,6-Tetraacetyl- $\alpha$ -D-allopyranosyl bromide was prepared by dissolution of  $\beta$ -D-allose pentaacetate (6) [6] in an excess of a 33 % soln of HBr in HOAc maintained at ambient temp. After dilution with CHCl<sub>3</sub> and washing with H<sub>2</sub>O and then with saturated NaHCO<sub>3</sub> soln, the organic phase was dried and concd to a colourless, immobile gum which was characterized by the presence, in the <sup>1</sup>H NMR spectrum, of a clean one-proton doublet (H-1) centred at  $\delta$  6.50 (J = 7.5 Hz).

 $\beta$ -D-Allopyranosyl-2,4,6-trimethoxybenzene tetraacetate (5). From a suspension of dry ZnO (3.00 g) and Drierite (3.50 g) in dry CCl<sub>4</sub> (75 ml) containing trimethylphloroglucinol (1.03 g, 6.13 mmol) was distilled *ca* 30 ml solvent. To the remaining refluxing mixture was added, dropwise over 15 min, a soln of 2,3,4,6-tetraacetyl- $\alpha$ -D-allopyranosyl bromide [prepared from 0.60 g (1.45 mmol)  $\beta$ -D-allopyranose pentaacetate] in dry CCl<sub>4</sub> (10 ml). After refluxing for 2.5 hr, the mixture was filtered through Celite, the inorganic materials were washed with hot CCl<sub>4</sub> (2 × 50 ml), and the combined organic phases coned to a dark red oil (1.31 g) which slowly solidified. The total product was subjected to prep. TLC on silica gel and eluted with 40% EtOAc-toluene. Trimethylphloroglucinol (0.76 g) was recovered from the most mobile species,  $R_f$  ca 0.9. A band visible in UV light at  $R_f$  0.6 was eluted with EtOAc and concd to an immobile gum (0.41 g, 57°) and purified by further chromatography on silica gel to give 0.34 g of  $\beta$ -D-allopyranosyl-2,4,6-trimethoxy-benzene tetraacetate (5) as a colourless, immobile gum,  $[\alpha]_D$  + 9.8° (c 1.2 in CHCl<sub>3</sub>) (Found: C, 55.3; H, 6.2. C<sub>13</sub>H<sub>30</sub>O<sub>12</sub> requires: C, 55.4; H, 6.1%). <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.77 (s, 3H, OAc), 2.04 (s, 6H, OAc), 2.21 (s, 3H, OAc), 3.78 (s, 3H, OMe), 3.80 (s, 6H, OMe), 5.31 (d, 1H, J = 10.0 Hz, H-1 allose), 6.09 (s, 2H, aromatic). Signals attributable to the remaining allopyranosyl ring protons appeared as broad multiplets centred at  $\delta$  4.19 (3H), 5.10 (1H) and 5.80 (2H).

 $\beta$ -D-Allopyranosyl-2,4,6-trimethoxybenzene (4). A soln of 5 (0.40 g, 0.80 mmol) in dry MeOH (20 ml) containing a catalytic amount (ca 0.05 g) of Na was stirred for 0.5 hr at ambient temp. before being neutralized with ion exchange resin (Dowex 50W-X8, H). Concn of the supernatant organic phase furnished a colourless, immobile gum (0.23 g) which was purified by prep. TLC in Me<sub>2</sub>CO-EtOAc-H<sub>2</sub>O (5:4:1). The band at  $R_f$  0.5 gave  $\beta$ -D-allopyranosyl-2,4,6-trimethoxybenzene (4) (0.20 g, 75 %) as a colourless gum,  $[\alpha]_D$  1.1 (c 1.183 in MeOH); <sup>1</sup>H NMR (d<sub>6</sub>-DMSO): & 3.69 (s, 6H, OMe), 3.74 (s, 3H, OMe), 4.93 (d, 1H, J = 9.9 Hz, anomeric proton), 6.16 (s, 2H, aromatic). Signals attributable to the remaining allopyranosyl ring protons were not clearly recognizable but appeared as a complex multiplet between  $\delta$  3.17 and 4.35. <sup>13</sup>C NMR (*d*<sub>6</sub>-DMSO);  $\delta$  160.5 (C-2, 4, 6); 108.8 (C-1); ca 92 br (C-3, 5); 76.2, 71.7, 68.5, 67.8, 62.2 (allosyl carbons); 56.0 (2 and 6-OMe); 55.2 (4-OMe).

In some instances the C-alloside tetraacetate (5) was contaminated after prep. TLC with an equally mobile impurity (believed to be a disaccharide, cf. ref. [7]), which was removed by deacetylation and prep. TLC on silica gel, eluting with  $Me_2CO-EtOAc-H_2O$  (5:4:1) as described above.

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