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Stereoselective Synthesis of (Z)-1,2-Diarylthio-1-alkene via the Reaction of Diaryl Disulfides with Terminal Alkynes Catalyzed by Cesium Hydroxide

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Abstract: In the presence of a catalytic amount of cesium hydroxide under a nitrogen atmosphere, terminal alkynes reacted with diaryl disulfides at room temperature in dimethylformamide to give almost exclusively (Z)-1,2-diarylthio-1-alkene in good yields, but under an air atmosphere, the reaction gave a mixture of alkynyl sulfide and (Z)-1,2-diarylthio-1-alkene.

Keywords: Cesium hydroxide, diaryl disulfide, (Z)-1,2-diarylthio-1-alkene, terminal alkyne

Many different classes of organochalcogen compounds have been prepared and studied to date. For their syntheses, vinyl sulfides have been used extensively because vinyl sulfides can be conveniently converted to functional alkenes^[1,2] stereoselectively. Among the common vinyl sulfides, 1,2-dithio-1-alkenes are of much importance because of their use as bidentate ligands in coordination chemistry, precursors of new optical materials for microelectronics,^[1] and important intermediates for

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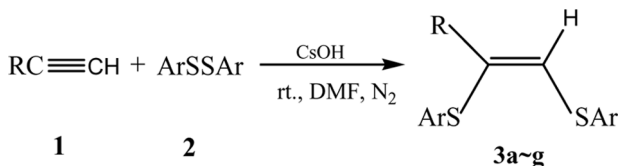
organic synthesis. Traditionally there are two kinds of paths for the preparation of 1,2-dithio-1-alkenes: one involves the addition of disulfides to terminal alkynes^[3–11] and the other is the addition of thiol to alkynyl sulfides.^[12–17] In most cases, both methods need transition-metal catalysis. In this article, we report a stereoselective synthesis of (Z)-1,2-diarylthio-1-alkenes via cesium hydroxide-catalyzed reaction of disulfides with terminal alkynes.

Cesium base has received increasing interest in the past few years. It has been shown recently that cesium hydroxide could catalyze the reaction of terminal alkynes with aldehydes and ketones to give propargyl alcohols, with nitriles to give unsaturated adduct,^[18] and with phenylselenyl bromide to give alkynyl selenides.^[19] Our group also reported that cesium hydroxide-catalyzed addition of diaryldiselenides to terminal alkynes under N₂ gave (Z)-1,2-bis(arylseleno)-1-alkenes^[20] selectively and the reaction of diarylditelluride with terminal alkynes under an air atmosphere gave alkynyl tellurides.^[21] Lothar et al. have also reported that in the presence of K₂CO₃ and a catalytic amount of copper(I), diphenyl disulfide and ditelluride reacted with terminal alkynes to give alkynyl sulfides and tellurides.^[22] These results prompted us to investigate the reaction of diaryl disulfides with terminal alkynes in the presence of a catalytic amount of cesium hydroxide.

We first tested the reactions of 3-methoxy propyne with diphenyl disulfide. The experiment showed that under an air atmosphere, 3-methoxy propyne (2.2 mmol) reacted with diphenyl disulfide (1.0 mmol) in the presence of cesium hydroxide (0.2 mmol) at room temperature in dimethylformamide (DMF) to give a 6:4 mixture of 3-methoxy propynyl phenylsulfide and Z-1,2-diphenylthio-3-methoxy propene in 81% yield. Though the yield is satisfactory, the selectivity is not. However, much to our delight, it gave almost exclusively (Z)-1,2-diphenylthio-3-methoxy propene in 80% yield when the reaction

Table 1. Reaction of terminal alkynes with diaryl disulfides to synthesize (Z)-1,2-diarylthio-1-alkenes

Entry	R	Ar	Yields (%)
3a	HO(CH ₃) ₂ C	Ph	61
3b	HOCH ₂	<i>p</i> -CH ₃ C ₆ H ₄	69
3c	CH ₃ OCH ₂	Ph	82
3d	CH ₃ OCH ₂	<i>p</i> -CH ₃ C ₆ H ₄	77
3e	PhOCH ₂	Ph	75
3f	PhOCH ₂	<i>p</i> -CH ₃ C ₆ H ₄	73
3g	Ph	Ph	71

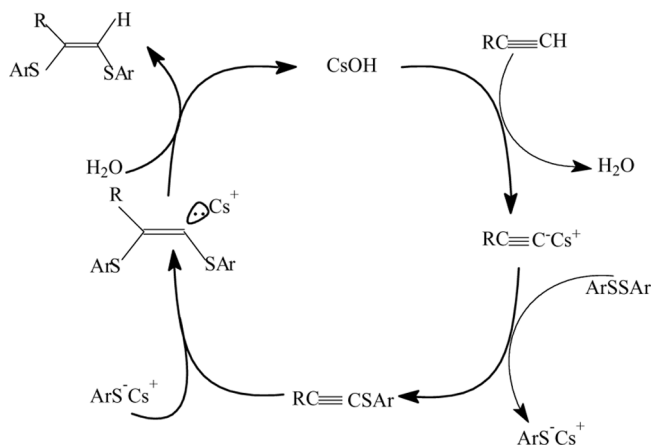


Scheme 1. The synthesis of (Z)-1,2-diarylthio-1-alkenes.

