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## Chinese Chemical Letters

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Original article

## A two-photon mitotracker based on a naphthalimide fluorophore: Synthesis, photophysical properties and cell imaging

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## ARTICLE INFO

## Article history:

Received 27 March 2014

Received in revised form 26 April 2014

Accepted 29 April 2014

Available online xxx

## Keywords:

Mitotracker

Mitochondria

Naphthalimide

Two-photon

Fluorescence imaging

## ABSTRACT

PAHPN, a naphthalimide-based mitotracker with reasonable two-photon excitation emission activity and polarity-sensitive fluorescence properties has been efficiently synthesized and studied in two-photon, co-localization, and FLIM imaging.

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## 1. Introduction

Mitochondria, as cell's power producers, are important organelles for cellular respiration that ultimately generates fuel for the cellular activities [1]. In addition, mitochondria are involved in other tasks such as signaling, cellular differentiation, cell death, as well as the control of the cell cycle and cell growth [2]. In particular, mitochondria's key role in activating apoptosis has attracted much attention [3]. Mitochondria are highly dynamic organelles that continuously move, divide and fuse in a highly regulated fashion during various cellular processes [4]. For example, mitochondrial fission accompanies apoptotic cell death and appears to be important for the progression of the apoptotic pathways [5].

Mitotracker [6], namely mitochondria targeted fluorescent probes, have proved valuable tools to visualize mitochondria's dynamic changes during apoptosis and other cellular processes. However, the common mitotracker, e.g., Rhodamine 123 and tetramethylrhodamine methyl ester (TMRM), are not efficient two-photon fluorophores, which restricted their application in TPEE (two photon excitation emission) microscopy. This newly

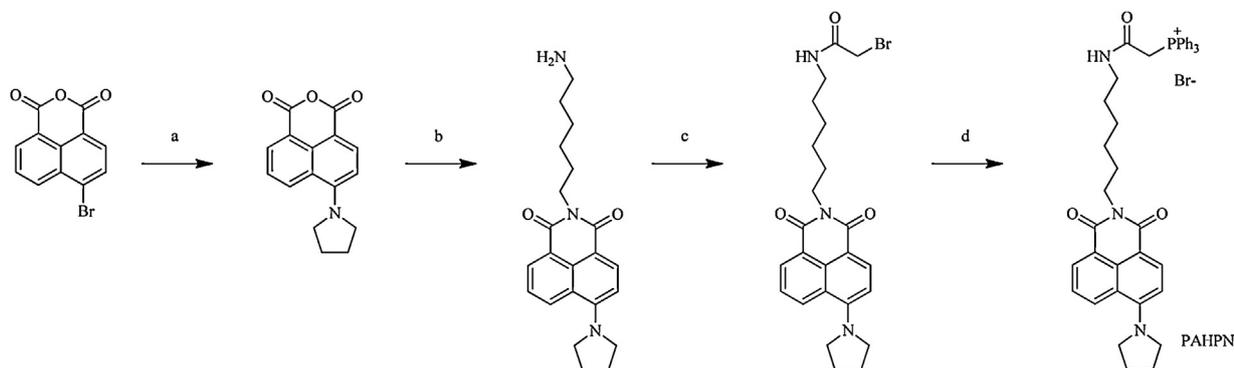
emerging imaging technique using pulsed NIR excitation can be a superior alternative to confocal microscopy (one photon imaging) due to its deeper tissue penetration (>500 μm), efficient light detection, and reduced phototoxicity [7,8]. Due to its imaging mechanism, it also provides advantages such as highly localized excitation and prolonged observation time. These are desired imaging properties for trackers and the development of two-photon mitotracker represents a critical priority.

Herein, we have developed PAHPN, a new mitotracker with reasonable TPEE activity. There are two key aspects in our design. Firstly, we chose 4-pyrrolidino-1,8-naphthalimide as the fluorophore because its TPEF activity is foreseeable since another 4-amino naphthalimide probe showing strong TPEE has been reported by ourselves very recently [9]. Secondly, a triphenylphosphonium (TPP) moiety has been attached to the naphthalimide fluorophore via a flexible long alkyl chain, which generates a positively charged but highly lipophilic molecule that tends to accumulate in mitochondrial inner membranes at negative potentials [10,11].

