

Intramolecular C-Arylation of Benzylated Sugars. The Unexpected Double C-Glycosideration of Tris-*O*-*m*-methoxybenzyl- β -D-ribofuranose Derivatives

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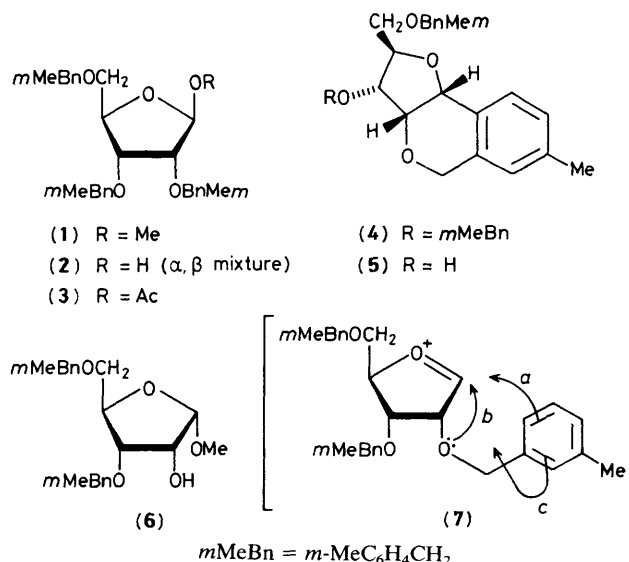
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Upon treatment with tin(IV) chloride, D-ribofuranose derivatives bearing activated *O*-benzyl groups were found to undergo single or double intramolecular C-arylation, with *m*-methyl- and *m*-methoxy-substituted benzyl groups respectively.

We recently reported¹ that benzylated glycofuranosyl acetates readily undergo intramolecular Friedel–Crafts alkylation of the 2-*O*-benzyl group upon treatment with a Lewis acid. In order to explore the synthetic potential of this unusual C-glycosylation reaction, the behaviour of several types of benzylated sugars in the presence of a Lewis acid is being actively studied. We now report some remarkable results obtained from D-ribofuranose derivatives bearing 'activated' benzyl groups.

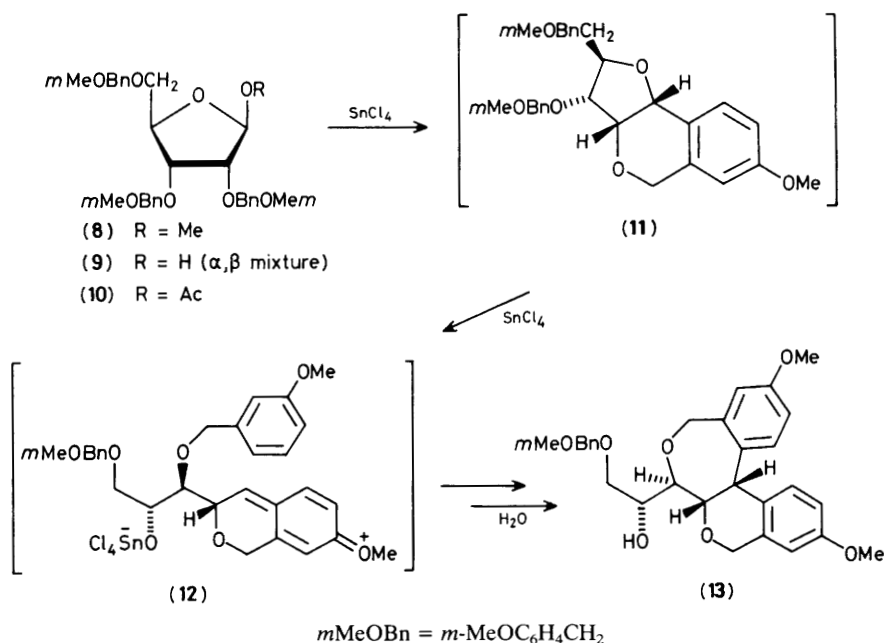
As expected, tris-*O*-*m*-methylbenzyl- β -D-ribofuranosyl acetate (3), prepared from methyl β -D-ribofuranoside by way of (1) and (2) using standard procedures,² gave compound (4)[†] (syrup; $[\alpha]_D^{20} + 70.0^\circ$, c 1.2, CHCl_3) as the major product (59%[‡]) upon treatment with tin(IV) chloride; a small amount (11%) of the 3-*O*-debenzylated analogue of (4) was also isolated [compound (5), m.p. 82.5–84 °C, $[\alpha]_D^{20} 0^\circ$, c 0.7, CHCl_3], as well as a trace (8%) of a compound tentatively identified as the isomer of (4) resulting from the sterically less favourable alkylation of the benzyl group at C-2 [pathway *c* in (7)]. Under the same conditions however, the corresponding methyl glycoside (1) led to a mixture of (4) (30%) and methyl 3,5-bis-*O*-*m*-methylbenzyl- α -D-ribofuranoside (6) (49%) (syrup; $[\alpha]_D^{20} + 108.1^\circ$, c 1.48, CHCl_3), identified on the basis of the $J_{1,2}$ coupling constant (4.0 Hz) and of the change of chemical shift of H-2 upon acetylation. The formation of this unexpected product is attributable to the ambident nucleophilic reactivity of the neighbouring benzyloxy group: reaction at the phenyl group [pathway *a* in (7)] leads indeed irreversibly to C-arylation, whereas participation of the ether oxygen (pathway *b*) gives a benzyloxonium ion from which the debenzilation product (6) probably arises. The factors responsible for the difference of behaviour between the methyl glycoside (1) and the glycosyl acetate (3) are not yet clearly understood.

