

Tetrahedron: Asymmetry 12 (2001) 937-942

# An expedient synthesis of D-callipeltose

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Received 9 March 2001; accepted 19 March 2001

Abstract—Methyl D-callipeltose 12 and D-callipeltose 4 were synthesized from D-glucal 5 in 10 and 11 steps, respectively. The synthesis features an azide displacement reaction of an  $\alpha$ -nosyloxy ketone 7 and a highly diastereoselective C-methylation of  $\alpha$ -azido ketone 8. © 2001 Elsevier Science Ltd. All rights reserved.

#### 1. Introduction

Callipeltoside A 1 belongs to a new class of cyclodepsipeptides called callipeltins. The molecules were isolated in 1996 from the shallow water lithistid sponge *Callipelta* sp., a native of New Caledonia.<sup>1,2</sup> They were found to inhibit in vitro proliferation of KB and P338 cells and to protect cells infected with HIV. Structurally, 1 consists of a macrocyclic lactone linked to a unique dienyne cyclopropane side chain and the deoxy amino sugar, callipeltose, 4 (Fig. 1). The absolute stereochemistry of the molecule has not been established.

There has been considerable synthetic interest in 1 although no total synthesis has been reported to date.<sup>3–7</sup>

The deoxy amino sugar **4** contains an unusual fivemembered oxazolidinone ring fused to the sugar backbone at positions 3 and 4. Both enantiomers of **4** have been synthesized—methyl L-callipeltose in 10 steps from L-rhamnose<sup>8</sup> and methyl D-callipeltose in 14 steps from methyl D-mannose.<sup>9</sup> Herein we report an expedient synthesis of  $\alpha$ -D-callipeltose **4** from D-glucal **5**.

### 2. Results and discussion

The  $\alpha,\beta$ -unsaturated ketone **6** was synthesized from D-glucal in three steps (Scheme 1).<sup>10</sup> A sequence of tosylation (TsCl, pyr.)<sup>11</sup> and reduction with LiAlH<sub>4</sub><sup>12,13</sup> effected deoxygenation of the 6-position of D-glucal **5**.



Figure 1. Structures of callipeltosides A-C, 1-3.

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Scheme 1. Synthesis of D-callipeltose. Reagents and conditions: (a) TsCl, py,  $CH_2Cl_2$ ,  $0^{\circ}C \rightarrow rt$ , 78%; (b) LiAlH<sub>4</sub>, THF,  $\Delta$ , 59%; (c) MnO<sub>2</sub>,  $CH_2Cl_2$ , rt, 66%; (d) NsCl, py,  $CH_2Cl_2$ ,  $0^{\circ}C \rightarrow rt$ , 80%; (e) *n*-Bu<sub>4</sub>NN<sub>3</sub>,  $CH_2Cl_2$ ,  $0^{\circ}C \rightarrow rt$ , 95%; (f) MeLi, THF, -100°C, 77%; (g) *m*-CPBA, NaHCO<sub>3</sub>, MeOH,  $0^{\circ}C \rightarrow rt$ , 60%; (h) MeI, Ag<sub>2</sub>O, Et<sub>2</sub>O,  $\Delta$ , 72%; (i) H<sub>2</sub>, 10% Pd/C, Boc<sub>2</sub>O, EtOAc, rt, 62%; (j) *t*-BuOK, THF,  $0^{\circ}C \rightarrow rt$ , 69%; (k) 2 M H<sub>2</sub>SO<sub>4</sub> in 1:1 H<sub>2</sub>O/dioxane, 60–70°C, 48%. Rt=room temperature, NsCl=nosyl chloride, py=pyridine, *m*-CPBA=*m*-chloroperbenzoic acid, Boc=*t*-butoxycarbonyl.

The resulting D-rhamnal was oxidized with  $MnO_2^{14}$  to give  $\alpha,\beta$ -unsaturated ketone 6. The next task was to introduce a nitrogen atom at the 4-position of the sugar by azide displacement of a suitable leaving group.

