Intramolecular Dehydration of β-Hydroxyalkylphosphonic Acid Monoesters. A Novel Type of Olefin Formation

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The title reaction using dicyclohexylcarbodiimide (DCC) gave stereospecifically the corresponding olefins in good yields via tetracoordinate 1,2-oxaphosphetanes. Use of more than one equivalent of DCC afforded better yields of the olefin.

β-Lactones and β-sultines have been known as strained molecules to give olefins on thermolysis. ¹⁾ On the other hand, tetracoordinate 1,2-oxaphosphetanes, their phosphorus analogue, have been synthesized as stable compounds, ²⁾ but most of them have not been characterized unambiguously in contrast to pentacoordinate species. ³⁾ In fact, revised structures have been proposed for some compounds. ^{2b,2c,4)} In this paper we wish to report the title reaction as a novel mode of olefin formation via a tetracoordinate 1,2-oxaphosphetane.

 β -Hydroxyalkylphosphinic acid monoesters $1^{5)}$ were prepared in moderate to good yields by treatment of the corresponding diester 4 with NaI in boiling acetonitrile. A direct method for the preparation of 1 g was also developed as shown in the following scheme. The polyether moiety of alkylphosphonic acid monoester 5 is necessary for effective dilithiation, otherwise no carbonyl adduct was obtained.

MeO O O-P-Et
$$\xrightarrow{1) 2n\text{-BuLi}}$$
 $\xrightarrow{H_3O^+}$ MeO O O-P \xrightarrow{Ph} Ph HO Me

Table 1. Synthesis of β -Hydroxyalkylphosphonic Acid Monoesters 1 and Olefi

						Yields ^{a)} /%		
Rur	l	R	R^1	\mathbb{R}^2	\mathbb{R}^3	R^4	1	3
1	a:	Me	Н	Н	CH ₂ Ph	CH ₂ Ph	98	91 ^{b)}
2	b:	Me	Н	Н	Н	$n\text{-}\!\mathrm{C}_{11}\mathrm{H}_{23}$	89	71
3	c:	Me	Н	H	Н	$p\text{-NO}_2\text{C}_6\text{H}_4$	34	76
4	d:	Me	Me	Me	Н	Ph	46	82 ^{c)}
5	e:	Me	Н	Me	Н	Ph	62 ^{d)}	72 ^{c,e)}
6	f:	Me	Н	Me	Н	<i>n</i> -C ₆ H ₁₃	76 ^{f)}	77 ^{c,g)}
7	g: Me	e(OCH ₂ CH	2) ₂ H	Me	Н	Ph	49 ^{h)}	91 ^{c,i)}

a) Isolated yields based on **4** and **1** for **1** and **3**, respectively. b) When 0.6, 1.0, 1.5, and 2 equivalents of DCC were used, yields of **3a** were 55-56, 60-66, 91, and 90%, respectively. c) Determined by 1 H NMR spectroscopy. d) *erythro:threo*=78:22. e) (*Z*): (*E*)= 77:23. f) *erythro:threo*=76:24. g) (*Z*): (*E*)=72:28. h) Isolated yield based on **5**. *erythro:threo*=60:40. i) (*Z*): (*E*)=58:42.

Treatment of β -hydroxyalkylphosphonic acid monoesters 1 with dicyclohexylcarbodiimide (DCC) in CH₂Cl₂ at 0 °C then at room temperature for 1 h gave the corresponding olefins 3 along with dicyclohexylurea after chromatography on silica gel. The results using 2 equivalents of DCC are shown in Table 1. The reaction was monitored by ³¹P NMR spectroscopy and showed that the signal due to the starting material (δ_p 33.4) disappeared rapidly to give a new signal (δ_p 24.3), which decreased gradually over -30 °C to afford finally complex signals. The value of δ_p and nonequivalency of two protons of α -methylene group (¹H NMR) and that of two benzyl groups (¹H and ¹³C NMR) are consistent with a structure of intermediate 2a.⁵⁾ As shown in the footnote of Table 1 use of more than one equivalent of DCC gave better yields of 3a, indicating that DCC probably plays an important role at the stage of olefin formation from the intermediate.⁶⁾

