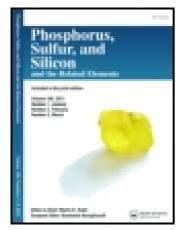
This article was downloaded by: [University of Connecticut]

On: 14 October 2014, At: 08:31 Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH,

UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information:

http://www.tandfonline.com/loi/gpss20

COPPER-CATALYSED REARRANGEMENT OF 4SUBSTITUTED-2,3-1HBENZOXAZINE-1-THIONES

Hikmet Aĝirbaş ^a & Selahattin Güner ^a Department of Chemistry , Kocaeli University , 41300, Izmit, Turkey Published online: 04 Oct 2006.

To cite this article: Hikmet Aĝirbaş & Selahattin Güner (2000) COPPER-CATALYSED REARRANGEMENT OF 4-SUBSTITUTED-2,3-1H-BENZOXAZINE-1-THIONES, Phosphorus, Sulfur, and Silicon and the Related Elements, 161:1, 257-263, DOI: 10.1080/10426500008042112

To link to this article: http://dx.doi.org/10.1080/10426500008042112

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or

indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at http://www.tandfonline.com/page/terms-and-conditions

COPPER-CATALYSED REARRANGEMENT OF 4-SUBSTITUTED-2,3-1H-BENZOXAZINE-1-THIONES

HİKMET AĞIRBAŞ* and SELAHATTİN GÜNER

Department of Chemistry, Kocaeli University, 41300 Emit, Turkey

(Received November 02, 1999)

4-Substituted-2,3-1H-benzoxazine-1-thiones were prepared by the treatment of the corresponding benzoxazine-1-ones with P₂S₅. The thermal rearrangement of 4-substituted-2,3-1H-benzoxazine-1-thiones, catalysed by metallic copper, yielded 4-substituted-2,3-1H-benz-thiazine-1-ones.

Keywords: 2,3-1H-Benzoxazine-1-one; 2,3-1H-benzoxazine-1-thione; 2,3-1H-benzthi-azine-1-one

Uncatalysed thion-thiol rearrangements of acyclic compounds have been extensively studied. ¹⁻¹⁰ The temperature of these rearrangements ranges from 25°C to about 300°C depending on the structure of the thiones. The thion-thiol rearrangements of 1,2,4-oxadiazole-5-thiones were also studied as five membered heterocyclic ring systems. ¹¹⁻¹⁵ Uncatalysed rearrangements of 1,2,4-oxadiazole-5-thiones were found to take place considerable less readily than copper-catalysed ones.

In this paper, we report the thion-thiol rearrangement of 4-substituted-2,3-1H-benzoxazine-1-thiones as six membered heterocyclic ring systems 4-Substituted-2,3-1H-benzoxazine-1-thiones(3) were prepared from thionation of the corresponding 2,3-benzoxazine-1-ones which were synthesized from the cyclication of o-benzoylbenzoic acids with hydroxylamine hydrochloride as described in the literature ¹⁶ (Scheme 1).

The thermal rearrangement of 4-substituted-2,3-1H-benzox-azine-1-thiones(3) were carried out with a catalytic amount of copper powder in xylene and yielded the corresponding 4-substituted-2,3-

^{*} Corresponding Author.

1H-benzthiazine-1-ones(4). The reaction was normally completed in 8 h at 140°C. The analyses of compounds (3and 4) were carried out by following the disappearance of the C=O frequency(ir) at 1700–1800 cm⁻¹ and the appearance of the C=S vibration in the region of 1200–1400 cm⁻¹. The mechanism of the copper-catalysed rearrangement of 4-substituted-2,3–1H-benzoxazine-1-thione very likely follows the mechanism of the copper-catalysed rearrangement of five membered heterocyclic ring system. ^{12,13} as shown below (scheme 2).

SCHEME 1

SCHEME 2

In order to test whether the rearrangement occurs without copper catalysis, compound(3a) was heated for 8h at 140°C in n-pentadecane, but it remained unchanged. However, when the same compound was submitted to the same treatment for 2h at 180°C, it was all converted into compoud(4a). Therefore, an alternative reasonable mechanism for the uncatalysed thermal rearrangement of 2,3-benzoxazine-1-thiones can be considered to proceed through a free radical intermediate as shown below (Scheme 3).

