

# The use of $\pi$ -allyltricarbyliron lactone complexes in the synthesis of the resorcylic macrolides $\alpha$ - and $\beta$ -zearalenol

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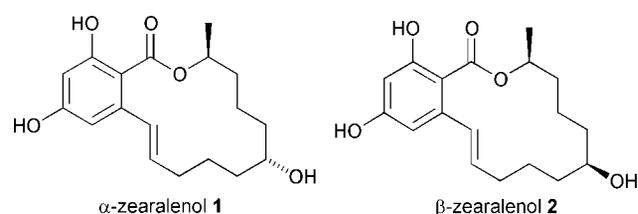
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A highly stereoselective synthesis of  $\alpha$ - and  $\beta$ -zearalenol **1** and **2** is accomplished utilising  $\pi$ -allyltricarbyliron lactone complexes **5** and **6** to establish the 1,5-stereochemical relationship of oxygen functionalities present in the natural products.

The 14-membered resorcylic macrolides  $\alpha$ - and  $\beta$ -zearalenol **1** and **2** are estrogenic mycotoxins produced by certain species



of the fungus *Fusarium*.<sup>1</sup> Their hormonal activity is linked to the close spatial similarity to 17 $\beta$ -estradiol,<sup>2</sup> with the  $\alpha$ -isomer **1** being three to four times as active as the  $\beta$ -isomer **2**.<sup>3</sup> While several total syntheses of the parent compound zearalenone were accomplished over the last 30 years,<sup>4</sup> to our knowledge no independent synthesis of **1** or **2** has been reported so far. Here we report the first enantioselective preparation of **1** and **2** employing  $\pi$ -allyltricarbyliron lactone complexes as key intermediates.

We have previously shown that organoaluminium reagents possessing active  $\beta$ -hydrogens, like tripropyl- and triisobutylaluminium, reduce carbonyl groups appended to the allyl ligand of  $\pi$ -allyltricarbyliron lactone complexes with excellent diastereoselectivity.<sup>5</sup> Also, we recently reported that sodium triacetoxyborohydride efficiently decomplexes  $\pi$ -allyltricarbyliron lactone complexes bearing a hydroxy group in the side-chain to afford stereodefined 1,5-diols.<sup>6</sup> By exploiting this methodology in this work, we show that  $\pi$ -allyltricarbyliron lactone complexes can be used to set up the relative oxygen atom stereochemistry present in the natural products.

The route to  $\alpha$ - and  $\beta$ -zearalenol **1** and **2** relied upon the formation of the  $\pi$ -allyltricarbyliron lactone intermediates **5** and **6**, respectively, whose preparation is delineated in Scheme 1. Reduction of the ester **3**<sup>7</sup> using lithium aluminium hydride followed by Swern oxidation and Horner–Wadsworth–Emmons homologation with the phosphonate **18**,<sup>†</sup> prepared according to the methodology of Grieco,<sup>8</sup> gave the corresponding (*E*)-enone in 83% yield over three steps. Deprotection of the acetonide under acidic conditions and transformation of the liberated diol to the cyclic sulfite using thionyl chloride<sup>9</sup> afforded the compound **4** in 82% overall yield. Treatment of **4** with diironnonacarbonyl in benzene under sonication conditions<sup>10</sup> provided the two diastereoisomeric  $\pi$ -allyltricarbyl-

iron lactone complexes, *endo*-**5** and *exo*-**6**, in 70% combined yield and in a ratio of *ca.* 1:1. Separation of the two isomers **5** and **6** was readily achieved by flash column chromatography.

With these key intermediates in hand we were able to proceed to the target molecules  $\alpha$ - and  $\beta$ -zearalenol **1** and **2** (Scheme 1). For example, for  $\alpha$ -zearalenol **1**, reduction of the side-chain ketone in the *endo* complex **5** was achieved in 94% yield using tripropylaluminium<sup>5</sup> to give **7**, as the sole product as determined by 600 MHz <sup>1</sup>H NMR analysis. Treatment of **7** with sodium triacetoxyborohydride in tetrahydrofuran<sup>6</sup> resulted in a highly stereoselective decomplexation to afford, after TBDMS-protection and hydrogenation, the alcohol **8**. Swern oxidation of **8** provided the corresponding aldehyde which in turn was transformed into the vinylstannane **9** by applying the procedure developed by Hodgson<sup>11</sup> utilising chromium(II) chloride and Bu<sub>3</sub>SnCH<sub>2</sub> in *N,N*-dimethylformamide. Stille coupling of the stannane **9** with the known aromatic iodide **10**<sup>4f</sup> using Farina's catalyst<sup>12</sup> provided the coupled product **11** in 82% yield. Treatment of **11** with HF·pyridine followed by hydrolysis of the methyl ester functionality using aqueous potassium hydroxide in ethane-1,2-diol at 120 °C provided the seco acid **12** in 83% yield over two steps.