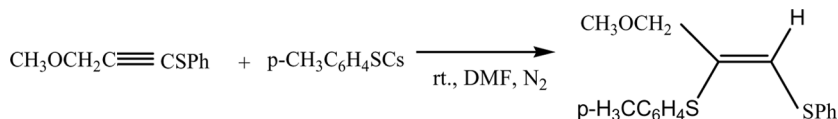
was run under N₂. We reasoned that if the reaction was performed under an air atmosphere, some sulfide ions would be transformed into the starting disulfides because of air oxidation, and the reaction would give a mixture of alkynyl sulfide and (Z)-1,2-diarylthio-1-alkene. Therefore it is important to run the reaction in the absence of air. With the fine-tuned method in hand, we set out to investigate the scope of the reaction. A variety of (Z)-1,2-diarylthio-1-alkenes were synthesized by reacting different alkynes with diphenyldisulfide or di(*p*-methylphenyl)disulfide (Scheme 1), and the results are summarized in Table 1.

A possible reaction mechanism for the reaction of terminal alkynes with diaryl disulfides is depicted in Scheme 2. First, terminal alkynes reacted with cesium hydroxide to give cesium alkynides and H₂O. Second, the alkynides reacted with diaryl disulfides to form alkynyl sulfides and cesium arylsulfides. Finally, the cesium arylsulfides underwent nucleophilic addition to the alkynyl sulfides to give the desired (Z)-1,2-diarylthio-1-alkenes.

To shed some light on the reaction mechanism, the reaction of *p*-CH₃C₆H₄SCs with 3-methoxy propargyl phenylsulfide in DMF at room



Scheme 2. The mechanism for the reaction of terminal alkynes with diaryl disulfides.



Scheme 3. The addition of *p*-CH₃C₆H₄SCS to 3-methoxy propynyl phenylsulfide.

temperature was performed. The fact that 1-phenylthio-2-(*p*-methylphenyl)thio-3-methoxy propene was isolated in almost 100% yield proved that (Z)-1,2-diarylthio-1-alkene was indeed produced through the alkynylsulfide intermediate (Scheme 3).

The analysis of ¹H NMR and ¹³C NMR spectra showed that all vinyl sulfides prepared presented data in full agreement with their assigned structures. The stereochemistry of trisubstituted alkenes were determined by nuclear Overhauser effect (NOE) experiments. For example, when compound **3c** was irradiated at the peak of vinylic proton (6.99 ppm), 6% enhancement of CH₂ (3.94 ppm) was observed, showing a *cis* relation between them.

In conclusion, a mild, efficient, and selective method for the stereoselective synthesis of (Z)-1,2-diarylthio-1-alkenes was developed. Compared with the known methods, our methods use only a catalytic amount of cesium hydroxide and require neither the preparation of alkynyl sulfides nor the use of expensive transition metals. It could provide a new and expedient way for the preparation of (Z)-1,2-diarylthiol-1-alkenes.

EXPERIMENTAL

¹H NMR and ¹³C NMR spectra were recorded on a Bruker Advance 400 spectrometer in CDCl₃ with tetramethylsilane (TMS) as the internal standard, chemical shifts are quoted and *J* values are given in hertz, and elemental analyses were performed on a Yanaco MT-3CHN elemental analyser. Mass spectra were determined on an HP5989A spectrometer. Reactions were monitored by thin-layer chromatography (TLC) using silica-gel F₂₅₄ aluminum sheets.

General Procedure for the Reaction of Terminal Alkyne with Diaryl Disulfide to Synthesize (Z)-1,2-Diarylthio-1-alkene

Diaryl disulfides (1.0 mmol) and cesium hydroxide (0.2 mmol) were added to a solution of alkyne (2.2 mmol) in DMF (5.0 mL). The reaction mixture was stirred at room temperature under a nitrogen atmosphere for

30 h. Then 20 mL of water was added to the residue and extracted with petroleum ether (3×10 ml). The combined organic phase was dried over anhydrous Na_2SO_4 and concentrated under vacuum. The residue was purified by flash chromatography on silica gel, giving **3a–3g**.

Data

Compound **3a**

(*Z*)-1,2-Bis(phenylthio)-3-hydroxy-3-methyl-1-butene: ^1H NMR 1.46 (s, 6H), 2.28 (s, 1H), 7.12–7.42 (m, 11H); ^{13}C NMR 29.3, 75.1, 125.4, 127.0, 127.5, 128.8, 129.0, 130.2, 134.8, 134.9, 135.1, 137.7; Ms (*m/z*): 302 (M^+). Anal. calcd. for $\text{C}_{17}\text{H}_{18}\text{OS}_2$: C, 67.5; H, 6.00; found: C, 67.3; H, 6.01.

