# New Strategy for Carbohydrate-Based Syntheses of Multichiral Arrays: Pyranosidic Homologation. <sup>1</sup> 3<sup>2</sup>

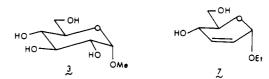
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Abstract: A process of pyranosidic homologation has been developed whereby "satellite" pyranosides are attached to the front and rear of a "backbone" pyranoside. The resulting manifold has capacity for eight contiguous chiral centers, two of which surive from the starting material, levoglucosan (1,6-anhydro- $\alpha$ -D-glucopyranose). The other six are introduced rationally and with complete stereo- and/or regioselectivity in each case, and configurational assignments are readily made by 200-MHz <sup>1</sup>H NMR. The study is outlined with respect to a key intermediate for the ansa chain of rifamycin, but the general applicability to other "pseudo" higher carbon sugars, as well as their "authentic" counterparts, is clearly obvious.

Macrolides and ionophores have, in the past few years, provided a stimulating milieu for the development of synthetic methodologies.4 Their profound resemblance to sugar derivatives was recognized early, 5.6 and carbohydrate-based maneuvers have frequently proved to be extremely appropriate. 7-10 In this context, many of these polyhydroxylated compounds may be regarded as "pseudo" higher carbon sugars, 11 and these targets present specific challenges—as, indeed, do their "authentic" counterparts. 12 We have been developing a simple, versatile, and reliable strategy for "higher carbon" sugars, "authentic" or "pseudo", particularly relevant to those possessing multiple contiguous chiral centers, 13 of which an essential feature is the capability to establish, ipso facto, all stereocenters as they are created and not having to rely on correlation with the synthetic target for authentication.<sup>14</sup> In

this article we outline the salient features of the new strategy with reference to the ansa chain of rifamycin S, 2,15-19 and report the preparation of a key intermediate, 1.



(1) Portions of this work were presented at EUCHEM Conference, "Methods in Organic Synthesis" Lovain-la-Neuve, Belgium, July 1982; XIII IUPAC Symposium on the Chemistry of Natural Products, Pretoria, South Africa, Aug 1982; IV International Conference on Organic Synthesis, Tokyo Japan, Aug 1982; and the 28th National Organic Chemistry Symposium, Bozeman, Montana, June 1983.

(2) Part 1: Molino, B. F.; Magdzinski, L.; Fraser-Reid, B. Tetrahedron Lett. 1983, 24, 5819. Part 2: Magdzinski, L.; Cweiber, B.; Fraser-Reid, B. Tetrahedron Lett. 1983, 24, 5823.

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(4) For some leading references see: Masamune, S.; Bates, G. S.; Corcoran, J. W. Angew Chem., Intl. Ed. Engl. 1977, 16, 585. Wierenga, W. "The Total Synthesis of Ionophores" in "The Total Synthesis of Natural Products"; ApSimon J., Ed.; Wiley: New York and Toronto (1981); Vol. 4 pp 263-351. Kishi, Y. Aldrichimica Acta 1980, 13, 23. Evans, D. A. Ibid. 1982, 15, 23.

(5) Miljkovic, M.; Gligorijevic, M.; Satoh, T.; Miljkovic, D. J. Org. Chem. 1974, 39, 1830. Miljkovic, M., Glissin, D. Ibid 1975, 40, 3357

(6) Hanessian, S.; Rancourt, G. Can J. Chem. 1977, 55, 111. Hanessian, S.; Rancourt, G.; Guidon, Y.; Ibid. 1978, 55, 1843.

