

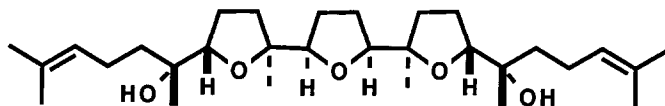
**A SHORT STEP SYNTHESIS OF TEURILENE. STEREOCONTROLLED SEQUENTIAL DOUBLE
 CYCLIZATION OF THE C₃₀-TETRAENETETRAOL TO THE TANDEM TETRAHYDROFURAN SYSTEM**

Masaru Hashimoto, Hiroko Harigaya, Mitsutoshi Yanagiya,
 and Haruhisa Shirahama*

Department of Chemistry, Faculty of Science, Hokkaido University,
 Sapporo 060, Japan

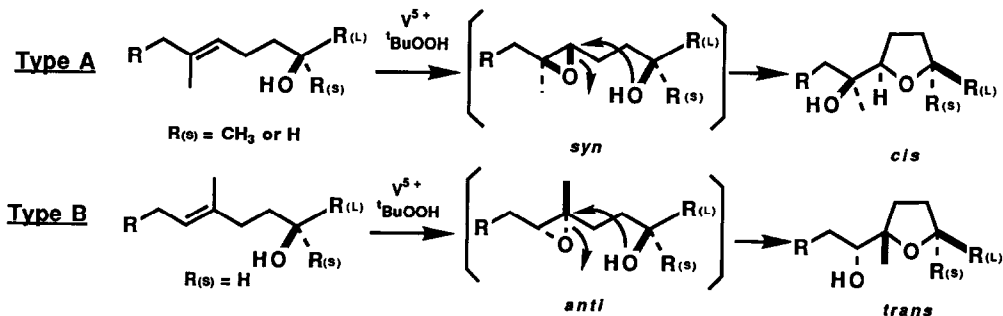
A *meso*-triterpene ether, teurilene was stereoselectively synthesized through one step formation of a link of two tetrahydrofurans by V⁵⁺ catalyzed oxidation of a C₃₀-tetraenetetraol derivative.

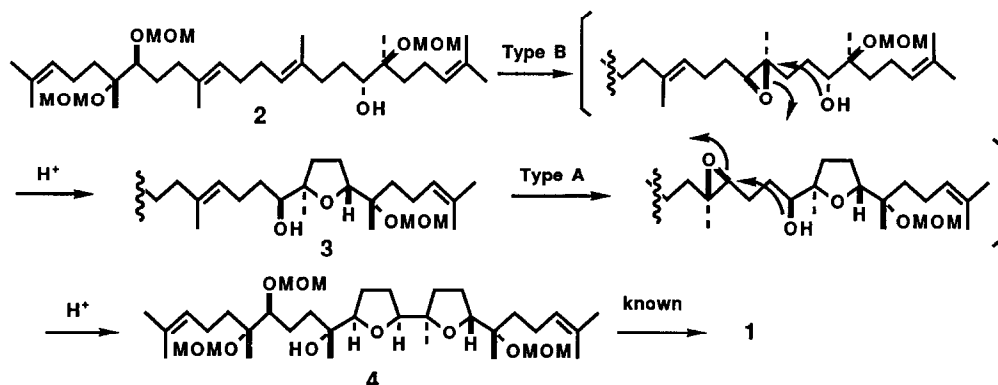
A marine triterpene, teurilene (1)¹⁾ is characterized by beautiful arrangement of eight asymmetric carbons and a link of three tetrahydrofurans in the center of its molecule. The decorous molecule arouses special interest in its synthesis and conformational properties.²⁾ We should like to report a short step synthesis of teurilene (1) by stereocontrolled sequential double cyclization of a C₃₀-tetraenetetraol derivative 2.



teurilene (1)

In the course of the studies on total synthesis of thyriferol,³⁾ we found a "rule" of V⁵⁺ catalyzed oxidation-cyclization of bishomoallyl alcohol system, i.e. a 5-substituted 4-en-1-ol gave a *cis*-2,5-disubstituted tetrahydrofuran through a *syn*-epoxide (Type A), while a 4-substituted 4-en-1-ol gave a *trans*-2,5,5-trisubstituted tetrahydrofuran through anti-epoxide (Type B)⁴⁾. This "rule" was extremely valuable to assemble stereoselectively 2,5-disubstituted tetrahydrofuran moieties and its usefulness was successfully demonstrated by application to the present synthesis.



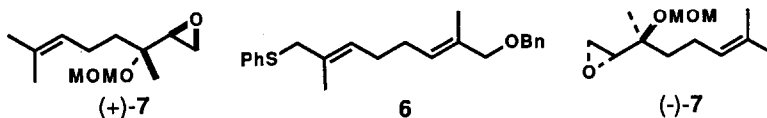


The synthesis was outlined above. A C₃₀-tetraenetetraol derivative **25**) was treated with VO(acac)₂ (0.01 eq.) and *t*-butylhydroperoxide (3 eq.) in the presence of AcOH (0.1 eq.) in benzene at 47 °C for 7 hr. The sequential epoxidation-cyclization of Type A and B proceeded smoothly and stereoselectively afforded a tandem tetrahydrofuran system **4** in a single step.⁶⁾ Assembling the third tetrahydrofuran ring through the known procedure^{2a)} furnished teurilene (**1**) which was completely identical with natural product by direct comparison of HPLC retention time, m.p., [α]_D, 400 MHz ¹H-NMR and IR.

Acknowledgement We are most grateful to Prof. Etsuro Kurosawa (Hokkaido University) for a generous gift of a sample of natural teurilene (**1**).

References and Notes

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- 2) a) M. Hashimoto, M. Yanagiya, and H. Shirahama, *Chem. Lett.*, **1988**, 645;
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- 3) M. Hashimoto, T. Kan, M. Yanagiya, H. Shirahama, and T. Matsumoto, *Tetrahedron Lett.*, **28**, 5665 (1987); M. Hashimoto, T. Kan, K. Nozaki, M. Yanagiya, H. Shirahama, and T. Matsumoto, *ibid.*, **29**, 1143 (1988).
- 4) Selectivity of "Type B" has been known: T. Fukuyama, B. Vranesic, D. P. Negri, and Y. Kishi, *Tetrahedron Lett.*, 2741 (1978).
- 5) Compound **2** was prepared from **6** by the following treatment, i) *n*-BuLi/TMEDA/(+)-**7**,^{2a)} ii) MOMCl/*i*-Pr₂NEt, iii) Na/*i*-PrOH/Δ, iv) Li/NH₃, v) PPh₃/CCl₄, vi) NaSPh, vii) *n*-BuLi/TMEDA/(-)-**7**,^{2a)} viii) Na/*i*-PrOH/Δ.



- 6) Bicyclic compound **4** was obtained in 25% yield with monocyclic compound **3** (30%) and polar materials. Longer time oxidation of **2** at higher temperature or employing larger amount of catalyst reduced considerably the yield of desired compound **4** (<10%).

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