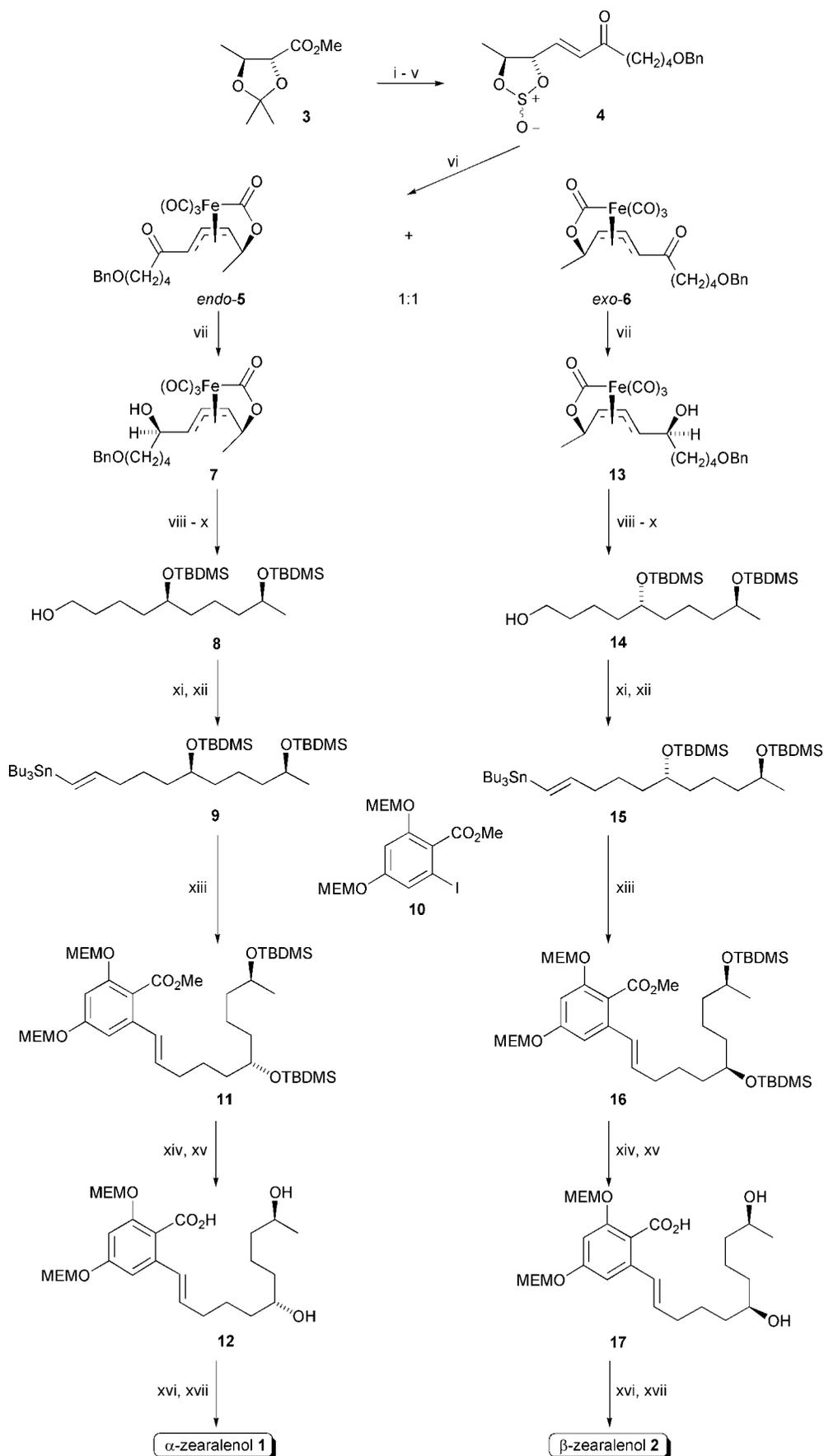
Cyclisation of **12** using Mukaiyama's protocol<sup>13</sup> afforded the desired MEM-protected  $\alpha$ -zearalenol in 64% yield. Final deprotection of the MEM-ethers with aqueous hydrochloric acid in tetrahydrofuran at 40 °C provided  $\alpha$ -zearalenol **1** in 93% yield and with a *de* of 94% as determined by 600 MHz <sup>1</sup>H NMR analysis { $[\alpha]_D^{25}$  –93.6 (*c* 0.55 in acetone) [optical rotation obtained on an authentic sample ‡  $[\alpha]_D^{25}$  –97.3 (*c* 0.55 in acetone)]}.

