A Direct Synthesis of 3-(Pyrrolidin-3-yl)indoles for Use As Conformationally Restricted Analogs of Tryptamines

John E. Macor,*¹ David H. Blank, Kevin Ryan, Ronald J. Post Department of Medicinal Chemistry, Central Research Division, Pfizer Inc, Groton, Connecticut 06340, USA Received 8 July 1996; revised 24 September 1996

An efficient, two step synthesis of 3-(pyrrolidin-3-yl)indoles 4 is described. Indoles react with maleimides in refluxing acetic acid affording 3-(indol-3-yl)succinimides 6. Reaction times and yields depend on the substituents on the indole 5. Direct reduction of the succinimides 6 with LAH affords the desired conformationally restricted tryptamine derivatives 4.

The use of conformationally restricted analogs of medicinally important molecules as a means to better understand or improve the pharmacology of those agents is a common theme in medicinal chemistry. Although often an empirical approach, the use of conformationally restricted analogs reveals positive or negative information concerning the active conformation of the biologically relevant molecule under study. In many cases, the conformationally restricted targets themselves can be synthetic challenges, and the use of new or improved methodology is often required to achieve the desired conformationally restricted target.

Tryptamines are a class of indole derivatives which have extensive medicinal utility. The best known tryptamine is serotonin (5-hydroxytryptamine, 5-HT, 1a), which is a ubiquitous neurotransmitter in the mammalian central nervous system (CNS). Since there are a large number of receptor proteins (fourteen at the present time) which bind this small, but powerful molecule, serotonergic receptor dysfunction has been implicated in a wide array of disease states.² Accordingly, conformational restriction of the 3-(2-aminoethyl) sidechain of serotonin has been examined as an approach to 5-HT analogs with receptor subtype selectivity, ideally affording drugs with specific therapeutic activity and reduced side effects. Our laboratories³⁻⁸ and others⁹ have been actively pursuing such studies via the restriction of the 3-(2-aminoethyl) sidechain of serotonin (1a). The structure of indoles 1-4 denotes the various themes employed thus far in both the primary and patent literature. While compounds of generic structures 2^{3-8} and 3^9 have been documented in the primary literature, the 3-(pyrrolidin-3-yl)indoles 4 have received almost no attention. While there has been some reference to 3-(pyrrolidin-3-yl)indoles 4 in the serotonergic patent literature, actual synthetic details of these conformationally restricted tryptamines are often lacking. Where synthetic details are given, the universal approach to these compounds 4 is via a Fischer indolization reaction utilizing an appropriate arylhydrazine and an aliphatic aldehyde (Scheme 1, path a). The syntheses of these components are multistep, and the yields from the Fischer indole reaction are highly variable. In short, a concise and versatile synthesis of 4 has not been found in the literature.

1

$$R_5$$
 R_5
 R_5

Furthermore, in a series of indole derivatives 1b, 2b, and 4b in which the C5 substituent was held constant and the C3 substituent integrated, various degrees of conformation restriction of the 2-(aminoethyl) sidechain, the

Scheme 1

444 Papers SYNTHESIS

3-(pyrrolidin-3-yl)indole analog **4b** was found to impart the greatest selectivity for the 5-HT_{1D} receptor, ⁶ a serotonin receptor subtype believed to play a role in migraine headaches. ¹¹ Hoping to optimize this receptor subtype selectivity, we sought a rapid, general synthesis of 3-(pyrrolidin-3-yl)indoles **4** capable of allowing extensive functionalization at C5 of the indole derivative. In this report, we detail such a synthesis.

Scheme 1 (paths b and d) summarizes retrosynthetically our pathway to the desired 3-(pyrrolidin-3-yl)indoles 4. Reduction of 3-(indol-3-yl)succinimides 6 using lithium aluminum hydride to form 3-(pyrrolidin-3-yl)indoles 4 has been previously reported by two groups in 1964 and 1966 (Scheme 1, paths b and c)¹⁰. They arrived at the succinimides 6 through a series of transformations which commenced with indoles and concluded with the cyclization of 2-(indol-3-yl)succinic acids with ammonia. This particular approach to 4 was limited by the cumbersome route used to obtain the 3-(indol-3-yl)succinimides 6.

However, from a single report from 1962 on the reaction of 2-phenylindole with N-methylmaleimide, 12 we saw this type of Michael reaction as an ideal approach to 3-(indol-3-yl)succinimides 6, incorporating in a single step the desired carbon framework of the 3-(pyrrolidin-3-yl)indoles 4 (Scheme 1, path d). To our knowledge, only two other examples of this indole/maleimide Michael reaction have been subsequently reported, both involving 2-(indol-2-yl)indole and N-methylmaleimide, and both during the pursuit of an alternative product. One group observed the reaction under BF₃ · Et₂O catalysis at room temperature, 13 while the other group only observed the reaction after refluxing the two components at 100 °C for 7 days. 14 Also, indole magnesium halides have been reacted with 3,4-dibromomaleimides affording 3,4-bisindol-3-ylmaleimides. 15,16 Therefore, the reaction of indoles with maleimides appeared to be a reasonable approach to the desired carbon framework of our targets, but the generality of the reaction was unexplored.

We saw the opportunity to combine the Michael reaction of indoles with maleimides forming 3-(indol-3-yl)succinimides 6 with the direct reductions of these succinimides with LiAlH₄ as an expeditious route to 3-(pyrrolidin-3yl)indoles 4, (Scheme 1, paths b and d and Scheme 2). Accordingly, Table 1 summarizes our detailed study of this general synthesis of these conformationally restricted analogs of tryptamines 4. The Michael reaction between 5-methoxyindole (5d) and three equivalents of N-methylmaleimide did not proceed in refluxing dioxane (101 °C), and even after seven days at reflux no addition product was detected. However, use of acetic acid as the reaction solvent led to complete consumption of 5-methoxyindole after 72 hours reflux affording the desired succinimide 6f in 80 % yield. This indicated that the Michael reaction was greatly facilitated by the acidic nature of the solvent used. This result, coupled with the ease of reaction between 2-(indol-2-yl)indole and N-methylmaleimide using BF₃, ¹³ suggests that maleimide is activated in these reactions by the presence of acid coordination to the amide oxygens which is needed for increased Michael acceptor reactivity.