For this purpose, several leaving groups, including mesylate,<sup>15</sup> triflate<sup>15</sup> and imidazolyl sulfonate,<sup>16</sup> were examined, but only the nosylate group<sup>17</sup> could be displaced at an acceptable rate without decomposition. The nosylate 7 was obtained by the action of nosyl chloride and pyridine on 6 in 80% yield along with ca. 3-5% of its C-(4) epimer. Sodium azide was found to be too basic for the displacement reaction; stirring 7 overnight in DMSO with NaN<sub>3</sub> at room temperature gave, in addition to recovered starting material, ca. 1:1 mixture of both epimers of 8, presumably due to product enolization. However, tetra-n-butylammonium azide<sup>18,19</sup> furnished the desired azide 8 in 90% yield along with 5-10% of its C-(4) epimer.<sup>20</sup> The C-methylation with methyllithium in THF at -100°C gave the tertiary alcohol 9 in 77% yield as a single diastereomer.14,21 The observed selectivity is rationalized through a synergy of steric and stereoelectronic considerations, as described in Fig. 2: the axially disposed azide group prevents attack from the Re face of the carbonyl, and the electronegativity of the nitrogen favors attack from the opposite (Si) face.

The next objective was to install the final two stereocenters of the sugar. Epoxidation accompanied by concomitant stereo- and regioselective opening of the epoxide is a useful method to functionalize the double bond in glucals.<sup>22–28</sup> Thus, reaction of **9** with *m*-CPBA in methanol in the presence of NaHCO<sub>3</sub> led to the stereoselective formation of **10** in 60% yield. Methylation of the newly created alcohol with MeI and Ag<sub>2</sub>O<sup>29</sup> cleanly alkylated the secondary hydroxyl group. Subsequent palladium-catalyzed hydrogenation reduced the azide to the corresponding amine, which, in the pres-



Figure 2. Preferred approach of nucleophile in the conversion of 8 to 9.

ence of Boc-anhydride, furnished amide 11,<sup>30</sup> setting the stage for the oxazolidinone ring closure. Thus, exposure of the tertiary alcohol 11 to potassium *t*-butoxide in THF, following the procedure of Davies et al., furnished the methyl glycoside 12.<sup>31</sup> Finally, the glycoside was hydrolyzed with 2 M sulfuric acid in water/dioxane (1:1) at 60°C,<sup>32,33</sup> concluding the synthesis of D-callipeltose 4 in 11 steps from commercially available D-glucal 5.

#### 3. Conclusion

In conclusion, methyl D-callipeltose 12 and D-callipeltose 4 were synthesized in a highly selective manner starting from D-glucal 5 in 10 and 11 steps, respectively. Key features of this synthesis, the shortest reported to date for 4, include: (i) readily available starting materials; (ii) use of an  $\alpha$ -nosyloxy ketone 7 en route to azide 8; and (iii) a highly diastereoselective *C*-methylation rationalized in Fig. 2.

#### 4. Experimental

All reactions were carried out under an argon atmosphere with dry solvents under anhydrous conditions, if

necessary. Anhydrous solvents were obtained by passing them through commercially available activated alumina columns. Yields refer to chromatographically and spectroscopically (<sup>1</sup>H NMR) homogeneous materials, unless otherwise stated. All reagents were purchased at highest commercial quality and used without further purification, unless otherwise stated. All reactions were monitored by thin-layer chromatography carried out on 0.25 mm E. Merck silica gel plates (60F-254) using UV light as visualizing agent and acidic PMA (2.5% phosphomolybdic acid, 1.5% ophosphoric acid and 5% sulfuric acid in water) or 1% aqueous potassium permanganate solution and heat as developing agents. E. Merck silica gel (60, particle size 0.040–0.063 mm) was used for flash column chromatography. NMR spectra were recorded on Bruker DRX-600, DRX-500 or AMX-400 instruments and calibrated using residual undeuterated solvents as an internal reference (CDCl<sub>3</sub> 7.26 ppm, MeOH 3.34 ppm). The following abbreviations are used to explain the multiplicities: s = singlet; d = doublet; t = triplet;q = quartet; m = multiplet; b = broad. IR spectra were recorded on a Perkin-Elmer 1600 series FT-IR spectrometer. Optical rotations were recorded on a Perkin-Elmer 241 polarimeter. High resolution mass spectra (HRMS) were recorded on an IonSpec mass spectrometer under MALDI-FTMS conditions with NBA or DHB as the matrix. Low resolution mass spectra were recorded on a Hewlett Packard 5971A benchtop GC/ MS. Melting points (mp) are uncorrected and were recorded on a Thomas-Hoover Unimelt capillary melting point apparatus.

