# Organic & Biomolecular Chemistry

## PAPER

Cite this: Org. Biomol. Chem., 2013, 11, 5239

Received 26th April 2013, Accepted 21st June 2013 DOI: 10.1039/c3ob40859k

www.rsc.org/obc

## Introduction

 $\alpha$ -Pyrone, coumarin and their benzannulated scaffolds are key structural motifs found in a large number of biologically important natural alkaloids, isolated from an eclectic array of plant and marine sources.<sup>1,2</sup> Owing to their interesting photophysical properties<sup>3</sup> and the broad range of pharmacological activities<sup>4</sup> associated with them, they come under the category of privileged scaffolds in medicinal chemistry. As a consequence; numerous approaches to the synthesis of pyrones, coumarins and benzocoumarins have been reported either by the conventional approaches<sup>5</sup> or by procedures involving transition metal-catalyzed reactions.<sup>6,7</sup>

Undoubtedly these classical approaches play a crucial role in the realm of natural and medicinal chemistry; however in the era of high throughput screening and the rapid generation of libraries of small molecules, chemists need to keep up their pace to fulfil the demands of current research. Therefore new efficient and general synthetic methods, which produce privileged scaffolds such as  $\alpha$ -pyrones, coumarins and benzocoumarins under one set of reaction protocols, would be what is needed today, especially when the reported procedures usually

<sup>a</sup>Division of Medicinal and Process Chemistry, CSIR-Central Drug Research Institute, Lucknow, India. E-mail: atul\_goel@cdri.res.in

## Diversity-oriented general protocol for the synthesis of privileged oxygen scaffolds: pyrones, coumarins, benzocoumarins and naphthocoumarins†

Atul Goel,<br/>\*\* Gaurav Taneja,\* Ashutosh Raghuvanshi,\* Ruchir Kant<br/>b $^{\rm b}$  and Prakas R. Maulik $^{\rm b}$ 

A new general methodology for the synthesis of various functionalized privileged oxygen heterocyclic scaffolds, *viz.* pyrones, coumarins, and benzannulated coumarins, is developed. The synthesis proceeds through carbanion-induced ring transformation of lactones with various methylene carbonyl compounds followed by DDQ-mediated unprecedented oxidative cleavage of oxaylidenes intermediates. Studies of the mechanism of the conversions of oxaylidene intermediates into corresponding carbonyl compounds in the presence of DDQ revealed that the reactions took place *via* the formation of a Michael adduct instead of an intermolecular charge transfer complex. The methodology offers the fabrication of diverse privileged scaffolds with tolerance for many functional groups onto the oxygen heterocyclic molecular framework.

require expensive organometallic catalysts (Pd, Ru or Au), harsh reaction conditions and often lead to substances with traces of toxic metal impurities.<sup>6,7</sup> Herein we report a new protocol for the synthesis of various privileged heterocyclic scaffolds through carbanion-induced ring transformation of flexible or rigid lactones followed by DDQ-promoted unprecedented oxidative cleavage of oxaylidenes to their corresponding carbonyl compounds.

## **Results and discussion**

Over the last decade we have been exploring the reactivity of 2H-pyran-2-ones towards various C- and N-nucleophiles for generating diverse molecular architectures suitable for biological and material applications.8 We have recently demonstrated an efficient synthesis of the partially hydrogenated oxa[5]helicenes 1a, 2a and investigated their stereochemical behaviour in collaboration with Bringmann and co-workers.9 In our further efforts directed towards the oxidation of 9,10-dihydro-7-oxa[5]helicene-8-ylidenylacetonitrile 2a to produce oxa[5]helicene 3a with a higher inversion barrier, we reacted 2a (inversion barrier 78.2 kJ mol<sup>-1</sup>) with 2 equiv. of 2,3-dichloro-5,6dicyano-1,4-quinone (DDQ) using benzene as a solvent. Under this condition, we failed to obtain 3a. When we attempted the reaction of 5,6,9,10-tetrahydro-7-oxa[5]helicene-8-ylidenylacetonitrile 1a in dioxane using 4 equiv. of DDQ, we observed the formation of two new products 4a and 5a on TLC under UV exposure at 254 nm (Scheme 1). The structures of both the compounds were assigned by analyzing the spectroscopic data. The structure of 4a was unambiguously confirmed by single-

View Article Online View Journal | View Issue

**RSC**Publishing

<sup>&</sup>lt;sup>b</sup>Division of Molecular and Structural Biology, CSIR-Central Drug Research Institute, Lucknow, India

<sup>&</sup>lt;sup>†</sup>Electronic supplementary information (ESI) available: Copies of <sup>1</sup>H and <sup>13</sup>C NMR of the compounds prepared, X-ray data (CIF, **4a** and **12**). CCDC 877580 and 877581. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c3ob40859k

TT 1 10 (0/)



crystal X-ray analysis (**4a**: CCDC no. 877580; Scheme 1). An unusual chlorination of the ylidenic proton in **1a** led to the formation of **5a** in minor yield. Few unexpected chlorinations on different substrates have been reported earlier for DDQ-mediated oxidation reactions.<sup>10</sup> Interestingly the unexpected chlorination of the ylidenic proton is possible, however, the conversion of such an ylidenic group into corresponding carbonyl functionality promoted by DDQ has not been reported in the literature prior to this study. These new observations prompted us to investigate the possibility of using the DDQ for the cleavage of the ylidenic double bond.

Considering the fact that there are limited synthetic methods for the preparation of dinaphthopyranones,<sup>11</sup> we thus focused our attention on optimizing the reaction conditions for this unexpected transformation (Table 1). In order to improve the yield of **4a**, we investigated the reaction using 5,6,9,10-tetrahydro-7-oxa[5]helicene-8-ylidenylacetonitrile **1a** (0.2 mmol) and DDQ (0.8 mmol) by changing the reaction conditions as mentioned in Table 1. No product was observed in THF and MeOH (Table 1, entries 7 and 8) while only traces of cleaved products were formed in acetonitrile and acetone at room temp (Table 1, entries 5 and 6). When dioxane was used as a solvent we isolated **4a** in moderate yield (entries 1 and 2), together with a minor product **5a** (see the ESI; Scheme S1<sup>†</sup>).

However switching the solvent from dioxane to acetic acid significantly increased the yield of **4a** (entries 3 and 4) and only **4a** was isolated as the sole product. The best result was obtained with 4 equiv. of DDQ in AcOH at room temperature producing a 58% yield of **4a** (entry 4). It is worth noting that in the absence of DDQ, no reaction occurred suggesting the role of DDQ in the given transformation (entries 3 and 6).

With the optimized conditions in hand, a series of substituted (*Z*)-2-(5,6-dihydro-1*H*-benzo[*f*]naphtho-[2,1-*c*]chromen-4-(2*H*)-ylidene)acetonitriles<sup>9</sup> (**1a**-**f**) was subjected to oxidative cleavage using DDQ (4 equiv.) as an oxidant in acetic acid to afford partially hydrogenated 4*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4-ones (**4a**-**f**) in good yields (Scheme 2).

 Table 1
 Optimization of unusual transformation



|       |         |              |           |          | 1 leiu (%) |    |
|-------|---------|--------------|-----------|----------|------------|----|
| Entry | Solvent | DDQ (equiv.) | Temp (°C) | Time (h) | 4a         | 5a |
| -     | Dioxane | 4            | rt        | 24       | 15         | 5  |
| 2     | Dioxane | 4            | rx        | 4        | 43         | 12 |
|       | Dioxane | 0            | rx        | 4        | _          | _  |
| l.    | AcOH    | 4            | rt        | 1        | 58         | _  |
|       | AcOH    | 4            | rx        | 2        | 42         | _  |
| 5     | AcOH    | 0            | rx        | 2        | _          | _  |
| ,     | MeCN    | 4            | rt        | 24       | Trace      | _  |
| :     | Acetone | 4            | rt        | 24       | Trace      | _  |
| )     | THF     | 4            | rt        | 24       | _          | _  |
| .0    | MeOH    | 4            | rt        | 24       | —          | —  |
|       |         |              |           |          |            |    |

<sup>*a*</sup> Isolated yield, rt means room temperature, rx means reflux.

The compounds 4a,b were further converted into their fully aromatized products 4H-benzo[f]naphtho[2,1-c]chromen-4-ones (11a,b) using an excess amount of DDQ (15 equiv.) in refluxing benzene. Such dinaphthopyranones exist in configurationally labile helimeric equilibrium and they are important chemical scaffolds for the atroposelective synthesis of a broad range of naturally occurring biaryls and axially chiral auxiliaries.<sup>11</sup> The classical approaches for the synthesis of dinaphthopyranones



**Scheme 2** Synthesis of 4*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4-ones **4** and **11**.

Paper



Scheme 3 Proposed mechanism for the oxidative cleavage of the ylidenic bond of 2a.

mainly involve palladium-catalyzed intramolecular Heck arylation using harsh reaction conditions and limited only to less sterically hindered substrates.<sup>11</sup> Our approach provides an easy access to these interesting compounds under mild reaction conditions and without using any transition metal catalysts. In order to examine the reaction mechanism, a survey of the literature reveals examples of carbon-oxygen adduct formation through charge transfer complex between DDQ and aromatic hydrocarbons, heterocycles, and olefins.<sup>12</sup> Based on these literature reports, we proposed two possible pathways for the formation of 4a as depicted in Scheme 3, which revealed that the formation of 4a may be explained via a nonradical mechanism (path I) or through a radical mechanism (path II). In the nonradical mechanism (path I), a carbon-oxygen adduct (A) may be formed by Michael addition of the 5,6,9,10-tetrahydro-7-oxa[5]-helicene-8-ylidenyl-acetonitrile 2a to the 2,3dichloro-5,6-dicyano-1,4-quinone (DDQ) followed by intramolecular cyclization involving methyne carbon and the nitrile group of DDQ to produce intermediate B. This intermediate B under acidic conditions may lead to intermediate C, which on further internalization would afford the final product 4a together with the formation of 3-amino-benzofuran 12. Our presumption for the path I is supported by the isolation and characterization of a final product 4a together with the benzofuran derivative 12, whose structure was unambiguously confirmed using X-ray analysis of the air-sensitive single-crystals of 12 in acetonitrile (Fig. 1). It is worth noting that the formation of diffraction quality crystals of 12 was largely dependent on the acetonitrile solvent molecules, which formed intermolecular hydrogen bonding with the NH<sub>2</sub> group of the furan ring. As soon as the solvent evaporates during the crystal growth, the crystals become opaque revealing the importance of strong non-covalent interactions with acetonitrile solvent in the crystal formation.

Similarly a radical mechanism (path II) may be proposed *via* the formation of an intermolecular charge transfer (ICT) complex between **2a** and DDQ followed by a single electron transfer (SET),<sup>12</sup> which on recombination may form a carbon-oxygen adduct (**A**). To check the possibility of a single electron transfer (SET) mechanism (path II), we carried out an independent experiment by reacting **2a** and DDQ in acetic acid at room



Fig. 1 The crystal structure of 12 showing the network of intermolecular N–H…N and O–H…N interactions.

temperature. The absorption spectra of the reaction mixtures (~10<sup>-6</sup>  $\mu$ M) in acetic acid were monitored at different time intervals (Fig. 2).

Neither a coloured ICT complex nor any band in the range of 500–700 nm was observed during the reaction.<sup>12</sup> These results clearly suggested that the reaction follows non-ICT path I to produce compounds **4a** and **12**.

To further extend the scope of the above ring transformation and considering that 4-aryl-coumarins (neoflavones) are ubiquitous structural units in a plethora of natural products with interesting biological activities,<sup>13</sup> we designed a new protocol utilizing the reactivity of substituted resorcinols 13 towards 2H-pyran-2-ones 14 in the presence of a base (Scheme 4). Thus reaction of commercially available substituted resorcinols 13 (R<sup>1</sup>, R<sup>2</sup> = H, Me, Cl) and 6-aryl-4-(methylthio)-2-oxo-2H-pyran-3-carboxylate<sup>14</sup> 14 in NaH/DMF afforded methyl 2-(7-hydroxy-2H-chromen-2-ylidene)acetates 15 in good yields (up to 60-70%) as a mixture of geometrical isomers. As depicted in Scheme 4, the reaction is possibly initiated by a Michael addition of a conjugate base of resorcinol generated in situ in the presence of a base, at C6 of the 6-aryl-4-(methylthio)-2-oxo-2H-pyran-3-carboxylates 14 followed by ring opening to give intermediate I.



Fig. 2 Absorption spectra of the reaction mixtures (~10^{-6}  $\mu M)$  in acetic acid at different time intervals.

This Michael intermediate I may undergo an intramolecular cyclization involving the carbonyl functionality and C4 center of 14 with the elimination of carbon dioxide and methyl mercaptan to yield the (Z/E)-methyl 2-(7-hydroxy-2H-chromen-2-ylidene)acetates 15 in good yields. Furthermore these ylidenes 15a-f were converted to 4-aryl-7-hydroxy-coumarins 16a-f through oxidative cleavage using DDQ (2 equiv.) in dioxane at 80 °C for 1-2 h. The formation of carbomethoxysubstituted benzofuran 17 as a side product similar to the product 12 confirmed the possible route I for obtaining the coumarins 16a-f. Although numerous synthetic methodologies for 4-aryl/heteroaryl-coumarins are reported, most of them depend on organometal-catalyzed reactions via C-H or C-X activation on a preformed coumarin nucleus.15 Our novel methodology provides an easy access to 4-aryl-7-hydroxycoumarin derivatives under a transition-metal free environment using easily available precursors in good yields. After successful demonstration of this methodology for the preparation of 4-aryl-7-hydroxy-coumarins 16a-f, we extended the methodology to the synthesis of benzocoumarins viz. 5,6-dihydro-2*H*-benzo[*h*]chromen-2-ones (19) and 2*H*-benzo[*h*]chromen-2-ones (20) (Scheme 5). To synthesize these compounds, the requisite precursor methyl 2-(4-aryl-5,6-dihydro-2H-benzo[h]chromen-2-ylidene)acetates 18a-c were prepared according to improved methodology.<sup>16</sup> Treatment of 18a-c with DDQ (2 equiv.)/AcOH afforded the 4-aryl-5,6-dihydro-2Hbenzo[h]chromen-2-one **19a-c** in moderate yields (46-52%). The partially hydrogenated benzocoumarin 19a was aromatized using DDQ in benzene at reflux temperature to afford 4-(4-bromophenyl)-8-methoxy-2H-benzo[h]chromen-2-one (20a) in a 52% yield.