$$R_{5}$$
 R_{6}
 R_{7}
 R_{7}
 R_{7}

LIAIH₄
THF,
$$\Delta$$
 R_6
 R_6
 R_6

Scheme 2

A wide variety of indoles successfully participated in this Michael condensation with maleimide, N-methylmaleimide, and N-benzylmaleimide (Table 1). While there appeared to be little effect on the success or rate of the reaction depending on the maleimide used, the rate and efficiency of these condensation reactions in acetic acid appeared to be related to the electronic nature of the indole used. Comparison of both the reaction times and yields of 6 using indoles with electron-donating substituents versus using indoles with electron-withdrawing substituents showed that indoles with electron-withdrawing substituents at C5 required longer reaction times and produced lower yields of the desired succinimide 6. Electron-withdrawing substituents at C5 decrease the net contribution of the enamine-like resonance structure in the reactive character of the indole. Because of their reduced enamine-like nature, electron poor indoles participated poorly in these Michael reactions relative to electron neutral or electron rich indoles. Accordingly, 5-nitroindole, 5-cyanoindole, and indole-5-carboxylic acid afforded poor yields of 6 over the longer reaction times, and significant amounts of the starting indole were still present when these three reactions were terminated. For the other indoles, the yield of succinimide 6 for these reactions was moderate to good (Table 1), and the starting indoles were fully consumed.

Reduction of the (indol-3-yl)succinimides was accomplished using an excess of lithium aluminum hydride in refluxing THF affording the 3-(pyrrolidin-3-yl)indoles 4 in good to excellent yields. When functionality susceptible to reduction was present in the indole portion of 6 (e.g. 6k and 6m), reduction of that group occurred concomitantly. We found that the order of addition of reagents could have a significant effect on the efficiency of the reduction reaction. When LAH was added to a solution of 6 in THF, a greater occurrence of byproducts and reduced yields of 4 were often seen. Addition of the succinimide to a mixture of excess LiAlH₄ in THF at room temperature, followed by a short reflux consistently gave excellent yields of the conformationally restricted tryptamine 4 (Table 1).

Table 1. Preparation of 3-(Pyrrolidin-3-yl)indoles 4 via Succinimides 6 (see Scheme 2)

Starting Materials				Time	Inter-	Yield	Prod-	Yield	Overall Yield (%)
Maleimide, R	Indole, 5	R ⁵	R ⁶	— (h)	mediate 6	(%)	uct 4	(%)	1 1010 (%)
Me	5a	Н	H	30	6a	74	4a	88	65
Me	5b	F	\mathbf{H}	24	6b	87	4b	79	69
Me	5c	Br	H	24	6c	62	4c	48	30
H	5c	Br	H	72	6d	61	4d		· · · · · · · · · · · · · · · · · · ·
CH ₂ Ph	5c	Br	\mathbf{H}	144	6e	23	4e	_	
Me ˜	5d	OMe	H	72	6 f	80	4f	90	73
Н	5d	OMe	\mathbf{H}	69	6g	61	4g	70	42
CH ₂ Ph	5d	OMe	H	19	6ĥ	57	4h	98	56
Me	5e	OCH ₂ O		72	6i	60	4i	76	46
Me	5f	OCH ₂ C≡CH	\mathbf{H}	24	6 j	31	4j	-	_
Me	5g	NH(Č=O)Ph	H	26	6k	52	4k	76ª	40
Me	5h	Me	H	24	61	74	41	60	44
Me	5i	CO ₂ H	Ħ	36	6m	41	4m	73 ^b	30
Me	5j	CN ²	\mathbf{H}	96	6n	39	4n	_	_
Me	5k	NO_2	\mathbf{H}	97	6o	2	40	_	_

^a Isolated as 5-benzylamino-(3-N-methylpyrrolidin-3-yl)indole.

In conclusion, a two step synthesis of 3-(pyrrolidin-3yl)indoles 4 has been described (Table 1). This concise approach to these conformationally restricted analogs of tryptamine commences with the Michael reaction of a variety of indoles with a variety of maleimides in refluxing acetic acid to afford 3-(indol-3-yl)succinimides 6 (Table 1). It would appear that this reaction is general in nature, but indoles with electron-withdrawing groups at C5 appear to suffer in both reaction rate and efficiency. This is likely a result of their reduced enamine-like character in the pyrrole portion of the indole heterocycle due to increased resonance delocalization of the indole nitrogen lone pair of electrons into the benzene portion of the indole molecule. Reduction of these succinimides 6 is smoothly accomplished via the addition of the succinimide 6 to a solution/mixture of LiAlH₄ in THF followed by a short reflux affording the desired 3-(pyrrolidin-3yl)indoles 4 in generally good to excellent yields (Table 1).

Melting points were determined on a Thomas-Hoover open capillary melting point apparatus and are uncorrected. IR spectra were obtained on a Nicolet 510 FT-IR Spectrometer. NMR spectra were recorded on either a Bruker AM-300 (300 MHz) or Bruker AM-250 (250 MHz) spectrometer. NMR data are reported in parts per million (δ) and are referenced to the deuterium lock signal from the sample solvent. EI mass spectra were obtained on a Kratos Profile instrument, and FAB mass spectra were obtained on a Kratos Concept IS instrument ($\sigma = 5$ ppm for HRMS). FAB mass spectrum are noted by a [MH]⁺ or [M·NH₄]⁺ designation for molecular ions. Elemental analyses were performed at Schwarzkopf Microanalytical Laboratory in Woodside, New York. Commercial reagents (Aldrich Chemical Co.) were utilized without further purification, including Aldrich anhydrous solvents. Chromatography refers to column chromatography performed using 32-63 μ m silica gel (approx 50 g silica gel per gram of material to be chromatographed) and executed under N2 pressure (flash chromatography)

3-(Indol-3-yl)succinimides 6; General Procedure:

A solution of the indole (10.0 mmol) and the maleimide (30.0 mmol) in AcOH (50 mL) was heated at reflux under N2 for a period of time determined by the individual components and shown in Table 1. Upon consumption of the starting indole as observed by TLC (or after the time shown in Table 1), the resulting mixture was cooled. In some cases, the desired succinimide 6 precipitated directly from the reaction mixture. Otherwise, the solution was evaporated under reduced pressure, and the residue was dissolved in EtOAc (50 mL) and a saturated solution of NaHCO₃ (50 mL). The EtOAc layer was removed, and the aqueous layer was extracted with EtOAc $(2 \times 50 \text{ mL})$. The organic extracts were combined, dried (MgSO₄), and evaporated under reduced pressure. The extraction residue was either crystallized directly or chromatographed using silica gel (ca 200 g) and eluted with an appropriate solvent system to afford the product 6.

