

Easy Entry to Terminal 1,5- and 1,6-Enynes from (Indenyl)ruthenium(II) Allenylidenes and Grignard Reagents

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Summary: The allenylidene complexes $[\text{Ru}(\text{C}=\text{C}=\text{CR}^1\text{R}^2)(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2][\text{PF}_6]$ ($\text{R}^1 = \text{R}^2 = \text{Ph}$ (**1a**); $\text{R}^1 = \text{H}$, $\text{R}^2 = \text{Ph}$ (**1b**), $\text{C}(\text{Me})=\text{CPh}_2$ (**1c**); $\text{CR}^1\text{R}^2 = (1R)\text{-1,3,3-trimethylbicyclo[2.2.1]hept-2-ylidene}$ (**1d**)) react with Grignard reagents $\text{CH}_2=\text{CH}(\text{CH}_2)_n\text{MgBr}$ ($n = 1, 2$) to yield the σ -alkynyl derivatives $[\text{Ru}\{\text{C}\equiv\text{CCR}^1\text{R}^2(\text{CH}_2)_n\text{-CH}=\text{CH}_2\}(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2]$ (**2a–d**, **3a–c**). Protonation of **2a–d** and **3a–c** with $\text{HBF}_4\cdot\text{Et}_2\text{O}$ leads to the vinylidene complexes $[\text{Ru}\{\text{C}=\text{C}(\text{H})\text{CR}^1\text{R}^2(\text{CH}_2)_n\text{CH}=\text{CH}_2\}(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2][\text{BF}_4]$ (**4a–d** and **5a–c**), which can be easily demetalated with acetonitrile to afford the terminal enynes $\text{HC}\equiv\text{CCR}^1\text{R}^2(\text{CH}_2)_n\text{CH}=\text{CH}_2$ (**6a–d** and **7a–c**) and the nitrile complex $[\text{Ru}(\text{N}\equiv\text{CMe})(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2][\text{BF}_4]$ (**8**).

Introduction

The Nicholas reaction (Scheme 1; path A) has found elegant applications in organic synthesis, especially in the field of carbon–carbon bond formation.¹ This type of propargylic substitution proceeds via dicobalt hexacarbonyl– π -alkyne complexes containing coordinated propargyl cations (**A**).² These intermediates selectively react at C-3 with a wide variety of heteroatom- and carbon-centered nucleophiles to furnish functionalized propargylic derivatives after oxidative demetalation. In this regard, the cationic transition-metal allenylidene complexes $[\text{M}]^+=\text{C}=\text{C}=\text{CR}^1\text{R}^2$, readily available by dehydration of 2-propyn-1-ols upon coordination to an unsaturated metal center,³ can be considered as stabilized propargyl cations, due to the extensive contribution of the metal–alkynyl resonance form $[\text{M}]-\text{C}\equiv\text{CC}^+\text{R}^1\text{R}^2$.⁴ Although the reactivity of cationic allenylidenes is governed by the electron-deficient character of both the

C_α and C_γ atoms of the cumulenenic chain,⁵ regioselective nucleophilic additions at C_γ are usually observed with electron-rich and/or bulky metallic fragments, leading to a wide variety of the functionalized σ -alkynyl complexes $[\text{M}]-\text{C}\equiv\text{CC}(\text{Nu})\text{R}^1\text{R}^2$.⁴ On the basis of these nucleophilic attacks, we have recently developed an alternative procedure for the propargylic substitution of 2-propyn-1-ols mediated by the (indenyl)ruthenium(II) moiety $[\text{Ru}(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2]^+$ (Scheme 1; path B).⁶ Thus, in the first step allenylidene complexes **B** are formed and subsequently transformed into the corresponding σ -alkynyl derivatives **C** ($\text{Nu}^- = \text{HC}\equiv\text{C}^-$, $\text{R}^3\text{COCH}_2^-$, $(\text{CO})_5\text{W}=\text{C}(\text{OMe})\text{CH}_2^-$), which undergo a selective C_β protonation to afford the vinylidene complexes **D**.⁷ Finally, demetalation of **D** with acetonitrile leads to the functionalized terminal alkynes in excellent yields.

