

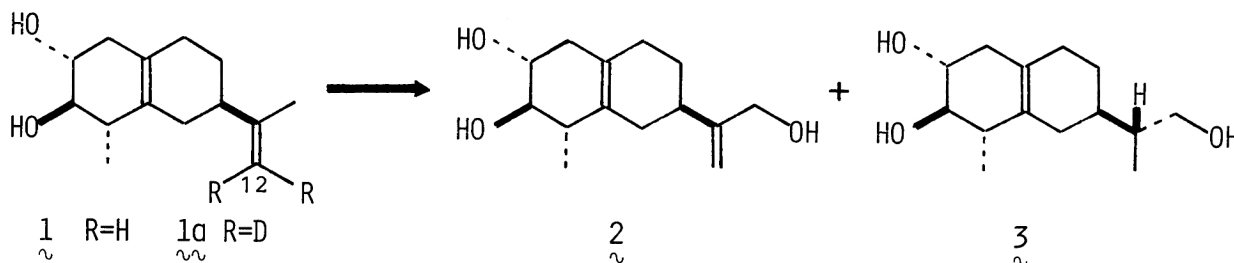
SYNTHESIS OF $(-)-[12,12-^2\text{H}_2]\text{RISHITIN}^{1)}$

Akio MURAI* and Tadashi MASAMUNE

Department of Chemistry, Faculty of Science,
Hokkaido University, Sapporo 060

For studies on the vital metabolic pattern of rishitin, a representative phytoalexin of diseased potatoes, $(-)-[12,12-^2\text{H}_2]\text{-rishitin}$ has been synthesized stereoselectively from (11S)-2 α -acetoxy-3-oxo-4 β ,5 α -eudesman-6 β ,12-olide.

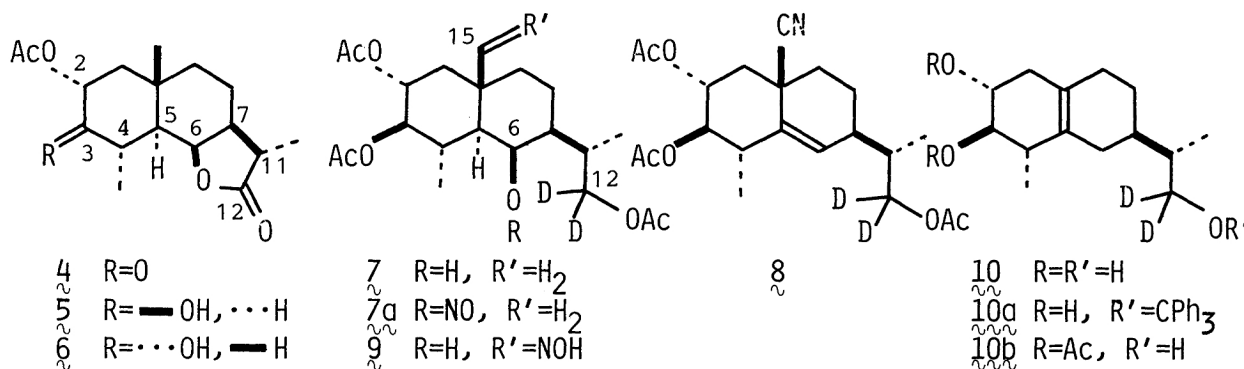
In the course of biosynthetic studies on phytoalexins isolated from diseased potatoes, we previously reported that rishitin (**1**) is metabolized in healthy potato tuber tissues into rishitin M-1 (**2**) and M-2 (**3**).²⁾ In order to clarify the mutual relationship in the metabolic pathway, we required hot rishitin, labelled with two deuterium atoms at C₁₂ position. This communication deals with the synthesis of the title compound, $(-)-[12,12-^2\text{H}_2]\text{rishitin}$ (**1a**), involving practical improvement of the synthetic pathway for cold rishitin, which has been published by us several years ago.³⁾