3-(Indol-3-yl)-N-methylsuccinimide (6a): Indole (5a) and N-methylmaleimide were used. Filtration of the cooled reaction mixture afforded 6a (74%) as a yellow solid (Tables 2, 3, 6).

3-(5-Fluoroindol-3-yl)-N-methylsuccinimide (6b): 5-Fluoroindole (5b) and N-methylmaleimide were used. Filtration of the cooled reaction mixture afforded 6b (87%) as a yellow solid (Tables 2, 6). 3-(5-Bromoindol-3-yl)-N-methylsuccinimide (6c): 5-Bromoindole (5c) and N-methylmaleimide were used. Filtration of the cooled reaction mixture afforded 6c (62%) as a yellow solid (Tables 2, 3, 6). 3-(5-Bromoindol-3-yl)succinimide (6d): 5-Bromoindole (5c) and maleimide were used. The reaction mixture was evaporated under reduced pressure, and the residual solid was suspended in CH₂Cl₂ (75 mL). Undissolved solid was filtered to afford 6d (61%) as a yellow powder (Tables 2, 3, 6).

N-Benzyl-3-(5-bromoindol-3-yl)succinimide (6e): 5-Bromoindole (5c) and N-benzylmaleimide were used. The resulting reaction mixture was evaporated under reduced pressure, and the residue was chromatographed using a 0-1% EtOAc gradient in CH₂Cl₂ to afford a gray foam. Crystallization of this foam from Et₂O/CH₂Cl₂ (9:1) afforded **6e** (23%) as a pale yellow solid (Tables 2, 3, 6).

3-(5-Methoxyindol-3-yl)-N-methylsuccinimide (6f): 5-Methoxyindole (5d) and N-methylmaleimide were used. The resulting reaction mixture was evaporated under reduced pressure, and the residue was chromatographed using EtOAc/hexanes (1:1) to afford a foam. Crystallization of this foam in EtOAc afforded 6f (80%) as a pale yellow solid (Tables 2, 3, 6).

3-(5-Methoxyindol-3-yl)succinimide (6g): 5-Methoxyindole (5d) and maleimide were used. The resulting reaction mixture was evaporated under reduced pressure, and the residue was crystallized from EtOAc to afford 6g (61%) as a yellow solid (Tables 2, 3, 6). N-Benzyl-3-(5-methoxyindol-3-yl) succinimide (6h): 5-Methoxyindole (5d) and N-benzylmaleimide were used. The resulting reaction

b Isolated as 5-hydroxymethyl-3-(N-methylpyrrolidin-3-yl)indole.

446 Papers SYNTHESIS

mixture was evaporated under reduced pressure, and the residue was chromatographed using EtOAc/hexanes (1:1) to afford a pale yellow foam. Crystallization of this foam from Et₂O/CH₂Cl₂ (19:1) afforded **6h** (57%) as a white powder (Tables 2, 3, 6).

N-Methyl-3-(5,6-methylenedioxyindol-3-yl)succinimide (6i): 5,6-Methylenedioxyindole (5e) and N-methylmaleimide were used. Filtration of the cooled reaction mixture afforded 6i (60%) as a tan solid (Tables 2, 3, 6).

N-Methyl-3-(5-(prop-2-ynoxy)indol-3-yl)succinimide (6j):⁵ 5-(Prop-2-ynynloxy)indole (5f) and N-methylmaleimide were used. The resulting reaction mixture was evaporated under reduced pressure, and the residue was crystallized from EtOAc to afford 6j (31%) as a pale yellow solid (Tables 2, 3, 6).

3-(5-Benzoylaminoindol-3-yl)-N-methylsuccinimide (6k): 5-Benzoylaminoindole (5g) and N-methylmaleimide were used. Chromatography of the crude product using a gradient of EtOAc/hexanes (1:3 to 1:1) afforded 6k (52%) as a white solid (Tables 2, 3, 6).

N-Methyl-3-(5-methylindol-3-yl)succinimide (61): 5-Methylindole (5h) and N-methylmaleimide were used. Filtration of the cooled mixture afforded 61 (74%) as a tan solid (Tables 2, 3, 6).

3-(5-Carboxyindol-3-yl)-N-methylsuccinimide (6m): Indole-5-carboxylic acid (5i) and N-methylmaleimide were used. Filtration of the cooled reaction mixture afforded 6m (41%) as a off-white solid (Tables 2, 3, 6).

3-(5-Cyanoindol-3-yl)-N-methylsuccinimide (6n): 5-Cyanoindole (5j) and N-methylmaleimide were used. The reaction mixture was evaporated under reduced pressure, and the residue was chromatographed using EtOAc to afford 6n (39%) as a yellow solid (Tables 2, 3, 6).

N-Methyl-3-(5-nitroindol-3-yl)succinimide (60): 5-Nitroindole (5k) and N-methylmaleimide were used. The solvent was evaporated under reduced pressure, and the residue was chromatographed using EtOAc to afford 60 (2%) as a yellow foam. An analytical sample was prepared via crystallization from CHCl₃ to afford a pale yellow solid (Tables 2, 3, 6).

3-(Pyrrolidin-3-yl)indoles 4; General Procedure:

To a stirred solution of LiAlH₄ (1.00 g, 26.4 mmol, 5 eq) in anhyd THF (50 mL) at 0 °C was added the succinimide 6 (5.00 mmol) as a solid portionwise. The resulting mixture was heated at reflux under N₂ for a period of time determined by the individual succinimide 6 (6–24 h). Upon consumption of 6 as determined by TLC, the resulting mixture was cooled, and Na₂SO₄ · 10 H₂O (ca 15 g) was added cautiously portionwise, followed by H₂O (1 mL), and then by EtOAc (100 mL). The resulting mixture was then stirred at r.t. under N₂ for 16 h. The mixture was then filtered through Celite, and the filtrate was evaporated under reduced pressure to afford the desired 3-(pyrrolidin-3-yl)indole 4. If necessary, column chromatography was carried out using silica gel (approx 50 g) and elution with CH₂Cl₂/MeOH/NH₄OH (9:1:0.1) or other appropriate solvent system to give 4.