#### 4.1. D-Rhamnal

To a stirred solution of D-glucal 5 (2.50 g, 17.1 mmol) in pyridine (42 mL) and  $CH_2Cl_2$  (42 mL), cooled at 0°C, was added tosyl chloride (4.89 g, 25.7 mmol) and the cooling bath was removed. After stirring for 2.5 h at ambient temperature, the reaction mixture was cooled again to 0°C, quenched with water (5 mL) and stirred for 30 min. More water (20 mL) was added, the organic layer was separated and washed with sat. aq. CuSO<sub>4</sub> (3×20 mL) and water (3×20 mL). The combined aqueous phases (excluding the first water layer) were extracted with  $CH_2Cl_2$  (2×20 mL) and those organic extracts were washed with water  $(2 \times 10)$ mL). The combined organic extracts were washed with brine (20 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated to provide crude 6-O-tosyl-D-glucal as a yellow oil (3.99 g, 78% yield).  $R_{\rm f}$  0.55 (EtOAc); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.81 (d, J=8.2 Hz, 2H), 7.36 (d, J=8.2 Hz, 2H), 6.24 (dd, J=6.1, 1.6 Hz, 1H), 4.74 (dd, J=6.1, 2.1 Hz, 1H), 4.48 (dd, J=11.4, 3.8 Hz,1H), 4.27 (app. d, J=11.4 Hz, 2H), 3.91 (ddd, J=10.0, 3.8, 2.1 Hz, 1H), 3.77 (dd, J = 10.0, 7.3 Hz, 1H), 2.45 (s, 3H).

To a stirred solution of 6-O-tosyl-D-glucal (3.99 g, 13.3 mmol) in THF (30 mL) at 0°C was added LiAlH<sub>4</sub> (0.9 M solution in THF, 44.2 mL, 39.9 mmol) dropwise and the reaction mixture was heated to reflux.

After stirring the reaction for 1 h, the mixture was cooled to 0°C and quenched slowly with H<sub>2</sub>O (1.5 mL), 15% NaOH (1.5 mL) and H<sub>2</sub>O (4.5 mL). The resulting slurry was diluted with Et<sub>2</sub>O, filtered through a pad of Celite (Et<sub>2</sub>O rinse) and concentrated. Purification by dry-column flash chromatography<sup>34</sup> (silica, 20–100% EtOAc/hexanes) provided D-rhamnal as a white solid (1.02 g, 59% yield).  $R_{\rm f}$  0.37 (EtOAc); mp 72–73°C (Et<sub>2</sub>O/hexanes) (lit.<sup>35</sup> 71–73°C (EtOAc/hexanes)); <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  6.32 (dd, J= 6.0, 1.6 Hz, 1H), 4.72 (dd, J=6.0, 2.2 Hz, 1H), 4.21 (app. d, J=6.3 Hz, 1H), 3.87 (dq, J=9.8, 6.3 Hz, 1H), 3.43 (app. t, J=8.6 Hz, 1H), 1.39 (d, J=6.3 Hz, 3H).

# 4.2. 1,5-Anhydro-2,6-dideoxy-D-*erythro*-hex-1-enit-ulose 6

To a stirred solution of D-rhamnal (100 mg, 0.768 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (7.5 mL) at room temperature was added MnO<sub>2</sub> (200 mg, 2.30 mmol). After 3 h, more MnO<sub>2</sub> (200 mg, 2.30 mmol) was added and the stirring was continued. After 15 h, the reaction mixture was diluted with CH<sub>2</sub>Cl<sub>2</sub> (10 mL), filtered through a pad of Celite (EtOAc rinse) and concentrated. Purification by flash chromatography (silica, 40–50% EtOAc/hexanes) provided 65 mg of **6** as a white volatile solid (66% yield).  $R_{\rm f}$  0.66 (EtOAc); mp 91–92°C (Et<sub>2</sub>O/hexanes) (lit.<sup>36</sup> 92–93°C); <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.39 (app. d, J=5.7 Hz, 1H), 5.46 (d, J=5.7 Hz, 1H), 4.20 (dqd, J=12.9, 6.4, 0.7 Hz, 1H), 3.54 (s, 1H), 1.57 (d, J=6.4 Hz, 3H).