(7) Ireland, R. E.; McGarvey, G. J.; Anderson, R. C.; Badond, R.; Fitzsimmons, B.; Thaisrivongs, S. J. Am. Chem. Soc. 1980, 102, 6178. Ireland, R. E.; Thrisrivongs, S.; Wilcox, C. W. Ibid. 1980, 102, 1155. Nicolaou, K. C.; Pavia, M. R.; Seitz, S. P. Ibid. 1982, 104, 2027. Nicolaou, K. C.; Clar-C.; Pavia, M. R.; Seitz, S. P. Ibid. 1982, 104, 2021. Nicolaou, K. C.; Claremon, D. A.; Papahatjis, D. P.; Magolda, R. L. Ibid. 1981 103, 6969. Tatsuta, K.; Amemiya, Y.; Kanemura, Y., Kinoshita, M. Tetrahedron Lett. 1981, 22, 2997. Nakahara, Y.; Beppu, Y.; Ogawa, T. Ibid. 1981, 22, 3197. Hanessian, S.; Tyler, P. C.; Demailly, G.; Chapleur, Y. J. Am. Chem. Soc. 1981, 103, 6243. Corey, E. J.; Weigel, L. O.; Chamberlin, A. R.; Cho, H.; Hua, D. H. Ibid. 1980, 102, 6615. Nakata, M.; Enrai, H.; Kinoshita, M. Bull, Chem. Soc.

Jpn. 1982, 55, 3283.(8) Nakata, M.; Takao, H.; Ikeyama, Y., Sakai, T.; Tatsuta, K.; Kinoshita,

M. Bull, Chem. Soc. Jpn. 1981, 54, 1749.

(9) Hannessian, S.; Pougny, J-R, Boessenkool, I. K. J. Am. Chem. Soc. 1982, 104, 6164.

(10) Mukaiyama, T.; Noyaic, R. Kagaku Zokan (Kyoto) 1981, 91, 101. (11) The name of the seco acid of erthronolide A is 2,4,7,8,10,14,15heptadeoxy-2,4,6,8,10,12-hexa-C-methyl-D-erythro-L-ido-pentadec-9-ulosonic

(12) For some typical systems see: Secrist, J. A.; Wu, S-R J. Org. Chem.
1974, 44, 1434. Secrist, J. A.; Barnes, K. D. Ibid. 1980, 45, 4526.
(13) (a) Fraser-Reid, B.; Magdzinski, L.; Molino, B. In "Current Trends in Organic Synthesis"; Nozaki, H., Ed.; Pergamon Press: Oxford, 183; pp 197-203. (b) Abstracts of the 28th National Organic Symposium, June 20-24, 1983, Bozeman MT.

#### The Problem

- (1) The crucial problem confronting sugar-based approaches to higher carbon "sugars", "pseudo" or "authentic", is exemplified with methyl  $\alpha$ -D-glucopyranoside, (3), a favorite precursor. This sugar has only five contiguous chiral centers, but since those at C-1 and C-5 give the template its stereochemical integrity, 20-22 synthetic manipulations are permissible only at C-2, C-3, and C-4. C-6 must be regarded as an "off-template" site and hence does not normally display high stereoselection in its reactions.
- (2) An equally debilitating defect concerns the assignment of configurations to such diastereomeric products, since they would not be amenable to the simple conclusive NMR analyses normally enjoyed by carbohydrate derivatives.21
- (3) The selection of protecting groups is of paramount importance with these polyhydroxylated compounds, and any reduction in the number of such groups which must be used is advantageous.

These problems are particularly menacing for arrays such as 2 for which there are no logical sites for disconnection. As a result,

(15) The first synthesis of rifamycin S was achieved by Kishi's group. 16 A carbohydrate-based total synthesis of and an intermediate for the ansa chain have been reported. For other work on the ansa chain see ref 17-19.

(16) Nagaoka, H.; Rutsch, W.; Schmid, G.; Iio, H.; Johnson, M. R.; Kishi, Y. J. Am. Chem. Soc. 1980, 102, 7962. Iio, H.; Nagaoka, H.; Kishi, Y. Ibid.

1980, 102, 7965

(17) Still, W. C.; Barrish, J. C. J. Am. Chem. Soc. 1983, 105 2487. (18) Masamune, S.; Imperiali, B.; Garvey, D. S. J. Am. Chem. Soc. 1982, 104, 5528.

(19) Corey, E. J.; Hase, T. Tetrahedron Lett. 1979, 335.

(20) Hanessian, S. Acc. Chem. Res. 1979, 12, 159

(21) Fraser-Reid, B.; Anderson, R. C. Prog. Chem. Org. Nat. Prod. 1980,

(22) Fraser-Reid, B. Acc. Chem. Res. 1975, 8, 192.

<sup>(14)</sup> In many instances, 8,9,15-19 the stereochemistry of linear arrays cannot be independently assigned with complete assurance but must be compared with known fragments or converted into ring systems.