3-(N-Methylpyrrolidin-3-yl)indole (4a): 3-(Indol-3-yl)-N-methyl-succinimide (6a) was used, and the reaction mixture was heated at reflux for 24 h. Evaporation of the filtrate (from the workup described above) under reduced pressure afforded 4a (88%) as a white solid (Tables 4, 5, 7).

5-Fluoro-3-(N-methylpyrrolidin-3-yl) indole (4b): 3-(5-Fluoroindol-3-yl)-N-methylsuccinimide (6b) was used, and the reaction mixture was heated at reflux for 4.5 h. Evaporation of the filtrate (from the

Table 2. ¹H NMR Spectral Data for 6

6n

60

	A TAMA Special Zala 191 C
Prod- uct	1 H NMR (Solvent/TMS) δ , J (Hz)
6a	(DMSO- d_6): 11.05 (br s, NH), 7.40–7.34 (m, 2H), 7.08 (ddd, J = 1.8, 7.9, 8.2, 1H), 6.97 (ddd, J = 1.0, 7.9, 8.2, 1H), 4.36 (dd, J = 5.1, 9.4, 1H), 3.23 (dd, J = 9.4, 18.0, 1H), 2.93 (s, 3H), 2.80 (dd, J = 5.1, 18.0, 1H)
6b	(DMSO- d_6): 11.2 (br s, NH), 7.43 (d, $J = 2.4$, 1H), 7.37 (dd, $J = 4.5$, 8.8, 1H), 7.21 (dd, $J = 2.5$, 10.0, 1H), 6.99–6.92 (m, 1H), 4.36 (dd, $J = 5.3$, 9.3, 1H), 3.22 (dd, $J = 9.6$, 17.9, 1H), 2.92 (s, 3H), 2.83 (dd, $J = 5.3$, 17.9, 1H)
6c	$(DMSO-d_6)$: 11.3 (br s, NH), 7.67 (d, $J=1.8$, 1H), 7.41 (d, $J=2.3$, 1H), 7.35 (d, $J=8.8$, 1H), 7.21 (dd, $J=1.9$, 8.8, 1H), 4.39 (dd, $J=5.3$, 9.3, 1H), 3.22 (dd, $J=9.3$, 17.9, 1H), 2.91 (s, 3H), 2.85 (dd, $J=5.3$, 17.9, 1H)
6d	$(DMSO-d_6)$: 11.25 (br s, NH), 7.64 (d, $J=1.7$, 1H), 7.42 (d, $J=2.4$, 1H), 7.35 (d, $J=8.7$, 1H), 7.21 (dd, $J=1.8$, 8.6, 1H), 4.37 (dd, $J=5.5$, 9.4, 1H), 3.37 (br s, NH), 3.17 (dd, $J=9.5$, 18.0, 1H), 2.81 (dd, $J=5.5$, 18.0, 1H)
6e	(CDCl ₃): 8.37 (br s, NH), $7.45-7.40$ (m, 3H), $7.37-7.29$ (m, 3H), 7.20 (dd, $J=1.8$, 8.6, 1H), 7.06 (d, $J=8.8$, 1H), 6.89 (d, $J=2.5$, 1H), 4.76 (s, 2H), 4.19 (dd, $J=5.2$, 9.5, 1H), 3.22 (dd, $J=9.5$, 18.4, 1H), 2.84 (dd, $J=5.2$, 18.4, 1H)
6f	$(DMSO-d_6)$: 10.9 (br s, NH), 7.29 (d, $J=1.9$, 1H), 7.27 (d, $J=8.4$, 1H), 6.88 (d, $J=2.4$, 1H), 6.77 (dd, $J=2.4$, 8.8, 1H), 4.33 (dd, $J=5.0$, 9.3, 1H), 3.74 (s, 3H), 3.24 (dd, $J=9.3$, 17.9, 1H), 2.93 (s, 3H), 2.80 (dd, $J=5.0$, 17.9, 1H)
6g	$(DMSO-d_6)$: 11.3 (br s, NH), 10.85 (br s, NH), 7.29–7.26 (m, 2H), 6.92 (d, $J=2.3$, 1H), 6.78 (dd, $J=2.4$, 8.8, 1H), 4.31 (dd, $J=5.3$, 9.4, 1H), 3.75 (s, 3H), 3.19 (dd, $J=9.5$, 18.0, 1H), 2.77 (dd, $J=5.3$, 18.0, 1H)
6h	(acetone- d_6): 10.06 (br s, NH), 7.40–7.34 (m, 2H), 7.34–7.25 (m, 5H), 6.87 (d, $J = 2.4$, 1H), 6.77 (dd, $J = 2.4$, 8.8, 1H), 4.71 (s, 2H), 4.44 (dd, $J = 5.0$, 9.5, 1H), 3.64 (s, 3H), 3.35 (dd, $J = 9.5$, 18.1, 1H), 2.88 (dd, $J = 5.0$, 18.1, 1H)
6i	$(DMSO-d_6)$: 10.8 (br s, NH), 7.14 (d, $J=2.1$, 1H), 6.89 (s, 1H), 6.88 (s, 1H), 5.93 (d, $J=0.7$, 1H), 5.92 (d, $J=0.7$, 1H), 4.27 (dd, $J=5.1$, 9.3, 1H), 3.20 (dd, $J=9.3$, 17.9, 1H), 2.90 (s, 3H), 2.77 (dd, $J=5.1$, 17.9, 1H)
6 j	(acetone- d_6): 10.15 (fr s, NH), 7.35 (d, $J = 8.8$, 1H), 7.32 (d, $J = 1.7$, 1H), 7.10 (d, $J = 2.4$, 1H), 6.86 (dd, $J = 2.4$, 8.8, 1H), 4.75 (d, $J = 2.4$, 2H), 4.38 (dd, $J = 5.0$, 9.4, 1H), 3.31 (dd, $J = 9.4$, 18.0, 1H), 3.04 (t, $J = 2.4$, 1H), 2.99 (s, 3H), 2.82 (dd, $J = 5.0$, 18.0, 1H)
6k	$(DMSO-d_6)$: 11.03 (br s, NH), 10.13 (s, NH), 7.95 (dd, $J=8.8$, 1.7, 2H), 7.82 (d, $J=1.4$, 1H), 7.59–7.45 (m, 4H), 7.33 (d, $J=8.6$, 1H), 7.33 (d, $J=2.4$, 1H), 4.32 (dd, $J=9.4$, 5.0, 1H), 3.22 (dd, $J=17.9$, 9.4, 1H), 2.94 (s, 3H), 2.74 (dd, $J=17.9$, 5.0, 1H)
61	$(DMSO-d_6)$: 10.9 (br s, NH), 7.27 (d, $J=2.2$, 1H), 7.26 (d, $J=8.3$, 1H), 7.17 (s, 1H), 6.93 (dd, $J=1.2$, 8.3, 1H), 4.32 (dd, $J=5.0$, 9.3, 1H), 3.22 (dd, $J=9.4$, 17.9, 1H), 2.93 (s, 3H), 2.78 (dd, $J=5.1$, 18.0, 1H), 2.36 (s, 3H)
6m	$(DMSO-d_6)$: 12.45 (br s, CO_2H), 11.4 (br s, NH), 8.16 (s, 1H), 7.73 (dd, $J=1.5$, 8.6, 1H), 7.46 (d, $J=2.1$, 1H), 7.42 (d, $J=8.6$,