Pursuing our studies aimed at exploiting the utility of this synthetic route, in this paper we report an easy entry to terminal 1,5- and 1,6-enynes (see Chart 1) based on the regioselective coupling of Grignard reagents $\text{CH}_2=\text{CH}(\text{CH}_2)_n\text{MgBr}$ ($n = 1, 2$) with the allenylidene chain in the complexes $[\text{Ru}(\text{C}=\text{C}=\text{CR}^1\text{R}^2)(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2][\text{PF}_6]$ ($\text{R}^1 = \text{R}^2 = \text{Ph}$; **1a**; $\text{R}^1 = \text{H}$, $\text{R}^2 = \text{Ph}$, $\text{C}(\text{Me})=\text{CPh}_2$; $\text{CR}^1\text{R}^2 = (1R)\text{-1,3,3-trimethylbicyclo[2.2.1]hept-2-ylidene}$ ($\text{C}(\text{C}_9\text{H}_{16})$), followed by the corresponding demetalation of the resulting σ -alkynyl complexes (Scheme 1; path B).

Results

The allenylidene complexes $[\text{Ru}(\text{C}=\text{C}=\text{CR}^1\text{R}^2)(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2][\text{PF}_6]$ ($\text{R}^1 = \text{R}^2 = \text{Ph}$ (**1a**);^{5a} $\text{R}^1 = \text{H}$, $\text{R}^2 =$

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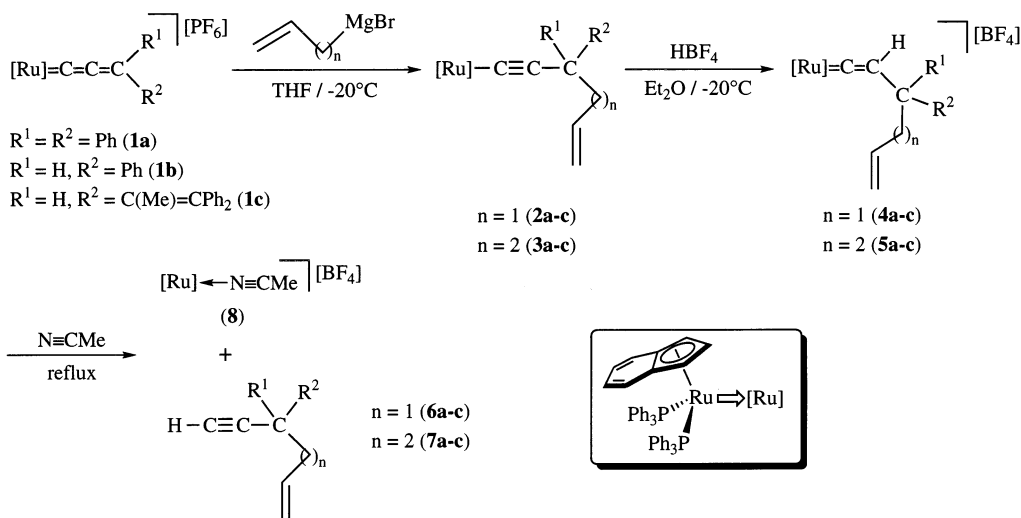
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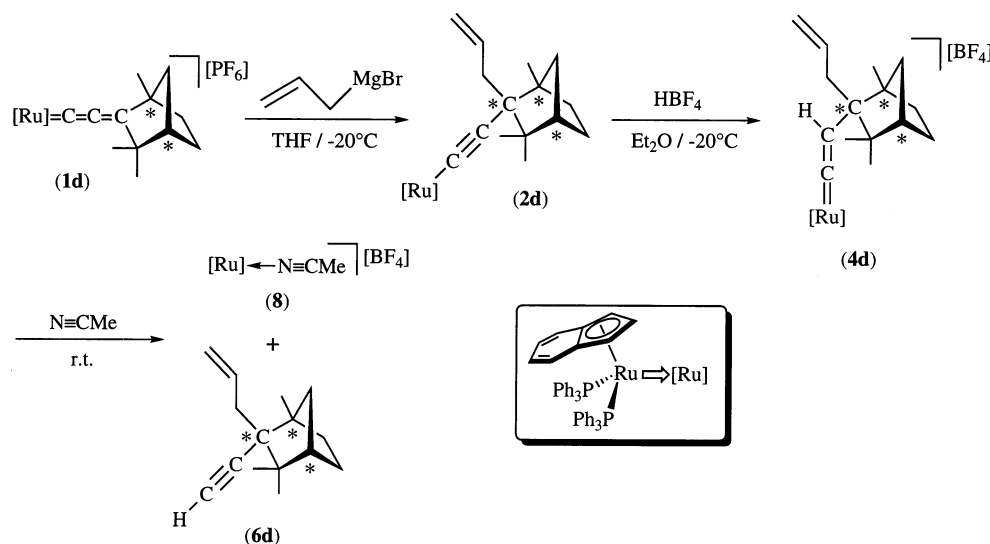
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Scheme 2