Scheme I

combination of two (or more) monosaccharide subunits must perforce occur at "off-template" sites where the formation of rich diastereomeric mixtures of unknown configurations threatens to compromise the stereocontrol which had been carefully crafted into the individual subunits.

### Retroanalysis

The retroanalysis requires the use of chemical models only, although computer modeling may facilitate some of the conformational evaluations.

The pivotal idea out of which this new strategy grew was the well-known fact that the anomeric carbon of a sugar may be readily converted into a methyl group.<sup>23</sup> Thus as indicated in Scheme I, each 1,4-related CH<sub>3</sub> and OH represent a "hidden" pyranoside as accentuated in the retroanalysis 4, 5, 6 shown in Scheme I.1 Five such potential pyranosides (I-V) exist; we must now narrow our choice(s) in a rational manner.

Consider first the "end" options I and II. In I, the C-2 appendage R2 has four "off-template" stereocenters that, as noted above, would be difficult to control and/or analyze. The same applies to option II. Both are therefore rejected.

For the second phase of the elimination process, options II, III, and IV are coiled into the conformations shown in Scheme I. It is seen that not one, but two, pyranose rings can be generated, VI, VII, and VIII, respectively. Option VIII is now eliminated on the ground that the "backbone" pyranoside (heavy lines) would require the use of an L-hexopyranoside as a precursor, and since these are usually expensive, option IV is unacceptable to us.

The remaining viable options are II and III. The obvious benefits accruing to their bipyranose equivalents, VI and VII, could

be enhanced even further by elaboration of a third pyranoside and it is here that the final selection of option III is mandated. Thus, further "pyranosidic homologation" of either VI and VII would sacrifice the C-24 CH3 in either case. However, in the case of VI, the anomeric center of the resulting upper "satellite" would require the sacrifice of a second stereocenter (C-20). Comparable loss would not occur with VII, since the C-28-CH<sub>2</sub> (which is achiral) would coincide with an anomeric center, which, being a masked aldehyde, is potentially achiral anyway! Thus, all eight contiguous chiral centers of  $2 (\equiv III)$  can be accommodated within the ten-carbon continuum, highlighted in the tricyclic construct

Retrosynthetic analysis of the neighboring CH<sub>3</sub> and OH groups in IX lead to the two olefinic sites in X, each of which would be generated and manipulated at different stages. The unsaturated rings in X are reminiscent of hex-2-enopyranosides such as 724 and would therefore inherit an extensive body of knowledge garnered in this<sup>22</sup> and other laboratories.<sup>25,26</sup>

#### Results and Discussion

The starting material for our work was the known oxirane 8,27 which can be prepared from levoglucosan<sup>28</sup> in four steps in 80%

<sup>(23)</sup> For a typical example of the transformation 6 to 4 see: Wolform, M. L., Lemieux, R. U.; Olin, S. M. J. Am. Chem. Soc. 1949, 71, 2780.

<sup>(24)</sup> Ferrier, R. J. Adv. Carbohydr. Chem. Biochem. 1969, 24, 199; 1965,

<sup>20, 67.</sup> (25) For a recent exhaustive reviews see: Holder, N. Chem. Rev. 1982,

<sup>(26)</sup> Achmatowicz, O. "An Approch to the Synthesis of Higher-carbon Sugars," In "Organic Synthesis - Today and Tomorrow,"; Trost, B. M., Hutchison, C. R., Eds.; Pergamon Press: Oxford, 1981; p 307. (27) Trana, T.; Cerny, M. Collect. Czech. Chem. Commun. 1971, 36,

<sup>(28)</sup> Levoglucosan is obtained in 30-40% by the vacuum pyrolysis of Ward, R. B. Methods Carbohydr. Chem. 1963, 2, 394 starch:

Scheme II<sup>24</sup>

Scheme III

overall yield. Reaction with the ethnylalane 9,30 followed by Lindlar hydrogenation, gave compound 10, which upon solvolysis with benzyl alcohol gave a mixture of the external and internal glucosides, 11 and 12a, respectively. It is the former that is

(29) All compounds in Scheme II gave satisfactory elemental analyses and/or HRMS data as well as consistent <sup>1</sup>H NMR spectra.

