

Trapping of Diaceptor-Substituted Methylenecyclopropanes with Isocyanides – A Further Application of the Principle of the Twofold Nucleophilic Substitution at a Cyclopropane

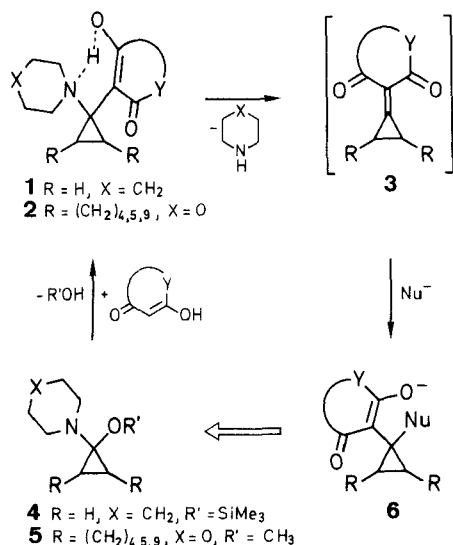
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Dedicated to Prof. Dr. H.J. Bestmann on the occasion of his 65th birthday

Unstable 2-cyclopropylidene-1,3-diones **3** can be trapped by isocyanides **8** to give [4+1]-cycloadducts **9** and **12**. In the case of Meldrum's acid, as the dioxo component, spirotricyclic imides **15** are isolated due to an easy degradation of the primary cycloaddition products **14** in methanol. Subsequent reactions are described for the [4+1]cycloadducts leading to a cyclopropanecarboxamide **19b** or to a phenylamino-substituted furan **20/21**. Trapping of **3** by isocyanides **8** represents a novel second step in the concept of the twofold nucleophilic substitution at a cyclopropane.

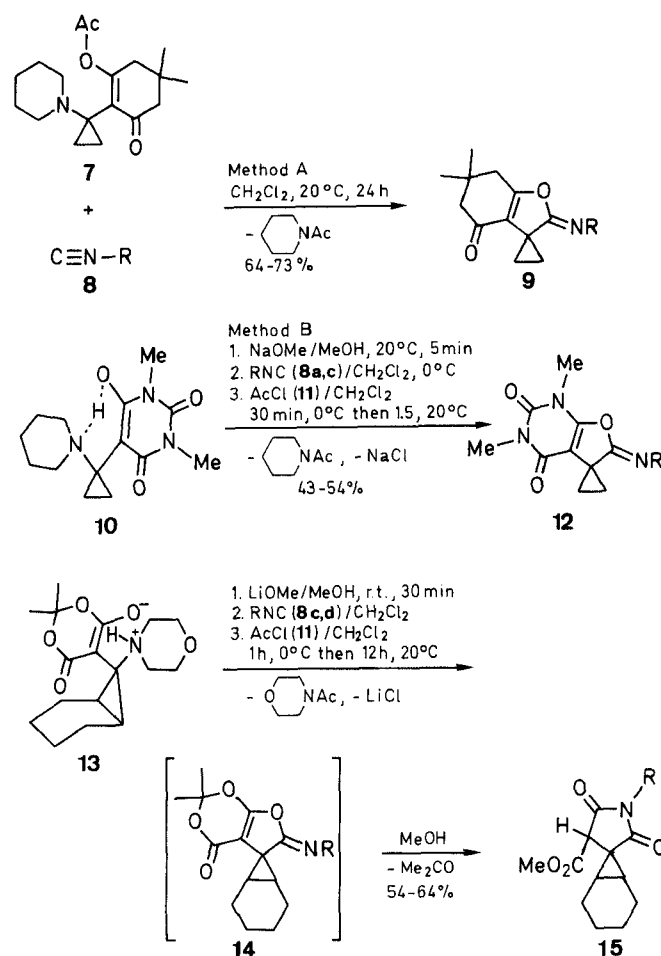
2-Cyclopropylidene-1,3-cycloalkanediones **3** represent a class of very reactive compounds which as yet have proved to be unisolable.¹ They are trapped, however, by a variety of reagents such as 1,3-dienes,² electron-rich alkenes,³ electron-rich alkynes,⁴ or nucleophiles (e.g. cyanide,⁵ hydride,⁵ amines,⁶ carbanions^{7–11}), leading to products **6**. Intermediates **3** are best generated from aminocyclopropylated CH-acids **1/2**. Since the latter are easily prepared from cyclopropanone *O,N*-acetals **4/5** and CH-acids, the sequence **4/5** → **1/2** → **6** corresponds to the overall result of a twofold nucleophilic substitution at a cyclopropane.



