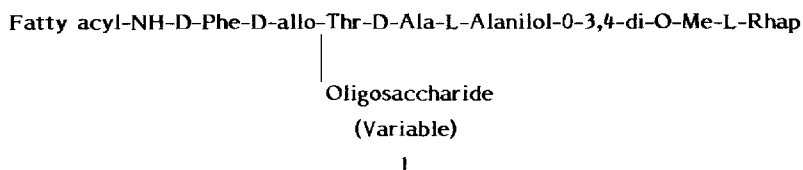


SYNTHETIC STUDIES ON THE BASIC GLYCOPEPTIDE STRUCTURE OF C-MYCOSIDES

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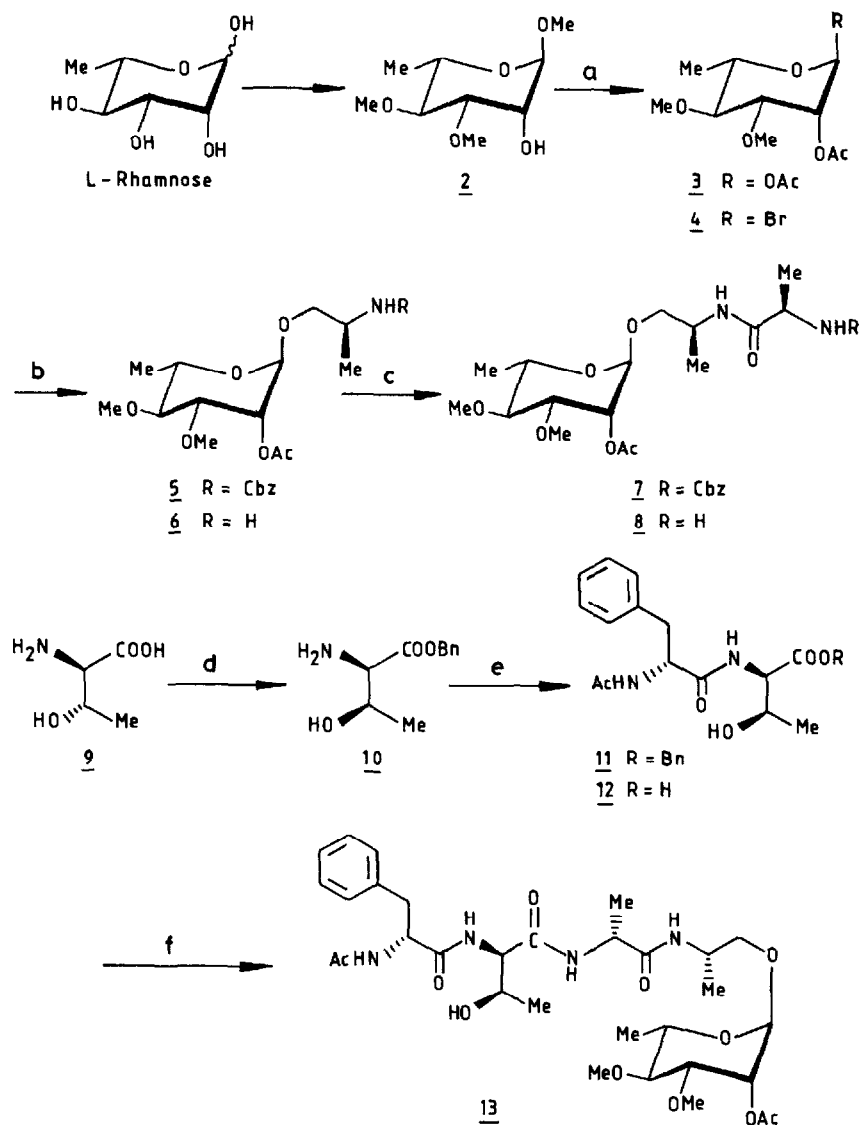
Abstract: The first synthetic approach towards **D**-Phe-**D**-allo-Thr-**D**-Ala-**L**-Alanilol-O-(3,4-di-O-Me-Rhap), the glycolipid of C-mycoside is described.