Table 1 shows that the present method can be used for the preparation of mono-, di-, and tri-substituted olefins, and is applicable also to 1c and 1d, the methyl ester of which (4c and 4d) did not give the olefin under reported basic conditions. As shown in Runs 5, 6, and 7 the reaction using an *erythro*-enriched diastereomeric mixture gave a (Z)-enriched olefin mixture. In each reaction the *erythro/threo* ratio is almost same as the (Z)/(E) ratio, indicating that the present reaction proceeds stereospecifically via syn elimination. Taking into account the above results and the fact that use of the sodium salt of 1a or diester 4a gave no olefin, it is most likely that the

reaction is initiated by acid-catalyzed addition of the acid monoester to the imino group of the carbodiimide, followed by back-side attack of the hydroxyl group to give intermediate 2 and dicyclohexylurea, the former of which undergoes a Wittig-like reaction to afford the olefin together with readily oligomerized metaphosphate.⁷⁾

$$\begin{array}{c|c}
 & O & OH \\
RO & R^4 \\
O & R^2 \\
R'NH & NR' & R^3 \\
\hline
 & R' = -H
\end{array}$$

$$\begin{array}{c|c}
 & O & R^4 \\
RO - P & R^3 \\
\hline
 & R^2
\end{array}$$

These tetracoordinate 1,2-oxaphosphetanes, which are intermediates of a novel type of olefin formation, are highly reactive in marked contrast to the previously reported ones. $^{2a,d,e,h)}$ Also, it is very interesting that the present olefin formation proceeds under milder reaction conditions compared with usual Horner-Emmons reaction from β -hydroxyalkylphosphonates $^{8)}$ or -phosphine oxides $^{9)}$ without any carbanion-stabilizing group at the α -position.

References

- β-Lactones, see: S. Searles, "Comprehensive Heterocyclic Chemistry," ed by A. K. Katritzky and C. W. Rees, Pergamon Press, Oxford (1984), Vol. 7, Chap. 5.13, pp. 363-402; W. Adam, J. Baeza, and J.-C. Liu, J. Am. Chem. Soc., 94, 2000 (1972); S. Maegeswaren and M. U. S. Sultsnbawa, J. Chem. Soc., Perkin Trans. 1, 1976, 884; J. Mulzer, A. Pointner, A. Chucholowski, and G. Bnütrup, J. Chem. Soc., Chem. Commun., 1979, 52. β-Sultines, see: J. W. Timberlake and E. S. Elder, "Comprehensive Heterocyclic Chemistry," ed by A. K. Katritzky and C. W. Rees, Pergamon Press, Oxford (1984), Vol. 7, Chap. 5.15, pp. 449-489; G. F. Jung, N. K. Sharma, and T. Durst, J. Am. Chem. Soc., 95, 3420 (1973); M. D. M. Gray, D. R. Russell, D. J. H. Smith, T. Durst, and B. Gimbarzevsky, J. Chem. Soc., Perkin Trans. 1, 1981, 1826.
- a) H. G. Henning and M. Morr, Chem. Ber., 101, 3963 (1968); b) M. Regitz, H. Scherer, W. Illger, and H. Eckes, Angew. Chem., Int. Ed. Engl., 12, 1010 (1973); c) S. Nakayama, M. Yoshifuji, R. Okazaki, and N. Inamoto, Bull. Chem. Soc. Jpn., 49, 1173 (1976); d) M. Constenla and K. Dimroth, Chem. Ber., 109, 3099 (1976); e) G. Maas and R. Hoge, Justus Liebigs Ann. Chem., 1980, 1028; f) M. Regitz and H. Eckes, Chem. Ber., 113, 3303 (1980); g) M. Regitz and H. Eckes, Tetrahedron, 37, 1039 (1981); h) H. Tomioka, M. Watanabe, N. Kobayashi, and K. Hirai, Tetrahedron Lett., 31, 5061 (1990).
- 3) For a recent review, see: B. E. Maryanoff and A. B. Reitz, Chem. Rev., 89, 863 (1989); For isolable