[4+1]-Cycloaddition reactions are reported to take place between isocyanides and a few α,β -unsaturated ketones.^{12–16} Thereby the α,β -unsaturated ketone either has to be activated by a further acceptor moiety (e.g. a –CF₃,¹⁴ or a C=O^{15,16} group) or diethylaluminum chloride is necessary as a catalyst.¹³ Our present investigations show that 2-cyclopropylidene-1,3-cycloalkanediones **3** are activated enough to be trapped by isocyanides **8** smoothly, in situ, using mild reaction conditions.

Thus [4+1]-cycloaddition products **9a–c** are obtained in 64–73% yield from an isocyanide **8** and the acylated dimedone derivative **7** after 24 h at room temperature in

dichloromethane as a solvent. The generation and the trapping of 5-cyclopropylidenebarbituric acid **3** is managed by addition of acetyl chloride to a suspension of the sodium salt of **10** and an isonitrile **8**. Compounds **12a** and **12c** were isolated in 43% and 54% yield, respectively. A similar procedure is applied to the bicyclic Meldrum's acid derivative **13** using its lithium salt. Here, however, the [4+1]-cycloaddition product **14** decomposes very easily to give a stable methyl ester **15** upon addition of methanol (overall yields: **15c**: 54%; **15d**: 64%). The degradation of the Meldrum's acid moiety in **14** is accompanied by an acylimide – imide rearrangement (example of a Dimroth rearrangement¹⁷). The ¹³C-NMR data clearly demonstrate the structure of the cycloadducts **9** and **12**, as well as the tricyclic imides **15**. The presence of a cyclopropane system is indicated by a singlet and one triplet in the case of **9** and **12**, and a



8,9,12,14,15	a	b	c	d
R	CH ₂ Ts	<i>t</i> -Bu	Ph	4-MeOC ₆ H ₄

singlet and two doublets in the case of **15**. The $^1J_{\text{CH}}$ coupling constants are characteristic of a cyclopropane system (see Table 2 and the experimental). The imideate ^{13}C -NMR signal ranges from $\delta = 156.6$ to 167.7 for **9a–c** and occur at $\delta = 157$ – 158 for **12a** and $\delta = 163.8$ for **12c**. In contrast, the signals for the enone system are almost constant for compounds **9** and **12**, respectively. This

behavior is consistent with the imideate structure **9** and **12** and would not be expected for a lactam-type structure **16**. The *N*-phenyl substituent gives further proof for the imideate unit in **9c** and **12c**. The C-1 atom of the *N*-phenyl group shows absorption at $\delta = 144.8$ for **9c** or $\delta = 143.4$ for **12c**. This value is in accordance with literature data for an isocyanide – $[4 + 1]$ cycloadduct (imideate – struc-

Table 1. ^1H -NMR Spectroscopic Data of Compounds **9** and **12** (CDCl_3/TMS , δ)