(30) For a comprehensive review of ethynyl alanes see: Brono, G. "The Use of Aluminum Alkyls in Organic Syntheses,"; 1969–1972 supplement; Ethyl Corporation: Baton Rouge, LA, 1973.

required for the bipyranoside IX, but compound 12 proved to be a better prospect<sup>31</sup> since it could be obtained as the exclusive product by treatment of 10 with trifluoroacetic acid. Accordingly, the preferred target is 1, rather than the stereochemically equivalent IX.

Epoxidation proceeded with greatest exoselectivity with the acetate 12b, and the product, 13, reacted smoothly with lithium

<sup>(31)</sup> For a fuller discussion of this aspect see ref 13b.

dimethylcuprate to give the alcohol 14a. Oxidation followed by immediate reduction after workup afforded the epimer 14b (87%) yield) whose stereochemistry was confirmed by the NOE shown in Scheme II. In order to prepare for the upper "satellite" some straightforward functional group manipulations were required to give 15 and thence 16. Of special note is the selective hydrolysis of the more-hindered isopropylidene ring engaging the two secondary hydroxyls of 15.

Elaboration of the upper "satellite" was problematic.31 Painstaking investigations educed (a) that a modified<sup>32</sup> Corey-Kim procedure 33 was the reagent of choice for selective oxidation of the primary alcohol to the aldehyde 16b (96%), (b) that the ethylenic acetal 17, of Sargent and co-workers<sup>34</sup> was the preferred version of the Trippett-Bestmann reagent<sup>35</sup> for obtaining the 4:1 Z/E mixture 16c, and (c) That both of these geometric isomers upon treatment with methanol and, specifically, Grieco's acid36 led directly to 18 as the only anomer in 86% Yield.

Epoxidation of the olefinic sugar 17 with MCPBA was extremely slow; however, aqueous NBS generated a mixture of bromohydrins, which gave a single epoxide 19, upon reaction with sodium hydride. The opening of epoxide 19 was effected best by Me<sub>2</sub>Mg in THF,<sup>37</sup> generated by mixing equimolar amounts of methyllithium and methylmagnesium chloride. The resulting product was methylated to afford the targeted intermediate 1.

The reactions in Scheme II proceed with very high yields, but the length of the sequence diminishes its attractiveness and hence efforts have been made to provide shorter pathways. We have had encouraging results with respect to the bipyanose 14b (Scheme

By application of Normant's chemistry<sup>38</sup> we obtained the vinyllithium derivative 20a which was converted into the Grignard 20b. The latter reacted with 8 to give 21 in 60% yield. This

(34) Cresp, T. M.; Sargent, M. V.; Vogel, P. J. Chem. Soc., Perkin Trans.

1 1974, 37.

moderate yield is not serious since the only other product is a bromohydrin that is recycled to 8. Solvolysis of 21 gave the  $\alpha$ -anomer 22 predominantly.

Hydroboration of 22 occurred from the exo face, giving 14a, making it possible to obtain 14b from 8 in five steps instead of the eight shown in Scheme II. Eight more steps are presently required to go from 14b to 1, but efforts are under way to reduce these.

#### Conclusion

There are several features about the new strategy outlined above that are noteworthy:

- (a) Each of the seven centers highlighted in the tripyranose 1, obtained by either route (Schemes II or III), has been created with complete stereo- and/or regioselectivity, and therefore no troublesome separations of isomers are required. This is due to the fact that each ring is a reliable pyranose and hence there are no "off-template" stereocenters! Problem 1 (vide supra) is therefore obviated.
- (b) Although each center was introduced rationally and was easily monitored, all protons in the entire assembly 1 can be assigned, readily and simply, by [only!] 200-MHz 1H NMR, as is apparent from the data shown in Scheme II. The easy proof of configuration common to carbohydrate derivatives (see problem 2 above) is therefore observed in 1.
- (c) The folding pattern adopted cuts down upon the number of "external" protecting groups that would be required by a linear array, since 1 (or IX) protects itself. Note also that 1 requires one less protecting group than IX.
- (d) Since the oxirane moeity is a synthon for "pyranosidic homologation" (e.g.,  $8 \rightarrow 12$ ), an iterative procedure should be applicable to either 13 or 19.
- (e) Retroanalysis of many other propionate-derived entities, e.g., the ansa chains of streptovaricin and maytansine, according to the five principles outlined above indicate other opportunities for the pyranosidic homologation appraoch.