Scheme 3



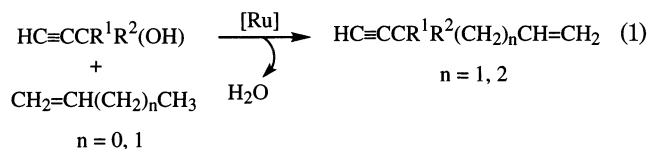
with HBF_4 ⁷ and fully characterized by means of standard spectroscopic techniques and elemental analysis.¹⁰ The most relevant spectroscopic features of **4a–c** and **5a–c** are (i) (1H NMR) the singlet (**4a** and **5a**) or doublet (**4b,c** and **5b,c**; $J_{HH} = 6.9–9.1$ Hz) signal for the acidic vinylidene proton $[Ru]=C=CH$ at δ 4.26–4.63 and (ii) ($^{13}C\{^1H\}$ NMR) the characteristic deshielded $Ru=C_\alpha$ resonance, which appears as a triplet (**4a** and **5a**) or a doublet of doublets (**4b,c** and **5b,c**) in the range 343.53–354.55 ppm ($^2J_{CP} = 15.9–17.4$ Hz). In a second step acetonitrile solutions of vinylidenes **4a–c** and **5a–c** were heated under reflux, affording the terminal enynes **6a–c** and **7a–c** and the cationic nitrile complex $[Ru(N\equiv CMe)(\eta^5-C_9H_7)(PPh_3)_2][BF_4]$ (**8**).^{6a} Enynes **6a–c** and **7a–c** were purified from the reaction mixture by column chromatography on silica gel (86–95% isolated yields) after filtering off the unsoluble complex **8**. Spectroscopic data of **6a–c** and **7a–c** are in agreement with the proposed structures.¹⁰ Thus, the IR spectra show the expected $\nu(C\equiv C)$ absorption in the range 2113–2132 cm^{-1} . The acetylenic $C\equiv CH$ proton appears, in the 1H NMR spectra, as a singlet (**6a** and **7a**) or a doublet (**6b,c** and **7b,c**; $J_{HH} = ca. 2$ Hz) at 1.92–2.66 ppm, and the $HC\equiv C$ carbons resonate, in the $^{13}C\{^1H\}$ NMR, in the ranges δ 70.32–74.14 ($HC\equiv$) and 85.29–87.77 ($\equiv C$).

The optically active 1,5-enyne $HC\equiv CC(C_9H_{16})CH_2-CH=CH_2$ (**6d**) has been also prepared by starting from the chiral allenylidene complex $[Ru\{=C=C=C(C_9H_{16})\}(\eta^5-C_9H_7)(PPh_3)_2][PF_6]$ (**1d**)^{6c} by reaction with allylmagnesium bromide (see Scheme 3). Compounds **2d**, **4d**, and **6d** were obtained in a diastereoselective manner as the result of the selective exo addition of the 2-propen-1-yl fragment to the allenylidene chain in **1d**. We have recently reported analogous stereoselective nucleophilic exo additions to **1d**.^{6c,d,8e} Since the analytical and spectroscopic data of these species are comparable to those observed for their achiral counterparts, **2a–c**, **4a–c**, and **6a–c** are not worth further discussion.

Discussion

As expected, nucleophilic addition of the Grignard reagents $CH_2=CH(CH_2)_nMgBr$ ($n = 1, 2$) to $[Ru(=C=C=CR^1R^2)(\eta^5-C_9H_7)(PPh_3)_2][PF_6]$ (**1a–d**) proceeds regioselectively at the C_γ atom of the allenylidene chain, affording the σ -alkynyl complexes $[Ru\{C\equiv CCR^1R^2(CH_2)_n-CH=CH_2\}(\eta^5-C_9H_7)(PPh_3)_2]$ (**2a–d** and **3a–c**). As we have extensively reported,^{6,8} this process is based on the ability of the moiety $[Ru(\eta^5-C_9H_7)(PPh_3)_2]$ to sterically protect the electrophilic C_α atom of the allenylidene chain, which allows the selective formation of σ -alkynyl