 $(DMSO-d_6)$: 11.65 (br s, NH), 8.09 (s, 1H), 7.58-7.53 (m, 2H), 7.46 (d, J=8.4, 1H), 4.46 (dd, J=5.6, 9.2, 1H), 3.24 (dd,

 $(DMSO-d_6)$: 11.83 (br s, NH), 8.58 (d, J = 2.1, 1H), 8.02 (dd, J = 2.2, 9.0, 1H), 7.65 (d, J = 2.0, 1H), 7.55 (d, J = 9.0, 1H), 4.56

1H), 4.44 (dd, J = 5.1, 9.3, 1H), 3.23 (dd, J = 9.3, 17.9, 1H), 2.90 (s, 3H), 2.90–2.82 (m, 1H)

(dd, J = 5.8, 9.1, 1H), 3.26 (dd, J = 9.3, 18.0, 1H), 2.96 (dd, J = 5.4, 18.0, 1H), 2.92 (s, 3H)

J = 9.2, 17.9, 1H), 2.92 (s, 3H), 2.90 (dd, J = 5.6, 17.9, 1H)

61

6m

6n

60

36.3, 24.7

Table 3. 13C NMR Spectral Data of 6

Prod- uct	13 C NMR (Solvent/TMS) δ , J (Hz)
6a	(DMSO- <i>d</i> ₆): 178.5, 176.7, 136.5, 126.0, 123.5, 121.4, 118.8, 118.4, 111.7, 110.8, 37.6, 36.2, 24.6
6c	(DMSO- <i>d</i> ₆): 178.2, 176.5, 135.1, 128.2, 124.9, 123.9, 121.0, 113.7, 111.4, 110.6, 37.3, 35.8, 24.6
6 d	(DMSO- <i>d</i> ₆): 179.7, 178.0, 135.2, 128.1, 124.8, 123.9, 121.0, 113.7, 111.4, 110.8, 38.9, 37.1
6e	(CDCl ₃): 177.9, 175.8, 135.7, 135.2, 128.9, 128.8, 128.2, 127.3, 125.6, 123.5, 121.0, 113.3, 113.1, 110.9, 42.8, 38.0, 36.3
6f	(DMSO- <i>d</i> ₆): 178.5, 176.8, 153.2, 131.6, 126.5, 123.9, 112.4, 111.3, 110.5, 100.5, 55.4, 37.6, 36.1, 24.6
6g	(DMSO- <i>d</i> ₆): 179.9, 178.1, 153.2, 131.6, 126.5, 123.8, 112.3, 111.3, 110.7, 100.6, 55.4, 38.9, 37.4
6h	(acetone-d ₆): 178.7, 176.8, 154.9, 137.7, 133.0, 129.3, 129.1, 128.4, 127.5, 124.3, 113.2, 112.9, 112.2, 101.1, 55.8, 42.8, 39.0, 37.0
6i	(DMSO- <i>d₆</i>): 178.4, 176.6, 144.2, 142.0, 131.3, 121.7, 119.9, 111.1, 100.2, 97.3, 92.4, 37.6, 36.0, 24.6
6j	(acetone- <i>d</i> ₆): 178.8, 176.9, 152.7, 133.4, 127.3, 124.5, 113.2, 113.1, 112.3, 103.5, 80.3, 76.4, 57.1, 38.7, 37.0, 24.8
6k	(DMSO-d ₆): 178.5, 176.6, 165.2, 135.5, 133.6, 131.2, 131.2,

128.3, 127.6, 125.6, 124.3, 116.5, 111.5, 110.9, 110.3, 37.7,

(DMSO-d₆): 178.5, 176.7, 134.9, 127.2, 126.3, 123.5, 123.0,

 $(DMSO-d_6)$: 178.3, 176.6, 168.3, 138.9, 134.6, 125.9, 124.9,

122.7, 121.4, 112.3, 111.4, 37.3, 35.9, 24.6 (DMSO-*d*₆): 178.0, 176.4, 140.6, 139.6, 127.0, 125.9, 116.8,

116.4, 113.8, 112.2, 37.1, 35.5, 24.7 (DMSO-*d*₆): 178.0, 176.4, 140.6, 139.5, 126.9, 125.9, 116.8,

118.0, 111.4, 110.2, 37.6, 36.2, 24.6, 21.3

116.3, 113.7, 112.1, 37.1, 35.5, 24.7

3H), 2.39-2.29 (m, 1H), 2.03-1.92 (m, 1H)