Compound **3b**

(*Z*)-1,2-Bis(*p*-methylphenylthio)-3-hydroxy-1-propene: ^1H NMR 2.18 (s, 1H), 2.33 (s, 3H), 2.34 (s, 3H), 4.10 (s, 2H), 6.98 (t, $J = 1.02$ Hz, 1H), 7.10 (d, $J = 8.0$ Hz, 2H), 7.12 (d, $J = 8.0$ Hz, 2H), 7.45 (d, $J = 8.0$ Hz, 2H), 7.47 (d, $J = 8.0$ Hz, 2H); ^{13}C NMR 21.1, 67.4, 124.6, 126.5, 130.1, 130.2, 132.2, 132.9, 133.4, 137.7, 137.9; Ms (*m/z*): 302 (M^+). Anal. calcd. for $\text{C}_{17}\text{H}_{18}\text{OS}_2$: C, 67.5; H, 6.00; found: C, 67.3; H, 6.02.

Compound **3c**

(*Z*)-1,2-Bis(phenylthio)-3-methoxy-1-propene: ^1H NMR 3.30 (s, 3H), 3.94 (s, 2H), 6.99 (s, 1H), 7.21–7.36 (m, 6H), 7.40 (d, $J = 8.0$ Hz, 2H), 7.45 (d, $J = 8.0$ Hz, 2H); ^{13}C NMR 58.0, 74.4, 126.8, 127.1, 127.3, 129.0, 129.1, 130.0, 130.3, 133.3, 134.6, 134.9; Ms (*m/z*): 288 (M^+). Anal. calcd. for $\text{C}_{16}\text{H}_{16}\text{OS}_2$: C, 66.6; H, 5.59; found: C, 66.7; H, 5.58.

Compound **3d**

(*Z*)-1,2-Bis(*p*-methylphenylthio)-3-methoxy-1-propene: ^1H NMR: 2.33 (s, 3H), 2.35 (s, 3H), 3.26 (s, 3H), 3.91 (s, 2H), 6.95 (t, $J = 1.02$ Hz, 1H), 7.10 (d, $J = 8.0$ Hz, 2H), 7.13 (d, $J = 8.4$ Hz, 2H), 7.45 (d, $J = 8.0$ Hz, 2H), 7.47 (d, $J = 8.4$ Hz, 2H); ^{13}C NMR 21.1, 57.8, 76.5, 124.9, 126.8, 129.0, 130.0, 130.1, 132.9, 133.1, 133.3, 137.6, 137.8; Ms (*m/z*): 316 (M^+). Anal. calcd. for $\text{C}_{18}\text{H}_{20}\text{OS}_2$: C, 68.3; H, 6.37; found: C, 68.5; H, 6.38.

Compound **3e**

(*Z*)-1,2-Bis(phenylthio)-3-phenoxy-1-propene: ^1H NMR 4.57 (s, 2H), 6.79 (d, $J=8.0$ Hz, 2H), 6.93 (t, $J=7.2$ Hz, 1H), 7.23 (t, $J=8.0$ Hz, 2H), 7.28–7.29 (m, 6H), 7.47 (s, 1H), 7.49–7.51 (m, 2H), 7.57–7.60 (m, 2H); ^{13}C NMR 72.0, 114.8, 121.1, 127.1, 127.5, 128.5, 129.2, 129.3, 129.3, 130.4, 132.6, 132.7, 134.0, 157.9; Ms (m/z): 350 (M^+). Anal. calcd. for $\text{C}_{21}\text{H}_{18}\text{OS}_2$: C, 72.0; H, 5.18; found: C, 71.9; H, 5.19.

Compound **3f**

(*Z*)-1,2-Bis(*p*-methylphenylthio)-3-phenoxy-1-propene: ^1H NMR 2.33 (s, 6H), 4.53 (s, 2H), 6.78 (d, $J=8.0$ Hz, 2H), 6.92 (t, $J=6.8$ Hz, 1H), 7.09 (d, $J=8.0$ Hz, 4H), 7.20–7.25 (m, 2H), 7.38 (m, 3H), 7.48 (d, $J=6.8$ Hz, 2H); ^{13}C NMR 21.4, 72.3, 115.2, 121.3, 124.9, 127.1, 127.4, 129.6, 130.4, 130.4, 133.3, 133.5, 133.8, 138.0, 138.1, 158.3; Ms (m/z): 378 (M^+). Anal. calcd. for $\text{C}_{23}\text{H}_{22}\text{OS}_2$: C, 72.9; H, 5.86; found: C, 72.8; H, 5.87.

Compound **3g**

(*Z*)-1,2-Bis(phenylthio)styrene: ^1H NMR 7.07 (s, 1H), 7.21–7.34 (m, 9H), 7.42–7.50 (m, 6H); ^{13}C NMR 126.0, 126.5, 127.4, 127.6, 128.2, 128.6, 128.8, 129.2, 129.3, 130.5, 134.5, 135.2, 136.4, 138.7; Ms (m/z): 320 (M^+). Anal. calcd. for $\text{C}_{20}\text{H}_{16}\text{S}_2$: C, 74.9; H, 5.03; found: C, 75.1; H, 5.27.

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