complexes. Taking advantage of this behavior and the well-established synthetic methodology of functionalized alkynes mediated by the (indenyl)ruthenium moiety $[\text{Ru}(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2]$ (path B; Scheme 1), 1,5- and 1,6-enynes $\text{HC}\equiv\text{CCR}^1\text{R}^2(\text{CH}_2)_n\text{CH}=\text{CH}_2$ ($n = 1, 2$; **6a–d** and **7a–c**) have been synthesized in high yields. This synthetic approach can be regarded as the result of the formal coupling of 2-propyn-1-ols with 1-propene or 1-butene (see eq 1).



The overall processes of Schemes 2 and 3 constitute an efficient alternative to the known coupling of propargylic alcohols with allylsilanes or -stannanes via $[\text{Co}_2(\text{CO})_6]$ -stabilized propargyl cations (Nicholas reaction).^{1,11a,b} Although the same number of steps are required in both propargylic substitution reactions (see Scheme 1), the quantitative recovery of the metal fragment as the acetonitrile complex $[\text{Ru}(\text{N}\equiv\text{CMe})(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2][\text{BF}_4]$ (**8**) represents an advantage compared with the classical Nicholas reaction in which the metal auxiliary cannot be recovered after the oxidative decomplexation step.¹² Further studies directed to the preparation of novel functionalized terminal alkynes using the allenylidene complexes $[\text{Ru}(\text{C}=\text{C}=\text{CR}^1\text{R}^2)(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2][\text{PF}_6]$ as synthons of propargyl cations $\text{HC}\equiv\text{CC}^+\text{R}^1\text{R}^2$ are in progress.

Experimental Section

The manipulations were performed under an atmosphere of dry nitrogen using vacuum-line and standard Schlenk techniques. All reagents were obtained from commercial suppliers and used without further purification. Solvents were dried by standard methods and distilled under nitrogen before use. Allenylidene complexes $[\text{Ru}(\text{C}=\text{C}=\text{CR}^1\text{R}^2)(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2][\text{PF}_6]$ ($\text{R}^1 = \text{R}^2 = \text{Ph}$ (**1a**);^{5a} $\text{R}^1 = \text{H}$, $\text{R}^2 = \text{Ph}$ (**1b**),^{5a} $\text{C}(\text{Me})=\text{CPh}_2$ (**1c**);^{8c} $\text{CR}^1\text{R}^2 = \text{C}(\text{C}_9\text{H}_{16})$ (**1d**)^{6c}) were prepared by following the methods reported in the literature.

Synthesis of the σ -Alkynyl Complexes $[\text{Ru}\{\text{C}\equiv\text{CCR}^1\text{R}^2(\text{CH}_2)_n\text{CH}=\text{CH}_2\}(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2]$ ($\text{R}^1 = \text{R}^2 = \text{Ph}$, $n = 1$ (**2a**), **2 (3a)**; $\text{R}^1 = \text{H}$, $\text{R}^2 = \text{Ph}$, $n = 1$ (**2b**), **2 (3b)**; $\text{R}^1 = \text{H}$, $\text{R}^2 = \text{C}(\text{Me})=\text{CPh}_2$, $n = 1$ (**2c**), **2 (3c)**; $\text{CR}^1\text{R}^2 = \text{C}(\text{C}_9\text{H}_{16})$, $n = 1$ (**2d**)). A solution of the corresponding allenylidene complex $[\text{Ru}(\text{C}=\text{C}=\text{CR}^1\text{R}^2)(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2][\text{PF}_6]$ (**1a–d**; 1 mmol) in 30 mL of THF was treated at -20°C with $\text{CH}_2=\text{CHCH}_2\text{MgBr}$

(1.0 M in Et_2O ; 1.1 mL, 1.1 mmol) or $\text{CH}_2=\text{CH}(\text{CH}_2)_2\text{MgBr}$ (0.5 M in THF; 2.2 mL, 1.1 mmol) for 30 min. The mixture was warmed to room temperature, and the solvent was then removed in vacuo. The resulting solid residue was dissolved in dichloromethane (ca. 5 mL) and transferred to an Al_2O_3 (neutral; activity grade I) chromatography column. Elution with a mixture of hexane and diethyl ether (3/1) gave an orange band from which complexes **2a–d** and **3a–c** were isolated after solvent removal. **2a**: yield 79% (0.768 g). **3a**: yield 81% (0.799 g). **2b**: yield 82% (0.735 g). **3b**: yield 88% (0.801 g). **2c**: yield 87% (0.881 g). **3c**: yield 76% (0.780 g). **2d**: yield 84% (0.791 g).