Finally, the synthesis of highly substituted  $\alpha$ -pyrones (24a–f) was explored as shown in Scheme 6. The precursors functionalized 2*H*-pyran-2-ylidene-acetonitriles 23a–f were prepared by reacting substituted 2*H*-pyran-2-ones 21a–f with substituted methylene carbonyl compounds 22a–f (see the

#### View Article Online

**Organic & Biomolecular Chemistry** 



Scheme 4 Synthesis of 4-arylcoumaryl-2-ylidenes 15 and 4-arylcoumarins 16.

Experimental section) and converted to highly substituted  $\alpha$ -pyrones **24a–f** through oxidative cleavage using DDQ (2 equiv.)/AcOH at room temperature in 5–6 h (Scheme 6).

It is important to note that highly hindered 2-(6-isopropyl-4-(4-methoxyphenyl)-3-methyl-5-phenyl-2*H*-pyran-2-ylidene)acetonitrile (**23f**) was successfully converted to the corresponding carbonyl compound 6-isopropyl-4-(4-methoxyphenyl)-3-methyl-5-phenyl-2*H*-pyran-2-one (**24f**) under the same reaction conditions. This new methodology offers a general synthesis of  $\alpha$ -pyrones with four point diversity.

### Conclusion

In summary, we have demonstrated a novel and general synthesis of highly functionalized  $\alpha$ -pyrones, coumarins and



Scheme 5 Synthesis of benzocoumarinyl-2-ylidenes 18 and benzocoumarins 19 and 20.



Scheme 6 Synthesis of pyran-2-ylidenes 23 and  $\alpha$ -pyrones 24.

benzannulated coumarins through the ring transformation of easily accessible lactones with various methylene carbonyl compounds followed by unprecedented DDQ-mediated oxidative cleavage of oxaylidene intermediates under mild reaction conditions. Studies of the synthesis mechanism of a benzonaphthocoumarin revealed that the reaction was initiated through Michael addition as examined using UV-vis absorption experiments. This methodology can be used for the synthesis of a variety of oxygen heterocycles with flexibility of substituents variation on to the molecular scaffolds. Interestingly, this diversity-oriented synthetic protocol may be useful to generate a small library of compounds for drug discovery perspectives.

## **Experimental section**

<sup>1</sup>H and <sup>13</sup>C NMR spectra were taken on a NMR spectrophotometer at 400, 300 and 200 MHz. CDCl<sub>3</sub> and DMSO-d<sub>6</sub> were used as the solvents. Chemical shifts are reported in parts per million ( $\delta$ -value) from Me<sub>4</sub>Si ( $\delta$  = 0 ppm for 1H) as an internal standard or based on the middle peak of the solvent (for CDCl<sub>3</sub>, 7.26 ppm for <sup>1</sup>H and 77.1 (±0.1) ppm for <sup>13</sup>C NMR; for DMSO-d<sub>6</sub>, 2.5 ppm for <sup>1</sup>H and 40.0 ppm for <sup>13</sup>C NMR). Signal patterns are indicate as s, singlet; d, doublet; dd, double doublet; t, triplet; m, multiplet. Coupling constants (*J*) are given in hertz. Infrared (IR) spectra were recorded on an IR spectrophotometer using KBr discs and reported in wave number (cm<sup>-1</sup>). MS (ESI) and HRMS (ESI) were recorded on a Mass Spectrophotometer with a Q-ToF mass analyzer.

### General procedure for the synthesis of 5,6-dihydro-1*H*-benzo-[*f*]naphtho[2,1-*c*]chromen-4(2*H*)-ylidene)acetonitrile 1a-f<sup>9</sup>

A mixture of the 4-morpholino-2-oxo-5,6-dihydro-2*H*-benzo[*h*]chromene-3-carbonitriles **9a–c** (1.0 mmol), the respective 2-tetralones **10** (1.2 mmol), and NaH (1.5 mmol) in THF was stirred for 1–2.5 h at ambient temperature. The reaction was monitored using TLC until completion. Excess THF was removed under reduced pressure and the reaction mixture was poured into ice-cool water with constant stirring followed by neutralization with 10% HCl. The precipitate thus obtained was filtered, washed with water, dried and purified with a neutral alumina column using 20% CHCl<sub>3</sub> in hexane as the eluent to afford **1a–f**.

### (Z)-2-(11-Methoxy-5,6-dihydro-1*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4(2*H*)-ylidene)acetonitrile (1b)

A mixture of 4-morpholino-2-oxo-5,6-dihydro-2*H*-benzo[*h*]chromene-3-carbonitrile (**9a**, 308 mg, 1.0 mmol, 1 equiv.), 8-methoxy-2-tetralone (**10**, 211 mg, 1.2 mmol, 1.2 equiv.) and NaH (60% dispersion in oil, 60 mg, 1.5 mmol, 1.5 equiv.) in dry THF (5 ml) was stirred at room temperature for 1.5 h. Using the general procedure described above, 208 mg (59%) of **1b** was obtained as a yellow solid:  $R_{\rm f} = 0.44$  (CHCl<sub>3</sub>–*n*-hexane, 7:3, v/v); mp 146–148 °C (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{\rm max}/{\rm cm}^{-1} = 2927$  (m), 2847 (w), 2190 (s), 1628 (s); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 2.42–3.11 (m, 11H), 4.43 (s, 1H), 6.54 (d, J = 8.2 Hz, 1H), 6.84–6.98 (m, 4H), 7.08–7.17 (m, 2H) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>):  $\delta$  = 22.5, 27.2, 27.8, 29.0, 53.9, 63.7, 108.8, 110.2, 118.8, 120.1, 121.3, 123.1, 123.4, 125.9, 127.1, 127.6, 134.1, 135.6, 136.6, 137.6, 154.4, 159.2, 165.2 ppm; m/z (ESI): 354 [M + H<sup>+</sup>]; HRMS (ESI) calcd for  $C_{24}H_{19}NO_2$  [M<sup>+</sup> + 1] 354.1494, found 354.1491.

### (Z)-2-(9-Methoxy-5,6-dihydro-1*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4(2*H*)-ylidene)acetonitrile (1e)

A mixture of 9-methoxy 4-morpholino-2-oxo-5,6-dihydro-2Hbenzo[h]chromene-3-carbonitrile (9e, 338 mg, 1.0 mmol, 1 equiv.), 2-tetralone (10, 0.146 ml, 1.2 mmol, 1.2 equiv.) and NaH (60% dispersion in oil, 60 mg, 1.5 mmol, 1.5 equiv.) in dry THF (5 ml) was stirred at room temperature for 1 h. Using the general procedure described above, 155 mg (44%) of 1e was obtained as a yellow solid:  $R_f = 0.47$  (CHCl<sub>3</sub>-*n*-hexane, 7:3, v/v); mp 162–164 °C (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{\text{max}}/\text{cm}^{-1}$  = 2932 (m), 2848 (w), 2199 (s), 1728 (w); <sup>1</sup>H NMR (300 MHz,  $CDCl_3$ :  $\delta$  = 2.08–2.27 (m, 1H), 2.40–2.53 (m, 1H), 2.72–2.86 (m, 5H), 3.01-3.18 (m, 1H), 3.44 (s, 3H), 4.49 (s, 1H), 6.54 (d, J = 1.7 Hz, 1H), 6.73–6.86 (m, 2H), 6.93–7.24 (m, 4H) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>):  $\delta$  = 23.0, 26.8, 27.3, 28.3, 55.1, 64.6, 110.8, 113.9, 115.2, 118.6, 125.6, 126.2, 126.3, 127.3, 127.6, 128.3, 129.6, 131.4, 131.6, 134.1, 135.3, 157.3, 159.1, 164.8 ppm; m/z (ESI): 354 [M + H<sup>+</sup>]; HRMS (ESI) calcd for  $C_{24}H_{19}NO_2[M^+ + 1]$  354.1494, found 354.1480.

#### General procedure for the synthesis of 4a-f

(*Z*)-2-(5,6-Dihydro-1*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4(2*H*)ylidene)acetonitriles (**1a**–**f**) were treated with 4.0 mmol of DDQ in AcOH at room temp. for 1 h. After completion, the reaction mixture was filtered and the filtrate was collected and evaporated to dryness. Thereafter, the crude obtained was purified through neutral alumina using 60% CHCl<sub>3</sub> in hexane as eluent to afford **4a–f**.

#### 5,6-Dihydro-4H-benzo[f]naphtho[2,1-c]chromen-4-one (4a)

(Z)-2-(5,6-dihydro-1H-benzo[f]naphtho[2,1-c]chromen-4 The (2H)-ylidene)acetonitrile (1a, 323 mg, 1.0 mmol, 1 equiv.) was treated with DDQ (908 mg, 4.0 mmol, 4 equiv.) in glacial acetic acid for 1 h. Using the general procedure described above, 172 mg (58%) of 4a was obtained as a light yellow solid:  $R_{\rm f}$  = 0.57 (CHCl<sub>3</sub>-n-hexane, 7:3, v/v); mp 204-206 °C (CHCl<sub>3</sub>-nhexane); IR (KBr)  $\nu_{\text{max}}$ /cm<sup>-1</sup> = 2931 (w), 2744 (w), 1704 (s), 1609 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  = 2.23–2.41 (m, 1H), 2.90-3.00 (m, 2H), 3.16-3.29 (m, 1H), 7.05-7.15 (m, 1H), 7.20-7.30 (m, 2H), 7.31-7.53 (m, 4H), 7.81-7.96 (m, 3H) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta$  = 22.8, 28.1, 112.4, 117.3, 125.4, 125.5, 125.8, 126.7, 127.2, 127.9, 128.6, 128.9, 129.0, 130.0, 131.1, 132.0, 132.4, 138.5, 145.9, 152.6, 161.5 ppm; m/z(ESI): 299  $[M^+ + H]$ ; HRMS (ESI) calcd for  $C_{21}H_{14}O_2 [M^+ + 1]$ 299.1072, found 299.1081.

The crystal data of **4a** (CCDC deposit no: 877580): C<sub>21</sub>H<sub>14</sub>O<sub>2</sub>, M = 298.32, Orthorhombic, *Pbca*, a = 15.050(5) Å, b = 8.960(3) Å, c = 21.187(7) Å, V = 2857(2) Å<sup>3</sup>, Z = 8,  $D_c =$  1.387 g cm<sup>-3</sup>,  $\mu$  (Mo-Kα) = 0.09 mm<sup>-1</sup>, F(000) = 1248, rectangular block, reddish, 17 307 reflections measured ( $R_{int}$  = 0.0861), 3513 unique, w $R_2$  = 0.2040 for all data, conventional  $R_1$  = 0.0630 for 2408  $F_0 > 4\sigma(F_0)$  and 0.1060 for all 3513 data, S = 1.083 for all data and 209 parameters.

#### 11-Methoxy-5,6-dihydro-4*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4-one (4b)

The (Z)-2-(11-methoxy-5,6-dihydro-1*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4(2H)-ylidene)acetonitrile (1b, 353 mg, 1.0 mmol, 1 equiv.) was treated with DDQ (908 mg, 4.0 mmol, 4 equiv.) in glacial acetic acid for 1 h. Using the general procedure described above, 157 mg (48%) of 4b was obtained as a light yellow solid:  $R_f = 0.47$  (CHCl<sub>3</sub>-*n*-hexane, 7:3, v/v); mp 191–193 °C (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{\text{max}}/\text{cm}^{-1}$  = 3020 (m), 2932 (w), 1725 (s), 1610 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  = 2.26-2.42 (m, 1H), 2.93-3.01 (m, 2H), 3.02 (s, 3H), 3.16-3.30 (m, 1H), 6.72 (d, J = 7.6 Hz, 1H), 6.84 (d, J = 7.6 Hz, 1H), 6.94-7.03 (m, 1H), 7.17-7.51 (m, 5H), 7.88 (d, J = 8.8 Hz, 1H) ppm; <sup>13</sup>C NMR (50.3 MHz, CDCl<sub>3</sub>)  $\delta$  = 22.4, 27.9, 53.6, 107.6, 117.5, 120.5, 121.3, 123.5, 124.0, 126.0, 126.2, 127.6, 128.7, 131.3, 132.4, 136.1, 152.5, 155.1, 161.7 ppm; m/z (ESI): 329  $[M^+ + H]$ ; HRMS (ESI) calcd for  $C_{22}H_{16}O_3$   $[M^+ + 1]$  329.1178, found 329.1177.

#### 14-Methoxy-5,6-dihydro-4*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4-one (4c)

The (*Z*)-2-(14-methoxy-5,6-dihydro-1*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4(2*H*)-ylidene)acetonitrile (**3c**, 353 mg, 1.0 mmol, 1 equiv.) was treated with DDQ (908 mg, 4.0 mmol, 4 equiv.) in glacial acetic acid for 1 h. Using the general procedure described above, 170 mg (52%) of **4c** was obtained as a light yellow solid:  $R_{\rm f} = 0.50$  (CHCl<sub>3</sub>-*n*-hexane, 7 : 3, v/v); mp 149–151 °C (CHCl<sub>3</sub>-*n*-hexane); IR (KBr)  $\nu_{\rm max}$ /cm<sup>-1</sup> = 2942 (m), 2841 (w), 1711 (s); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  = 2.23–2.42 (m, 1H), 2.91–3.02 (m, 2H), 3.17–3.30 (m, 1H), 4.03 (s, 3H), 6.79 (d, *J* = 7.7 Hz, 1H), 7.06–7.51 (m, 7H), 8.45 (d, *J* = 9.1 Hz, 1H) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta$  = 22.7, 28.0, 55.7, 103.6, 112.1, 116.2, 119.9, 123.0, 125.7, 125.9, 126.1, 126.6, 127.8, 129.0, 129.9, 130.2, 132.5, 138.4, 146.2, 152.9, 155.6, 161.6 ppm; *m*/*z* (ESI): 329 [M<sup>+</sup> + H]; HRMS (ESI) calcd for C<sub>22</sub>H<sub>16</sub>O<sub>3</sub> [M<sup>+</sup> + 1] 329.1178, found 329.1180.

