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First Synthesis of 4-Substituted Benzenesulfonylcyanoguanidines

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Abstract: N'-substituted-N-cyano-S-methylcarbamimidothioates react with 4-substitutedbenzene sulfonamides to give the corresponding sulfonylcyanoguanidines. Copyright © 1996 Elsevier Science Ltd

The cyanoguanidine group is regarded as a non-classical bioisostere of the urea and thiourea function¹. This concept has been developed for the gastric antisecretory N'-alkyl-N-imidazolylalkylthioureas to give cimetidine². Conversly, N'-alkyl-N-substituted pyridyl-thioureas and -ureas were found to be as active as pinacidil, an antihypertensive cyanoguanidine^{3,4}. In the present study, we have applied this strategy to hypoglycemic sulfonylureas to obtain original drugs bearing a new moiety: the sulfonylcyanoguanidine function.



Scheme 1 Reagents : i, Y-NH₂ (1.5 equiv), ethanol, 25°C; ii, NaOH (1 equiv), DMF, dioxane, reflux.

The N"[4-(substitutedphenyl)sulfonyl]-N'-substituted-N"cyanoguanidines **5-20** (table 1) were prepared by refluxing an excess (1.2 equiv) of N'-substituted-N-cyano-S-methylcarbamimidothioate **2** with the sodium salt of 4-methylbenzenesulfonamide 3^5 or the 4-substitutedcarboxamidoethylbenzenesulfonamide 4^6 (scheme 1). The carbamimidothioates **2** were synthesized by stirring a mixture of N-cyano-S,S'-dimethylimino carbonate 1^7 in ethanol with the required amine (Y-NH₂, 1.5 equiv).

The compounds 7, 8, 12 and 13 are the bioisosteres of tolbutamide, tolcyclamide, tolazamide and gliclazide, the main representatives of the hypoglycemic drugs of first chemical generation $(X = CH_3)^8$. The antidiabetic sulfonylureas bearing a substituted carboxamidoethyl side-chain belong to the second chemical generation. Compound 16 is the bioisostere of glibenclamide, one of the most potent hypoglycemic drug⁸. The electron-withdrawing groups (SO₂ and =N-CN) should maintain the acidity of the moiety as compared to the sulfonylurea function. The ¹H-NMR spectra⁹ of 7 ($\delta = 5.6$, 2H, br s) and 16 ($\delta = 4.7$, 2H, br s) revealed that both protons of the cyanoguanidine moiety were not distinguishable and located with the residual water contained in DMSO-d₆. On the contrary, tolbutamide ($\delta = 10.41$, 1H, br s; 6.30, 1H, br t) and glibenclamide ($\delta = 10.27$, 1H, br s; 6.25 1H, br d), their sulfonylurea counterparts, clearly exhibited three different signals corresponding to the two urea protons and residual water. These data suggest tautomeric forms of the sulfonylurea bioisosteres. The chemical shift values lead to the proposal that 7 is probably more acidic than 16, as it is

reported for tolbutamide (pKa = 5.3)¹⁰ compared to glibenclamide (pKa = 6.8)¹¹. All compounds were characterized by analytical and spectral methods⁹.

	X - SO ₂ NH-C-NH-Y			
N°	х	Y	Yield (%)	Mp (°C) ^a
5	CH ₃	С ₂ н ₅	17	131-133
6	CH ₃	(CH ₃) ₂ CH	72	132-134
7	CH ₃	CH ₃ (CH ₂) ₃	28	108-110
8	CH ₃	cyclohexyl	79	156-158
9	CH ₃	cycloheptyl	78	144-146
10	CH ₃	(±)4-CH ₃ cyclohexyl	52	163-165
11	CH ₃	(CH ₂) ₅ N	14	210-212
12	CH ₃	(CH ₂) ₆ N	22	193-195
13	CH ₃	1-azabicyclo[3,2,1]octane	19	186-188
14	C ₆ H ₅ CONHCH ₂ CH ₂	cyclohexyl	76	174-176
15	4-CH3OC6H4CONHCH2CH2	cyclohexyl	26	176-178
16	2-CH ₃ O,5-CIC ₆ H ₃ CONHCH ₂ CH ₂	cyclohexyl	50	92-95
17	2-CH ₃ O,5-CIC ₆ H ₃ CONHCH ₂ CH ₂	(±)4-CH ₃ cyclohexyl	64	85-87
18	2-furfurylCONHCH2CH2	cyclohexyl	55	146-148
19	C6H5NHCONHCH2CH2	cyclohexyl	37	108-110
20	4-CIC ₆ H ₄ NHCONHCH ₂ CH ₂	cyclohexyl	82	128-130

Table I : Synthesis of 4-substitutedbenzenesulfonylcyanoguanidines

^a All compounds were crystallized from ethanol.

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- 9 The elemental analyses for C,H,N,S are within 0.4% of the theorical values and were performed on a Carlo Erba 1108 analyzer. All new compounds gave IR and 1H-NMR spectra in accordance with their proposed structure. 7: 1H-NMR (80 MHz, d6-DMSO) δ 7.65 (2H, d, J=8Hz), 7.30 (2H, d, J=8Hz), 5.60 (2H, m), 3.15 (2H, br s), 2.30 (3H, s), 1.65-1.25 (4H, m), 0.8 (3H, t). IR (KBr) 2199 cm⁻¹ (C=N st). 16: ¹H-NMR (80 MHz, d6-DMSO) δ 8.18 (1H, m), 7.75 (2H, d), 7.58 (1H, d), 7.4 (3H, dd), 7.05 (1H, d), 4.70 (2H, br s), 3.72 (3H, s), 3.25-3.65 (3H, m), 2.95 (2H, m), 1.80-0.95 (10H, m). IR (KBr) 2188 cm⁻¹ (C=N st).
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