- compounds, see: F. Ramirez, C. P. Smith, and J. F. Pilot, J. Am. Chem. Soc., 90, 6726 (1968); H. J. Bestmann, K. Roth, E. Wilheim, R. Böhme, and H. Burzlaff, Angew. Chem., Int. Ed. Engl., 18, 87 (1979); T. Kawashima, K. Kato, and R. Okazaki, J. Am. Chem. Soc., 114, 4008 (1992); T. Kawashima, K. Kato, and R. Okazaki, Angew. Chem., Int. Ed. Engl., 32, 869 (1993). For spectral observations, see: A. B. Reitz, S. O. Nortey, A. D. Jordan, Jr., M. S. Mutter, and B. E. Maryanoff, J. Org. Chem., 51, 3302 (1986); B. E. Maryanoff, A. B. Reitz, M. S. Mutter, R. R. Inners, H. R. Almond, Jr., R. R. Whittle, R. A. Olofson, J. Am. Chem. Soc., 108, 7664 (1986); E. Vedejs, C. F. Marth, and R. Ruggeri, ibid., 110, 3940 (1988).
- 4) T. Kawashima, S. Nakayama, M. Yoshifuji, R. Okazaki, and N. Inamoto, *Bull. Chem. Soc. Jpn.*, **64**, 711 (1991); T. Kawashima and N. Inamoto, *ibid.*, **64**, 713 (1991).
- 5) Physical and spectral data of **1a** and **2a** are shown as typical examples: **1a**: mp 100.9-102.5 °C. 1 H NMR (500 MHz, CDCl₃): δ= 2.00 (2H, d, J_{HP} =19.1 Hz, PC \underline{H}_{2}), 2.83 (2H, d, J_{HH} =13.6 Hz, PCH₂(C \underline{H}_{2} H'Ph)₂), 2.97 (2H, dd, J_{HH} =13.6 Hz, J_{HP} =3.1 Hz, PCH₂(CH \underline{H}_{2} Ph)₂), 3.68 (3H, d, J_{HP} =11.4 Hz, POC \underline{H}_{3}), and 7.21-7.38 (12H, m, 2×OH, 2×C₆ \underline{H}_{5}). 13 C{ 1 H} NMR (125 MHz, CDCl₃): δ = 32.99 (d, J_{CP} =140.5 Hz, PC \underline{H}_{2}), 46.32 (d, J_{CP} =9.2 Hz, C \underline{H}_{2} Ph), 51.40 (d, J_{CP} =7.0 Hz, POC \underline{H}_{3}), 72.57 (d, J_{CP} =4.1 Hz, COH(CH₂Ph)₂), 126.53 (s), 128.12 (s), 130.74 (s), and 136.33 (s). 31 P NMR (36.3 MHz, CDCl₃): δ =33.4. Found: C, 63.56; H, 6.31%. Calcd for C₁₇H₂₁O₄P: C, 63.74; H, 6.61%. **2a**: 1 H NMR (500 MHz, CDCl₃, -33 °C): δ = 2.92 (1H, dd, J_{HH} =14.7 Hz, J_{HP} =19.2 Hz, PC \underline{H}_{1} H), 2.96 (1H, dd, J_{HH} =14.7 Hz, J_{HP} =18.6 Hz, PCH \underline{H}_{1}), 3.07 (1H, d, J_{HH} =14.1 Hz, C \underline{H}_{1} H'Ph), 3.12 (1H, d, J_{HH} =14.2 Hz, C \underline{H}_{1} H'Ph'), 3.13 (1H, d, J_{HH} =14.1 Hz, CH \underline{H}_{1} Ph), 3.16 (1H, d, J_{HH} =14.2 Hz, CDCl₃, -33 °C): δ = 39.76 (d, J_{CP} =97.6 Hz, PC \underline{H}_{2}), 43.91 (d, J_{CP} =5.9 Hz, C \underline{H}_{2} Ph), 44.74 (d, J_{CP} =7.6 Hz, CH₂Ph'), 53.33 (d, J_{CP} =6.5 Hz, POC \underline{H}_{3}), 76.91 (d, J_{CP} =20.0 Hz, POC(CH₂Ph)₂), 127.04 (s), 127.19 (s), 128.23 (s), 128.30 (s), 130.50 (s), 130.55 (s), 134.53 (s), and 134.77 (s). 31 P NMR (36.3 MHz, CDCl₃, -38 °C): δ = 24.3.
- 6) Isolation of stable tetracoordinate 1,2-oxaphosphetanes and effects of excess of DCC will be discussed in the following paper: T. Kawashima, H. Takami, and R. Okazaki, submitted to *Chem. Lett*.
- 7) For a review, see: F. H. Westheimer, Chem. Rev., 81, 313 (1981).
- 8) T. Kawashima, T. Ishii, and N. Inamoto, Chem. Lett., 1983, 1375; 1984, 1097.
- For examples, see: A. D. Buss and S. Warren, J. Chem. Soc., Perkin Trans. 1, 1985, 2307; A. D. Buss, W. B. Cruse, O. Kennard, and S. Warren, ibid., 1984, 243.

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