#### 12-Methoxy-5,6-dihydro-4*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4-one (4d)

The (*Z*)-2-(12-methoxy-5,6-dihydro-1*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4(2*H*)-ylidene)acetonitrile (**1d**, 353 mg, 1.0 mmol, 1 equiv.) was treated with DDQ (908 mg, 4.0 mmol, 4 equiv.) in glacial acetic acid for 1 h. Using the general procedure described above, 181 mg (55%) of **4d** was obtained as a light yellow solid:  $R_{\rm f} = 0.52$  (CHCl<sub>3</sub>-*n*-hexane, 7:3, v/v); mp 176-178 °C (CHCl<sub>3</sub>-*n*-hexane); IR (KBr)  $\nu_{\rm max}/{\rm cm}^{-1} = 3022$  (m), 1629 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta = 2.16-2.33$  (m, 1H), 2.84-3.14 (m, 2H), 3.12 (s, 1H), 3.37 (s, 3H), 6.96-7.04 (m, 1H), 7.05-7.39 (m, 6H), 7.67 (d, *J* = 8.9 Hz, 1H), 7.78 (d, *J* = 8.8 Hz, 1H) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta = 22.6$ , 28.1, 54.7, 107.4, 111.5, 114.7, 117.6, 125.6, 126.3, 126.4, 128.0, 129.3, 129.9, 130.0, 130.3, 131.7, 132.1, 138.7, 145.9, 153.2, 157.1, 161.5 ppm; m/z (ESI): 329 [M<sup>+</sup> + H]; HRMS (ESI) calcd for  $C_{22}H_{16}O_3$  [M<sup>+</sup> + 1] 329.1178, found 329.1176.

### 9-Methoxy-5,6-dihydro-4*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4-one (4e)

The (Z)-2-(9-methoxy-5,6-dihydro-1*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4(2H)-ylidene)acetonitrile (1e, 353 mg, 1.0 mmol, 1 equiv.) was treated with DDQ (908 mg, 4.0 mmol, 4 equiv.) in glacial acetic acid for 1 h. Using the general procedure described above, 187 mg (57%) of 4e was obtained as a light yellow solid:  $R_f = 0.45$  (CHCl<sub>3</sub>-*n*-hexane, 7:3, v/v); mp 163–165 °C (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{max}/cm^{-1}$  = 2930 (m), 2851 (w), 1708 (s); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  = 2.23–2.42 (m, 1H), 2.87-2.95 (m, 2H), 3.16-3.29 (m, 1H), 3.52 (s, 3H), 6.81 (d, J = 2.4 Hz, 1H), 6.92 (dd, J = 8.2, 2.6 Hz, 1H), 7.27–7.36 (m, 2H), 7.40-7.55 (m, 2H), 7.83-7.99 (m, 3H) ppm; <sup>13</sup>C NMR  $(75.5 \text{ MHz}, \text{CDCl}_3) \delta = 22.7, 27.1, 55.3, 112.2, 114.1, 114.2,$ 116.3, 117.3, 125.4, 125.4, 127.0, 127.4, 128.6, 128.7, 128.8, 130.6, 131.2, 132.0, 133.1, 139.2, 145.8, 152.5, 157.5, 161.4 ppm; m/z (ESI): 329 [M<sup>+</sup> + H]; HRMS (ESI) calcd for  $C_{22}H_{16}O_3 [M^+ + 1] 329.1178$ , found 329.1175.

### 8-Methoxy-5,6-dihydro-4*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4-one (4f)

The (Z)-2-(9-methoxy-5,6-dihydro-1*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4(2H)-ylidene)acetonitrile (1f, 353 mg, 1.0 mmol, 1 equiv.) was treated with DDQ (908 mg, 4.0 mmol, 4 equiv.) in glacial acetic acid for 1 h. Using the general procedure described above, 154 mg (47%) of 4f was obtained as a light yellow solid:  $R_f = 0.44$  (CHCl<sub>3</sub>-*n*-hexane, 7:3, v/v); mp 156–158 °C (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{\text{max}}/\text{cm}^{-1}$  = 2942 (m), 2842 (w), 1653 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  = 2.25–2.44 (m, 1H), 2.86-3.07 (m, 2H), 3.17-3.28 (m, 1H), 3.87 (s, 3H), 6.61-6.69 (m, 1H), 6.91 (s, 1H), 7.17-7.32 (m, 2H), 7.39-7.53 (m, 2H), 7.82–7.99 (m, 3H) ppm; <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta = 22.5, 28.5, 55.4, 111.3, 112.4, 113.2, 117.3, 124.5, 125.2,$ 125.3, 125.4, 127.1, 128.5, 129.1, 130.6, 131.1, 131.9, 140.7, 145.9, 152.5, 160.9, 161.6 ppm; m/z (ESI): 329 [M<sup>+</sup> + H]; HRMS (ESI) calcd for  $C_{22}H_{16}O_3 [M^+ + 1] 329.1178$ , found 329.1207.

#### 2-Chloro-2-(5,6-dihydro-4*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4-ylidene)acetonitrile (5a)

The (*Z*)-2-(5,6-dihydro-1*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4-(2*H*)-ylidene)acetonitrile (1a, 323 mg, 1.0 mmol, 1 equiv.) was treated with DDQ (908 mg, 4.0 mmol, 4 equiv.) in dioxane for 4 h at reflux temperature. After completion, reaction mixture was filtered and filtrate was collected and evaporated to dryness. Thereafter, the crude mixture was purified through neutral alumina using 10% CHCl<sub>3</sub> in hexane as the eluent to afford 43 mg (12%) of 5a {along with 128 mg (43%) of 4a}as a light yellow solid:  $R_{\rm f} = 0.71$  (CHCl<sub>3</sub>-*n*-hexane, 7:3 v/v); mp 178–180 °C (CHCl<sub>3</sub>-*n*-hexane); IR (KBr)  $\nu_{\rm max}$ /cm<sup>-1</sup> = 2924 (s), 2855 (m), 2193 (s), 1540 (s); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 2.32–2.55 (m, 1H), 2.92–3.03 (m, 2H), 3.65–3.76 (m, 1H), 7.04–7.09 (m, 2H), 7.20–7.51 (m, 5H), 7.66 (d, J = 8.6 Hz, 1H), 7.80–7.94 (m, 2H); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>):  $\delta = 24.0$ , 27.8, 113.5, 116.4, 118.2, 125.3, 125.7, 126.0, 127.0, 127.4, 127.9, 128.4, 128.6, 129.5, 131.2, 131.8, 132.5, 137.1, 137.3, 151.8, 158.1 ppm; m/z (ESI): 356 [M<sup>+</sup> + H]; HRMS (ESI) calcd for C<sub>23</sub>H<sub>14</sub>ClNO [M<sup>+</sup> + 1] 356.0842, found 356.0850.

#### General procedure for the synthesis of 8a-c<sup>17</sup>

A mixture of methyl 2-cyano-3,3-dimethylthioacrylate 7 (10.1 g, 0.05 mol), the respective 1-tetralone **6a–c** (7.3 ml, 0.05 mol) was stirred in the presence of powdered KOH (3.3 g, 0.06 mol) in DMSO (50 ml) for 5–6 h at ambient temperature. Completion of the reaction was monitored using TLC and on completion the reaction mixture was poured into ice-cool water with constant stirring and the resulting precipitate was filtered, washed with water, dried, and purified through a silica gel column using 2% MeOH in CHCl<sub>3</sub> as the eluent to afford **8a–c**.

#### Synthesis of 9-methoxy-4-(methylthio)-2-oxo-5,6-dihydro-2*H*benzo[*h*]chromene-3-carbonitrile (8c)

A mixture of methyl 2-cyano-3,3-dimethylthioacrylate 7 (10.1 g, 0.05 mol) and 7-methoxy-1-tetralone **6c** (8.8 g, 0.05 mol) was stirred in the presence of powdered KOH (3.3 g, 0.06 mol) in DMSO (50 ml) for 6 h. Using the general procedure described above, 13 g (87%) of **8c** was obtained as a yellow solid:  $R_{\rm f} = 0.56$  (MeOH-CHCl<sub>3</sub>, 1:25 v/v); mp 188–190 °C (CHCl<sub>3</sub>-*n*-hexane); IR (KBr)  $\nu_{\rm max}$ /cm<sup>-1</sup> = 3339 (m), 2212 (w), 1713 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 2.74-2.83$  (m, 2H), 2.84–2.93 (m, 2H), 2.99 (s, 3H), 3.84 (s, 3H), 6.92–7.02 (m, 1H), 7.14 (d, J = 8.4 Hz, 1H), 7.34–7.41 (m, 1H) ppm; <sup>13</sup>C NMR (50.3 MHz, CDCl<sub>3</sub>):  $\delta = 17.6$ , 22.0, 26.1, 55.7, 108.6, 119.3, 129.0, 130.6, 131.0, 138.4, 159.1 ppm; m/z (ESI): 300 [M<sup>+</sup> + H]; HRMS (ESI) calcd for C<sub>16</sub>H<sub>13</sub>NO<sub>3</sub>S [M<sup>+</sup> + 1] 300.0694, found 300.0695.

#### General procedure for the synthesis of 9a-c<sup>17</sup>

A mixture of **8** (2.7 g, 0.01 mol) and morpholine (1.3 ml, 0.012 mol) in methanol (25 ml) was refluxed for 5 h. The reaction mixture was cooled to room temperature, and the precipitate obtained was filtered, washed with methanol, dried, and crystallized from methanol to afford **9a–c**.

# Synthesis of 9-methoxy-4-morpholino-2-oxo-5,6-dihydro-2*H*-benzo[*h*]chromene-3-carbonitrile (9c)

A mixture of **8c** (3.0 g, 0.01 mol) and morpholine (1.3 ml, 0.012 mol) in methanol (25 ml) was refluxed for 5 h. The reaction mixture was cooled to room temperature, and the precipitate obtained was filtered, washed with methanol, dried, and crystallized from methanol to afford 3.01 g (89%) of **9c** as a yellow solid:  $R_{\rm f} = 0.49$  (MeOH–CHCl<sub>3</sub>, 1:25 v/v); mp 174–176 °C (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{\rm max}/{\rm cm}^{-1} = 3021$  (w), 2215 (w), 1635 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 2.56-2.72$  (m, 2H), 2.75–2.91 (m, 2H), 3.52–3.69 (m, 4H), 3.82 (s, 3H), 3.85–3.98 (m, 4H), 6.87–7.01 (m, 1H), 7.13 (d, J = 8.3 Hz, 1H), 7.28–7.39 (m, 1H) ppm; <sup>13</sup>C NMR (50.3 MHz, CDCl<sub>3</sub>):  $\delta = 23.8$ , 26.9, 51.7, 55.7, 67.1, 81.5, 108.7, 116.5, 118.7, 128.5, 129.9,

157.7, 159.1, 166.6 ppm; m/z (ESI): 339 [M<sup>+</sup> + H]; HRMS (ESI) calcd for C<sub>19</sub>H<sub>18</sub>N<sub>2</sub>O<sub>4</sub> [M<sup>+</sup> + 1] 339.1345, found 339.1336.

#### General procedure for the synthesis of 11a,b

5,6-Dihydro-4*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4-ones (4**a**,**b**) were treated with 15 mmol of DDQ in benzene and refluxed for 24 h. After completion, reaction mixture was filtered and filtrate was collected and evaporated to dryness. Thereafter, the crude obtained was purified through neutral alumina using 40% CHCl<sub>3</sub> in hexane as the eluent.

#### 4H-Benzo[f]naphtho[2,1-c]chromen-4-one (11a)

The 5,6-dihydro-4*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4-one (4a, 298 mg, 1.0 mmol, 1 equiv.) was treated with DDQ (3405 mg, 15.0 mmol, 15 equiv.) in refluxing benzene for 24 h. Using the general procedure described above, 177 mg (60%) of **11a** was obtained as an off-white solid:  $R_{\rm f} = 0.71$  (CHCl<sub>3</sub>–*n*-hexane, 7 : 3, v/v); mp 237–239 °C {lit.<sup>18</sup> 236–238 °C} (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{\rm max}/\rm cm^{-1} = 2924$  (m), 2855 (w), 1730 (s); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta = 7.32-7.55$  (m, 3H), 7.57–7.73 (m, 2H), 7.88–8.16 (m, 6H), 8.37 (d, J = 8.5 Hz, 1H) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta = 112.9$ , 117.2, 121.6, 124.2, 125.4, 125.5, 125.8, 127.2, 128.3, 128.4, 129.1, 129.4, 129.8, 131.2, 131.7, 134.9, 136.7, 150.5, 161.8 ppm; *m*/*z* (ESI): 297 [M<sup>+</sup> + H]; HRMS (ESI) calcd for C<sub>21</sub>H<sub>12</sub>O<sub>2</sub> [M<sup>+</sup> + 1] 297.0916, found 297.0900.

#### 11-Methoxy-4*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4-one (11b)

The 11-methoxy-5,6-dihydro-4*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4-one (**4b**, 328 mg, 1.0 mmol, 1 equiv.) was treated with DDQ (3405 mg, 15.0 mmol, 15 equiv.) in refluxing benzene for 24 h. Using the general procedure described above, 185 mg (57%) of **11b** was obtained as an off-white solid:  $R_f = 0.60$  (CHCl<sub>3</sub>-*n*-hexane, 7 : 3, v/v); mp 214–216 °C (CHCl<sub>3</sub>-*n*hexane); IR (KBr)  $\nu_{max}$ /cm<sup>-1</sup> = 2946 (m), 2835 (m), 1654 (m); <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>)  $\delta$  = 2.98 (s, 3H), 6.81 (d, *J* = 7.4 Hz, 1H), 7.28–7.38 (m, 1H), 7.52–7.64 (m, 4H), 7.79 (d, *J* = 8.5 Hz, 1H), 7.91–8.02 (m, 3H), 8.30 (d, *J* = 8.6 Hz, 1H) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta$  = 53.9, 106.6, 110.7, 117.5, 120.7, 123.3, 125.3, 126.0, 126.3, 128.0, 128.2, 128.7, 131.3, 134.7, 155.6 ppm; *m*/*z* (ESI): 327 [M<sup>+</sup> + H]; HRMS (ESI) calcd for C<sub>22</sub>H<sub>14</sub>O<sub>3</sub> [M<sup>+</sup> + 1] 327.1021, found 327.1016.

#### 3-Amino-6,7-dichloro-5-hydroxybenzofuran-2,4dicarbonitrile (12)

The (*Z*)-2-(5,6-dihydro-4*H*-benzo[*f*]naphtho[2,1-*c*]chromen-4ylidene)acetonitrile (**2a**, 321 mg, 1.0 mmol, 1 equiv.) was treated with DDQ (554 mg, 2.0 mmol, 2 equiv.) in glacial acetic acid for 4 h. After completion, the reaction mixture was filtered and filtrate was collected and evaporated to dryness. Thereafter, the crude obtained was purified with a neutral alumina column using 10% MeOH in chloroform as the eluent to afford 88 mg (33%) **12** (along with **4a**) as an off-white solid:  $R_{\rm f} = 0.31$  (MeOH–CHCl<sub>3</sub>, 1:4, v/v); mp > 250 °C (CH<sub>3</sub>CN); IR (KBr)  $\nu_{\rm max}/{\rm cm}^{-1} = 3489$  (w), 3374 (m), 3151 (m), 2222 (s); <sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>)  $\delta = 6.12$  (brs, 2H) ppm; <sup>13</sup>C NMR (100.6 MHz, DMSO-d<sub>6</sub>)  $\delta = 90.8$ , 111.7, 112.9, 114.3, 119.7, 122.3, 124.5, 141.0, 144.6, 155.2 ppm; m/z (ESI): 267 [M<sup>+</sup> + H]; HRMS (ESI) calcd for  $C_{10}H_3Cl_2N_3O_2$  [M<sup>+</sup> + 1] 267.9681, found 267.9641.