Application of the same sequence of reactions to the *exo* complex **6** afforded the diastereoisomeric  $\beta$ -zearalenol **2** in similar overall yield *via* the intermediates **13** to **17**, as shown in Scheme 1 {*de* >95% as determined by 600 MHz <sup>1</sup>H NMR analysis;  $[\alpha]_D^{25}$  –12.5 (*c* 1.00 in acetone) [optical rotation obtained on an authentic sample ‡  $[\alpha]_D^{25}$  –12.9 (*c* 1.00 in acetone)]}.

These highly stereoselective syntheses of  $\alpha$ - and  $\beta$ -zearalenol **1** and **2** clearly demonstrate the utility of carbonyl substituted  $\pi$ -allyltricarbyliron lactone complexes in organic synthesis. Using the *endo* complex **5** as well as the *exo* complex **6** we were able to set up the required 1,5-stereochemical relationship of oxygen functionalities present in the natural products.

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**Scheme 1** Reagents and conditions: i.  $\text{LiAlH}_4$ ,  $\text{Et}_2\text{O}$ ,  $0^\circ\text{C}$ , 2 h; ii.  $(\text{COCl})_2$ , DMSO,  $\text{Et}_3\text{N}$ , DCM,  $-78^\circ\text{C}$ , 3 h; iii.  $(\text{EtO})_2\text{P}(\text{O})\text{CH}_2\text{CO}(\text{CH}_2)_4\text{OBn}$  **18**,<sup>†</sup>  $\text{NaH}$ , THF,  $-78^\circ\text{C}$ , 1 h, 83% (over 3 steps); iv.  $\text{AcOH}-\text{H}_2\text{O}$  (1:1),  $40^\circ\text{C}$ , 24 h, 92%; v.  $\text{SOCl}_2$ ,  $\text{Et}_3\text{N}$ ,  $\text{Et}_2\text{O}$ ,  $0^\circ\text{C}$ , 30 min, 89%; vi.  $\text{Fe}_2(\text{CO})_9$ , benzene, sonication,  $30^\circ\text{C}$ , 3 h, 35% **5**, 35% **6**; vii.  $\text{AlPr}^n_3$ , DCM,  $0^\circ\text{C}$ , 94% (80%);<sup>§</sup> viii.  $\text{NaBH}(\text{OAc})_3$ , THF, 3 d, 75% (83%); ix.  $\text{TBDMSCl}$ , imidazole, DMF,  $0^\circ\text{C}$ , 30 min, then rt 24 h, 87% (85%); x.  $\text{Pd/C}$  (10%),  $\text{H}_2$ ,  $\text{EtOAc}$ , 30 min, 94% (93%); xi.  $(\text{COCl})_2$ , DMSO,  $\text{Et}_3\text{N}$ , DCM,  $-78^\circ\text{C}$ , 3 h, 86% (80%); xii.  $\text{Bu}_3\text{SnCHI}_2$ ,  $\text{CrCl}_2$ , DMF,  $0^\circ\text{C}$ , 67% (69%); xiii. methyl 4,6-bis[(2-methoxyethoxy)methoxy]-2-iodobenzoate **10**,  $\text{Pd}(\text{dba})_3$ ,  $\text{P}(\text{2-furyl})_3$ , toluene,  $100^\circ\text{C}$ , 4 h, 82% (85%); xiv.  $\text{HF}\cdot\text{pyridine}$ , pyridine, THF, 12 h, 95% (93%); xv. 10 M aqueous  $\text{KOH}$ , ethane-1,2-diol,  $120^\circ\text{C}$ , 4 h, 87% (91%); xvi. syringe pump addition of a solution of the seco acid and  $\text{Et}_3\text{N}$  in MeCN over 10 h to 1-methyl-2-chloropyridinium iodide, MeCN, reflux, 64% (62%); xvii. 1.5 M aqueous  $\text{HCl}$ , THF,  $40^\circ\text{C}$ , 93% (93%).

## Notes and references

† Compound **18** was prepared by alkylation of the dianion of diethyl (2-oxopropyl)phosphonate (NaH, BuLi, 0 °C) with benzyl 3-bromopropyl ether (82%).

‡ Authentic samples of  $\alpha$ - and  $\beta$ -zearealenol were purchased from Sigma Aldrich.

§ Yields given refer to the synthesis of  $\alpha$ -zearealenol **1**, while those given in parentheses correspond to the synthesis of  $\beta$ -zearealenol **2**.

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