As shown in Scheme 1, the synthesis of PAHPN can be achieved from 4-Br-1,8-naphthalic anhydride in four steps, namely, nucleophilic substitution, condensation, acylation and quaternization. Each reaction is facile under mild conditions, thanks to Qian's pioneering work on the modifications of the naphthalimide platform [12].

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**Scheme 1.** Synthesis of PAHPN. (a) Pyrrolidine, 2-methoxyethanol, reflux, 72%. (b) Hexamethylenediamine, ethanol, reflux, 80%. (c) Bromoacetic acid, DCC, dichloromethane, r.t., 94%. (d) Triphenylphosphine, dichloromethane, r.t., 35%.

## 2. Experimental

Culture of cells and fluorescent imaging: MCF7 (human breast carcinoma) cells, RAW 264.7 (macrophages cells) and COS-7 cells were obtained from Institute of Basic Medical Sciences (IBMS) of Chinese Academy of Medical Sciences (CAMS). All cell lines were maintained under standard culture conditions (atmosphere of 5% CO<sub>2</sub> and 95% air at 37 °C) in RPMI 1640 medium, supplemented with 10% FBS (fetal calf serum).

Cells were grown in the exponential phase of growth on 35 mm glass-bottom culture dishes (Ø20 mm) for 1–2 days to reach 70–90% confluency. These cells were used in co-localization experimentation. The cells were washed three times with RPMI 1640, and then incubated with 2 mL of RPMI 1640 containing probes (1 µmol/L) in an atmosphere of 5% CO<sub>2</sub> and 95% air for 3 min at 37 °C. Cells were washed twice with 1 mL of PBS at room temperature and observed under a confocal microscopy (Olympus FV1000).

The 400 MHz <sup>1</sup>H NMR and 100 MHz <sup>13</sup>C NMR spectra were collected at room temperature and were given in supporting information. Melting points were obtained with a capillary melting point apparatus in open-ended capillaries and were uncorrected. Chromatographic purifications were conducted using silica gel. All solvent mixtures are given as volume/volume ratios.

**Compound 1:** 4-Bromo-1,8-naphthalic anhydride (10 g, 36.2 mmol) was dissolved in ethylene glycol monomethyl ether under reflux. Pyrrolidine (5 mL, 66.4 mmol) was added in four portions in 2 h. After the addition of pyrrolidine, the mixture was refluxed for one more hour. After the reaction mixture was cooled to room temperature, the yellow solid was collected (7 g, 72.4%).

**Compound 2:** Compound 1 (1 g, 3.74 mmol) and hexamethylenediamine (1.3 g, 11.23 mmol) were refluxed in 10 mL ethanol. The reaction mixture was cooled to room temperature. After the ethanol was removed under reduced pressure, the residue was recrystallized from ethanol to give the compound 2 (1 g, 80%). Mp. 60–61.4 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 1.48 (m, 6H), 1.74 (s, 2H), 2.1 (d, 6H), 2.71 (t, 2H), 3.77 (s, 4H), 4.16 (t, 2H), 6.80 (d, 1H), 7.52 (t, 1H), 8.41 (d, 1H), 8.58–8.54 (m, 2H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 26.2, 26.7, 27, 28.2, 29.8, 33.2, 40.1, 42.1, 53.3, 108.7, 111, 122.8, 123.2, 131.1, 131.3, 132, 133.5, 152.8, 164.2, 165.0. HRMS (MALDI-TOF) (*m/z*): Calcd. for C<sub>22</sub>H<sub>27</sub>N<sub>3</sub>O<sub>2</sub>: 365.2103, found: 366.2192 ([M+H]<sup>+</sup>).

