

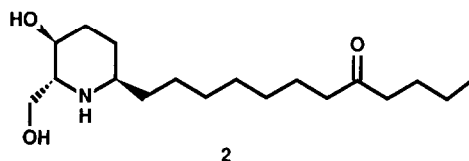
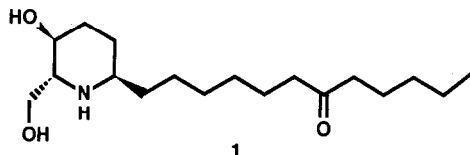
## SYNTHESIS OF (±)-ISOPROSOPININES A AND B

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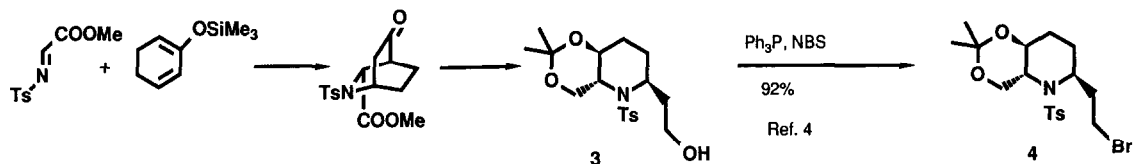
**Summary:** The total synthesis of the piperidine alkaloid isoprosopinine A (1) and an improved synthesis of isoprosopinine B (2) are reported. The key steps were alkylation of the anions of the sulphones (5) and (6) respectively with the common intermediate bromide (4) and the novel reductive cleavage of the sulphonamides (7) and (8) with sodium-amalgam.

The prosopis alkaloids form an interesting class of 2,3,6-trisubstituted piperidine derivatives exhibiting a range of biological activities.<sup>1</sup> We recently reported a synthesis of isoprosopinine B (2),<sup>2</sup> but comparison of the synthetic material with an authentic sample was complicated by the fact that (2) was isolated as an inseparable mixture with its isomer isoprosopinine A (1).<sup>3</sup> The previous synthesis of (2) relied on a relatively economical Wittig homologation reaction to introduce the side chain, but the yield was moderate. In this Letter we report a much improved method for side chain construction which has been applied to the first total synthesis of isoprosopinine A (1) and also to isoprosopinine B (2). During this synthesis a novel method for the reductive deprotection of N-tosylpiperidines has been discovered.



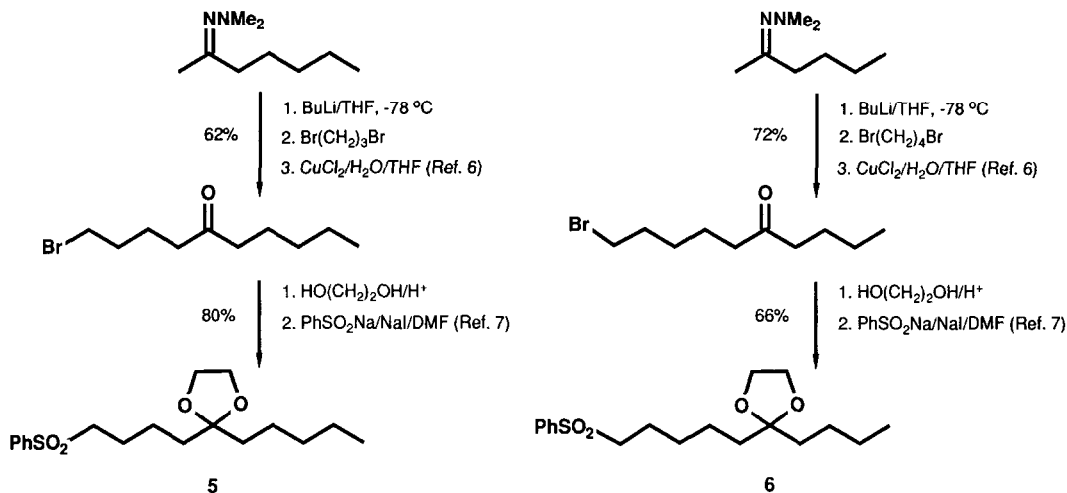
The common starting material for the syntheses is the bromide (4) which is readily prepared from the alcohol (3), itself available in five steps from simple starting materials (Scheme 1).<sup>2</sup>

SCHEME 1



The required side-chain sulphones (5) and (6) were prepared in good yield using Corey-Enders hydrazone alkylations of heptanone and hexanone respectively (Scheme 2).<sup>5</sup>