The crystal data for **12** (CCDC deposit no: 877581):  $C_{12}H_6Cl_2N_4O_2$ , M = 309.11, monoclinic, C2/c, a = 20.542(2) Å, b = 7.512(1) Å, c = 18.182(2) Å,  $\beta = 109.593(1)^\circ$ , V = 2643.2(5) Å<sup>3</sup>, Z = 8,  $D_c = 1.554$  g cm<sup>-3</sup>,  $\mu$  (Mo-K $\alpha$ ) = 0.50 mm<sup>-1</sup>, F(000) =1248, rectangular block, yellow, 8334 reflections measured ( $R_{int} = 0.0537$ ), 4295 unique, w $R_2 = 0.1660$  for all data, conventional  $R_1 = 0.0594$  for 2448  $F_o > 4\sigma(F_o)$  and 0.1133 for all 4295 data, S = 0.993 for all data and 183 parameters.

#### General procedure for the synthesis of 15a-f

A mixture of the methyl 4-(methylthio)-2-oxo-6-aryl-2*H*-pyran-3-carboxylates **14** (1.0 mmol), the respective resorcinol **13** (2.0 mmol) and NaH (60% dispersion in oil, 200 mg, 5.0 mmol, 5 equiv.) in dry DMF (5 ml) was stirred at a temperature of 60 °C for 3-4 h. Completion of the reaction was monitored using TLC and on completion the reaction mixture was poured into crushed ice with vigorous stirring and finally neutralized with 10% HCl. The precipitate obtained was filtered and purified on a silica gel column using 10% ethyl acetate in *n*-hexane as the eluent to afford **15a–f** as a mixture of *cis-* and *trans*-isomers.

#### (*Z/E*)-Methyl 2-(7-hydroxy-4-phenyl-2*H*-chromen-2-ylidene)acetate (15a)

A mixture of methyl 4-(methylthio)-2-oxo-6-phenyl-2H-pyran-3-carboxylate (14a, 276 mg, 1.0 mmol, 1 equiv.), resorcinol (13a, 220 mg, 2.0 mmol, 2 equiv.) and NaH (60% dispersion in oil, 200 mg, 5.0 mmol, 5 equiv.) in dry DMF (5 ml) was stirred at a temperature of 60 °C for 3 h. Using the general procedure described above, 200 mg (68%) of 15a was obtained as a mixture of *cis*- and *trans*-isomers, as a yellow solid:  $R_{\rm f} = 0.47$ (ethyl acetate-n-hexane, 1:9, v/v); mp 210-212 °C (ethyl acetate–*n*-hexane); IR (KBr)  $\nu_{max}/cm^{-1} = 3584$  (s), 1636 (w); <sup>1</sup>H NMR (300 MHz,  $CDCl_3 + DMSO-d_6$ ):  $\delta = 4.43$  (s, 3H, *cis*), 4.47 (s, 3H, trans), 5.72 (s, 1H, trans), 6.09 (s, 1H, cis), 6.83 (s, 1H, trans), 7.31-7.38 (m, 2H, cis + trans), 7.42 (d, J = 2.2 Hz, 1H, cis), 7.61 (d, J = 2.2 Hz, 1H, trans), 7.82-7.92 (m, 2H, cis + trans), 8.10-8.21 (m, 12H, cis + trans), 10.37 (s, 1H, trans), 10.38 (s, 1H, *cis*) ppm; <sup>13</sup>C NMR (75.5 MHz, DMSO-d<sub>6</sub>):  $\delta$  = 50.0, 89.8, 102.0, 111.2, 111.7, 112.6, 126.7, 127.8, 128.4, 128.6, 135.7, 143.8, 153.3, 160.3, 162.1, 166.5; m/z (ESI): 295 [M + H<sup>+</sup>]; HRMS (ESI) calcd for  $C_{18}H_{14}O_4$  [M<sup>+</sup> + 1] 295.0970; found 295.0965.

### (Z/E)-Methyl 2-(7-hydroxy-4-(naphthalen-2-yl)-2*H*-chromen-2-ylidene)acetate (15b)

A mixture of methyl 4-(methylthio)-6-(naphthalen-2-yl)-2-oxo-2*H*-pyran-3-carboxylate (**14b**, 326 mg, 1.0 mmol, 1 equiv.), resorcinol (**13a**, 220 mg, 2.0 mmol, 2 equiv.) and NaH (60% dispersion in oil, 200 mg, 5.0 mmol, 5 equiv.) in dry DMF (5 ml) was stirred at a temperature of 60 °C for 4 h. Using the general procedure described above, 227 mg (66%) of **15b** was obtained as a mixture of *cis*- and *trans*-isomers, as a yellow solid:  $R_{\rm f} = 0.56$  (ethyl acetate–*n*-hexane, 1:9, v/v); mp 224–226 °C (ethyl acetate–*n*-hexane); IR (KBr)  $\nu_{\rm max}/{\rm cm}^{-1} = 3201$  (w), 2857 (m), 1647 (m), 1562 (w); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub> + DMSO-d<sub>6</sub>):  $\delta = 4.09$  (s, 3H, *cis*), 4.13 (s, 3H, *trans*), 5.42 (s, 1H, *trans*), 5.77 (s, 1H, *cis*), 6.62 (s, 1H, *trans*), 6.98–7.05 (m, 2H, *cis* + *trans*), 7.10 (d, J = 1.74 Hz, 1H, *trans*), 7.27 (s, 1H, *trans*), 7.54–7.63 (m, 2H, *cis* + *trans*), 7.91–7.99 (m, 8H, *cis* + *trans*), 8.31–8.38 (m, 8H, *cis* + *trans*), 10.22 (s, 1H, *trans*), 10.22 (s, 1H, *cis*) ppm; <sup>13</sup>C NMR (75.5 MHz, DMSO-d<sub>6</sub>):  $\delta = 50.9$ , 90.9, 102.9, 112.3, 112.7, 113.9, 126.4, 127.2, 127.4, 127.8, 127.9, 128.0, 128.7, 128.8, 133.3, 133.4, 134.2, 144.7, 154.3, 161.3, 163.0, 167.5; *m/z* (ESI): 345 [M + H<sup>+</sup>]; HRMS(ESI) calcd for C<sub>22</sub>H<sub>16</sub>O<sub>4</sub> [M<sup>+</sup> + 1] 345.1126, found 345.1188.

### (Z/E)-Methyl 2-(4-(4-bromophenyl)-7-hydroxy-2*H*-chromen-2-ylidene)acetate (15c)

A mixture of methyl 6-(4-bromophenyl)-4-(methylthio)-2-oxo-2H-pyran-3-carboxylate (14c, 354 mg, 1.0 mmol, 1 equiv.), resorcinol (13a, 220 mg, 2.0 mmol, 2 equiv.) and NaH (60% dispersion in oil, 200 mg, 5.0 mmol, 5 equiv.) in dry DMF (5 ml) was stirred at a temperature of 60 °C for 4 h. Using the general procedure described above, 257 mg (69%) of 15c was obtained as a mixture of cis- and trans-isomers, as a yellow solid:  $R_f = 0.56$  (ethyl acetate-*n*-hexane, 1:9, v/v); mp 284–252 °C (ethyl acetate–*n*-hexane); IR (KBr)  $\nu_{\text{max}}/\text{cm}^{-1}$  = 3201 (w), 2857 (m), 1647 (m), 1562 (w); <sup>1</sup>H NMR (300 MHz, DMSO $d_{6+}$ CDCl<sub>3</sub>):  $\delta$  = 3.59 (s, 1H, *cis*), 3.60 (s, 1H, *trans*), 5.15 (s, 1H, trans), 5.32 (s, 1H, cis), 6.41 (s, 1H, trans), 6.60-6.65 (m, 2H, cis + trans), 7.05-7.10 (m, 1H, cis + trans), 7.37-7.46 (m, 2H, cis + trans), 7.51 (s, 1H, cis), 7.70-7.75 (m, 2H, cis + trans), 10.44 (s, 1H, OH, D<sub>2</sub>O exchange); <sup>13</sup>C NMR (75 MHz, DMSO-d<sub>6</sub>):  $\delta$  = 50.9, 79.6, 91.2, 102.9, 111.8, 112.7, 113.7, 123.0, 127.5, 130.1, 132.3, 135.9, 143.5, 154.2, 161.4, 162.8, 167.4; HRMS (ESI) calcd for  $C_{18}H_{13}BrO_4 [M^+ + 1] 373.0075$ , found 373.0066.

# (*Z*/*E*)-Methyl 2-(6-chloro-7-hydroxy-4-phenyl-2*H*-chromen-2-ylidene)acetate (15d)

A mixture of methyl 4-(methylthio)-2-oxo-6-phenyl-2H-pyran-3carboxylate (14a, 276 mg, 1.0 mmol, 1 equiv.), 4-chloro-resorcinol (13b, 286 mg, 2.0 mmol, 2 equiv.) and NaH (60% dispersion in oil, 200 mg, 5.0 mmol, 5 equiv.) in dry DMF (15 ml) was stirred at a temperature of 60 °C for 3 h. Using the general procedure described above, 230 mg (70%) of 15d was obtained as a mixture of *cis*- and *trans*-isomers, as a yellow solid:  $R_{\rm f}$  = 0.58 (ethyl acetate-n-hexane, 1:9, v/v); mp 234-236 °C (ethyl acetate–*n*-hexane); IR (KBr)  $\nu_{\text{max}}/\text{cm}^{-1}$  = 3600 (s), 1650 (m); <sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>):  $\delta$  = 3.59 (s, 3H, *cis*), 3.60 (s, 3H, trans), 5.20 (s, 1H, trans), 5.36 (s, 1H, cis), 6.49 (s, 1H, trans), 6.84 (s, 2H, cis + trans), 7.09 (s, 1H, trans), 7.11 (s, 1H, cis), 7.42–7.57 (m, 12H, *cis* + *trans*), 11.30 (brs, 2H, *cis* + *trans*) ppm; <sup>13</sup>C NMR (75.5 MHz, DMSO-d<sub>6</sub>):  $\delta$  = 51.1, 91.9, 104.1, 113.2, 114.6, 116.4, 126.7, 128.7, 129.4, 129.6, 129.8, 129.5, 136.0, 143.4, 152.5, 156.4, 162.4, 167.3; HRMS (ESI) calcd for  $C_{18}H_{13}ClO_4 [M^+ + 1] 329.0581$ , found 329.0574.

# (Z/E)-Methyl 2-(4-(4-chlorophenyl)-7-hydroxy-8-methyl-2*H*-chromen-2-ylidene)acetate (15e)

A mixture of methyl 6-(4-chlorophenyl)-4-(methylthio)-2-oxo-2H-pyran-3-carboxylate (14d, 310 mg, 1.0 mmol, 1 equiv.), 2-methyl-resorcinol (13c, 248 mg, 2.0 mmol, 2 equiv.) and NaH (60% dispersion in oil, 200 mg, 5.0 mmol, 5 equiv.) in dry DMF (5 ml) was stirred at a temperature of 60 °C for 3 h. Using the general procedure described above, 205 mg (60%) of 15e was obtained as a mixture of cis- and trans-isomers, as a yellow solid:  $R_f = 0.56$  (ethyl acetate-*n*-hexane, 1:9, v/v); mp 226–228 °C (ethyl acetate–*n*-hexane); IR (KBr)  $\nu_{\text{max}}/\text{cm}^{-1}$  = 3458 (w), 1728 (s), 1670 (s), 1586 (s); <sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>):  $\delta = 2.14$  (s, 3H, *cis*), 2.26 (s, 3H, *trans*), 3.59 (s, 3H, *cis*), 3.62 (s, 3H, trans), 5.14 (s, 1H, trans), 5.37 (s, 1H, cis), 6.40 (s, 1H, trans), 6.69 (d, J = 8.2 Hz, 2H, cis + trans), 6.92 (d, J = 8.4 Hz, 2H, cis + trans), 7.45-7.63 (m, 10H, cis + trans), 10.25 (s, 1H, trans), 10.35 (s, 1H, cis) ppm; <sup>13</sup>C NMR (75.5 MHz, DMSO-d<sub>6</sub>):  $\delta$  = 7.9, 50.4, 90.4, 111.0, 111.4, 112.8, 123.8, 128.8, 130.2, 133.8, 135.2, 143.5, 151.5, 158.7, 162.5, 167.0; HRMS (ESI) calcd for  $C_{19}H_{15}ClO_4 [M^+ + 1] 343.0737$ , found 343.0729.

# (Z/E)-Methyl 2-(7-hydroxy-4-(thiophen-2-yl)-2H-chromen-2-ylidene)acetate (15f)

A mixture of methyl 4-(methylthio)-2-oxo-6-(thiophen-2-yl)-2Hpyran-3-carboxylate (14e, 282 mg, 1.0 mmol, 1 equiv.), resorcinol (13a, 220 mg, 2.0 mmol, 2 equiv.) and NaH (60% dispersion in oil, 200 mg, 5.0 mmol, 5 equiv.) in dry DMF (15 ml) was stirred at a temperature of 60 °C for 4 h. Using the general procedure described above, 186 mg (62%) of 15f was obtained as a mixture of *cis*- and *trans*-isomers, as a yellow solid:  $R_{\rm f}$  = 0.58 (ethyl acetate-n-hexane, 1:9, v/v); mp 234-236 °C (ethyl acetate–*n*-hexane); IR (KBr)  $\nu_{\text{max}}/\text{cm}^{-1} = 3600$  (s), 1650 (m); <sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>):  $\delta$  = 3.62 (s, 3H, *cis* + *trans*), 5.20 (s, 1H, trans), 5.31 (s, 1H, cis), 6.57 (s, 1H, trans), 6.63-6.75 (m, 2H, cis + trans), 7.24-7.31 (m, 1H, cis + trans), 7.44-7.62 (m, 2H, cis + trans), 7.70-7.83 (m, 2H, cis + trans), 10.50 (s, 1H, OH,  $D_2O$  exchange); <sup>13</sup>C NMR (75.5 MHz, DMSO-d<sub>6</sub>):  $\delta$  = 51.0, 91.2, 103.1, 111.3, 112.8, 113.7, 127.5, 128.6, 128.8, 128.8, 129.4, 137.1, 137.5, 154.3, 161.5, 162.5, 167.4; HRMS (ESI) calcd for  $C_{16}H_{12}O_4S[M^+ + 1]$  301.0535, found 301.0528.