#### 4.3. 1,5-Anhydro-2,6-dideoxy-4-*O*-nosyl-D-*erythro*-hex-1-enit-3-ulose 7

To a stirred solution of 6 (0.212 g, 1.65 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (8.0 mL) at 0°C were added pyridine (0.667 mL, 8.25 mmol) and nosyl chloride (0.733 g, 3.31 mmol). The reaction mixture was allowed to warm to room temperature over 2 h, and after stirring the mixture for a further 3 h more nosyl chloride (0.367 g,1.66 mmol) and pyridine (0.667 mL, 8.25 mmol) were added. After 12 h the reaction mixture was cooled to 0°C and water (2 mL) was added. After stirring for 30 min more water (10 mL) was added, the aqueous layer was separated and extracted with  $CH_2Cl_2$  (2×10 mL), the combined organic layers were washed with brine (5 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated. Purification by dry-columm flash chromatography (silica, 15–30% EtOAc/hexanes) provided 7 as a yellow crystalline solid (0.411 g, 80% yield).  $R_{\rm f}$  0.60 (50% EtOAc/ hexanes); mp 93°C; [a]<sub>D</sub> +138.8 (c 1.07, CHCl<sub>3</sub>); IR (film) 3108, 1695, 1598, 1533, 1353, 1256, 1188, 1026, 854, 780 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 8.39 (app. d, J=9.2 Hz, 2H), 8.19 (app. d, J=9.2 Hz, 2H), 7.35 (d, J = 5.9 Hz, 1H), 5.37 (d, J = 5.9 Hz, 1H), 5.02 (d, J=12.1 Hz, 1H), 4.54 (dq, J=12.1, 6.2 Hz, 1H), 1.63 (d, J = 6.2 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  180.0, 163.5, 150.8, 142.1, 129.7, 124.1, 105.0, 79.6, 77.4, 17.4; HRMS (MALDI) calcd for  $C_{12}H_{11}NO_7S$  $(M^+)$  m/z: 313.0256, found 313.0286.

# 4.4. 1,5-Anhydro-4-azido-2,6-dideoxy-D-*threo*-hex-1enit-3-ulose 8

To a stirred solution of 7 (190 mg, 0.606 mmol) in  $CH_2Cl_2$  (3.3 mL) at 0°C was added *n*-Bu<sub>4</sub>NN<sub>3</sub> (0.651 g, 2.29 mmol) [CAUTION]<sup>19</sup> in CH<sub>2</sub>Cl<sub>2</sub> (0.5 mL) via cannula. The reaction mixture was allowed to warm to room temperature over 1.5 h and stirred at that temperature for 1 h. Water (5 mL) was added, the aqueous layer was extracted with  $CH_2Cl_2$  (2×10 mL), the combined organic extracts were washed with brine (5 mL) and dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated. Purification by dry-column flash chromatography (silica, 10-20% EtOAc/hexanes) provided 8 as a yellow oil (88 mg, 95% yield).  $R_{\rm f}$  0.43 (33% EtOAc/hexanes);  $[\alpha]_{\rm D}$  -78.0 (c 0.383, CHCl<sub>3</sub>); IR (film) 3378, 2919, 2106, 1675, 1593, 1414, 1274, 1050 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$ 7.37 (d, J = 6.0 Hz, 1H), 5.49 (dd, J = 6.0, 0.9 Hz, 1H), 4.53 (qd, J = 6.6, 3.1 Hz, 1H), 3.81 (d, J = 3.1 Hz, 1H), 1.48 (d, J = 6.6 Hz, 3H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 187.1, 163.6, 104.9, 76.9, 63.8, 15.3; HRMS (MALDI) calcd for  $C_6H_8NO_2$  (MH-N<sub>2</sub><sup>+</sup>) m/z: 126.0550, found 126.0554.