Com- pound	Cyclopropane (m_c) ^a	Bicyclic System		R
		CH_3 (s)	CH_2 (s) ^a	
9a	1.27, 1.79	1.12 ^b	2.22, 2.43	2.43 (s, 3H), 4.73 (s, 2H), 7.34, 7.78 (AA'XX'-system, 4H)
9b	1.10, 1.66	1.15 ^b	2.23, 2.49	1.26 (s, 9H)
9c	1.46, 1.91	1.14 ^b	2.27, 2.49	7.11 (m_c , 3H), 7.32 (m_c , 2H)
12a	1.32, 1.85	3.30 ^c , 3.43 ^c	–	2.46 (s, 3H), 4.76 (s, 2H), 7.36, 7.79 (AA'XX'-system, 4H)
12c	1.52, 1.92	3.31 ^c , 3.38 ^c	–	7.12 (m_c , 3H), 7.34 (m_c , 2H)

^a Each signal corresponds to 2H.

^b 6H.

^c 3H.

Table 2. ^{13}C -NMR Spectroscopic Data of Compounds **9** and **12** (CDCl_3/TMS , δ)

Com- pound	Cyclopropane		Condensed Bicyclic System						R
	(s)	(t) ^a	Enone (s)	(s)	O–C=N (s)	CH ₃ (q)	CH ₂ (t)	C _q (s)	
9a	24.7	17.7 [167]	169.0, 192.7	116.1	167.7	28.5	36.7, 51.2	34.1	21.5 (q), 68.0 (t), 128.9 (d), 129.3 (d), 134.4 (s), 144.6 (s)
9b	24.2	16.6 [166]	169.9, 192.7	115.3	156.6	28.7	37.3, 51.3	33.9	30.0 (q), 53.6 (s)
9c	24.9	17.7 [165]	169.7, 192.7	115.8	161.5	28.5	37.0, 51.3	34.0	122.6 (d), 124.1 (d), 128.5 (d), 144.8 (s)
12a	24.9	16.5 [166]	157.8 ^b , 156.6 ^b , 150.6	89.6	163.8 ^b	27.6, 29.5	–	–	21.4 (q), 68.5 (t), 129.0 (d), 129.6 (d), 134.3 (s), 145.2 (s)
12c	25.1	16.7 [167]	157.9 ^b , 157.0 ^b , 150.7	89.6	157.3 ^b	27.7, 29.7	–	–	122.7 (d), 125.0 (d), 128.9 (d), 143.4 (s)

^a $^1J_{\text{CH}}$ coupling constant (Hz) in [parenthesis].

^b Unequivocal assignment not possible.

Table 3. Products **9** and **12** from the Cycloaddition of Isocyanides **8** with 2-Cyclopropylidene-1,3-cycloalkanediones **3** ($\text{R} = \text{H}$)

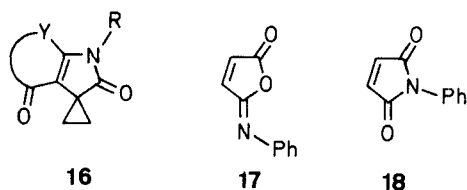
Prod- uct	Method	Recrystallization Solvent	Yield ^a (%)	mp (°C)	Molecular Formula ^b	IR (KBr) $\nu_{\text{C=O}}$, $\nu_{\text{C=N}}$, $\nu_{\text{C=C}}$ (cm^{-1})
9a	A	acetone/ H_2O (1 : 1)	0.60 g (64%)	135	$\text{C}_{20}\text{H}_{23}\text{NO}_4\text{S}$ (373.5)	1715, 1665, 1595
9b	A	acetone/ H_2O (1 : 1)	0.48 g (73%)	96	$\text{C}_{16}\text{H}_{23}\text{NO}_2$ (261.4)	1730, 1655, 1635
9c	A	MeOH^c	0.49 g (70%)	112	$\text{C}_{18}\text{H}_{19}\text{NO}_2$ (281.4)	1700, 1650, 1585
12a	B	MeCN	0.42 g (43%)	183	$\text{C}_{18}\text{H}_{19}\text{N}_3\text{O}_5\text{S}$ (389.4)	1740, 1700–1670 (br), 1645, 1590
12c	B	MeOH	0.40 g (54%)	170	$\text{C}_{16}\text{H}_{15}\text{N}_3\text{O}_3$ (297.3)	1740, 1700, 1665 (br), 1590

^a Yield of isolated product **9** and **12** based on **7** and **10** as starting material, respectively.