EXPERIMENTAL

IR spectra were taken on a Simadzu FTIR-821PC Fourier Transform IR spectrometer. ¹H NMR spectra were recorded on a Brucker AC 200-L (200 MHz) spectrometer. Mass spectra were run at 70 eV by direct inlet on a VCZAP SPEC instrument. Silica Gel (Fluka or Merck) were used for thin layer chromatography. Solvents were dried and purified by standard methods prior to use. Melting points were determined on a Büchi apparatus and are uncorrected.

o-Aroylbenzoic acids were prepared from the reaction of phthalic anhydride with substituted benzene as described in the literature 16.

SCHEME 3

4-(p-Tolyl)-2,3-1H-benzoxazine-1-one, 2a

General Procedure

A mixture of o-(4-methyl)benzoic acid (3.05g, 13mmole), hydroxylamine hydrochloride (2.01g, 29 mmole) and anhydrous sodium acetate (1.25g, 15mmole) in ethanol (40ml) was refluxed for 2h and cooled slowly to room temperature. The resultant white crystalline solid was filtered and washed repeatedly with water, then dried in air and recrystallized from ethanol to give compound (2a) (2.57g, 84%); mp 158–160°C (lit. 17160°C).

4-(p-Tolyl)-2,3-1H-benzoxazine-1-thione, 3a

General Procedure

4-(p-Tolyl)-2,3-1H-benzoxazine-1-one (2a) (2.6g, 11mmole) and phosphorus pentasulphide (2.44g, 11mmole) were refluxed in xylene for 8h. The hot solution was filtered and xylene was evaporated under reduced pressure. The remaining solid was extracted with ether. The solvent was

evaporated and the product was recrystallized from ethanol to give compound (3a) (1.63g, 63%); mp 112-114°C;

 $IR(KBr): 1514(C=N), 1278 \text{ cm}^{-1}(C=S);$

¹H NMR(DMSO-d₆): δ 2.45(s, 3H, CH₃), 7.26(m, 4 aromatic H), 7.80(m, 4 aromatic H);

MS (EI, 70eV): m/z $253(M^+)$.

4-Phenyl-2,3-1H-benzoxazine-1-thione, 3b

The compound was chromatographed on Silica Gel HF_{254} layer with ethylacetate: Light petroleum(40–60) (1:4) (Rf: 0.86) to give an oily compound (3b) (34 mg, 57%).

 $IR(KBr): 1589(C=N),^{1} 1317 \text{ cm}^{-1}(C=S);$

¹H NMR(DMSO-d₆): δ 7.45(m, 5 aromatic H), 7.95(4 aromatic H); MS (EI, 70eV): m/z 239(M⁺).

4-(3,4-Dimethylphenyl)-2,3-1H-benzoxazine-1-thione, 3c

The compound was chromatographed on Silica Gel HF_{254} layer with ethylacetate: light petroleum(40–60) (1:4) (Rf: 0.73) to give an oily compound (3c) (28 mg, 52%).

IR(KBr): 1608(C=N), 1446 cm⁻¹(C=S);

 1 H NMR(CDCl₃): δ 2.25(s, 3H, CH₃), 2.27(s, 3H, CH₃), 6.95(m, 2 aromatic H), 7.24(s, 1 aromatic H), 7.64(2 aromatic H), 7.87(2 aromatic H);

MS (EI, 70eV): m/z $267(M^+)$.

$\hbox{$4$-(p-Isopropylphenyl)-2,3$-1$H-benzoxazine-1-thione, 3d}$

The compound was chromatographed on Silica Gel HF_{254} layer with ethylacetate: light petroleum(40–60) (1:4) (Rf: 0.76) to give an oily compound (3d) (38 mg, 60%).

IR(KBr). 1510(C=N), 1315 cm⁻¹(C=S);

¹H NMR(CDCl₃): δ 1.24(d, 6H, 2CH₃), 2.94(sep., 1H, CH), 7.15(m, 2 aromatic H), 7.31(m, 2 aromatic H), 7.70(m, 2 aromatic H), 7.88(m, 2 aromatic H),

MS (EI, 70eV): m/z 281 (M⁺).

4-(p-Tolyl)-2,31H-benzothiazine-1-one, 4a

General Procedure

4-(p-Tolyl)-2,3-1H-benzoxazine-1-thione (3a)(0.56g, 2.2mmole) was refluxed in xylene for 8h in the presence of a catalytic quantity of copper powder. Xylene was evaporated under reduced pressure to give a crude solid. The solid product was extracted with ether and filtered. The solvent was evaporated and the remaining solid was recrystallized from ethanol to give compound (4a)(0.33g, 59%); mp 130-132°C; IR(KBr): 1512(C=N), 1745 cm⁻¹(C=O)

 1 H NMR(DMSO-d₆): δ 2.50(s, 3H, CH₃), 7.32(m, aromatic 4H), 7.94(m, aromatic 4H).

MS (EI, 70eV): m/z $253(M^+)$.