Glycopeptides is one area which has seen a strong emergence of interest because they are the defining factors in biological recognition and selectivity¹. Marked attention² has been focussed on synthesis with a view to understanding structure-activity relationships coupled with conformational analysis by modern spectroscopic techniques. The glycopeptidolipids present³ on the cell wall surface of C-mycosides such as *Mycobacterium* (*M.*) *avium*, *M. scrofulaceum*, *M. intracellulare*, *M. paratuberculosis* etc. are characterised by the presence of a basic structure - glycotetrapeptide residue (1) having a variable oligosaccharide chain attached directly to the hydroxyl group of **D**-allo-threonine⁴. Although its structure has undergone many modifications, the most likely structure as described in (1) has now been widely accepted⁵. The biosynthetic route⁶ towards the glycopeptide (1) has been studied however, to the best of our knowledge, this communication constitutes the first synthetic study towards (1). The strategy forwarded herein involved independent preparation of **D**-Ala-**L**-Alanilol-O-(3,4-OMe₂-**L**-Rhap) (8) and **D**-Phe-**D**-allo-Thr (12) derivatives followed by coupling reaction leading to the glycotetrapeptide (13).



The prerequisite sugar unit (4) was synthesised from **L**-rhamnose. The known⁷ methyl 3,4-di-O-methyl- α -**L**-rhamnopyranoside (2) was hydrolysed with 3N sulfuric acid at 100° and then acetylated in the presence of acetic anhydride-pyridine to afford the diacetate (3). Treatment of 3 with HBr in acetic acid at 0° provided the rhamnosyl bromide (4) representing the glycosylating agent.

The O-glycosylation of (4) with N-Cbz-**L**-alanilol was conducted in the presence of mercuric cyanide-mercuric bromide in CH₂Cl₂ at ambient temperature to give (5) (81%). The presence of a participating acetyl group at O-2 assured the α -configuration⁸ at the anomeric carbon. In addition, ¹H-NMR and ¹³C-NMR spectra of (5) were in complete agreement with the assigned structure⁹. Deprotection of Cbz group under neutral condition by hydrogenol-



(a) i. 3N H_2SO_4 , Dioxane, Δ , 6h; ii. Ac_2O , Py, DMAP (Cat.), RT, 18h; iii. HBr, AcOH, 0° , CH_2Cl_2 , 3h; (b) i. N-Cbz-L-alaninol, $HgBr_2$, $Hg(CN)_2$, CH_2Cl_2 , RT, 18h; ii. Pd-C, H_2 , MeOH, RT, 1 atm, 1h; (c) i. N-Cbz-D-alanine, DCC (2 eq), HOBT (4 eq), CH_2Cl_2 , RT, 18h; ii. Pd-C, H_2 , MeOH, RT, 1 atm, 1h; (d) Ref.11; (e) i. N-Ac-D-Phenylalanine, DCC (2 eq), HOBT (4 eq), CH_2Cl_2 , RT, 18h; ii. Pd-C, H_2 , MeOH, RT, 1 atm, 6h; (f) 8, DCC (2 eq), HOBT (4 eq), CH_3CN , RT, 18h.

lysis over 10% Pd-C at normal pressure and temperature gave (6) in almost quantitative yield. The product (6), without purification, was subjected to peptide bond formation¹⁰ with N-Cbz-D-alanine in the presence of DCC and HOBT in dry CH₂Cl₂. The structure and stereochemical integrity of the dipeptide (7) was established by ¹H-NMR spectrum. Subsequent deprotection of N-Cbz group by hydrogenolysis furnished the required segment (8).

The other dipeptide unit (12) containing D-allo-threonine, which is prone to elimination of water, had to be synthesised with care². The benzyl ester as a protecting group of the amino acid was considered most suitable because of the ease with which it could be removed under neutral conditions. Accordingly D-threonine (9) was converted¹¹ into D-allo-threonine benzyl ester (10) by the standard procedure. Treatment of (10) with N-acetyl-D-phenylalanine with DCC and HOBT as condensation catalysts afforded (11) (70%) whose benzyl ester group was subsequently cleaved by hydrogenolysis to generate (12).