# 4.5. 1,5-Anhydro-4-azido-2,6-dideoxy-3-C-methyl-Dlyxo-hex-1-enitol 9

To a stirred solution of 8 (95 mg, 0.620 mmol) in THF (3.1 mL) at -100°C was added MeLi (1.5 M solution in Et<sub>2</sub>O, 0.500 mL, 0.744 mmol) dropwise. After 1.5 h, sat. aq. NH<sub>4</sub>Cl (1.0 mL) was added and the reaction mixture was allowed to warm to room temperature. The aqueous layer was extracted with EtOAc ( $2 \times 5$  mL), the combined organic extracts were washed with brine (5 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated. Purification by dry-column flash chromatography (silica, 10-25% EtOAc/hexanes) provided 9 as a yellow oil (81 mg, 77% yield).  $R_f 0.15$  (20% EtOAc/hexanes);  $[\alpha]_D + 84.2$  (c 0.483, CHCl<sub>3</sub>); IR (film) 3413, 2919, 2097, 1637, 1449, 1378, 1232, 1132, 1049, 944, 820, 750 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  6.27 (d, J=6.3 Hz, 1H), 4.70 (dd, J = 6.3, 2.2 Hz, 1H), 4.19 (qd, J = 6.5, 0.9 Hz, 1H), 3.34 (br s, 1H), 2.24 (br s, 1H) 1.45 (d, J = 6.6 Hz, 3H), 1.43 (s, 3H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  143.6, 104.8, 72.1, 68.8, 68.4, 29.5, 18.0. Low resolution mass spectrum (GC/MS) calcd for  $C_7H_{11}N_3O_2$  (M<sup>+</sup>) 169, found 169.

# 4.6. Methyl 4-azido-4,6-dideoxy-3-C-methyl-α-D-*talo*pyranoside 10

To a stirred solution of **9** (100 mg, 0.591 mmol, 100 mol%) in MeOH (3.1 mL) at 0°C were added NaHCO<sub>3</sub> (153 mg, 1.86 mmol, 315 mol%) and *m*-CBPA (55–60%, 0.321 g, 1.86 mmol, 315 mol%). After stirring for 1 h sat. aq. NaHCO<sub>3</sub> (2 mL) was added, the aqueous layer was extracted with EtOAc (2×10 mL), the combined organic extracts were washed with brine (5 mL) and dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated. Purification by flash chromatography (silica, 20–30% EtOAc/hexanes) provided **10** as an oil (76 mg, 60% yield).  $R_{\rm f}$  0.28 (33% EtOAc/hexanes); [ $\alpha$ ]<sub>D</sub> +143.8 (*c* 0.417, CHCl<sub>3</sub>); IR (film) 3449, 2921, 2109, 1449, 1343, 1261, 1126, 1061,

997, 944, 814 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  4.76 (s, 1H), 4.03 (qd, *J*=6.5, 1.2 Hz, 1H), 3.62 (br s, 1H), 3.36 (s, 3H), 3.30 (d, *J*=11.0 Hz, 1H), 3.24 (s, 1H), 2.71 (d, *J*=11.0 Hz, 1H), 1.39 (s, 3H), 1.34 (d, *J*=6.5 Hz, 3H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  102.0, 73.2, 70.0, 64.2, 55.5, 23.5, 18.1; HRMS (MALDI) calcd for C<sub>8</sub>H<sub>15</sub>N<sub>3</sub>O<sub>4</sub>Na (M+Na<sup>+</sup>) *m/z*: 240.0955, found 240.0956.

#### 4.7. Methyl 4-azido-4,6-dideoxy-3-*C*,2-*O*-dimethyl-α-D*talo*-pyranoside

To a stirred solution of 10 (70 mg, 0.322 mmol) in Et<sub>2</sub>O (2.0 mL) at room temperature were added MeI (0.200 mL, 3.22 mmol) and freshly prepared  $Ag_2O^{29}$  (0.224 g, 0.967 mmol). The reaction mixture was protected from light and heated to reflux. After stirring under reflux for 5 h, the reaction was diluted with EtOAc (10 mL), filtered through a pad of Celite (EtOAc rinse) and concentrated. Purification by flash chromatography (silica, 5-15% EtOAc/hexanes) provided the azidomethyl glycoside as an oil (53 mg, 72% yield).  $R_f$  0.33 (33%) EtOAc/hexanes);  $[\alpha]_{D}$  +164.8 (c 0.708, CHCl<sub>3</sub>); IR (film) 3483, 2931, 2105, 1455, 1349, 1102, 1055, 597 cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  4.79 (s, 1H), 4.00 (qd, J=6.5, 1.6 Hz, 1H), 3.72 (d, J=1.0 Hz, 1H), 3.49(s, 3H), 3.36 (s, 3H), 3.11 (br s, 1H) 2.90 (s, 1H), 1.40 (d, J=1.0 Hz, 3H), 1.35 (d, J=6.5 Hz, 3H); <sup>13</sup>C NMR  $(125 \text{ MHz}, \text{CDCl}_3) \delta$  98.4, 82.0, 70.1, 69.6, 64.3, 59.8, 55.2, 24.5, 17.9; HRMS (MALDI) calcd for C<sub>9</sub>H<sub>17</sub>N<sub>3</sub>O<sub>4</sub>Na  $(M+Na^+)$  m/z: 254.1111, found 254.1112.