4-Phenyl-2,3-1H-benzothiazine-1-one, 4b

The compound was chromatographed on Silica Gel HF_{254} layer with ethylacetate: light petroleum(40–60) (1:4) (Rf: 0.63) to give an oily compound (4b) (20 mg, 45%).

IR(KBr): 1515(C=N), 1716 cm⁻¹(C=O);

¹H NMR(DMSO-d₆): δ 7.38(m, aromatic 2H), 7.42(m, aromatic 3H), 7.94(m aromatic 5H);

MS (EI, 70eV): m/z $239(M^+)$.

4-(3,4-Dimethylphenyl)-2,3-1H-benzothiazine-1-one, 4c

The compound was chromatographed on Silica Gel HF_{254} layer with ethylacetate: light petroleum(40–60) (1:4) (Rf: 0.55) to give an oily compound (4c) (36 mg, 68%).

IR(KBr): 1504(C=N), 1720 cm⁻¹(C=O);

 1 H NMR(CDCl₃): δ 2.24(s, 6H, 2CH₃), 7.10(m, aromatic 2H), 7.22(m, aromatic 1H) 7.73(m, aromatic 2H), 7.88(m, aromatic 2H); MS (EI, 70eV): m/z 267(M⁺).

4-(p-Isopropylphenyl)-2,3-1H-benzothiazine-1-one, 4d

The compound was chromatographed on Silica Gel HF_{254} layer with ethylacetate: light petroleum(40–60) (1:4) (Rf: 0.59) to give an oily compound (4d) (32 mg, 53%).

 $IR(KBr): 1514(C=N), 1710 \text{ cm}^{-1}(C=O);$

¹H NMR(CDCl₃); δ 1.21(d, 6H, 2CH₃), 2.90(sep., 1H, CH), 7.19(m, aromatic 2H), 7.28(m, aromatic 2H), 7.72(m, aromatic 2H), 7.89(m, aromatic 2H);

 $MS (EI, 70eV): m/z 281(M^+).$

Acknowledgements

The financial support of Kocaeli University Research Fund is gratefully acknowledged.

References

- 1. A. Schönberg and L. Vargha, Chem. Ber. 63, 178(1930).
- 2. H.R. Al-Kazimi, D.S. Tarbel and D. Plant, J. Am. Chem. Soc., 77, 2479(1955).
- 3. D.H. Powers and D.S. Tarbell, J. Am. Chem. Soc., 78,70(1956).
- 4. M.S. Newman and H.A. Karnes, J. Org. Chem., 31, 3980(1966).
- 5. K. Miyazaki, Tetrahedron Lett., 2793(1968).
- 6. B. Cross, R.J.G. Searle and R.E. Woodall, J. Chem. Soc. (C), 1833(1971).
- R.F. Hudson, A.J. Lawson and E.A.C. Lucken, J. Chem. Soc. Chem. Commun., 807(1971).
- W.B. Ankers, C Brown, R.F. Hudson and A.J. Lawson, J. Chem. Soc. Chem. Commun., 935(1972).
- R.F. Hudson, A.J. Lawson and K.A.F. Record, J. Chem. Soc., Perkin Trans. 2, 869(1974).
- 10. W. Hanefeld and Z.E. Güneş, Liebigs Ann. Chem., 1140(1986).
- 11. A. Pelter and D. Sümengen, Tetrahedron Lett., 22, 1945(1977).
- 12. D. Sümengen and A. Pelter, J. Chem. Soc. Perkin 1, 687(1983).
- Y. Dürüst, H. Ağirbaş and D. Sümengen, Phosphorous, Sulfur and Silicon, 62,47(1991).
- H. Ağirbaş, Y. Dürüst and D. Sümengen, Phosphorous Sulfur and Silicon, 66, 321(1992).
- 15. H. Ağirbaş and K. Kahraman, Phosphorous, Sulfur and Silicon, 134/135, 381(1998).
- 16. A.I. Vogel, Practical Organic Chemistry, p. 739(1972).
- 17. R.K. Bansal, G. Kumar and B.K. Puri, Chim. Acta Turc., 18, 295(1990).