The final condensation of (8) with (12) in the presence of DCC and HOBT in CH₃CN gave the peptide (13) (43%). The structure of (13) was supported by ¹H-NMR and Mass spectra. In the e.i. and c.i. mass spectrum, the fragmentation pattern conclusively confirmed the structure (13)¹². A series of fragmentation peaks as described in Fig. A were observed.

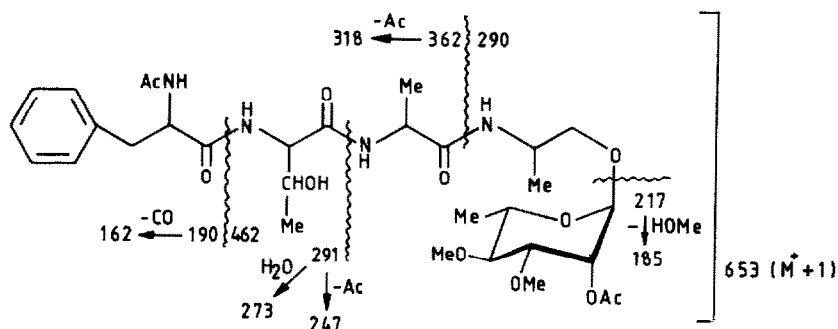


FIG. A

References and Notes

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9. NMR spectral data for some novel compounds:

- 5** ^{13}C -NMR (CDCl_3): δ 17.66, 20.87, 29.53, 46.52, 57.34, 60.70, 66.50, 67.73, 68.03, 70.81, 79.26, 81.58, 97.78, 127.94, 128.37, 136.40, 155.65, 170.18.
- 7** ^1H -NMR (CDCl_3): δ 1.10 (d, 3H, J = 6.0 Hz), 1.21 (d, 3H, J = 6.4 Hz), 1.29 (d, 3H, J = 6.4 Hz), 2.04 (s, 3H), 3.00 (t, 1H, J = 9.0 Hz), 3.29 (s, 3H), 3.48 (s, OMe), 3.9 (m, 2H), 4.61 (s, 1H), 5.06 (ABq, 2H), 5.14 (m, 1H), 5.50 (d, 1H, J = 8.5 Hz), 6.16 (d, 1H, J = 8.5 Hz), 7.28 (s, 5H).
- 11** ^1H -NMR (CDCl_3): δ 1.17 (d, 3H, J = 6.0 Hz), 1.94 (s, 3H), 2.5 (bs, 1H), 3.07 (dq, 2H, J = 5.6, 13.2 Hz), 4.33 (m, 1H), 4.60 (dd, 1H, J = 1.9, 7.5 Hz), 4.80 (dd, 1H, J = 7.5, 13.2 Hz), 5.20 (s, 2H), 6.30 (d, 1H, J = 7.5 Hz), 6.94 (d, 1H, J = 7.5 Hz), 7.1-7.5 (m, 10H).
- 13** ^1H -NMR (CDCl_3): δ 1.05 (d, 3H, J = 6.7 Hz), 1.17 (d, 3H, J = 6.0 Hz), 1.25 (d, 3H, J = 6.0 Hz), 1.34 (d, 3H, J = 6.5 Hz), 1.91 (s, 3H), 2.09 (s, 3H), 3.03 (m, 3H), 3.28 (m, 1H), 3.35 (s, 3H), 3.5 (m, 1H), 3.52 (s, 3H), 3.75 (m, 1H), 4.15 (m, 1H), 4.40 (t, 1H, J = 6.2 Hz), 4.62 (s, 1H), 5.15 (m, 1H), 6.08 (d, 1H, J = 7.8 Hz), 6.60 (d, 1H, J = 8.9 Hz), 6.76 (d, 1H, J = 7.8 Hz), 6.94 (d, 1H, J = 8.5 Hz), 7.2 (m, 5H).
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