**Compound 3:** Compound 2 (0.2 g, 0.55 mmol), bromoacetic acid (47.24 µL, 0.66 mmol) and DDC (160 mg, 0.66 mmol) were stirred in CH<sub>2</sub>Cl<sub>2</sub> at room temperature for 6 h. The insoluble materials were filtered off and the filtrate was evaporated to provide the compound 3 (0.25 g, 94.0%). Mp 79–80.2 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 1.15 (m, 2H), 1.74 (m, 4H), 1.93 (d, 2H), 2.11 (s, 4H), 3.29 (m, 2H), 3.79 (s, 4H), 3.9 (s, 2H), 4.18 (t, 2H), 6.69 (s, 1H), 6.82

(d, 1H, *J* = 8.8 Hz), 7.54 (t, 1H), 8.42 (d, 1H), 8.58 (t, 2H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 25.1, 25.7, 26.3, 28, 29.1, 29.5, 34, 39.9, 40.2, 49.4, 53.3, 108.3, 110.6, 122.7, 123.2, 131.2, 132.1, 133.2, 152.6, 164.3, 165, 165. HRMS (MALDI-TOF) (*m/z*): Calcd. for C<sub>24</sub>H<sub>28</sub>BrN<sub>3</sub>O<sub>2</sub>: 485.1314, found: 486.1394 ([M+H]<sup>+</sup>).

**PAHPN:** a mixture of compound 3 (0.12 g, 0.247 mmol) and triphenylphosphine was stirred in CH<sub>2</sub>Cl<sub>2</sub> at room temperature for 5 h. After the CH<sub>2</sub>Cl<sub>2</sub> was removed under reduced pressure, the residue was purified by silica gel column chromatography using eluent CH<sub>2</sub>Cl<sub>2</sub>/MeOH (20/1, v/v). A yellow solid was obtained (0.065 g, 35.2%). Mp 94–96 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 1.31–1.16 (m, 6H), 1.62 (m, 2H), 2.09 (s, 4H), 3.05 (d, 2H, *J* = 6 Hz), 3.76 (s, 4H), 4.09 (t, 2H), 5.02 (d, 2H, *J* = 14 Hz), 6.79 (d, 1H, *J* = 8.8 Hz), 7.50 (t, 1H), 7.64 (m, 6H), 7.82 (m, 9H), 8.38 (d, 1H, *J* = 8.8 Hz), 8.55 (m, 2H), 9.29 (s, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 26.1, 26.8, 26.9, 28.1, 29, 31.9, 32.5, 40.1, 40.2, 53.2, 108.6, 110.7, 118.1, 119, 122.5, 123.1, 130.1, 130.2, 131.0, 133.4, 134.1, 134.3, 135.0, 152.7, 162.1, 164, 164.8. HRMS (MALDI-TOF) (*m/z*): Calcd. for C<sub>42</sub>H<sub>43</sub>BrN<sub>3</sub>O<sub>3</sub>P: 747.2225, found: 668.2996 ([M–Br]<sup>+</sup>).

## 3. Results and discussion

PAHPN exhibits polarity-sensitive fluorescence properties. Its absorption and emission spectra in various solvents are shown in Fig. 1, and the basic data are listed in Table 1. Briefly, with the increase in polarity, the fluorescence spectra red-shift to longer wavelength range and the fluorescence quantum yields decrease sharply. For example, in toluene, the emission maximum is at 500 nm and the quantum yield is 0.96, while in acetonitrile, the emission peak moves to 541 nm and the quantum yield declines to 0.14. However, the absorption properties are less dependent on solvent polarity. Although the absorption spectra also shift in various solvents, the difference in molar extinction coefficients is not as significant as that of fluorescence quantum yields. The sharply different fluorescence is an advantage for PAHPN's application in mitochondria imaging since those PAHPN molecules localized in mitochondrial inner membrane would emit strong fluorescence due to the nonpolar lipophilic environment while the background noise from some PAHPN distributed in other aqueous intracellular compartments would be low. Its fluorescence life times ( $\tau$ ) in various solvents are given in Fig. 1, and the average  $\tau$  values are listed in Table 1. Fluorescence life time shows a decrease with the increase in polarity, which can be used to measure the polarity of mitochondria.