^b Satisfactory microanalyses obtained: $\text{C} \pm 0.37$, $\text{H} \pm 0.22$, $\text{N} \pm 0.29$.

^c The filtrate from the crystallization is concentrated and the solid is recrystallized from $\text{MeOH}/\text{H}_2\text{O}$ (1 : 1).

ture¹⁵). Furthermore, it fits very well with the analogous signal of imideate **17** ($\delta = 145.0$ ¹⁸) and contrasts strongly with the *N*-phenyl *ipso*-C signal of imide **18** ($\delta = 131.3$;¹⁹ at least **9** can be regarded as a vinylogous imide compound). On the other hand, the absorption of the *N*-phenyl singlet (*C-ipso*) unequivocally establishes a phenyl-imide structure **15** which stems from an imino-imide rearrangement (**15c**: $\delta = 131.8$, **15d**: $\delta = 124.4$, an upfield shift of 7.7 ppm²⁰ with respect to **15c** is expected for this signal due to the OCH₃-group at the 4-position). Compounds **15c** and **15d** are formed as one pure isomer. The configuration of **15c** and **15d** is not clear, but according to other trapping experiments of bicycloheptylidene Meldrum's acid intermediate **3**, a $1\alpha,6\alpha,7\alpha$ -structure is most likely.^{1,2,5-7,11}



The isocyanide trapping products can be used for subsequent reactions. Thus hydrolysis of **9b** leads to a substituted cyclopropanecarboxamide **19** (83% yield). Aluminum chloride induced ring opening transfers the phenyl substituted cycloadducts **9a** and **12a** into the aminofuran derivatives **20** (82% yield) and **21** (51% yield), respectively. The structure of **20** and **21** can be deduced from the ¹³C-NMR data: the high field shifting of the phenyl *ortho* and *para*-C-atoms (**20**: $\delta = 114.1$ and 120.1; **21**: $\delta = 114.0$ and 120.5) exclude an isomeric pyrrole unit in which interaction of the nitrogen lone pair and the phenyl unit is strongly decreased (*N*-phenylpyrrole:²¹ phenyl-C-2/6 signals: $\delta = 126.1$, phenyl-C-4 signal: $\delta = 127.8$). The CH₂Cl-moiety is indicated by a triplet with a coupling constant characteristic of an adjacent chlorine substituent²² (**20**: $\delta = 44.6$, $^1J_{CH} =$

153 Hz; **21**: $\delta = 44.0$, $^1J_{CH} = 153$ Hz). Addition of silver nitrate solution gives the information about the absence of ionic chloride; this is in accordance with the chloroethyl structural unit.

The trapping reactions of 2-cyclopropylidene-1,3-cycloalkanediones **3** with isocyanides **8** perform the connection of a d¹-carbon unit to a cyclopropane system. Thus, this sequence represents a further example for the application of the principle of twofold nucleophilic substitution at a cyclopropane.

All reagents were of commercial quality; CH₂Cl₂ and CHCl₃ were dried over Sicapent; TOSMIC is commercially available. The ¹H- and ¹³C-NMR spectra were recorded with a Bruker AM 400 spectrometer. The designation *m_c* is used to denote a complex multiplet. The IR spectra were measured on a Perkin-Elmer 397 Infrared Spectrophotometer. Microanalyses were performed with a Perkin-Elmer 2400 Elemental Analyzer.