The synthesis was commenced with (11S)-2 α -acetoxy-3-oxo-4 β ,5 α -eudesman-6 β ,12-olide (**4**),³⁾ readily derived from santonin, a totally synthesized sesquiterpene. The compound (**4**) was treated with the *t*-butylamine-borane complex⁴⁾ in aqueous methanol (0 °C, 10 min) to give a separable mixture of 3 β - and 3 α -hydroxy-2-acetates (**5** and **6**)³⁾ in 60 and 38% yields, respectively. The compound (**6**) could easily be reconverted to the starting material (**4**) by Jones oxidation in 97% yield. The yield of the desired 3 β -hydroxy derivative (**5**) amounted to 83% by repeated oxidation and reduction of the 3 α -ol (**6**). Incorporation of deuterium into the compound (**5**) was carried out by treatment with lithium aluminium deuteride (Merck, ²H-content min 98%) (in THF, reflux, 24 h) followed by acetylation (Ac₂O in Py, 23 °C, 20 h) to afford [12,12-²H₂]-2,3,12-triacetate (**7**), mp 99-100 °C, in 89% yield, which was converted (NOCl in Py, -20--30 °C, 20 min) quantitatively into the 6 β -nitrite (**7a**). Photolysis of the nitrite (**7a**) (200-W Hanovia high pressure mercury lamp in C₆H₆, 20 °C, 3.5 h; THF-*i*-PrOH, reflux, 2.5 h) and subsequent dehydration (POCl₃ in Py, 23 °C, 3 d; MsCl in Py, 23 °C, 39 h; collidine, 190 °C, 3 h) were performed under almost the same conditions as those of the cold rishitin

synthesis³⁾ to give $[12,12\text{-}^2\text{H}_2]\Delta^5\text{-10-nitrile}$ (**8**), mp 106-107 °C, in 40% overall yield, via $[12,12\text{-}^2\text{H}_2]\text{-15-oxime}$ (**9**), amorphous. Reductive decyanation³⁾ of the compound **8** (Na in toluene-EtOH, reflux, 1 h) afforded $[12,12\text{-}^2\text{H}_2]\Delta^5\text{-triol}$ (**10**), mp 106.5-107 °C in 81% yield.

Protection of 2,3-dihydroxyl groups in compound **10** was improved as follows.⁵⁾ Tritylation of **10** $[(\text{C}_6\text{H}_5)_3\text{CCl}, \text{Et}_3\text{N}$ and DMAP in DMF, 23 °C, 4 d]⁶⁾ afforded its 12-monotriphenylmethyl ether (**10a**), oil, in 91% yield. The compound (**10a**) was acetylated (Ac_2O in Py, 20 °C, 24 h) and then hydrolyzed with acid (a catalytic amount of TsOH in MeOH, 23 °C, 3 h) to give $[12,12\text{-}^2\text{H}_2]\text{-2,3-diacetoxy-12-ol}$ (**10b**), oil, in 96% overall yield. Successive treatment of **10b** (TsCl in Py, 23 °C, 26 h; NaI in acetone, 80 °C, 24 h; 5% KOH in MeOH, reflux, 2 h) provided **1a**, oil, in 66% overall yield: $[\alpha]_{\text{D}}^{26} = -30.6^\circ$ (c 2.1, EtOH), (lit.,⁷⁾ -35.1° for natural rishitin). $(-)\text{-}[12,12\text{-}^2\text{H}_2]\text{Rishitin}$ thus obtained was identical with cold rishitin in respects of R_f -values on TLC and $^1\text{H-NMR}$ spectra except two protons at C_{12} . The $^1\text{H-NMR}$ (400 MHz) and EI-MS spectra⁸⁾ of **1a** showed that the $^2\text{H}_2$ content of the synthetic rishitin (**1a**) amounted to more than 99.4%.



References

- 1) Part IL of "Studies on the Phytoalexins" and Part XX of "Synthetic Studies of Rishitin and Related Compounds." Part XXXIX and XIX of the respective series, A. Murai, S. Sato, and T. Masamune, Bull. Chem. Soc. Jpn., in press.
- 2) A. Murai, N. Katsui, F. Yagihashi, T. Masamune, Y. Ishiguri, and K. Tomiyama, J. Chem. Soc., Chem. Commun., 1977, 670.
- 3) A. Murai, K. Nishizakura, N. Katsui, and T. Masamune, Tetrahedron Lett., 1975, 4399; Bull. Chem. Soc. Jpn., 50, 1206 (1977).
- 4) G. C. Andrews and T. C. Crawford, Tetrahedron Lett., 21, 693 (1980).
- 5) All attempts to form the 2,3-acetonide of **10** failed, even with the Wako gel Q-23 available recently.³⁾
- 6) Cf., S. K. Chaudhary and O. Hernandez, Tetrahedron Lett., 1979, 95.
- 7) T. Masamune, A. Murai, M. Takasugi, A. Matsunaga, N. Katsui, N. Sato, and K. Tomiyama, Bull. Chem. Soc. Jpn., 50, 1201 (1977).
- 8) MS, m/z 224 (M^+), 206, 205, and 191; IR (CCl_4), 3410, and 1075 cm^{-1} ; $^1\text{H-NMR}$ (CDCl_3 , 400 MHz), δ 1.14 (3H, d, $J = 6.8$ Hz), 1.73 (3H, s), 3.20 (1H, t, $J = 9.0$ Hz), and 3.64 (1H, dt, $J = 6.4$ and 9.8 Hz).

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