With the more reactive *m*-methoxybenzyl substituents, both the methyl glycoside (8) and the corresponding glycosyl acetate (10), also prepared by standard procedures,² gave the same result upon treatment with tin(IV) chloride, namely the formation of one homogeneous product which could be crystallized directly from the processed reaction mixture (61%) (m.p. 108–109.5 °C; $[\alpha]_D^{20} + 26.3^\circ$, c 1.0, CHCl_3). Mass spectral and elemental analysis of this compound indicated a composition ($\text{C}_{29}\text{H}_{32}\text{O}_7$) consistent with the expected C-arylated structure (11); however, as shown by its i.r. spectrum, this compound also contains a free hydroxy group, suggesting that a second intramolecular C-arylation might have taken place, thus bringing about the opening of the tetrahydrofuran ring of (11) and the creation of a hydroxy function at C-4. In view of the high reactivity of the *p*-methoxybenzyl ether function present in (11), the Lewis acid-promoted cleavage of the C-1–O-4 bond of (11) to an



[†] Satisfactory microanalysis and spectral data were obtained for all new compounds.

[‡] All yields are after isolation by column chromatography.



oxocarbenium ion-type intermediate (12) is easily understandable; intramolecular alkylation of the nearest *m*-methoxybenzyl substituent then follows, leading to the unusual polyheterocyclic compound (13). The structure of this product was firmly established on the basis of its high-field ^1H n.m.r. spectrum (300 MHz); in particular, the values of the coupling constants in the saturated heterocyclic system indicated that the benzoannulated oxepane ring of (13) adopted a chair-type conformation,³ and that the fusion between the two oxygenated heterocycles was exclusively *cis*. Thus, the second intramolecular *C*-arylation occurred stereospecifically with *retention* of configuration at C-1 (sugar numbering); these findings support the two-steps mechanism proposed for the formation of (13) from (11).

The conversion of (8) [or (10)] into (13) in one stage is a fascinating reaction in which both of the C–O bonds at the anomeric centre of a cyclic sugar derivative are replaced by carbon–carbon bonds: in this respect, it can be considered as a *double C-glycosidation*. Previous examples of dialkylation or diarylation at C-1 are scarce^{4,5} and most of them involved a sugar lactone as starting material. The formation of (13) is the first example of an intramolecular double *C*-glycosidation, which provides a stereospecific and high-yielding route to a novel heterocyclic system from easily accessible *O*-benzylated

glycosides. Also, because of their acetal-like reactivity, *C*-*p*-methoxyphenyl derivatives of type (11) constitute attractive precursors for further double *C*-glycosidation reactions; investigations on compounds of this type are currently under progress and the results will be reported shortly.

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References

- O. R. Martin, *Tetrahedron Lett.*, 1985, **26**, 2055.
- R. Barker and H. G. Fletcher, Jr., *J. Org. Chem.*, 1961, **26**, 4605; T. Iwashige and H. Saeki, *Chem. Pharm. Bull.*, 1967, **15**, 1803; M. Kawana, H. Kuzuhara, and S. Emoto, *Bull. Chem. Soc. Jpn.*, 1981, **54**, 1492.
- D. R. Boyd in 'Comprehensive Heterocyclic Chemistry,' eds. A. R. Katritzky and C. W. Rees, Pergamon Press, Oxford, 1984, vol. 7, pp. 547–592.
- W. A. Bonner, *Adv. Carbohydr. Chem.*, 1951, **6**, 251.
- C. S. Wilcox, G. W. Long, and H. Suh, *Tetrahedron Lett.*, 1984, **25**, 395.