### General procedure for the synthesis of 16a-f

(Z/E)-Methyl 2-(7-hydroxy-2*H*-chromen-2-ylidene)-acetates (**15a-f**) were treated with 2.0 mmol of DDQ in dioxane at 80 °C for 1–2 h. After completion, the reaction mixture was filtered and the filtrate was collected and evaporated to dryness. Thereafter, the crude obtained was purified through neutral alumina using 10% acetone–*n*-hexane as the eluent to afford **16a–f**.

### 7-Hydroxy-4-phenyl-2H-chromen-2-one (16a)

The methyl 2-(7-hydroxy-4-phenyl-2*H*-chromen-2-ylidene)acetate (**15a**, 147 mg, 0.5 mmol, 1 equiv.) was treated with DDQ (227 mg, 1.0 mmol, 2 equiv.) in dioxane (5 ml) at 80  $^{\circ}$ C for 2 h. Using the general procedure described above, 66 mg (54%) of **16a** was obtained as a yellow solid:  $R_{\rm f}$  = 0.56 (acetone*n*-hexane, 1:10, v/v); mp 234–236 °C {lit.<sup>15</sup> 210–211 °C} (acetone-*n*-hexane); IR (KBr)  $\nu_{\rm max}/{\rm cm}^{-1}$  = 3102 (m), 1695 (s), 1595 (s); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub> + DMSO-d<sub>6</sub>)  $\delta$  = 6.03 (s, 1H), 6.65–6.79 (m, 2H), 7.24 (d, *J* = 13.0 Hz, 1H), 7.36–7.52 (m, 5H), 10.27 (s, 1H); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub> + DMSO-d<sub>6</sub>)  $\delta$  = 102.6, 110.0, 110.7, 112.9, 127.7, 127.9, 128.4, 129.1, 135.1, 155.5, 155.6, 160.5, 161.3; HRMS (ESI) calcd for C<sub>15</sub>H<sub>10</sub>O<sub>3</sub> [M<sup>+</sup> + 1] 239.0708, found: 239.0706.

#### 7-Hydroxy-4-(naphthalen-2-yl)-2H-chromen-2-one (16b)

The methyl 2-(7-hydroxy-4-(naphthalen-2-yl)-2H-chromen-2-ylidene)acetate (15b, 172 mg, 0.5 mmol, 1 equiv.) was treated with DDQ (227 mg, 1.0 mmol, 2 equiv.) in dioxane (5 ml) at 80 °C for 2 h. Using the general procedure described above, 80 mg (55%) of **16b** was obtained as a yellow solid:  $R_{\rm f} = 0.48$ (acetone-n-hexane, 1:10, v/v); mp 214-216 °C (acetone-nhexane); IR (KBr)  $\nu_{\text{max}}/\text{cm}^{-1}$  = 3402 (m), 1629 (s); <sup>1</sup>H NMR  $(300 \text{ MHz}, \text{CDCl}_3 + \text{DMSO-d}_6) \delta = 6.20 \text{ (s, 1H)}, 6.75 \text{ (dd, } J = 8.6,$ 2.2 Hz, 1H), 6.86 (d, J = 2.2 Hz, 1H), 7.34 (d, J = 8.7 Hz, 1H), 7.51-7.62 (m, 3H), 7.94-8.03 (m, 4H), 10.18 (s, 1H); <sup>13</sup>C NMR  $(75 \text{ MHz}, \text{ CDCl}_3 + \text{DMSO-d}_6) \delta = 102.7, 110.3, 110.9, 112.9,$ 125.2, 126.4, 126.7, 127.3, 127.4, 127.7, 127.8, 128.0, 132.4, 132.7, 132.9, 155.5, 155.6, 160.7, 161.2; HRMS (ESI) calcd for  $C_{19}H_{12}O_3[M^+ + 1]$  289.0865, found: 289.0859.

#### 4-(4-Bromophenyl)-7-hydroxy-2H-chromen-2-one (16c)

The methyl 2-(4-(4-bromophenyl)-7-hydroxy-2*H*-chromen-2ylidene)acetate (**15c**, 185 mg, 0.5 mmol, 1 equiv.) was treated with DDQ (227 mg, 1.0 mmol, 2 equiv.) in refluxing dioxane (5 ml) at 90 °C for 1 h. Using the general procedure described above, 84 mg (53%) of **16c** was obtained as a yellow solid:  $R_{\rm f}$  = 0.46 (acetone–*n*-hexane, 1 : 10, v/v); mp 246–248 °C (acetone–*n*hexane); IR (KBr)  $\nu_{\rm max}$ /cm<sup>-1</sup> = 3231 (m), 1705 (s), 1621 (s); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub> + DMSO-d<sub>6</sub>)  $\delta$  = 6.18 (s, 1H), 6.74–6.82 (m, 2H), 7.25 (d, *J* = 8.6 Hz, 1H), 7.47 (d, *J* = 8.4 Hz, 2H), 7.76 (d, *J* = 8.3 Hz, 2H); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub> + DMSO-d<sub>6</sub>)  $\delta$  = 103.2, 110.9, 111.0, 113.8, 123.6, 128.5, 131.0, 132.3, 134.8, 154.7, 156.0, 160.5, 162.0; HRMS (ESI) calcd for C<sub>15</sub>H<sub>9</sub>BrO<sub>3</sub> [M<sup>+</sup> + 1] 316.9813, found: 316.9802.

### 6-Chloro-7-hydroxy-4-phenyl-2H-chromen-2-one (16d)

The methyl 2-(6-chloro-7-hydroxy-4-phenyl-2*H*-chromen-2ylidene)acetate (**15d**, 164 mg, 0.5 mmol, 1 equiv.) was treated with DDQ (227 mg, 1.0 mmol, 2 equiv.) in dioxane (5 ml) at 80 °C for 1 h. Using the general procedure described above, 71 mg (52%) of **16d** was obtained as a yellow solid:  $R_{\rm f} = 0.48$ (acetone–*n*-hexane, 1 : 10, v/v); mp 226–228 °C {Llit.<sup>19</sup> 281 °C (ethyl acetate–heptane)} (acetone–*n*-hexane); IR (KBr)  $\nu_{\rm max}/$ cm<sup>-1</sup> = 3201 (w), 1691 (s); <sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>)  $\delta$  = 6.24 (s, 1H), 7.00 (s, 1H), 7.28 (s, 1H), 7.48–7.60 (m, 5H), 11.52 (brs, 1H, OH, D<sub>2</sub>O exchange); <sup>13</sup>C NMR (75 MHz, DMSO-d<sub>6</sub>)  $\delta$  = 104.4, 111.9, 112.0, 117.5, 127.4, 128.8, 129.4, 130.3, 135.0, 154.2, 154.8, 157.0, 160.1; HRMS (ESI) calcd for C<sub>15</sub>H<sub>9</sub>ClO<sub>3</sub> [M<sup>+</sup> + 1] 273.0318; found: 273.0311.

### 4-(4-Chlorophenyl)-7-hydroxy-8-methyl-2*H*-chromen-2-one (16e)

The methyl 2-(4-(4-chlorophenyl)-7-hydroxy-8-methyl-2*H*chromen-2-ylidene)acetate (**15e**, 171 mg, 0.5 mmol, 1 equiv.) was treated with DDQ (227 mg, 1.0 mmol, 2 equiv.) in dioxane (5 ml) at 80 °C for 1 h. Using the general procedure described above, 60 mg (42%) of **16e** was obtained as a yellow solid:  $R_{\rm f}$  = 0.48 (acetone–*n*-hexane, 1 : 10, v/v); mp 248–250 °C (acetone–*n*hexane); IR (KBr)  $\nu_{\rm max}/\rm cm^{-1}$  = 3398 (w), 1682 (s); <sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>)  $\delta$  = 2.20 (s, 3H), 6.16 (s, 1H), 6.84 (d, *J* = 8.6 Hz, 1H), 7.09 (d, *J* = 8.4 Hz, 1H), 7.48–7.64 (m, 4H), 10.58 (s, 1H); <sup>13</sup>C NMR (75.5 MHz, DMSO-d<sub>6</sub>)  $\delta$  = 8.0, 110.0, 110.5, 111.4, 112.0, 124.7, 128.8, 130.3, 134.0, 134.3, 153.3, 154.8, 159.2, 160.4; HRMS (ESI) calcd for C<sub>16</sub>H<sub>11</sub>ClO<sub>3</sub> [M<sup>+</sup> + 1] 287.0475; found: 287.0477.

### 7-Hydroxy-4-(thiophen-2-yl)-2H-chromen-2-one (16f)

The methyl 2-(7-hydroxy-4-(thiophen-2-yl)-2*H*-chromen-2ylidene)acetate (**15f**, 150 mg, 0.5 mmol, 1 equiv.) was treated with DDQ (227 mg, 1.0 mmol, 2 equiv.) in refluxing dioxane (5 ml) at 90 °C for 2 h. Using the general procedure described above, 110 mg (45%) of **16f** was obtained as a yellow solid:  $R_f =$ 0.63 (acetone–*n*-hexane, 1 : 10, v/v); mp 236–238 °C (acetone–*n*hexane); IR (KBr)  $\nu_{max}/cm^{-1} = 3413$  (s), 1730 (w); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub> + DMSO-d<sub>6</sub>)  $\delta = 6.27$  (s, 1H), 6.87 (d, *J* = 7.2 Hz, 2H), 7.28–7.35 (m, 1H), 7.53 (d, *J* = 2.3 Hz, 1H), 7.69–7.74 (m, 1H), 7.81 (d, *J* = 9.3 Hz, 1H), 7.98 (s, 1H), 10.40 (s, 1H, OH, D<sub>2</sub>O exchange); <sup>13</sup>C NMR (CDCl<sub>3</sub> + DMSO-d<sub>6</sub>)  $\delta =$ 103.2, 110.3, 113.4, 127.8, 128.2, 128.7, 129.4, 136.2, 148.2, 155.9, 160.5, 161.9; HRMS (ESI) calcd for C<sub>13</sub>H<sub>8</sub>O<sub>3</sub>S [M<sup>+</sup> + 1] 245.0272; found: 245.0266.

### Methyl 3-amino-6,7-dichloro-4-cyano-5-hydroxybenzofuran-2-carboxylate (17)

The methyl 2-(7-hydroxy-4-phenyl-2*H*-chromen-2-ylidene) acetate (**15a**, 147 mg, 0.5 mmol, 1 equiv.) was treated with DDQ (227 mg, 1.0 mmol, 2 equiv.) in dioxane (5 ml) at 80 °C for 2 h. Using the general procedure described above, 47 mg (31%) of **17** was obtained as an off-white solid:  $R_{\rm f}$  = 0.52 (MeOH–CHCl<sub>3</sub>, 1:4, v/v); mp > 250 °C (CHCl<sub>3</sub>–MeOH); IR (KBr)  $\nu_{\rm max}/{\rm cm}^{-1}$  = 3374 (m), 3206 (w), 2229 (m), 1684 (s); <sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>)  $\delta$  = 3.87 (s, 3H), 5.65 (s, 1H) ppm; <sup>13</sup>C NMR (50.3 MHz, DMSO-d<sub>6</sub>)  $\delta$  = 51.5, 90.7, 101.9, 113.8, 119.6, 128.9, 137.3, 150.8, 154.0 ppm; *m*/*z* (ESI): 301 [M<sup>+</sup> + H]; HRMS (ESI) calcd for C<sub>11</sub>H<sub>6</sub>Cl<sub>2</sub>N<sub>2</sub>O<sub>4</sub> [M<sup>+</sup> + 1] 300.9783, found 300.9760.

#### General procedure for the synthesis of 18a-c

A mixture of the methyl 4-(methylthio)-2-oxo-6-aryl-2*H*-pyran-3-carboxylates **14** (1 mmol), the respective 1-tetralone (1.2 mmol) and KOH (2.0 mmol) in dry DMF (10 ml) was stirred at 90 °C for 12–24 h. Completion of the reaction was monitored using TLC and on completion the reaction mixture was poured into crushed ice with vigorous stirring and finally neutralized with 10% HCl. The precipitate obtained was filtered and purified on a silica gel column using 20%  $CHCl_3$ in *n*-hexane as the eluent to afford **18a–c** as a mixture of *cis*and *trans*-isomers.

#### (*Z*/*E*)-Methyl 2-(4-(4-bromophenyl)-8-methoxy-5,6-dihydro-2*H*benzo[*h*]chromen-2-ylidene)acetate (18a)

A mixture of methyl 6-(4-bromophenyl)-4-(methylthio)-2-oxo-2H-pyran-3-carboxylate (14c, 353 mg, 1.0 mmol, 1 equiv.), 6-methoxy 1-tetralone (6c, 211 mg, 1.2 mmol, 1.2 equiv.) and KOH (112 mg, 2.0 mmol, 2 equiv.) in dry DMF (10 ml) was stirred at 90 °C for 12 h. Using the general procedure described above, 153 mg (35%) of 18a was obtained as a mixture of *cis*- and *trans*-isomers, as a red oil:  $R_{\rm f} = 0.54$ (CHCl<sub>3</sub>-*n*-hexane, 7:3, v/v); IR (KBr)  $\nu_{max}/cm^{-1} = 2952$  (m), 2861 (w), 2337 (m), 1605 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta =$ 2.42-2.51 (m, 4H, cis + trans), 2.71-2.80 (m, 4H, cis + trans), 3.68 (s, 3H, cis), 3.74 (s, 3H, trans), 3.84 (s, 6H, cis + trans), 4.93 (s, 3H, trans), 5.31 (s, 1H, cis), 6.71 (s, 2H, cis + trans), 6.78-6.92 (m, 1H, cis + trans), 7.12-7.24 (m, 2H, cis + trans), 7.51-7.68 (m, 4H, *cis* + *trans*) ppm; <sup>13</sup>C NMR (75.5 MHz,  $CDCl_3$ :  $\delta$  = 26.6, 28.1, 51.8, 55.3, 87.3, 112.1, 112.1, 113.5, 116.2, 124.3, 127.2, 129.5, 129.7, 130.2, 130.6, 131.4, 131.6, 140.0, 144.8, 146.4, 157.6, 159.3, 168.4 ppm; m/z (ESI): 439  $[M + H^{+}]$ ; HRMS (ESI) calcd for C<sub>23</sub>H<sub>19</sub>BrO<sub>4</sub>  $[M^{+} + 1]$  439.0545, found 439.0507.