The sensitivity of fluorescence toward polarity could be explained by TICT (twisted intramolecular charge transfer, or twisted ICT) mechanism [13]. PAHPN's fluorophore, 4-pyrrolidino-1,8-naphthalimide accords with the standard TICT structure: the strong electron donor (pyrrolidino) is connected to the strong

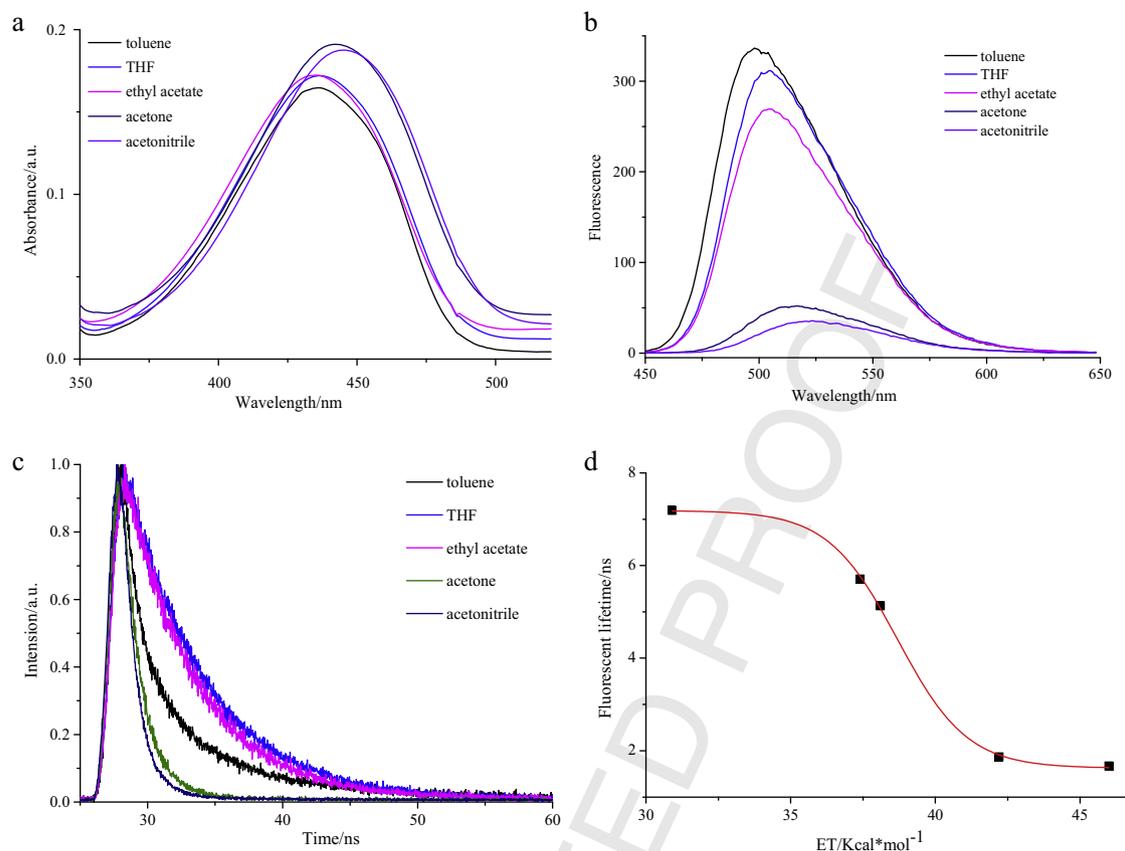


Fig. 1. Absorption (a) and emission (b) spectra of PAHPN in solvents of different polarities; fluorescence life time of PAHPN in solvents of different polarities (c) and (d).

electron acceptor (naphthalimide) by a single bond. Upon excitation of such a fluorophore, the single bond will be twisted quickly so that the donor plane and the acceptor plane become perpendicular to each other. Such a twisted configuration in excited state means a charge separation between the donor and acceptor, or a complete charge transfer. This movement of the single bond consumes the excitation energy and thus quenches the fluorescence sharply. As we know, highly polar solvents can stabilize and thus promote TICT state, while nonpolar solvents destabilize and disfavor TICT.

PAHPN demonstrates a reasonable two-photon excitation activity. As shown in Fig. 2, the two-photon excitation spectrum of PAHPN in toluene exhibits a broad band from 700 to 1000 nm and the maximum is at 820 nm, which is approximately the double of the one-photon absorption maximum. The TPEE activity across section ( $\Phi \times \delta$ , the product of  $\Phi$  and  $\delta$ ) at 820 nm is around 90 GM, which is a reasonable value and higher than that of several common fluorophores. In other word, the intrinsic brightness of PAHPN's TPEE is sufficient for two-photon microscopic applications.