Table 5. 13C NMR Spectral Data of 4

Prod-

¹³C NMR (Solvent/TMS)

uct	δ
4a	(DMSO- <i>d</i> ₆): 136.7, 126.4, 121.1, 120.9, 118.9, 118.1, 111.5,
4	62.7, 55.9, 42.2, 34.8, 32.0
4c	(DMSO- <i>d</i> ₆): 135.4, 128.2, 123.4, 122.9, 121.1, 118.3, 113.5, 110.8, 62.6, 56.0, 42.1, 34.4, 32.2
4f	(CD ₃ OD): 154.8, 133.8, 128.4, 122.5, 118.0, 113.1, 112.6, 101.9, 63.3, 57.0, 56.4, 42.5, 36.4, 32.8
4h	(CDCl ₃): 153.7, 139.3, 131.9, 128.9, 128.2, 127.2, 126.9, 121.0, 119.7, 112.0, 111.8, 101.7, 60.8, 56.0, 54.3, 34.6, 31.6
4i	(DMSO- <i>d</i> ₆): 143.8, 141.5, 131.5, 120.2, 119.6, 118.6, 100.0, 97.7, 92.2, 62.6, 56.0, 42.1, 34.8, 32.1
4k	(acetone-d ₆): 142.6, 141.9, 128.9, 128.2, 127.2, 121.2, 118.5, 112.3, 112.3, 101.7, 63.5, 56.9, 49.6, 42.4, 36.3, 33.7
41	(DMSO- <i>d</i> ₆): 135.1, 126.7, 126.4, 122.5, 121.1, 118.5, 117.5, 111.2, 62.7, 55.9, 42.2, 34.8, 32.0, 21.4
4m	(CDCl ₃): 136.3, 132.1, 127.0, 122.0, 120.6, 119.7, 118.3, 111.3, 66.2, 62.7, 56.2, 42.4, 35.0, 32.1

Table 4. ¹H NMR Spectral Data of 4

¹H NMR (Solvent/TMS)

uct	$\delta, J ext{ (Hz)}$
4a	(DMSO- d_6): 10.8 (br s, NH), 7.58 (d, $J = 7.8$, 1H), 7.33 (d, $J = 8.1$, 1H), 7.13 (d, $J = 2.1$, 1H), 7.08–7.03 (m, 1H), 6.98–6.93 (m, 1H), 3.59–3.48 (m, 1H), 2.95 (t, $J = 8.3$, 1H), 2.74–2.66 (m, 1H), 2.57–2.42 (m, 2H), 2.31 (s, 3H), 2.32–2.19 (m, 1H),
	1.95–1.83 (m, 1H)
4b	$(CDCl_3)$: 8.36 (brs, NH), 7.30 (dd, $J = 2.5$, 9.8, 1H), 7.24 (dd, $J = 4.3$, 8.8, 1H), 7.03 (d, $J = 2.1$, 1H), 6.99–6.89 (m, 1H),
	3.68-3.57 (m, 1H), 3.08 (t, $J = 9.0$, 1H), $2.89-2.81$ (m, 1H), $2.71-2.64$ (m, 1H), 2.60 (dd, $J = 8.0$, 9.0 , 1H), 2.45 (s, 3H), $2.43-2.33$
	(m, 1H), 2.05–1.94 (m, 1H)
4c	$(DMSO-d_6)$: 11.0 (br s, 1H), 7.74 (d, $J = 2.3$, 1H), 7.29 (d, $J = 8.8$, 1H), 7.20 (s, 1H), 7.15 (dd, $J = 1.7$, 8.6, 1H), 3.55–3.45 (m,
	1H), 2.88 (t, $J = 8.2$, 1H), 2.65–2.42 (m, 3H), 2.29 (s, 3H), 2.29–2.15 (m, 1H), 1.93–1.77 (m, 1H)
4f	(CD_3OD) : 7.20 (d, $J = 8.8$, 1H), 7.02 (s, 1H), 7.01 (d, $J = 2.4$, 1H), 6.74 (dd, $J = 2.3$, 8.8, 1H), 4.90 (s, DHO), 3.80 (s, 3H),
	3.67-3.55 (m, 1H), 3.13 (dd, $J = 8.0, 9.2, 1$ H), $2.94-2.86$ (m, 1H), $2.67-2.53$ (m, 2H), 2.41 (s, 3H), $2.41-2.29$ (m, 1H), $2.07-1.95$
	(m, 1H)
4g	(CD_3OD) : 7.21 (d, $J = 8.8$, 1H), 7.02 (d, $J = 2.0$, 1H), 6.75 (dd, $J = 2.4$, 8.8, 1H), 4.89 (s, approx 2 exchangeable H), 3.80 (s,
	3H), $3.49 - 3.36$ (m, $1H$), $3.33 - 3.28$ (m, $1H$), $3.12 - 2.96$ (m, $2H$), 2.84 (dd, $J = 8.2$, 10.6 , $1H$), $2.29 - 2.18$ (m, $1H$), $2.00 - 1.87$ (m, $1H$)
4h	$(CDCl_3)$: 8.09 (br s, 1H), 7.39–7.23 (m, 5H), 7.19 (d, $J = 9.0$, 1H), 7.14 (d, $J = 1.9$, 1H), 6.92 (d, $J = 2.4$, 1H), 6.85 (dd, $J = 2.3$,
	9.0, 1H), 3.85 (s, 3H), 3.75 (d, $J = 13.1$, 1H), 3.69 (d, $J = 12.9$, 1H), 3.70–3.59 (m, 1H), 3.14 (t, $J = 8.4$, 1H), 2.93–2.84 (m, 1H),
	2.75-2.66 (m, 1H), 2.64 (t, $J=8.6$, 1H), $2.42-2.28$ (m, 1H), $2.05-1.92$ (m, 1H)
4i	$(DMSO-d_6)$: 10.55 (br s, NH), 7.05 (s, 1H), 6.95 (d, $J=2.1$, 1H), 6.85 (s, 1H), 5.90 (s, 2H), 3.49–3.38 (m, 1H), 2.89 (t, $J=8.3$,
41	1H), $2.67-2.58$ (m, 1H), $2.58-2.50$ (m, 1H), 2.42 (t, $J=8.1$, 1H), 2.29 (s, 3H), $2.27-2.15$ (m, 1H), $1.89-1.76$ (m, 1H)
4k	(acetone- d_6): 9.53 (br s, NH), 7.44 (d, $J = 7.4$, 2H), 7.29 (t, $J = 7.4$, 2H), 7.19 (t, $J = 7.2$, 1H), 7.10 (d, $J = 8.7$, 1H), 6.95 (s, 1H),
	6.83 (d, $J = 2.1$, 1H), 6.63 (dd, $J = 8.6$, 2.2, 1H), 3.51–3.43 (m, 1H), 2.87 (t, $J = 8.7$, 1H), 2.71–2.62 (m, 1H), 2.51–2.39 (m, 2H), 2.20 (2.11), 2.21 (2.22) (m, 2H), 2.22 (m, 2H
43	2H), 2.29 (s, 3H), 2.34–2.23 (m, 1H), 1.96–1.83 (m, 1H)
41	(DMSO- d_6): 10.6 (br s, NH), 7.34 (s, 1H), 7.21 (d, $J = 8.2$, 1H), 7.07 (d, $J = 2.1$, 1H), 6.88 (dd, $J = 2.1$, 8.2, 1H), 3.55–3.47 (m,
	1H), 2.96 (t, $J = 8.2$, 1H), $2.74 - 2.67$ (m, 1H), $2.55 - 2.47$ (m, 1H), 2.43 (t, $J = 8.3$, 1H), 2.37 (s, 3H), 2.30 (s, 3H), $2.30 - 2.17$ (m,
	1H), 1.93–1.82 (m, 1H)
4m	$(CDCl_3)$: 8.33 (br s, NH), 7.57 (s, 1H), 7.31 (d, $J = 8.3$, 1H), 7.19 (dd, $J = 1.5$, 8.3, 1H), 6.97 (d, $J = 1.7$, 1H), 4.74 (s, 2H),
	3.68-3.57 (m, 1H), 3.06 (t, $J=8.8$, 1H), 2.95 (brs, OH), $2.86-2.79$ (m, 1H), $2.65-2.56$ (m, 1H), 2.56 (t, $J=8.8$, 1H), 2.39 (s, $2.20-2.20$ (c, $3.20-2.20$ (c, $3.20-2.20$ (c, $3.20-2.20$ (c, $3.20-2.20$ (c, $3.20-2.20$ (c)

