April 1996 SYNTHESIS 465

Regiospecific Synthesis of 6-Alkylated Lumazine Derivatives Using Silylenol Ethers

Mamoru Igarashi, Masaru Tada*

Department of Chemistry, School of Science and Engineering, Waseda University, Ohkubo, Shinjuku-ku, Tokyo 169, Japan Fax +81(3)32082735

Received 16 May 1995; revised 1 October 1995

The reaction of oxadiazinone with silylenol ethers gave regiospecifically 6-substituted lumazines by an hetero Diels-Alder addition followed by decarboxylation and silanol elimination.

Most pteridines are synthesized by the Gabriel-Isay reaction¹ in which the easily available 5,6-diaminopyrimidines are condensed with α -dicarbonyl compounds. The condensation of pyrimidines with unsymmetrical α -dicarbonyl compounds, however, occurs non-regiospecifically giving a mixture of 6- and 7-substituted pteridines. Many natural products of biological importance (e.g. folic acid and biopterin) contain the pteridine ring having an alkyl side-chain at the 6-position, ² and hence a versatile and regiospecific synthesis of 6-substituted pteridines is desirable.

We have recently reported the regioselective synthesis of 6-alkylated lumazine derivatives using 5,6,7,8-tetrahydro-5,7-dimethyl-3,6,8-trioxo-3*H*-pyrimido[5,4-*c*] [1,2,5]oxadiazine (1) and the enamines derived from aldehydes.³ This reaction is highly regioselective, but the yields were not satisfactory. In this paper, we report the reaction of oxadiazinone 1 with silylenol ethers instead of enamines (Scheme 1). The reaction proceeds in the same fashion and gives the same products as the reaction of oxadiazinone 1 with enamines. The yields of 6-alkylated lumazine derivatives were much improved by the new procedure.

Scheme 1

The starting silylenol ethers (ca. 1:1 mixture of E- and Z-isomers) were prepared from the corresponding aldehydes by the procedure described by House et al.⁴ A mixture of oxadiazinone 1 and one of the silylenol ethers $2\mathbf{a} - \mathbf{h}$ was reacted under nitrogen under the reaction conditions shown in the Table to give the products $3\mathbf{a} - \mathbf{h}$, which were identified by comparison with the authentic samples of $3\mathbf{a} - \mathbf{h}$ obtained in the reaction of 1 with enamines.³

As compared to the reaction of oxadiazinone 1 and enamines, the present method using silylenol ethers gave much improved yields of the products 3. Particularly, Method C gave satisfactory yields with almost all silylenol ethers used.

In the case of silylenol ether 2a (R = Me), heating was required to obtain a reasonable yield (Table), but all other reactions proceeded smoothly at room temperature giving high yields of the products.

The present reaction is initiated by an hetero Diels-Alder reaction of reversed electron demand⁵ as in the reaction

Scheme 2

Table. Reaction of Oxadiazinone 1 with Silylenol Ethers 2a-h

Product ^a	R	Yield (%)			mp (°C)	
		Method A ^b	Method B ^c	Method C ^d	found	reported
3a	Me	18	20	15 (53(°)	197.0-198.5	201.0-203.07
3b	Et	30	35	82	141.0-143.0	$139.0 - 141.0^8$
3c	Pr	51	57	78	113.5-114.0	$113.5 - 114.0^3$
3d	i-Pr	50	46	90	119.0-119.3	$119.0 - 119.3^3$
3e	$c - C_6 H_{11}$	47	34	93	157.5-158.2	$157.5 - 158.2^3$
3f	t-Bu 11	79	70	93	138.0-139.2	$142.0 - 143.0^9$
3g	Ph	74	70	85	248.6 - 249.2	$251.0 - 253.0^{10}$
3h	Bn	31	32	65	130.0-130.7	$130.0 - 130.7^3$

^a The spectral data (IR, NMR, MS) were identical with the authentic sample of 3a-h reported in Ref. 3.

b A 1:2 mixture of oxadiazinone 1 (0.2 mmol) and silylenol ether 2 (0.4 mmol) in THF was stirred at r.t. for 12 h.