# (*Z*/*E*)-Methyl 2-(4-(furan-2-yl)-5,6-dihydro-2*H*-benzo[*h*] chromen-2-ylidene)acetate (18b)

A mixture of methyl 6-(furan-2-yl)-4-(methylthio)-2-oxo-2Hpyran-3-carboxylate (14f, 266 mg, 1.0 mmol, 1 equiv.), 1-tetralone (6a, 0.146 ml, 1.2 mmol, 1.2 equiv.) and KOH (112 mg, 2.0 mmol, 2 equiv.) in dry DMF (10 ml) was stirred at 90 °C for 15 h. Using the general procedure described above, 102 mg (32%) of 18b was obtained as a mixture of cis- and transisomers, as a red oil:  $R_f = 0.52$  (CHCl<sub>3</sub>-*n*-hexane, 7:3, v/v); IR (KBr)  $\nu_{\text{max}}/\text{cm}^{-1} = 2932$  (m), 2854 (w), 1734 (m), 1632 (w); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 2.89 (s, 8H, *cis* + *trans*), 3.72 (s, 3H, trans), 3.75 (s, 3H, cis), 5.01 (s, 1H, trans), 5.32 (s, 1H, cis), 6.48-6.61 (m, 2H, *cis* + *trans*), 6.68 (d, J = 3.4 Hz, 1H, *trans*), 6.74 (d, J = 3.4 Hz, 1H, cis), 7.14-7.38 (m, 6H, cis + trans), 7.54 (d, J = 1.3 Hz, 2H, cis + trans), 7.64-7.71 (m, 1H, cis), 8.13 (d, J = 5.1 Hz, 2H, *cis* + *trans*) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>):  $\delta$ = 23.0, 27.4, 50.5, 87.8, 109.9, 111.9, 112.4, 113.8, 122.4, 126.9, 127.3, 128.5, 129.5, 134.4, 136.7, 143.8, 150.1, 151.1, 164.4, 168.4 ppm; m/z (ESI): 321 [M + H<sup>+</sup>]; HRMS (ESI) calcd for  $C_{20}H_{16}O_4 [M^+ + 1] 321.1127$ , found 321.1119.

### (Z/E)-Methyl 2-(4-phenyl-5,6-dihydro-2*H*-benzo[*h*]chromen-2-ylidene)acetate (18c)

A mixture of methyl 4-(methylthio)-2-oxo-6-phenyl-2*H*-pyran-3carboxylate (**14a**, 276 mg, 1.0 mmol, 1 equiv.), 1-tetralone (**6a**, 0.146 ml, 1.2 mmol, 1.2 equiv.) and KOH (112 mg, 2.0 mmol, 2 equiv.) in dry DMF (10 ml) was stirred at 90 °C for 24 h. Using the general procedure described above, 113 mg (34%) of **18c** was obtained as a mixture of *cis*- and *trans*-isomers, as a red oil:  $R_{\rm f} = 0.57$  (CHCl<sub>3</sub>-*n*-hexane, 7 : 3, v/v); IR (KBr)  $\nu_{\rm max}/{\rm cm}^{-1} =$  2926 (s), 2857 (m), 1647 (m), 1562 (w); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 2.47–2.58 (m, 4H, *cis* + *trans*), 2.75–2.84 (m, 2H, *cis* + *trans*), 3.68 (s, 3H, *cis*), 3.76 (s, 3H, *trans*), 4.96 (s, 1H, *trans*), 5.34 (s, 1H, *cis*), 7.14–7.44 (m, 8H, *cis* + *trans*), 7.70–7.78 (m, 1H, *cis* + *trans*) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>):  $\delta$  = 23.1, 27.6, 50.5, 87.5, 112.0, 117.0, 122.5, 126.9, 127.5, 127.8, 128.0, 128.4, 128.5, 128.6, 129.6, 137.1, 137.2, 147.2, 151.0, 165.3, 168.9 ppm; *m*/*z* (ESI): 331 [M + H<sup>+</sup>]; HRMS (ESI) calcd for C<sub>22</sub>H<sub>18</sub>O<sub>3</sub> [M<sup>+</sup> + 1] 331.1334, found 331.1372.

#### General procedure for the synthesis of 19a-c

(Z/E)-Methyl 2-(4-aryl-5,6-dihydro-2*H*-benzo[*h*]chromen-2ylidene)acetates (**18a–c**) were treated with 2.0 mmol of DDQ in AcOH at room temp for 1–2 h. After completion, the reaction mixture was filtered and the filtrate was collected and evaporated to dryness. Thereafter, the crude obtained was purified through a silica gel column using 40% CHCl<sub>3</sub> in hexane as the eluent to afford **19a–c**.

#### 4-(4-Bromophenyl)-8-methoxy-5,6-dihydro-2*H*-benzo[*h*]chromen-2-one (19a)

The (*Z*/*E*)-methyl 2-(4-(4-bromophenyl)-8-methoxy-5,6-dihydro-2*H*-benzo[*h*]chromen-2-ylidene)-acetate (**18a**, 219 mg, 0.5 mmol, 1 equiv.) was treated with DDQ (227 mg, 1.0 mmol, 2 equiv.) in glacial acetic acid (5 ml) for 2 h. Using the general procedure described above, 99 mg (52%) of **19a** was obtained as a yellow solid:  $R_{\rm f} = 0.70$  (CHCl<sub>3</sub>–*n*-hexane, 7 : 3, v/v); mp 176–178 °C (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{\rm max}/{\rm cm}^{-1} = 2925$  (w), 2856 (w), 1712 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta = 2.51$ –2.60 (m, 2H), 2.76–2.85 (m, 2H), 3.85 (s, 3H), 6.09 (s, 1H), 6.73 (s, 1H), 6.85 (d, *J* = 8.7 Hz, 1H), 7.14–7.24 (m, 2H), 7.56–7.63 (m, 2H) 7.85 (d, *J* = 8.6 Hz, 1H) ppm; <sup>13</sup>C NMR (50.3 MHz, CDCl<sub>3</sub>)  $\delta = 23.0$ , 27.9, 55.4, 109.3, 111.2, 112.2, 113.6, 121.1, 123.5, 125.6, 129.3, 131.9, 135.4, 139.6, 155.9, 157.5, 161.5 ppm; *m*/z (ESI): 383 [M<sup>+</sup> + H]; HRMS (ESI) calcd for C<sub>20</sub>H<sub>15</sub>BrO<sub>3</sub> [M<sup>+</sup> + 1] 383.0283, found 383.0250.

#### 4-(Furan-2-yl)-5,6-dihydro-2*H*-benzo[*h*]chromen-2-one (19b)

The (*Z/E*)-methyl 2-(4-(furan-2-yl)-5,6-dihydro-2*H*-benzo[*h*]chromen-2-ylidene)acetate (**18b**, 160 mg, 0.5 mmol, 1 equiv.) was treated with DDQ (227 mg, 1.0 mmol, 2 equiv.) in glacial acetic acid (5 ml) for 2 h. Using the general procedure described above, 54 mg (41%) of **19b** was obtained as a yellow solid:  $R_{\rm f} = 0.39$  (CHCl<sub>3</sub>–*n*-hexane, 7 : 3, v/v); mp 144–146 °C (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{\rm max}$ /cm<sup>-1</sup> = 2924 (m), 1630 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  = 2.94–3.00 (m, 4H), 6.57 (s, 2H), 6.86 (d, *J* = 3.0 Hz, 1H), 7.21–7.35 (m, 3H), 7.60–7.64 (m, 1H), 7.90 (d, *J* = 4.6 Hz, 1H) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta$  = 23.1, 27.3, 108.3, 109.6, 112.3, 114.3, 123.7, 127.1, 127.5, 128.3, 130.3, 136.9, 144.9, 148.9, 155.4, 162.0 ppm; *m/z* (ESI): 265 [M<sup>+</sup> + H]; HRMS (ESI) calcd for C<sub>17</sub>H<sub>12</sub>O<sub>3</sub> [M<sup>+</sup> + 1] 265.0865, found 265.0868.

#### 4-Phenyl-5,6-dihydro-2H-benzo[h]chromen-2-one (19c)

The(Z/E)-methyl 2-(4-phenyl-5,6-dihydro-2H-benzo[h]chromen-2-ylidene)acetate (**18c**, 165 mg, 0.5 mmol, 1 equiv.) was treated

with DDQ (227 mg, 1.0 mmol, 2 equiv.) in glacial acetic acid (5 ml) for 1 h. Using the general procedure described above, 65 mg (48%) of **19c** was obtained as a yellow solid:  $R_{\rm f} = 0.50$  (CHCl<sub>3</sub>–*n*-hexane, 7 : 3, v/v); mp 154–156 °C (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{\rm max}/{\rm cm}^{-1} = 2922$  (m), 2858 (w), 1750 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta = 2.56-2.65$  (m, 2H), 2.79–2.88 (m, 2H), 6.20 (s, 1H), 7.16–7.49 (m, 8H), 7.88–7.94 (m, 1H) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta = 23.0$ , 27.6, 111.7, 112.8, 123.7, 127.2, 127.7, 128.3, 128.7, 129.2, 130.3, 136.4, 137.3, 155.1, 158.6, 161.9 ppm; *m*/*z* (ESI): 275 [M<sup>+</sup> + H]; HRMS (ESI) calcd for C<sub>19</sub>H<sub>14</sub>O<sub>2</sub> [M<sup>+</sup> + 1] 275.1072, found 275.1067.

# 4-(4-Bromophenyl)-8-methoxy-2*H*-benzo[*h*]chromen-2-one (20a)

4-(4-Bromophenyl)-8-methoxy-5,6-dihydro-2H-benzo[h]chromen-2-one (19a, 76 mg, 0.2 mmol, 1 equiv.) was refluxed with 6 mmol of DDQ (272 mg, 1.2 mmol, 6 equiv.) in benzene (5 ml) at 90 °C for 8 h. After completion, the reaction mixture was filtered and the filtrate was collected and evaporated to dryness. The crude 20a obtained was purified by column chromatography on a silica gel using 30% CHCl<sub>3</sub> in n-hexane as the eluent to afford 40 mg (52%) of 4-(4-bromophenyl)-8-methoxy-2H-benzo[h]chromen-2-one (20a) was obtained as an off-white solid:  $R_f = 0.53$  (CHCl<sub>3</sub>-*n*-hexane, 7:3, v/v); mp 194–196 °C (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{max}/cm^{-1}$  = 3013 (w), 1718 (w), 1634 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  = 3.96 (s, 3H), 6.36 (s, 1H), 7.15 (d, J = 2.2 Hz, 1H), 7.27–7.33 (m, 1H), 7.34–7.41 (m, 3H), 7.51 (d, J = 8.8 Hz, 1H), 7.69 (d, J = 8.4 Hz, 1H), 8.51 (d, *J* = 9.2 Hz, 1H) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta = 55.4, 106.2, 112.2, 113.2, 118.2, 119.6, 122.7, 123.0, 124.0,$ 124.5, 130.1, 132.1, 134.6, 136.8, 151.8, 152.7, 155.5, 160.2, 160.7 ppm; m/z (ESI): 381 [M<sup>+</sup> + H]; HRMS (ESI) calcd for  $C_{20}H_{13}BrO_3 [M^+ + 1] 381.0126$ , found 381.0112.

#### General procedure for the synthesis of 23a-f

A mixture of the 4-(methylthio)-2-oxo-6-aryl-2*H*-pyran-3-carbonitriles **21** (1 mmol), substituted acetophenones **22** (1.2 mmol) and KOH (1.5 mmol) in dry DMF (15 ml) were stirred at 90 °C for 24 h. Completion of the reaction was monitored using TLC and on completion the reaction mixture was poured into crushed ice with vigorous stirring and finally neutralized with 10% HCl. The precipitate obtained was filtered and purified on a silica gel column using 30% CHCl<sub>3</sub> in *n*-hexane as the eluent to afford **23a–f** as a mixture of *cis-* and *trans*-isomers.

# (*Z/E*)-2-(6-(Naphthalen-1-yl)-4-phenyl-2*H*-pyran-2-ylidene)-acetonitrile (23a)

A mixture of 4-(methylthio)-2-oxo-6-phenyl-2*H*-pyran-3-carbonitrile (**21a**, 243 mg, 1.0 mmol, 1 equiv.), 1-(naphthalen-1-yl) ethanone (**22a**, 170 mg, 1.0 mmol, 1 equiv.) and KOH (84 mg, 1.5 mmol, 1.5 equiv.) in dry DMF (15 ml) was stirred at 90 °C for 24 h. Using the general procedure described above, 122 mg (38%) of **23a** was obtained as a mixture of *cis*- and *trans*isomers, as a red oil:  $R_{\rm f} = 0.59$  (CHCl<sub>3</sub>–*n*-hexane, 7:3, v/v); IR (KBr)  $\nu_{\rm max}/{\rm cm}^{-1} = 2926$  (w), 2215 (s), 1649 (s); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 4.39$  (s, 1H, *trans*), 4.61 (s, 1H, *cis*), 6.49 (s, 1H, *cis*), 6.52 (s, 1H, *trans*), 6.56 (s, 1H, *trans*), 7.05 (s, 1H, *cis*), 7.45–7.68 (m, 18H, *cis* + *trans*), 7.90–8.01 (m, 4H, *cis* + *trans*), 8.15 (d, J = 7.6 Hz, 1H, *trans*), 8.29 (d, J = 8.2 Hz, 1H, *trans*) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>):  $\delta = 66.4$ , 67.2, 105.9, 106.0, 112.1, 112.9, 113.2, 117.8, 119.1, 124.7, 124.7, 125.0, 125.9, 126.1, 126.4, 126.4, 126.5, 127.2, 127.4, 128.7, 129.1, 129.9, 130.0, 130.1, 130.2, 130.3, 130.4, 131.0, 133.7, 133.8, 135.5, 135.6, 142.9, 143.7, 158.0, 158.2, 166.3, 168.0 ppm; *m/z* (ESI): 322.2 [M + H<sup>+</sup>]; HRMS (ESI) calcd for C<sub>23</sub>H<sub>15</sub>NO [M<sup>+</sup> + 1] 322.1232, found 322.1237.