Bicyclic Imideates **9** and **12**; General Procedure:

Method A for Imideates **9:** A solution of acylated dimedone derivative **7³** (0.76 g, 2.5 mmol) and isocyanide **8** (2.5 mmol; **8a**: 0.49 g, **8b**²³: 0.21 g, **8c**²⁴: 0.26 g) in CH₂Cl₂ is stirred at 20°C for 24 h. Removal of the solvent *in vacuo* and crystallization of the residue gives crystalline imideates **9** (see Table 3.).

Method B for Imideates **12:** A solution of NaOMe in MeOH (2 M, 2.5 mL) is added to a suspension of aminocyclopropylated barbituric acid **10⁸** (1.40 g, 2.5 mmol) in MeOH (15 mL) and stirred for 5 min at 20°C. The solvent is removed *in vacuo*. The residue is stirred in Et₂O (25 mL) for 24 h, then isolated by suction filtration, washed with Et₂O (2 × 20 mL) and dried *in vacuo*. A mixture of this salt, isocyanide **8** (2.5 mmol; **8a**: 0.49 g, **8c**²⁴: 0.26 g) and CH₂Cl₂ (30 mL) is stirred at 0°C. A solution of AcCl (**11**; 0.22 g, 2.75 mmol) in CH₂Cl₂ is slowly added over 30 min. Stirring is continued for 1.5 h at 20°C. Filtration and evaporation of the solvent gives crude **12** which is recrystallized.

Methyl 1'-Aryl-2',5'-dioxospiro[bicyclo[4.1.0]heptane-7,3'-pyrrolidine]-4'-carboxylates **15**; General Procedure:

Meldrum's acid derivative **13²⁵** (0.65 g, 2.0 mmol) is added to a solution of LiOMe in MeOH (0.1 M LiOMe; 20 mL). Stirring for 30 min and removing the solvent *in vacuo* gives the crude lithium salt which is purified by trituration with Et₂O (30 mL) and washing with pentane (2 × 15 mL). A solution of AcCl (**11**; 0.14 mL, 2.0 mmol) in CH₂Cl₂ (20 mL) is added to a stirred mixture of dried lithium salt (12 h, 20°C/0.01 mbar) and isocyanide **8** (2.0 mmol; **8c**²⁴: 0.21 g; **8d**²⁶: 0.27 g) in CH₂Cl₂ (40 mL) at 0°C. Stirring is continued for 1 h at 0°C and 12 h at 20°C. Filtration and removal of the solvent *in vacuo* lead to an oil which gives a crystalline compound upon dissolving in MeOH (10 mL), addition of water till turbidity and storage at 4°C for 24 h. Recrystallization from MeOH/H₂O yields pure products **15c** and **15d**.

Methyl 2',5'-Dioxo-1'-phenylspiro[bicyclo[4.1.0]heptane-7,3'-pyrrolidine]-4'-carboxylate (15c**):** Yield: 0.34 g (54%); mp 124°C.

C₁₈H₁₉NO₄ calc. C 69.00 H 6.11 N 4.47 (313.4) found 68.4 6.1 4.3

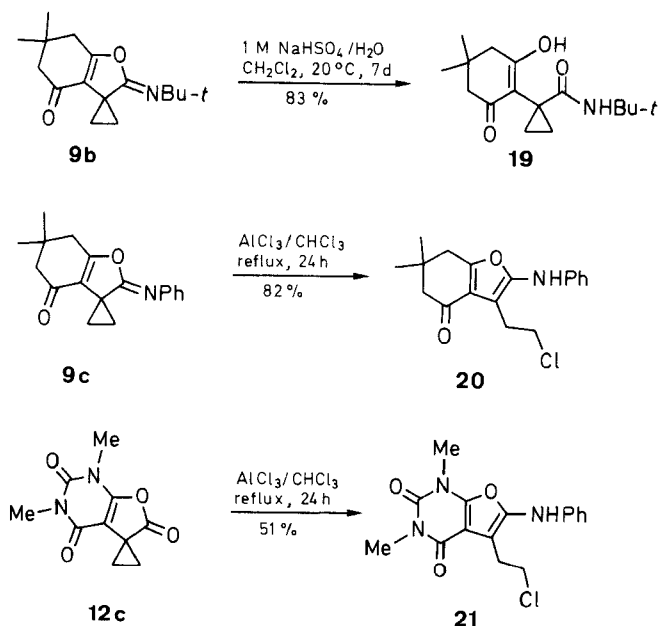
IR (KBr): $\nu = 1770, 1730, 1700 \text{ cm}^{-1}$ (C=O).