^c A 1:2 mixture of oxadiazinone 1 (0.2 mmol) and silylenol ether 2 (0.4 mmol) was refluxed in THF for 12 h.

^d A 1:5 mixture of oxadiazinone 1 and silylenol ether 2 was stirred at r.t. for 12 h without solvent.

The reaction mixture was heated at 80°C for 12 h.

466 Short Papers SYNTHESIS

of oxadiazinone 1 with enamines.³ Thus, the electron rich silylenol ether adds regioselectively to the electron poor oxadiazinone 1. The loss of carbon dioxide from the adduct followed by aromatization due to the elimination of trimethylsilanol produces the products 3 (Scheme 2). Polarities of oxadiazinone 1³ and the silylenol ether as shown in Scheme 2 define the regiospecificity of the first cycloaddition and therefore the final products.

In conclusion, the present reaction provides a convenient and versatile method for the preparation of 6-alkylated lumazine derivatives.

Melting points were recorded on a Yamato apparatus model MP-21 and are uncorrected. The ¹H NMR spectra were recorded on a Hitachi R-90, JEOL EX-270 and JEOL GSX-400 spectrometers in CDCl₃. The ¹³C NMR spectra were recorded on a JEOL EX-270 and JEOL GSX-400 spectrometers in CDCl₃.

Diazinone 1 was prepared by the reaction of 1,3-dimethyl-4-amino-5-nitrosouracil with bis(trichloromethyl) carbonate³ by modification of the original procedure.⁶ Silylenol ethers (2a-h) were prepared from the corresponding aldehydes by reaction with Me₃SiCl as reported by House et al.⁴ All the starting materials gave correct spectral data.

Reaction of Diazinone 1 with Silylenol Ethers 2; General Procedure: An appropriate amount (see Table) of diazinone 1 and one of the silylenol ethers 2a-h was dissolved in THF (1 mL) and the mixture was stirred under N_2 for the period recorded in the Table. The mixture was condensed in vacuo and the product was separated by

preparative TLC on silica gel plate $(20 \times 20 \times 0.2 \text{ cm})$ developed by EtOAc/CHCl₃ (1:1). The spectral data of products thus obtained were identical in every respect with the authentic samples reported in Ref. 3.

The present study was supported by the Grant-in-Aid for Scientific Research, Annual Project organized by Waseda University, and the Sasagawa Scientific Research Grant.

- Pfleiderer, W. In Comprehensive Heterocyclic Chemistry, Vol. 3; Katritzky, A. R.; Rees, C. W. Eds.; Pergamon: Oxford, 1984; p 263.
- (2) Brown, D. J. In *The Chemistry of Heterocyclic Compound*, Vol. 24; Taylor, E. C.; Weissberger, A. Eds.; Wiley: New York, 1988; p 511.
- (3) Igarashi, M.; Tada, M. J. Heterocycl. Chem. 1995, 32, 807.
- (4) House, H.O.; Czuba, L.J.; Gall, M.; Olmstead, H.D. J. Org. Chem. 1969, 34, 2324.
- (5) Sauer, J.; Wiest, H. Angew. Chem. 1962, 74, 353; Angew. Chem., Int. Ed. Engl. 1962, 1, 268; Boger, D. L. Chem. Rev. 1986, 86, 781.
- (6) Pfleiderer, W.; Kempter, F.E. Angew. Chem. 1967, 79, 234; Angew. Chem., Int. Ed. Engl. 1967, 6, 259.
- (7) Heckel, A.; Pfleiderer, W. Helv. Chim. Acta 1986, 69, 1095.
- (8) Taylor, E.C.; Inbasekaran, M. Heterocycles 1978, 10, 37.
- Tada, M.; Ito, T.; Ohshima, J. J. Heterocycl. Chem. 1986, 23, 1893.
- (10) Dick, G.P.G.; Wood, H.C.S.; Logan, W.R. J. Chem. Soc. 1956, 2131.