# (*Z/E*)-2-(4,6-Bis(4-bromophenyl)-2*H*-pyran-2-ylidene) acetonitrile (23b)

A mixture of 6-(4-bromophenyl)-4-(methylthio)-2-oxo-2H-pyran-3-carbonitrile (21b, 320 mg, 1.0 mmol, 1 equiv.), 1-(4-bromophenyl)ethanone (22b, 199 mg, 1.0 mmol, 1 equiv.) and KOH (84 mg, 1.5 mmol, 1.5 equiv.) in dry DMF (15 ml) was stirred at 90 °C for 24 h. Using the general procedure described above, 204 mg (48%) of 23b was obtained as a mixture of cis- and trans-isomers, as a red oil:  $R_f = 0.64$  (CHCl<sub>3</sub>-n-hexane, 7:3, v/v); IR (KBr)  $\nu_{\text{max}}/\text{cm}^{-1} = 2925$  (w), 2872 (w), 2195 (m), 1649 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 4.38$  (s, 1H, *cis*), 4.67 (s, 1H, trans), 6.48 (s, 1H, cis), 6.54 (s, 1H, trans), 6.58 (s, 1H, cis), 6.93 (s, 1H, trans), 7.41-7.63 (m, 14H, cis + trans), 7.73 (d, J = 8.6 Hz, 2H, *cis*) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>):  $\delta$  = 67.2, 68.1, 100.6, 100.9, 112.7, 113.7, 117.7, 118.8, 125.3, 126.6, 127.2, 127.4, 127.5, 127.6, 129.4, 129.8, 132.2, 132.2, 132.4, 134.0, 134.2, 141.9, 142.4, 155.9, 156.2, 165.5, 167.1 ppm; *m*/*z* (ESI): 427  $[M^+ + H]$ ; HRMS (ESI) calcd for  $C_{19}H_{11}Br_2NO$  $[M^+ + 1]$  427.9286, found 427.9328.

#### (Z/E)-2-(4,6-Di-p-tolyl-2H-pyran-2-ylidene)acetonitrile (23c)

A mixture of 4-(methylthio)-2-oxo-6-(p-tolyl)-2H-pyran-3-carbonitrile (21c, 257 mg, 1.0 mmol, 1 equiv.), 1-(4-methylphenyl) ethanone (22c, 0.133 ml, 1.0 mmol, 1 equiv.) and KOH (84 mg, 1.5 mmol, 1.5 equiv.) in dry DMF (15 ml) was stirred at 90 °C for 24 h. Using the general procedure described above, 111 mg (37%) of 23c was obtained as a mixture of cis- and transisomers, as a red oil:  $R_f = 0.64$  (CHCl<sub>3</sub>-*n*-hexane, 7:3, v/v); IR (KBr)  $\nu_{\text{max}}/\text{cm}^{-1} = 2923$  (m), 2856 (w), 2193 (m), 1646 (w); <sup>1</sup>H NMR (300 MHz,  $CDCl_3$ ):  $\delta = 2.40$  (s, 6H), 4.29 (s, 1H, *cis*), 4.57 (s, 1H, trans), 6.43 (s, 1H, cis), 6.58 (s, 1H, trans), 6.90 (s, 1H, *cis*), 6.91 (s, 1H, *trans*), 7.20–7.29 (m, 10H, *cis* + *trans*), 7.42-7.54 (m, 4H, cis + trans), 7.62 (d, J = 8.1 Hz, 2H, cis), 7.76 (d, J = 8.1 Hz, 2H, *trans*) ppm;  ${}^{13}$ C NMR (75.5 MHz, CDCl<sub>3</sub>):  $\delta$  = 21.3, 21.3, 21.4, 21.5, 65.3, 66.1, 100.0, 100.3, 111.2, 112.1, 118.5, 119.6, 125.1, 125.1, 125.9, 126.1, 128.4, 128.7, 129.5, 129.6, 129.6, 129.8, 132.9, 133.1, 140.4, 140.5, 141.1, 141.1, 143.3, 143.9, 156.8, 157.1, 166.4, 168.1 ppm; m/z (ESI): 300  $[M^{+} + H]$ ; HRMS (ESI) calcd for C<sub>21</sub>H<sub>17</sub>NO  $[M^{+} + 1]$  300.1388, found 300.1377.

# (*Z/E*)-2-(4,6-Di(naphthalen-2-yl)-2*H*-pyran-2-ylidene)acetonitrile (23d)

A mixture of 4-(methylthio)-6-(naphthalen-2-yl)-2-oxo-2*H*-pyran-3-carbonitrile (**21d**, 293 mg, 1.0 mmol, 1 equiv.),

1-(naphthalen-2-yl)ethanone (22d, 170 mg, 1.0 mmol, 1 equiv.) and KOH (84 mg, 1.5 mmol, 1.5 equiv.) in dry DMF (15 ml) was stirred at 90 °C for 24 h. Using the general procedure described above, 149 mg (40%) of 23d was obtained as a mixture of *cis*- and *trans*-isomers, as a red oil:  $R_f = 0.51$ (CHCl<sub>3</sub>-*n*-hexane, 7:3, v/v); IR (KBr)  $\nu_{\text{max}}/\text{cm}^{-1}$  = 2928 (w), 2860 (w), 2194 (s), 1647 (m); <sup>1</sup>H NMR (300 MHz,  $CDCl_3$ ):  $\delta =$ 4.36 (s, 1H, cis), 4.70 (s, 1H, trans), 6.55 (s, 1H, cis), 6.85 (s, 2H, *cis* + *trans*), 7.06 (s, 1H, *trans*), 7.48–7.65 (m, 10H, *cis* + *trans*), 7.79-8.10 (m, 16H, cis + trans), 8.23 (s, 1H, trans), 8.39 (s, 1H, *cis*) ppm; <sup>13</sup>C NMR (50.3 MHz,  $CDCl_3$ ):  $\delta$  = 66.1, 67.0, 100.8, 101.0, 112.2, 113.2, 118.3, 119.3, 121.6, 121.7, 122.9, 123.0, 123.6, 124.2, 125.0, 125.3, 125.7, 125.8, 126.7, 126.9, 127.2, 127.4, 127.5, 127.7, 128.0, 128.2, 128.5, 128.5, 128.6, 128.7, 128.9, 129.0, 132.5, 132.7, 132.8, 133.1, 133.8, 133.9, 134.0, 142.7, 143.3, 156.3, 156.5, 165.9, 167.6 ppm; m/z (ESI): 372  $[M^{+} + H]$ ; HRMS (ESI) calcd for C<sub>27</sub>H<sub>17</sub>NO  $[M^{+} + 1]$  372.1388, found 372.1387.

# (Z/E)-2-(4,6-Di(thiophen-2-yl)-2H-pyran-2-ylidene)acetonitrile (23e)

A mixture of 4-(methylthio)-2-oxo-6-(thiophen-2-yl)-2H-pyran-3-carbonitrile (21e, 248 mg, 1.0 mmol, 1 equiv.), 1-(thiophen-2-yl)ethanone (22e, 0.107 ml, 1.0 mmol, 1 equiv.) and KOH (84 mg, 1.5 mmol, 1.5 equiv.) in dry DMF (15 ml) was stirred at 90 °C for 24 h. Using the general procedure described above, 110 mg (39%) of 23e was obtained as a mixture of cisand *trans*-isomers, as a red oil:  $R_{\rm f} = 0.48$  (CHCl<sub>3</sub>-*n*-hexane, 7:3, v/v); IR (KBr)  $\nu_{\text{max}}/\text{cm}^{-1}$  = 2922 (m), 2859 (w), 2191 (w), 1629 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 4.31 (s, 1H, *cis*), 4.58 (s, 1H, trans), 6.39–6.44 (m, 3H, cis + trans), 6.85 (d, J = 1.3 Hz, 1H, trans), 7.07-7.16 (m, 4H, cis + trans), 7.38-7.48 (m, 6H, cis + trans), 7.64 (d, J = 2.8 Hz, 2H, trans) ppm; <sup>13</sup>C NMR (50.3 MHz,  $CDCl_3$ ):  $\delta$  = 66.2, 67.0, 98.7, 98.9, 109.0, 110.0, 126.6, 127.4, 128.3, 128.5, 128.6, 134.6, 139.1, 152.3, 165.1 ppm; m/z (ESI): 284 [M<sup>+</sup> + H]; HRMS (ESI) calcd for  $C_{15}H_9NOS_2[M^+ + 1]$  284.0204, found 284.0208.

# 6-(4-Methoxyphenyl)-5-methyl-4-morpholino-2-oxo-2*H*-pyran-3-carbonitrile (I)

In order to synthesize **23f**, the methylsulfanyl group of 6-(4methoxyphenyl)-5-methyl-4-(methylthio)-2-oxo-2*H*-pyran-3carbonitrile **21f**<sup>20</sup> was replaced by morpholine by reacting lactone **21f** (287 mg, 1.0 mmol) with morpholine (0.1 ml, 1.2 mmol) in methanol (10 ml) at reflux temperature for 8 h to give the precursor 6-(4-methoxyphenyl)-5-methyl-4-morpholino-2-oxo-2*H*-pyran-3-carbonitrile (I) in a quantitative yield of 310 mg (95%).

Yellow solid:  $R_{\rm f} = 0.56$  (MeOH–CHCl<sub>3</sub>, 1:25 v/v); mp 164–166 °C (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{\rm max}$ /cm<sup>-1</sup> = 3024 (w), 2215 (w), 1703 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 2.10 (s, 3H), 3.58–3.75 (m, 4H), 3.82–3.94 (m, 7H), 6.95 (d, *J* = 8.8 Hz, 2H), 7.53 (d, *J* = 8.8 Hz, 2H); <sup>13</sup>C NMR (50.3 MHz, CDCl<sub>3</sub>):  $\delta$  = 17.7, 51.8, 55.5, 67.0, 80.8, 106.6, 113.9, 116.5, 124.0, 131.5, 161.9, 162.2, 168.1 ppm; *m*/*z* (ESI): 327 [M<sup>+</sup> + H]; HRMS (ESI) calcd for C<sub>18</sub>H<sub>18</sub>N<sub>2</sub>O<sub>4</sub> [M<sup>+</sup> + 1] 327.1345, found 327.1344.

# 2-(6-Isopropyl-4-(4-methoxyphenyl)-3-methyl-5-phenyl-2*H*-pyran-2-ylidene)acetonitrile (23f)

A mixture of 6-(4-methoxyphenyl)-5-methyl-4-morpholino-2-oxo-2H-pyran-3-carbonitrile (I, 326 mg, 1.0 mmol, 1 equiv.), 3-methyl-1-phenyl-butan-2-one (22f, 0.162 ml, 1.0 mmol, 1 equiv.) and powdered KOH (84 mg, 1.5 mmol, 1.5 equiv.) in dry DMF (5 ml) was stirred at 90 °C for 24 h. Using the general procedure described above, 175 mg (49%) of 23f was obtained as a yellow solid:  $R_f = 0.49$  (CHCl<sub>3</sub>-*n*-hexane, 7:3, v/v); mp 183–185 °C (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{max}/cm^{-1} = 3022$  (w), 2930 (w), 2199 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  = 1.21 (d, J = 6.8 Hz, 6H), 1.67 (s, 3H), 2.55-2.65 (m, 1H), 3.70 (s, 3H), 4.24 (s, 1H), 6.61-6.76 (m, 4H), 6.82-6.91 (m, 2H), 7.08-7.16 (m, 3H) ppm; <sup>13</sup>C NMR (50.3 MHz, CDCl<sub>3</sub>)  $\delta$  = 14.8, 18.4, 20.2, 30.2, 55.0, 62.9, 113.2, 116.9, 119.9, 126.9, 127.9, 128.9, 130.0, 130.3, 135.3, 158.6, 160.1, 167.4 ppm; *m*/*z* (ESI): 358 [M<sup>+</sup> + H]; HRMS (ESI) calcd forC<sub>24</sub>H<sub>23</sub>NO<sub>2</sub> [M<sup>+</sup> + 1] 358.1807, found 358.1800.

#### General procedure for the synthesis of 24a-f

The functionalized 2*H*-pyran-2-ylidene-acetonitriles (23a-f) were treated with 2.0 mmol of DDQ in AcOH at room temp for 5–6 h. After completion, the reaction mixture was filtered and the filtrate was collected and evaporated to dryness. Thereafter, the crude obtained was purified through column chromatography on silica gel using 50% CHCl<sub>3</sub> in hexane as the eluent to afford highly functionalized 2*H*-pyran-2-ones **24a–f**.

#### 6-(Naphthalen-1-yl)-4-phenyl-2H-pyran-2-one (24a)

The (Z/E)-2-(6-(naphthalen-1-yl)-4-phenyl-2*H*-pyran-2-ylidene)acetonitrile (**23a**, 161 mg, 0.5 mmol, 1 equiv.) was treated with DDQ (227 mg, 1.0 mmol, 2 equiv.) in glacial acetic acid (5 ml) for 6 h. Using the general procedure described above, 70 mg (47%) of **24a** was obtained as a yellow solid:  $R_{\rm f} = 0.40$  (CHCl<sub>3</sub>*n*-hexane, 7 : 3, v/v); mp 126–128 °C (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{\rm max}/{\rm cm}^{-1} = 2926$  (m), 1715 (s), 1627 (m), 1538 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta = 6.58$  (d, J = 1.3 Hz, 1H), 6.85 (d, J =1.3 Hz, 1H), 7.48–7.79 (m, 9H), 7.88–8.01 (m, 2H), 8.22–8.29 (m, 1H) ppm; <sup>13</sup>C NMR (50.3 MHz, CDCl<sub>3</sub>)  $\delta = 106.5$ , 109.3, 124.9, 125.0, 126.5, 126.8, 127.4, 127.8, 128.7, 129.3, 130.4, 130.8, 131.2, 133.8, 135.8, 155.4, 161.8, 163.0 ppm; *m/z* (ESI): 299 [M<sup>+</sup> + H]; HRMS (ESI) calcd for C<sub>21</sub>H<sub>14</sub>O<sub>2</sub> [M<sup>+</sup> + 1] 299.1072, found 299.1078.

#### 4,6-Bis(4-bromophenyl)-2H-pyran-2-one (24b)

The (*Z*/*E*)-2-(4,6-bis(4-bromophenyl)-2*H*-pyran-2-ylidene)acetonitrile (23b, 213 mg, 0.5 mmol, 1 equiv.) was treated with DDQ (227 mg, 1.0 mmol, 2 equiv.) in glacial acetic acid (5 ml) for 5 h. Using the general procedure described above, 83 mg (41%) of **24b** was obtained as a yellow solid:  $R_{\rm f} = 0.48$  (CHCl<sub>3</sub>*n*-hexane, 7 : 3, v/v); mp 216–218 °C (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{\rm max}/{\rm cm}^{-1} = 2920$  (w), 1691 (s), 1535 (w); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta = 6.45$  (s, 1H), 6.88 (s, 1H), 7.45–7.68 (m, 6H), 7.75 (d, *J* = 8.6 Hz, 2H) ppm; <sup>13</sup>C NMR (50.3 MHz, CDCl<sub>3</sub>)  $\delta = 101.1$ , 109.7, 125.7, 127.2, 128.0, 128.2, 129.6, 130.3, 132.3, 132.6, 134.3, 137.2, 154.2, 159.6, 162.0 ppm; HRMS (ESI) calcd for  $C_{17}H_{10}Br_2O_2 [M^+ + 1]$  404.9126, found 404.9313.