Table 6. Compounds 6a-o Prepareda

Prod- uct	Yield (%)	mp (°C)	IR (KBr) ν (cm ⁻¹)	MS m/z (%)
6a	74	235 (dec.)	1681, 1652, 1618	228 (M ⁺ , 75), 143 (100), 83 (91)
6b	87	200 (dec.)	1769, 1683, 1582, 1486, 1459, 1454, 1440, 1418	246 (M ⁺ , 24), 161 (100), 133 (34)
6c	62	198199	1771, 1724, 1691, 1473, 1462, 1453, 1435, 1423	308 (M ⁺ with ⁸¹ Br, 38), 306 (M ⁺ with ⁷⁹ Br, 40), 223 (93), 221 (100)
6d	61	198-200	1777, 1709, 1461, 1420	294 (M ⁺ with ⁸¹ Br, 94), 292 (M ⁺ with ⁷⁹ Br, 100), 223 (97), 221 (97)
6e	23	167–170	1774, 1698, 1494, 1455, 1435, 1423, 1401	402 ($[M \cdot NH_4]^+$ with ⁸¹ Br, 100), 400 ($[M \cdot NH_4]^+$ with ⁷⁹ Br, 88)
6f	80	146–148	1775, 1699, 1625, 1585, 1549, 1486, 1438	259 (21), 258 (M ⁺ , 100), 243 (6), 173 (47), 158 (35), 130 (24)
6g	61	180 (dec.)	1779, 1714, 1487, 1456, 1442, 1417	245 (14), 244 (M ⁺ , 100), 173 (91), 158 (40)
6h	57	139.5–141.5	1698, 1582, 1487, 1456, 1429, 1399	336 (22), 335 (MH ⁺ , 100)
6i	60	260-262	1768, 1751, 1687, 1634, 1612, 1555, 1507	273 (15), 272 (M ⁺ , 100), 240 (100), 187 (61), 137 (46), 81 (100)
6j	31	179–180	2125, 1772, 1689, 1683, 1484, 1467, 1461, 1442, 1409	283 (11), 282 (M ⁺ , 54), 243 (100), 215 (34), 197 (34), 158 (20)
6k	52	226.5-227.5	3270 (br), 1690, 1645, 1490	348 (MH ⁺ , 100), 332 (2), 275 (4), 263 (5)
61	74	180-182	1767, 1745, 1688, 1583, 1488, 1440, 1415	243 (11), 242 (M ⁺ , 71), 157 (100), 83 (91)
6m	41	290 (dec.)	1767, 1718, 1675, 1622, 1481, 1459, 1439, 1412	273 (13), 272 (M ⁺ , 71), 187 (100), 170 (13)
6n	39	260 (dec.)	2220, 1771, 1694, 1633, 1619, 1470, 1440	254 (17), 253 (M ⁺ , 77), 168 (100)
60	2	242–245	1773, 1693, 1626, 1581, 1548, 1520, 1473, 1440, 1415	274 (13), 273 (M ⁺ , 55), 188 (25), 137 (21), 81 (100)

^a Satisfactory microanalyses (C \pm 0.35, H \pm 0.32, N \pm 0.48) and HRMS values (\pm 0.0066 amu) obtained.

Table 7. Compounds 4 Prepared^a

Prod- act	Yield (%)	mp (°C)	IR (KBr) v (cm ⁻¹)	MS m/z
1a	88	99-102	1620, 1603, 1579, 1545, 1502, 1479, 1453, 1418	201 (20), 200 (M ⁺ , 100), 156 (55), 143 (68), 130 (54), 115 (36)
lb	79	98.5-101.5	1629, 1613, 1582, 1498, 1484, 1483, 1470, 1449, 1416	219 (14), 218 (M ⁺ , 100), 200 (71), 169 (53)
1c	48	163-164		280 (M ⁺ with ⁸¹ Br, 99), 278 (M ⁺ with ⁷⁹ Br, 100), 223 (17), 221 (17)
lf	90	73-74	1624, 1583, 1483, 1451	231 (13), 230 (M ⁺ , 100), 215 (5), 173 (23), 156 (10)
lg	70	100 (dec.)	1682, 1624, 1582, 1483, 1454	217 (16), 216 (M ⁺ , 100), 187 (52), 174 (80), 160 (36)
h	98	foam	1630, 1481, 1460	306 (M ⁺ , 31), 163 (32), 159 (28), 133 (69), 91 (100)
li	76	110-112.5	1562, 1548, 1514, 1499, 1481, 1466, 1419	245 (14), 244 (M ⁺ , 100), 187 (41), 161 (43), 57 (70)
k	76	foam	3405–3027 (br), 1627, 1585, 1481, 1468, 1452	306 (MH ⁺ , 100), 263 (4), 248 (4), 223 (8)
1	60	105–107	1628, 1606, 1599, 1583, 1545, 1478, 1451, 1443, 1417	214 (M ⁺ , 43), 171 (26), 170 (29), 157 (21), 144 (18), 57 (100)
m	73	210 (dec.)	3231 (br), 1583, 1481, 1449	232 (67), 231 (MH ⁺ , 100)

^a Satisfactory microanalyses (C \pm 0.46, H \pm 0.27, N \pm 0.36, exception: 4i, H - 0.58) and HRMS values (\pm 0.0060 amu) obtained.

workup described above) under reduced pressure afforded 4b (79%) as a crystalline, tan solid (Tables 4, 7).