# 4.8. Methyl 4-(*tert*-butoxycarbonylamino)-4,6-dideoxy-3-*C*,2-*O*-dimethyl-α-D-*talo*-pyranoside 11

To a stirred solution of the methyl glycoside (26 mg, 0.112 mmol) and Boc<sub>2</sub>O (49 mg, 0.225 mmol) in EtOAc (0.50 mL) at room temperature under argon was carefully added Pd/C (10%, 2 mg). The reaction flask was evacuated and placed under an H<sub>2</sub> atmosphere. After 4 h more Pd/C (10%, 2 mg) was added. After stirring for 1 h the reaction flask was purged with argon, the mixture was filtered through a pad of Celite and concentrated. Purification by flash chromatography (silica, 10–20% EtOAc/hexanes with 2% Et<sub>3</sub>N) provided 11 as an oil (21 mg, 62% yield). R<sub>f</sub> 0.28 (33% EtOAc/hexanes);  $[\alpha]_{D}$  +59.4 (c 0.500, CHCl<sub>3</sub>) IR (film) 3429, 2929, 1716, 1503, 1170, 1060 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  5.30 (d, J=10.4 Hz, 1H), 4.73 (s, 1H), 4.01 (qd, J=6.4, 1.7 Hz, 1H), 3.49 (dd, J=10.4, 1.7 Hz, 1H), 3.47 (s, 3H), 3.37 (s, 3H), 3.23 (s, 1H) 2.91 (s, 1H), 1.44 (s, 9H), 1.41 (s, 3H), 1.15 (d, J=6.4 Hz, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 156.5, 98.6, 83.0, 79.3, 68.0, 65.0, 59.4, 58.4, 55.2, 28.3, 23.8, 17.1; HRMS (MALDI) calcd for  $C_{14}H_{27}NO_6Na$  (M+Na<sup>+</sup>) m/z: 328.1730, found 328.1731.

# 4.9. α-D-Methyl callipeltose 12

To a stirred solution of 11 (28 mg, 0.0917 mmol) in THF (0.50 mL) at 0°C was added *t*-BuOK (0.9 M solution in THF, 0.204 mL, 0.183 mmol) dropwise and

the cooling bath was removed. After 4 h at room temperature more t-BuOK (0.051 mL, 0.46 mmol) was added. After 1 h sat. aq. NH<sub>4</sub>Cl (1 mL) was added, the aqueous layer was extracted with EtOAc ( $2 \times 5$  mL), the combined organic extracts were washed with brine (5 mL) and dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated. Purification by flash chromatography (silica, 1-3% $MeOH/CH_2Cl_2$ ) provided 15 mg of 12 as a white solid (69% yield). R<sub>f</sub> 0.28 (5% MeOH/CH<sub>2</sub>Cl<sub>2</sub>); mp 145-146°C (hexanes) (lit.<sup>8</sup> 147–148°C);  $[\alpha]_{D}$  +91.6 (*c* 1.06, CHCl<sub>3</sub>) (lit.<sup>9</sup>  $[\alpha]_{D}$  +76 (*c* = 1.0, CHCl<sub>3</sub>)); IR (film) 3295, 2931, 1747, 1376, 1267, 1104, 1060, 670 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, MeOH- $d_4$ )  $\delta$  4.48 (d, J=6.1 Hz, 1H), 3.94 (qd, J=6.5, 2.0 Hz, 1H), 3.52 (s, 3H), 3.45 (d, J=2.0Hz, 1H), 3.41 (s, 3H), 3.38 (d, J = 6.1 Hz, 1H), 1.50 (s, 3H), 1.12 (d, J=6.5 Hz, 3H); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  6.69 (br s, 1H), 4.60 (d, J=5.6 Hz, 1H), 3.89 (qd, J=6.5, 2.0 Hz, 1H), 3.53 (s, 3H), 3.41 (s, 3H), 3.37(d, J=2.0 Hz, 1H), 3.19 (d, J=5.6 Hz, 1H), 1.54 (s, 3H), 1.17 (d, J = 6.5 Hz, 3H); <sup>13</sup>C NMR (100 MHz,  $CDCl_3$ )  $\delta$  159.1, 101.6, 81.9, 81.1, 63.7, 61.4, 60.6, 55.0, 23.3, 15.6; HRMS (MALDI) calcd for C<sub>10</sub>H<sub>17</sub>NO<sub>5</sub>Na (M+Na<sup>+</sup>) m/z: 254.0999, found 254.1001.