¹H-NMR (CDCl₃): $\delta = 1.04$ (*m_c*, 1 H), 1.28 (*m_c*, 1 H), 1.41 (*m_c*, 4 H), 1.75 (*m_c*, 1 H), 1.91 (*m_c*, 1 H), 2.11 (*m_c*, 1 H), 2.22 (*m_c*, 1 H), 3.72 (s, 1 H), 3.80 (s, 3 H), 7.30–7.47 (*m*, 5 H).

¹³C-NMR (CDCl₃): $\delta = 18.7$ (t), 19.4 (t), 20.6 (t), 20.9 (t), 21.5 (d), $^1J_{CH} = 166$ Hz), 25.2 (d, $^1J_{CH} = 166$ Hz), 31.5 (s), 49.3 (d), 53.0 (q), 126.2 (d), 128.5 (d), 128.9 (d), 131.8 (s), 167.8 (s), 171.4 (s), 178.2 (s).

Methyl 1'-(4-Methoxyphenyl)-2',5'-dioxospiro[bicyclo[4.1.0]heptane-7,3'-pyrrolidine]-4'-carboxylate (15d**):** Yield: 0.44 g (64%); mp 154°C.

C₁₉H₂₁NO₅ calc. C 66.46 H 6.16 N 4.08 (343.4) found 66.4 6.1 4.0



IR (KBr): $\nu = 1770, 1700 \text{ cm}^{-1}$ (C=O).

$^1\text{H-NMR}$ (CDCl_3): $\delta = 1.02$ (m, 1 H), 1.25 (m, 1 H), 1.41 (m, 4 H), 1.72 (m, 1 H), 1.91 (m, 1 H), 2.09 (m, 1 H), 2.20 (m, 1 H), 3.70 (s, 1 H), 3.79 (s, 3 H), 3.80 (s, 3 H), 6.95, 6.97 (2 H), 7.21, 7.23 (2 H) (AA'XX'-system).

$^{13}\text{C-NMR}$ (CDCl_3): $\delta = 18.7$ (t), 19.4 (t), 20.6 (t), 20.9 (t), 21.4 (d), $^1J_{\text{CH}} = 163 \text{ Hz}$, 25.1 (d, $^1J_{\text{CH}} = 165 \text{ Hz}$), 31.4 (s), 49.2 (d), 53.0 (q), 55.3 (q), 114.2 (d), 124.4 (s), 127.5 (d), 159.3 (s), 167.9 (s), 171.6 (s), 178.5 (s).

N-tert-Butyl-1-(2-hydroxy-4,4-dimethyl-6-oxo-1-cyclohexen-1-yl)cyclopropanecarboxamide (**19**):

A mixture of imide **9b** (0.26 g, 1.0 mmol), aq NaHSO_4 solution (1 M, 25 mL) and CH_2Cl_2 (25 mL) is stirred at 20°C for 7 d. The CH_2Cl_2 layer is separated and the aqueous layer extracted with CH_2Cl_2 ($2 \times 20 \text{ mL}$). Evaporation of the CH_2Cl_2 gives crude **19** which is recrystallized from acetone/ H_2O (1 : 1). Yield: 0.23 g (83 %), mp 134°C .

$\text{C}_{16}\text{H}_{25}\text{NO}_3$ calc. C 68.79 H 9.02 N 5.01
(279.4) found 68.7 8.9 5.2

IR (KBr): $\nu = 3380$ (OH), 1620, 1600 cm^{-1} (C=O, C=C).