#### 4,6-Di-p-tolyl-2H-pyran-2-one (24c)

The (*Z*/*E*)-2-(4,6-di-*p*-tolyl-2*H*-pyran-2-ylidene)acetonitrile (23c, 150 mg, 0.5 mmol, 1 equiv.) was treated with DDQ (227 mg, 1.0 mmol, 2 equiv.) in glacial acetic acid (5 ml) for 5 h. Using the general procedure described above, 65 mg (47%) of 24c was obtained as a yellow solid:  $R_{\rm f} = 0.38$  (CHCl<sub>3</sub>-*n*-hexane, 7 : 3, v/v); mp 163–165 °C (CHCl<sub>3</sub>-*n*-hexane); IR (KBr)  $\nu_{\rm max}/{\rm cm}^{-1} = 2928$  (w), 1696 (m), 1630 (m), 1512 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta = 2.41$  (s, 3H), 2.43 (s, 3H), 6.43 (s, 1H), 6.92 (s, 1H), 7.22–7.35 (m, 4H), 7.55 (d, *J* = 8.1 Hz, 1H), 7.79 (d, *J* = 8.1 Hz, 1H) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta = 21.4$ , 21.5, 100.6, 108.1, 125.7, 126.6, 128.9, 129.7, 129.9, 133.2, 141.1, 141.4, 155.5, 160.5, 163.0 ppm; *m*/z (ESI): 277 [M<sup>+</sup> + H]; HRMS (ESI) calcd for C<sub>19</sub>H<sub>16</sub>O<sub>2</sub> [M<sup>+</sup> + 1] 277.1229, found 277.1234.

#### 4,6-Di(naphthalen-2-yl)-2H-pyran-2-one (24d)

The (Z/E)-2-(4,6-di(naphthalen-2-yl)-2*H*-pyran-2-ylidene)acetonitrile (**23d**, 186 mg, 0.5 mmol, 1 equiv.) was treated with DDQ (227 mg, 1.0 mmol, 2 equiv.) in glacial acetic acid (5 ml) for 6 h. Using the general procedure described above, 75 mg (43%) of **24d** was obtained as a yellow solid:  $R_{\rm f} = 0.36$  (CHCl<sub>3</sub>*n*-hexane, 7 : 3, v/v); mp 166–168 °C (CHCl<sub>3</sub>–*n*-hexane); IR (KBr)  $\nu_{\rm max}$ /cm<sup>-1</sup> = 2920 (m), 2853 (m), 1708 (m), 1531 (w); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  = 6.64 (d, *J* = 1.2 Hz, 1H), 7.24 (d, *J* = 1.2 Hz, 1H), 7.54–7.63 (m, 4H), 7.72–7.80 (m, 1H), 7.84–8.03 (m, 7H), 8.20 (s, 1H), 8.53 (s, 1H) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta$  = 101.7, 107.9, 108.2, 116.0, 122.6, 129.9, 129.7, 130.1, 132.2, 138.2, 138.5, 139.7, 141.0, 152.5, 153.2, 153.3, 158.8, 159.5, 169.2 ppm.

#### 4,6-Di(thiophen-2-yl)-2H-pyran-2-one (24e)

The (Z/E)-2-(4,6-di(thiophen-2-yl)-2*H*-pyran-2-ylidene)acetonitrile (**23e**, 142 mg, 0.5 mmol, 1 equiv.) was treated with DDQ (227 mg, 1.0 mmol, 2 equiv.) in glacial acetic acid (5 ml) for 6 h. Using the general procedure described above, 50 mg (38%) of **24e** was obtained as a yellow solid:  $R_{\rm f} = 0.32$  (CHCl<sub>3</sub>*n*-hexane, 7 : 3, v/v); mp 158–160 °C {lit.<sup>21</sup> 158 °C} (CHCl<sub>3</sub>-*n*hexane); IR (KBr)  $\nu_{\rm max}/{\rm cm}^{-1} = 2913$  (m), 2139 (w), 1705 (m), 1651 (m); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta = 6.39$  (d, J = 1.3 Hz, 1H), 6.76 (d, J = 1.3 Hz, 1H), 7.11–7.22 (m, 2H), 7.48 (d, J =5.0 Hz, 1H), 7.51–7.59 (m, 2H), 7.64–7.69 (m, 1H) ppm; <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta = 99.1$ , 105.5, 127.7, 128.0, 128.4, 128.7, 129.1, 129.8, 135.1, 138.8, 148.1, 156.0, 161.8 ppm; *m/z* (ESI): 261 [M<sup>+</sup> + H]; HRMS (ESI) calcd for C<sub>13</sub>H<sub>8</sub>O<sub>2</sub>S<sub>2</sub> [M<sup>+</sup> + 1] 261.0044, found 261.0046.

#### 6-Isopropyl-4-(4-methoxyphenyl)-3-methyl-5-phenyl-2*H*-pyran-2-one (24f)

The (*Z*/*E*)-2-(6-isopropyl-4-(4-methoxyphenyl)-3-methyl-5-phenyl-2*H*-pyran-2-ylidene)acetonitrile (**23f**, 179 mg, 0.5 mmol, 1 equiv.) was treated with DDQ (227 mg, 1.0 mmol, 2 equiv.) in glacial acetic acid (5 ml) for 5 h. Using the general procedure described above, 65 mg (39%) of **24f** as a yellow solid:  $R_{\rm f} = 0.40$  (CHCl<sub>3</sub>-*n*-hexane, 7 : 3, v/v); mp 172–174 °C (CHCl<sub>3</sub>-*n*-hexane); IR (KBr)  $\nu_{max}$ /cm<sup>-1</sup> = 2974 (w), 2928 (w), 1707 (s); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  = 1.19 (d, *J* = 6.8 Hz, 6H), 1.92 (s, 3H), 2.60–2.72 (m, 1H), 3.71 (s, 3H), 6.65–6.92 (m, 4H), 7.83–7.92 (m, 2H), 7.11–7.17 (m, 3H) ppm; <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>)  $\delta$  = 14.5, 20.3, 30.4, 30.9, 55.2, 113.1, 113.9, 117.9, 118.8, 127.0, 127.8, 128.0, 128.3, 128.9, 129.6, 130.4, 130.7, 137.4, 141.3, 152.2, 158.7 ppm; *m*/*z* (ESI): 335 [M<sup>+</sup> + H]; HRMS (ESI) calcd for C<sub>22</sub>H<sub>22</sub>O<sub>3</sub> [M<sup>+</sup> + 1] 335.1647, found 335.1642.

### Acknowledgements

We gratefully acknowledge financial support by the CSIR under Network Projects THUNDER (BSC0102) and HOPE (BSC0114), New-Delhi, India. The spectroscopic data provided by SAIF, CDRI is gratefully acknowledged. The CSIR-CDRI communication number for the manuscript is 8477.

### Notes and references

- (a) R. D. H. Murray, J. Mendez and S. A. Brown, in *The Natural Coumarins: Occurrence, Chemistry, and Biochemistry,* Wiley, New York, 1982; (b) A. Goel and V. J. Ram, *Tetrahe dron*, 2009, **65**, 7865–7913.
- 2 D. J. Hadjipavlou-Litina, K. E. Litinas and D. N. Nicolaides, *Curr. Pharm. Des.*, 2004, **10**, 3813–3833.
- 3 (a) S. R. Trenor, A. R. Shultz, B. J. Love and T. E. Long, *Chem. Rev.*, 2004, **104**, 3059–3077; (b) J. Gordo, J. Avo,
  A. J. Parola, J. C. Lima, A. Pereira and P. S. Branco, *Org. Lett.*, 2011, **13**, 5112–5115.
- 4 (a) P. N. P. Rao, M. J. Uddin and E. E. Knaus, J. Med. Chem., 2004, 47, 3972–3990; (b) M. Riveiro, K. N. De, A. Moglioni, R. Vázquez, F. Monczor, C. Shayo and C. Davio, Curr. Med. Chem., 2010, 17, 1325–1328; (c) S. B. Combes, P. Barbier, S. Douillard, A. McLeer-Florin, V. R. Bourgarel-Rey, J.-T. Pierson, A. Y. Fedorov, J.-P. Finet, J. Boutonnat and V. Peyrot, J. Med. Chem., 2011, 54, 3153–3162.
- 5 (a) R. K. Dieter and J. R. Fishpaugh, J. Org. Chem., 1983, 48, 4439-4441; (b) G. Jones, Org. React., 1967, 15, 204-599;
  (c) G. Brufola, F. Fringuelli, O. Piermatti and F. Pizzo, Heterocycles, 1996, 43, 1257-1266; (d) J. Smodiš and B. Stanovnik, Tetrahedron, 1998, 54, 9799-9810.
- 6 (a) T. Luo and S. L. Schreiber, *Angew. Chem., Int. Ed.*, 2007,
  46, 8250–8253; (b) L. Ackermann, J. Pospech, K. Graczyk and K. Rauch, *Org. Lett.*, 2012, 14, 930–933.
- 7 (a) A. K. Chatterjee, F. D. Toste, S. D. Goldberg and R. H. Grubbs, *Pure Appl. Chem.*, 2003, 75, 421–425;
  (b) B. M. Trost, F. D. Toste and K. Greenman, *J. Am. Chem. Soc.*, 2003, 125, 4518–4526; (c) K. Nakai, T. Kurahashi and S. Matsubara, *J. Am. Chem. Soc.*, 2011, 133, 11066–11068;
  (d) S. Kim, D. Kang, C.-H. Lee and P. Ho. Lee, *J. Org. Chem.*, 2012, 77, 6530–6537; (e) D. K. Rayabarapu, P. Shukla and C. H. Cheng, *Org. Lett.*, 2003, 5, 4903–4906; (f) T. N. Van,

S. Debenedetti and N. D. Kimpe, *Tetrahedron Lett.*, 2003, 44, 4199–4201.

- 8 (a) A. Goel, V. Kumar, S. P. Singh, A. Sharma, S. Prakash, C. Singh and R. S. Anand, J. Mater. Chem., 2012, 22, 14880– 14888; (b) A. Goel, V. Kumar, P. Nag, V. Bajpai, B. Kumar, C. Singh, S. Prakash and R. S. Anand, J. Org. Chem., 2011, 76, 7474–7481; (c) A. Goel, V. Kumar, S. Chaurasia, M. Rawat, R. Prasad and R. S. Anand, J. Org. Chem., 2010, 75, 3656–3662; (d) A. Goel, S. P. Singh, A. Kumar, R. Kant and P. R. Maulik, Org. Lett., 2009, 11, 5122–5125; (e) A. Goel, F. V. Singh, V. Kumar, M. Reichert, T. A. M. Gulder and G. Bringmann, J. Org. Chem., 2007, 72, 7765–7768.
- 9 A. Goel, D. Verma, R. Pratap, G. Taneja, Y. Hemberger, M. Knauer, R. Raghunandan, P. R. Maulik, V. J. Ram and G. Bringmann, *Eur. J. Org. Chem.*, 2011, 2940–2947.
- 10 (a) P. V. P. Pragnacharyulu and E. Abushanab, *Tetrahedron Lett.*, 1997, 38, 3683–3686; (b) L. Minuti, A. Taticchi, A. Marrocchi, D. Lanari, E. Gacs-Baitzb and A. Gomoryb, *Tetrahedron Lett.*, 2005, 46, 949–950.
- 11 (a) G. Bringmann, B. Schoener, O. Schupp, K. Peters, E.-M. Peters and H. G. van Schnering, *Liebigs Ann. Chem.*, 1994, 91–97; (b) G. Bringmann, T. Gulder, T. A. M. Gulder and M. Breuning, *Chem. Rev.*, 2011, 111, 563–639.
- 12 (a) A. Bhattacharya, L. M. DiMichele, U.-H. Dolling,
  E. J. J. Grabowski and V. J. Grenda, *J. Org. Chem.*, 1989, 54, 6118–6120; (b) K. Tanemura, T. Suzuki and T. Horaguchi,

Bull. Chem. Soc. Jpn., 1993, 66, 1235–1238;
(c) U. Kucklaender, R. Bollig, W. Frank, A. Gratz and J. Jose, Bioorg. Med. Chem., 2011, 19, 2666–2674; (d) Ji-Y. Shin, B. O. Patrick and D. Dolphin, Org. Biomol. Chem., 2009, 7, 2032–2035.

- 13 M. M. Garazd, Y. L. Garazd and V. P. Khilya, *Chem. Nat. Compd.*, 2003, **39**, 54–121.
- 14 (a) A. Goel, M. Dixit, S. Chaurasia, A. Kumar, R. Raghunandan, P. R. Maulik and R. S. Anand, Org. Lett., 2008, 10, 2553–2556; (b) A. Goel, S. Chaurasia, M. Dixit, V. Kumar, S. Prakash, B. Jena, J. K. Verma, M. Jain, R. S. Anand and S. S. Manoharan, Org. Lett., 2009, 11, 1289–1292; (c) Y. Tominaga, Trends Heterocycl. Chem., 1991, 2, 43–83.
- 15 Y. Li, Z. Qi, H. Wang, X. Fu and C. Duan, *J. Org. Chem.*, 2012, 77, 2053–2057.
- 16 V. J. Ram and A. Goel, J. Chem. Res. (S), 1997, 460-461.
- 17 R. Pratap and V. J. Ram, *Tetrahedron Lett.*, 2007, **48**, 2755–2759.
- 18 (a) S. Oi, K. Kawagoe and S. Miyano, *Chem. Lett.*, 1993, 79–80; (b) G. Bringmann, M. Heubes, M. Breuning, L. Göbel, M. Ochse, B. Schfner and O. Schupp, *J. Org. Chem.*, 2000, 65, 722–728.
- 19 L. L. Woods and J. Sapp, *J. Org. Chem.*, 1962, 27, 3703-3704.
- 20 F. V. Singh, V. Kumar, B. Kumar and A. Goel, *Tetrahedron*, 2007, **63**, 10971–10978.
- 21 K. Sorensen, Acta Chem. Scand., 1970, 24, 343-345.