The two tasks remaining for conversion of 1 into 2 are deconvoluting the tricycle and adding the C-24 CH<sub>3</sub>. Procedures for carrying these out are being developed.

Registry No. 1, 88392-88-5; 8, 33208-47-8; 9, 88412-16-2; 10, 88392-78-3; 11, 88392-79-4; 12a, 88392-80-7; 12b, 88412-17-3; 13, 88412-18-4; 14a, 88392-81-8; 14b, 88424-60-6; 15, 88392-82-9; 16a, 88392-83-0; 16b, 88392-84-1; 16c, 88392-85-2; 17, 78950-65-9; 18, 88392-86-3; 19, 88392-87-4; 20a, 88392-89-6; 21, 88412-19-5; 22, 88392-90-9; rifmaycin S, 13553-79-2; levoglucosan, 498-07-7.

Supplementary Material Available: <sup>1</sup>H NMR data for compounds 1, 10, 12, 13, 14a, 14b, 16b, 16c, 18, 19, 21, and 22 (6 pages). Ordering information in given on any current masthead

## Robustadials A and B from Eucalyptus robusta

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Abstract: The structures of two new compounds, robustadial A and B, isolated from the active fraction of the antimalarial extract of Eucalyptus robusta leaves, a plant used in Chinese herbal medicine to treat malaria, have been determined as 2 and 3. The structures were resolved by using spectroscopic methods, especially two-dimensional NMR and difference nuclear Overhauser enhancement techniques. The monoterpenoid moiety containing the spiro linkage has not been encountered previously.

The leaves of Eucalyptus robusta Sm. (Myrtaceae) are used in Chinese herbal medicine for the treatment of dysentery, malaria, and other bacterial diseases. The benzene-soluble fraction of the crude 95% ethanol extract of the leaves showed significant inhibition against Plasmodium berghei, a malaria-inducing protozoan. From this fraction, an active compound, robustaol A (1) was isolated.1

<sup>(32)</sup> N-Chlorosuccinimide (800 mg, 6.0 mmol) was added to a solution of thioanisole (0.78 mL, 818 mg, 6.6 mmol) in 20 mL of dry CH<sub>2</sub>Cl<sub>2</sub> under argon at -20 °C. After 30 min, a solution of the diol 16a (1.1220 g, 3.0 mmol) in 10 mL of CH<sub>2</sub>Cl<sub>2</sub> was added dropwise. After an additional 30 min, diisopropylethylamine (1.05 mL, 6.0 mmol) was added dropwise at -20 °C. After stirring at -20 °C for 20 min, the reaction was warmed to 0 °C. TLC indicated the starting diol 16a ( $R_f$  0.24, 5% MeOH-CH<sub>2</sub>Cl<sub>2</sub>) to have formed a clean product ( $R_f$  0.40). The reaction was diluted with CH<sub>2</sub>Cl<sub>2</sub> (50 mL), successively washed with sat NaHCO<sub>3</sub> (30 mL), 2% HCl (30 mL), and sat NaHCO<sub>3</sub> (30 mL), dried over MgSO<sub>4</sub>, filtered, and evaporated.

(33) Corey, E. J.; Kim, C. U. J. Org. Chem. 1973, 38, 1233.

<sup>(35)</sup> Bestmann, H. J.; Roth, K.; Ettlinger, M. Chem. Ber. 1982, 115, 161; Angew. Chem., Int. Ed. Engl. 1979, 18, 687. Trippett, S.; Walder, D. M. J. Chem. Soc. 1961, 1266.

<sup>(36)</sup> Miyashita, M.; Yoshikoshi, A.; Grieco, P. A. J. Org. Chem. 1977, 42,

<sup>(37)</sup> Parker, K. A.; Babine, R. E. Tetrahedron Lett. 1982 23, 1763.

<sup>(38)</sup> Cahiez, G.; Bernard D.; Normant, J. F. Synthesis 1976, 245.

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