SCHEME 2

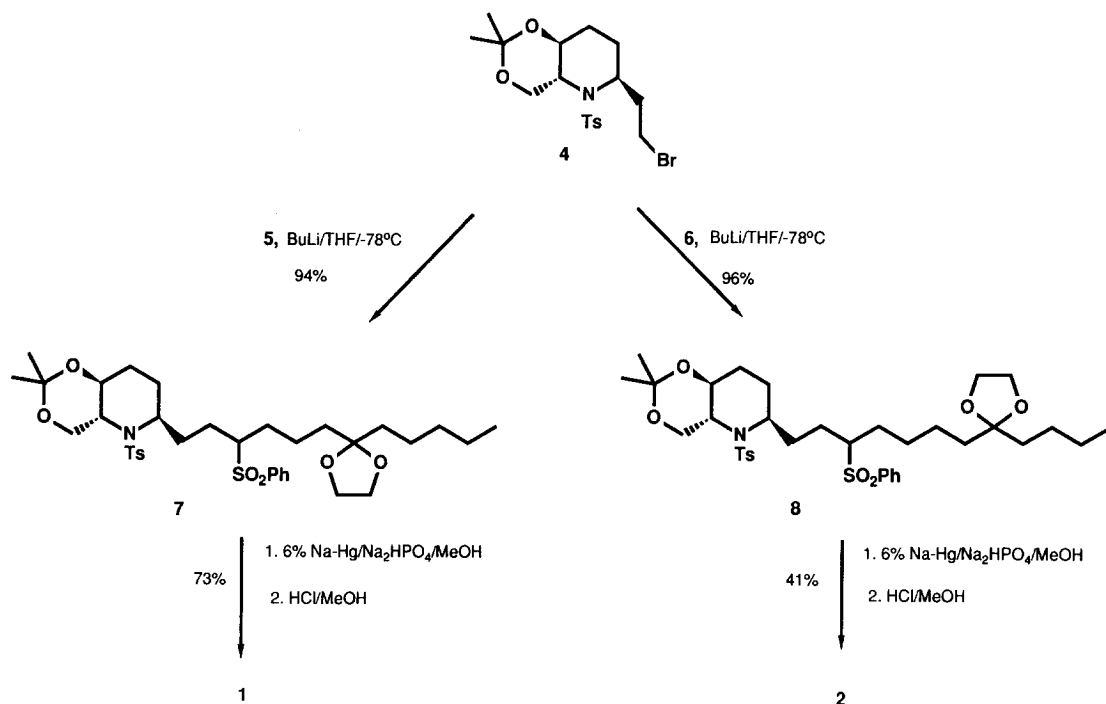


Coupling of the bromide (4) with the anions of the sulphones (5) and (6) gave the compounds (7) and (8) respectively, in high yield (Scheme 3). Attempts to remove the sulphone and *N*-tosyl groups from the piperidine (7) using either sodium in liquid ammonia<sup>8</sup> or sodium naphthalenide<sup>9</sup> gave complex mixtures. However the use of a large excess (20–30 eq.) of freshly prepared 6% sodium-amalgam in phosphate-buffered methanol removed (after two cycles) both the sulphone<sup>10</sup> and the *N*-tosyl protecting group cleanly, and the remaining protecting groups were removed with methanolic HCl to give (±)-isoprosopinine A (1) in good yield. This is, to our knowledge, the first report of the use of sodium-amalgam for sulphonamide cleavage, and may have wider application as a selective reagent for such deprotections. Similar treatment of the intermediate (8) gave, after removal of the other protecting groups (±)-isoprosopinine B (2) in slightly lower yield. This suggests that the success of the sodium-amalgam method for removal of *N*-tosyl groups is dependent on the age of the amalgam, and that optimum results require fresh

reagent.