#### 4.10. α-D-Callipeltose 4

A solution of 12 (4.3 mg, 0.186 mmol) in  $H_2SO_4$  (2 M in H<sub>2</sub>O/1,4-dioxane (1:1), 0.20 mL) was heated to  $60^{\circ}$ C for 30 h. Saturated aq. NaHCO<sub>3</sub> (1 mL) was added, the solution was saturated with NaCl and extracted with 25% EtOH/CHCl<sub>3</sub> (20×1 mL). The combined organic extracts were washed with brine and dried  $(Na_2SO_4)$ , filtered and concentrated. Purification by flash chromatography (silica, 5% MeOH/CH<sub>2</sub>Cl<sub>2</sub>) afforded D-callipeltose 4 as a film (1.9 mg, 48% yield).  $R_{\rm f}$  0.13 (5% MeOH/CH<sub>2</sub>Cl<sub>2</sub>); [*α*]<sub>D</sub> +25.3 (*c* 0.150, CHCl<sub>3</sub>); IR (film) 3313, 2924, 1737, 1449, 1273, 1261, 1061, 667 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  5.56 (s, 1H), 5.13 (d, J = 5.9Hz, 1H), 4.06 (qd, J=6.5, 1.9 Hz, 1H), 3.60 (s, 3H), 3.38 (d, J=1.9 Hz, 1H), 3.24 (d, J=5.9 Hz, 1H), 2.99 (br s, 1H), 1.56 (s, 3H), 1.17 (d, J=6.5 Hz, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 158.3, 95.0, 81.9, 81.8, 63.6, 61.3, 60.8, 23.3, 15.8; HRMS (MALDI) calcd for  $C_9H_{15}NO_5Na$  (M+Na<sup>+</sup>) m/z: 240.0842, found 240.0843.

#### Acknowledgements

We thank Dr. D. H. Huang and Dr. G. Siuzdak for NMR spectroscopic and mass spectrometric assistance, respectively. Financial support for this work was provided by The Skaggs Institute for Chemical Biology, The Ministry of Education Graduate School Program (Finland) and the Emil Aaltonen Foundation (both to A.J.P.).

#### References

 Zampella, A.; D'Auria, M. V.; Minale, L.; Debitus, C.; Roussakis, C. J. Am. Chem. Soc. 1996, 118, 11085–11088.