5-Bromo-3-(N-methylpyrrolidin-3-yl)indole (4c): 3-(5-Bromoindol-3-yl)-N-methylsuccinimide (6c) was used. The reaction mixture was heated at reflux for 3 h. The filtrate (from the procedure described above) was evaporated under reduced pressure. The residue was chromatographed using elution with 1% $\rm Et_3N$ in MeOH to afford 4c (48%) as an tan solid (Tables 4, 5, 7).

5-Methoxy-3- (N-methylpyrrolidin-3-yl) indole (4f): 3- (5-Methoxyin-1) indo

dol-3-yl)-N-methylsuccinimide (6f) was used. The reaction mixture was heated at reflux for 7 h. The filtrate (from the procedure described above) was evaporated under reduced pressure to afford 4f (90%) as a white solid (Tables 4, 5, 7).

5-Methoxy-3-(pyrrolidin-3-yl)indole (4g): 3-(5-Methoxyindol-3-yl)succinimide (6g) was used. The reaction mixture was heated at reflux for 6.5 h. The filtrate (from the procedure described above) was evaporated under reduced pressure to afford 4g (70%) as a white foam (Tables 4, 5, 7).

- 3-(N-Benzylpyrrolidin-3-yl)-5-methoxyindole (4h): N-Benzyl-3-(5-methoxyindol-3-yl)succinimide (6h) was used. The reaction mixture was heated at reflux for 6 h. The filtrate (from the procedure described above) was evaporated under reduced pressure. The residue was chromatographed using elution with EtOAc to afford 4h (98%) as a pale green foam (Tables 4, 5, 7).
- 5,6-Methylenedioxy-3-(N-methylpyrrolidin-3-yl)indole (4i): N-Methyl-3-(5,6-methylenedioxyindol-3-yl)succinimide (6i) was used. The reaction mixture was heated at reflux for 6 h. The filtrate (from the procedure described above) was evaporated under reduced pressure to afford 4i (76%) as a pale red foam (Tables 4, 5, 7).
- 5-Benzylamino-3-(N-methylpyrrolidin-3-yl)indole (4k): 3-(5-Benzo-ylaminoindol-3-yl)-N-methylsuccinimide (6k) was used. The reaction mixture was heated at reflux for 6 h. Chromatography using elution with EtOAc/MeOH/Et₃N (9:0:1 to 8:1:1) afforded 4k (76%) as a pale yellow foam (Tables 4, 5, 7).
- 5-Methyl-3-(N-methylpyrrolidin-3-yl)indole (41): N-Methyl-3-(5-methylindol-3-yl)succinimide (61) was used. The reaction mixture was heated at reflux for 6 h. The filtrate (from the procedure described above) was evaporated under reduced pressure to afford 41 (60%) as a white foam (Tables 4, 5, 7).
- 5-Hydroxymethyl-3-(N-methylpyrrolidin-3-yl)indole (4m): 3-(5-Carboxyindol-3-yl)-N-methylsuccinimide (6m) was used. The reaction mixture was heated at reflux for 12 h. The filtrate (from the procedure described above) was evaporated under reduced pressure to afford 4m (73%) as a white foam (Tables 4, 5, 7).
- Present Address: Bristol-Myers Squibb Pharmaceutical Research Institute, P.O. Box 4000, Princeton, NJ 08543-4000, USA.

- (2) Glennon, R.A. Neuroscience and Behavioral Reviews 1990, 14, 35
- (3) Macor, J.E.; Blake, J.; Fox, C.A.; Johnson, C.; Koe, B.K.; Lebel, L.A.; Morrone, J. M.; Ryan, K.; Schmidt, A. W.; Schulz, D. W.; Zorn, S. H. J. Med. Chem. 1992, 35, 4503.
- (4) Macor, J.E.; Blank, D.H.; Post, R.J.; Ryan, K. Tetrahedron Lett. 1992, 33, 8011.
- (5) Macor, J.E.; Blank, D.H.; Post, R.J. Tetrahedron Lett. 1994, 35, 45.
- (6) Macor, J.E.; Blank, D.H.; Fox, C.B.; Lebel, L.A.; Newman, M.E.; Post, R.J.; Ryan, K.; Schmidt, A.W.; Schulz, D.W.; Koe, B.K. J. Med. Chem. 1994, 37, 2509.
- (7) Macor, J.E.; Blank, D.H.; Desai, K.; Fox, C.B.; Koe, B.K.; Lebel, L.A.; Post, R.J.; Schmidt, A.W.; Schulz, D.W.; Seymour, P.A. Biorg. Med. Chem. Lett. 1995, 5, 2391.
- (8) Macor, J.E.; Olgilvie, R.J.; Wythes, M.J. Tetrahedron Lett. 1996, 37, 4289.
- (9) King, F.D.; Brown, A.M.; Gaster, L.M.; Kaumann, A.J.; Medhurst, A.D.; Parker, S.G.; Parsons, A.A.; Patch, T.L.; Raval, P. J. Med. Chem. 1993, 36, 1918.
- (10) Julia, M.; Bagot, J. Bull. Soc. Chim. France 1964, 8, 1924.
 Perron, Y.G.; Minor, W.F.; Bierwagen, M.E.; Ridlon, S.A.;
 Pindell, M.H. J. Med. Chem. 1966, 9, 136.
- (11) Peroutka, S.J. Pharmacol. Rev. 1991, 43, 579
- (12) Colonna, M.; Monti, A. Gazz. Chim. Ital. 1962, 92, 1401.
- (13) Somei, M.; Kodama, A. Heterocycles 1992, 34, 1285.
- (14) Barry, J. F.; Wallace, T. W.; Walshe, N. D. A. *Tetrahedron* **1995**, 51, 12797, and references cited therein.
- (15) Harris, W.; Hill, C.H.; Keech, E.; Malsher, P. Tetrahedron Lett. 1993, 34, 8361.
- (16) Xie, G.; Lown, J.W. Tetrahedron Lett. 1994, 35, 5555.