The spectra of synthetic isoprosopinines A and B were very similar and resembled those of the mixed sample derived from natural sources.<sup>3,11</sup> The main difference between (1) and (2) can be detected in the <sup>13</sup>C NMR spectrum where methylene resonances in the side-chain differ slightly.<sup>12</sup>

SCHEME 3



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## REFERENCES

1. Y. Saitoh, Y. Moriyama, T. Takahashi, and Q. Khuong-Huu, Tetrahedron Lett., 1980, **21**, 75; Y. Saitoh, Y. Moriyama, H. Hirota, T. Takahashi, and Q. Khuong-Huu, Bull. Chem. Soc. Jpn., 1981, **54**, 488; M. Natsume and M. Ogawa, Heterocycles, 1981, **16**, 973.
2. A. B. Holmes, J. Thompson, A. J. G. Baxter, and J. Dixon, J. Chem. Soc., Chem. Commun., 1985, 37.
3. Q. Khuong-Huu, G. Ratle, X. Monseur, and R. Goutarel, Bull. Soc. Chim. Belg., 1972, **81**, 443.
4. S. Hanessian, M. M. Ponpipom, and P. Lavallee, Carbohydr. Res., 1972, **24**, 45.
5. E. J. Corey and D. Enders, Chem. Ber., 1978, **111**, 1377.
6. E. J. Corey and S. Knapp, Tetrahedron Lett., 1976, **17**, 3667.
7. J. S. Meek and J. S. Fowler, J. Org. Chem., 1968, **33**, 3422.
8. V. du Vigneaud and O. K. Behrens, J. Biol. Chem., 1937, **117**, 27.
9. G. Wittig, W. Joos, and P. Rathfelder, Liebigs Ann. Chem., 1957, **610**, 180.
10. B. M. Trost, H. C. Arndt, P. E. Strege, and T. R. Verhoeven, Tetrahedron Lett., 1976, **17**, 3477.
11. Q. Khuong-Huu, X. Monseur, M. J. Gasic, P. M. Wovkulich, and E. Wenkert, J. Chem. Soc. Pak., 1982, **4**, 267.
12. Data for (4): m.p., 93.5-94.5 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 250 MHz) δ 7.8-7.2 (4H), 4.43 (1H, m), 4.31 (1H, dd, J 11.6, 10.8 Hz), 3.99 (1H, dd, J 11.6, 4.7 Hz), 3.61 (1H, ddd, J 10.6, 10.4, 4.4 Hz), 3.38 (2H, t, J 7.1 Hz), 3.17 (1H, ddd, J 10.8, 10.4, 4.7), 2.43 (3H, s), 2.41-1.4 (6H, m), 1.39 (3H, s), 1.34 (3H, s).  
 Data for (1): m.p. 86-7 °C.  
<sup>1</sup>H NMR (CDCl<sub>3</sub>, 250 MHz) δ 3.65 (1H, dd, J 10.6, 7.7 Hz), 3.60 (1H, dd, J 10.6, 5.5 Hz), 3.53 (1H, ddd, J 6.5, 5.5, 4.2 Hz), 2.82 (1H, dt, J 7.7, 5.5 Hz), 2.75 (1H, m), 2.37 (4H, t, J 7.5 Hz), 2.16 (3H, broad), 1.7-1.1 (20H, m), 0.87 (3H, t, J 6.8 Hz).  
<sup>13</sup>C NMR (CDCl<sub>3</sub>, 62.5 MHz) δ 211.5, 68.0, 62.3, 58.2, 49.8, 42.8, 42.7, 33.7, 31.4, 29.4, 29.2, 28.6, 27.4, 26.1, 23.7, 23.6, 22.4, 13.8.  
 IR (CHCl<sub>3</sub>) 3610 (m), 3420 (m, broad) 2935 and 2860 (m), 1710 (s) cm<sup>-1</sup>.  
 Data for (2): m.p. 90-1 °C.  
<sup>1</sup>H NMR (CDCl<sub>3</sub>, 250 MHz) δ 3.6 (1H, dd J 10.8, 7.6), 3.63 (1H, dd, J 10.8, 5.5 Hz), 3.55 (1H, ddd, J 6.5, 5.5, 4.2 Hz), 2.94 (1H, dt, J 7.7, 5.5 Hz), 2.83 (1H, m), 2.66 (3H, broad), 2.40 (4H, t, J 7.5 Hz), 1.8-1.1 (20H, m), 0.90 (3H, t, J 6.8 Hz).  
<sup>13</sup>C NMR (CDCl<sub>3</sub>, 62.5 MHz) δ 211.4, 67.9, 62.9, 58.4, 49.9, 42.7, 42.5, 33.8, 29.4, 29.3, 29.2, 28.6, 27.3, 26.2, 26.1, 23.8, 22.4, 13.8.  
 IR (CHCl<sub>3</sub>) 3620 (m), 3460 (broad), 2930 and 2860 (s), 1710 (s) cm<sup>-1</sup>.

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