- Zampella, A.; D'Auria, M. V.; Minale, L.; Debitus, C. *Tetrahedron* 1997, 53, 3243–3248.
- 3. Hoye, T. R.; Zhao, H. Org. Lett. 1999, 1, 169-177.
- 4. Velázquez, F.; Olivo, H. F. Org. Lett. 2000, 2, 1931–1933.
- Olivo, H. F.; Velázquez, F.; Trevisan, H. C. Org. Lett. 2000, 2, 4055–4058.
- Paterson, I.; Davies, R. D. M.; Marquez, R. Angew. Chem., Int. Ed. 2001, 40, 603–607.
- 7. Evans, D. A.; Burch, J. D. Org. Lett. 2001, 2, 503-505.
- Smith, G. R.; Finley, IV, J. J.; Giuliani, R. M. Carbohydr. Res. 1998, 308, 223–227.
- 9. Gurjar, M. K.; Reddy, R. Carbohydr. Lett. 1998, 3, 169–172.
- 10. On a larger scale, D-glucal can be prepared from tri-O-acetyl-D-glucal by saponification of the three acetyl groups with  $K_2CO_3$  in MeOH.
- 11. Brimacombe, J. S.; Da'aboul, I.; Tucker, L. C. N. Carbohydr. Res. 1971, 19, 276–280.
- 12. Boivin, J.; Montagnac, A.; Monneret, C.; Païs, M. Carbohydr. Res. 1980, 85, 223-242.
- Fraser-Reid, B.; Kelly, D. R.; Tulshian, D. B.; Ravi, P. S. J. Carbohydr. Chem. 1983, 2, 105–114.
- 14. Thiem, J.; Elvers, J. Chem. Ber. 1981, 114, 1442-1454.
- Csuk, R.; Huegener, M.; Vasella, A. Helv. Chim. Acta 1988, 71, 609–618.
- Hanessian, S.; Vatèle, J.-M. Tetrahedron Lett. 1981, 22, 3579–3582.
- Nicolaou, K. C.; Li, J.; Zenke, G. Helv. Chim. Acta 2000, 83, 1977–2006.
- Preparation of tetra-n-butylammonium azide: Brändström, A.; Lamm, B.; Palmertz, I. Acta Chem. Scand. B 1974, 28, 699–701. Commercially available from TCI America. The commercial material was used in this synthesis.
- Solutions of tetra-n-butylammonium azide in CH<sub>2</sub>Cl<sub>2</sub> are reported to form explosive products on storage. If stored for longer times, the compound should be dissolved in toluene, which gives a stable solution: Hansson, T. G.; Kihlberg, J. O. J. Org. Chem. 1986, 51, 4490–4492.
- β-Keto azides have been found to be prone to epimerization: Evans, D. A.; Britton, T. C.; Ellman, J. A.; Dorow, R. L. J. Am. Chem. Soc. 1990, 112, 4011-4030.
- 21. The C-(4) epimer was also removed at this stage.
- Nicolaou, K. C.; Groneberg, R. D. J. Am. Chem. Soc. 1990, 112, 4085–4086.
- 23. Barili, P. L.; Berti, G.; Catelani, G.; D'Andrea, F. Tetrahedron Lett. 1991, 32, 959–962.
- 24. Bellucci, G.; Catelani, G.; Chiappe, C.; D'Andrea, F. *Tetrahedron Lett.* **1994**, *35*, 8433–8436.
- Upreti, M.; Vishwakarma, R. A. *Tetrahedron Lett.* 1999, 40, 2619–2622.
- 26. Saleh, T.; Rousseau, G. Synlett 1999, 617-619.
- Rainier, J. D.; Allwein, S. P.; Cox, J. M. Org. Lett. 2000, 2, 231–234.
- Rainier, J. D.; Allwein, S. P.; Cox, J. M. J. Org. Chem. 2001, 66, 1380–1386.
- Silver(I) oxide was prepared according to: Pearl, I. A. Org. Syn. Coll. Vol. IV, 972–973, washed with MeOH and Et<sub>2</sub>O and dried in vacuo.
- Saito, S.; Nakajima, H.; Inaba, M.; Moriwake, T. *Tetra*hedron Lett. **1989**, 30, 837–838.
- 31. Bull, S. D.; Davies, S. G.; Jones, S.; Sanganee, H. J. J.

*Chem. Soc., Perkin Trans. 1* **1999**, 387–389. Giuliani et al. exploited a similar cyclization for the corresponding Cbz-protected amide using NaH.<sup>8</sup>

- 32. Brimacombe, J. S.; Rahman, K. M. M. J. Chem. Soc., Perkin Trans. 1 1985, 1067–1072.
- 33. Jütten, P.; Scharf, H.-D. Carbohydr. Res. 1991, 212,

93-108.

- 34. Harwood, L. M. Aldrichim. Acta 1985, 18, 25.
- Horton, D.; Issa, M.; Priebe, W.; Szhaidman, M. L. Carbohydr. Res. 1993, 246, 105–118.
- Czernecki, S.; Vijayakumaran, K.; Ville, G. J. Org. Chem. 1986, 51, 5472–5475.