$^1\text{H-NMR}$ ($\text{CD}_3\text{ONa}/\text{CD}_3\text{OD}$ solution, 0.5 M): $\delta = 0.65$ (m, 2 H), 1.03 (s, 6 H), 1.15 (m, 2 H), 1.27 (s, 9 H), 2.19 (s, 4 H).

$^{13}\text{C-NMR}$ ($\text{CD}_3\text{ONa}/\text{CD}_3\text{OD}$ solution, 0.5 M): $\delta = 15.4$ (t, $^1J_{\text{CH}} = 163 \text{ Hz}$), 23.5 (s), 29.16 (q), 29.19 (q), 32.8 (s), 51.1 (t), 51.5 (t), 111.8 (s), 177.9 (s), 194.5 (s).

Ring Opening Reactions of **9c** and **12c**; General Procedure:

A suspension of imide **9c** (0.28 g, 1.0 mmol) or **12c** (0.30 g, 1.0 mmol) and AlCl_3 (0.13 g, 1.0 mmol) in CHCl_3 (20 mL) is refluxed for 24 h. Filtration and evaporation of the solvent gives crude **20** or **21** which are crystallizing upon the addition of cold MeOH.

3-(2-Chloroethyl)-6,7-dihydro-6,6-dimethyl-2-phenylamino-benzofuran-4(5H)-one (**20**): Yield: 0.26 g (82 %); mp 123°C .

$\text{C}_{18}\text{H}_{20}\text{ClNO}_2$ calc. C 68.03 H 6.34 N 4.41
(317.8) found 68.0 6.4 4.3

IR (KBr): $\nu = 1655, 1640, 1600, 1570 \text{ cm}^{-1}$ (C=O, C=C).

$^1\text{H-NMR}$ (CDCl_3): $\delta = 1.14$ (s, 6 H), 2.37 (s, 2 H), 2.68 (s, 2 H), 2.99 (t, 2 H), 3.79 (t, 2 H), 5.69 (s, 1 H, NH), 6.69 (m, 2 H), 6.85 (m, 1 H), 7.20 (m, 2 H).

$^{13}\text{C-NMR}$ (CDCl_3): $\delta = 27.0$ (t), 28.5 (q), 34.9 (s), 37.2 (t), 44.6 (t), 52.2 (t), 110.5 (s), 114.1 (d), 119.0 (s), 120.1 (d), 129.2 (d), 144.6 (s), 146.7 (s), 162.9 (s), 194.4 (s).

5-(2-Chloroethyl)-1,3-dimethyl-6-phenylamino-furo[2,3-d]pyrimidine-2,4(1H,3H)-dione (**21**): Yield: 0.17 g (51 %); mp 140°C (dec).

$\text{C}_{16}\text{H}_{16}\text{ClN}_3\text{O}_3$ calc. C 57.58 H 4.83 N 12.59
(333.8) found 57.6 4.8 12.6

IR (KBr): $\nu = 1695, 1655, 1645, 1600, 1520 \text{ cm}^{-1}$ (C=O, C=C).

$^1\text{H-NMR}$ (CDCl_3): $\delta = 3.05$ (t, 2 H), 3.40 (s, 3 H), 3.49 (s, 3 H), 3.91 (t, 2 H), 5.68 (s, 1 H, NH), 6.71 (m, 2 H), 6.90 (m, 1 H), 7.24 (m, 2 H).

$^{13}\text{C-NMR}$ (CDCl_3): $\delta = 27.1$ (t), 28.1 (q), 29.4 (q), 44.0 (t), 96.1 (s), 113.7 (s), 114.0 (d), 120.5 (d), 129.4 (d), 143.0 (s), 144.3 (s), 150.6 (s), 152.6 (s), 158.5 (s).

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