Synthesis of the Vinylidene Complexes $[\text{Ru}\{\text{C}=\text{C}(\text{H})\text{-CR}^1\text{R}^2(\text{CH}_2)_n\text{CH}=\text{CH}_2\}(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2][\text{BF}_4]$ ($\text{R}^1 = \text{R}^2 = \text{Ph}$, $n = 1$ (**4a**), **2 (5a)**; $\text{R}^1 = \text{H}$, $\text{R}^2 = \text{Ph}$, $n = 1$ (**4b**), **2 (5b)**; $\text{R}^1 = \text{H}$, $\text{R}^2 = \text{C}(\text{Me})=\text{CPh}_2$, $n = 1$ (**4c**), **2 (5c)**; $\text{CR}^1\text{R}^2 = \text{C}(\text{C}_9\text{H}_{16})$, $n = 1$ (**4d**)). A diluted solution of $\text{HBF}_4\cdot\text{Et}_2\text{O}$ in diethyl ether was added dropwise at -20°C to a stirred solution of the corresponding σ -alkynyl complex $[\text{Ru}\{\text{C}\equiv\text{CCR}^1\text{R}^2(\text{CH}_2)_n\text{CH}=\text{CH}_2\}(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2]$ (**2a–d** and **3a–c**; 1 mmol) in 100 mL of diethyl ether. Immediately, an insoluble solid precipitated but the addition was continued until no further solid was formed. The solution was then decanted and the brown solid washed with diethyl ether (3×20 mL) and dried in vacuo. **4a**: yield 93% (0.986 g). **5a**: yield 95% (1.020 g). **4b**: yield 89% (0.876 g). **5b**: yield 85% (0.848 g). **4c**: yield 89% (0.979 g). **5c**: yield 87% (0.969 g). **4d**: yield 95% (0.978 g).

Synthesis of Terminal Enynes $\text{HC}\equiv\text{CCR}^1\text{R}^2(\text{CH}_2)_n\text{CH}=\text{CH}_2$ ($\text{R}^1 = \text{R}^2 = \text{Ph}$, $n = 1$ (**6a**), **2 (7a)**; $\text{R}^1 = \text{H}$, $\text{R}^2 = \text{Ph}$, $n = 1$ (**6b**), **2 (7b)**; $\text{R}^1 = \text{H}$, $\text{R}^2 = \text{C}(\text{Me})=\text{CPh}_2$, $n = 1$ (**6c**), **2 (7c)**; $\text{CR}^1\text{R}^2 = \text{C}(\text{C}_9\text{H}_{16})$, $n = 1$ (**6d**)). A solution of the corresponding vinylidene complex $[\text{Ru}\{\text{C}=\text{C}(\text{H})\text{CR}^1\text{R}^2(\text{CH}_2)_n\text{CH}=\text{CH}_2\}(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2][\text{BF}_4]$ (**4a–d** and **5a–c**; 1 mmol) in acetonitrile (30 mL) was heated under reflux for 30 min. The solution was then evaporated to dryness and the resulting solid residue extracted with diethyl ether (ca. 50 mL) and filtered. A yellow solid containing the nitrile complex $[\text{Ru}(\text{N}\equiv\text{CMe})(\eta^5\text{-C}_9\text{H}_7)(\text{PPh}_3)_2][\text{BF}_4]$ (**8**) remains insoluble. The extract was evaporated to dryness and the crude product purified by column chromatography on silica gel with hexane as eluent. Evaporation of the solvent gave terminal enynes **6a–d** and **7a–c** as colorless oils. **6a**: yield 88% (0.204 g). **7a**: yield 90% (0.222 g). **6b**: yield 86% (0.134 g). **7b**: yield 89% (0.151 g). **6c**: yield 95% (0.259 g). **7c**: yield 93% (0.266 g). **6d**: yield 98% (0.198 g).

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Supporting Information Available: Text giving analytical and spectroscopic data for compounds **2a–d** through **7a–c**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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