Table 1  
Spectral data of PAHPN.

Solvent	$\lambda_{ab}$ (nm)	$\lambda_{em}$ (nm)	$\Phi^a$	$\epsilon$	$\tau$ (ns)
Toluene	436	500	0.96	15,400	7.19
THF	436	505	0.98	15,900	5.71
Ethyl acetate	435	506	0.97	15,900	5.13
Acetone	443	517	0.20	17,700	1.87
CH <sub>3</sub> CN	446	523	0.14	17,400	1.67

<sup>a</sup> Quinine sulfate ( $\Phi=0.55$ ) as reference.

The applicability of PAHPN as a two-photon fluorescent cell marker has been confirmed by two-photon microscopy imaging experiments. As shown in Fig. 3, strong TPEE has been observed in MCF-7 cells stained with PAHPN. The distribution of fluorescence clearly indicates that the PAHPN molecules are not uniformly dispersed but localized in certain areas. Mitochondria usually assemble into networks, although single ones have an ellipsoid shape [6]. In Fig. 3, fluorescent networks are easily visualized with high resolution because the areas surrounding these networks are almost 'dark' (without background noise). Thus, the quality of two-photon imaging is high.

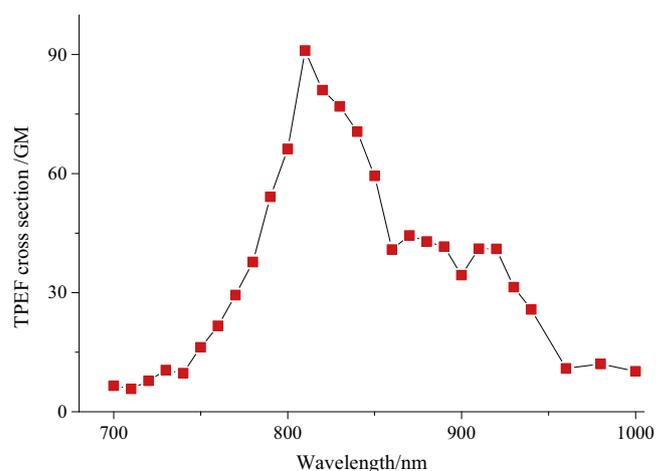
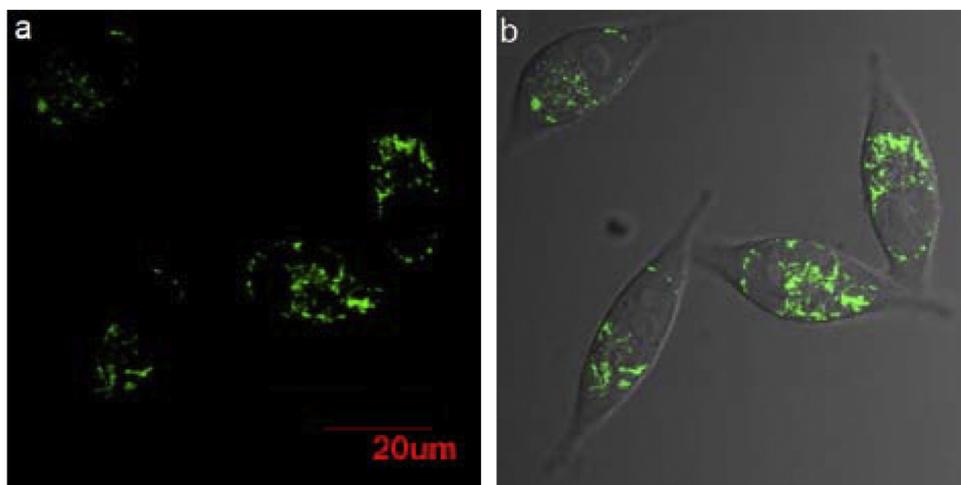


Fig. 2. Two-photon excitation spectrum of PAHPN in toluene.



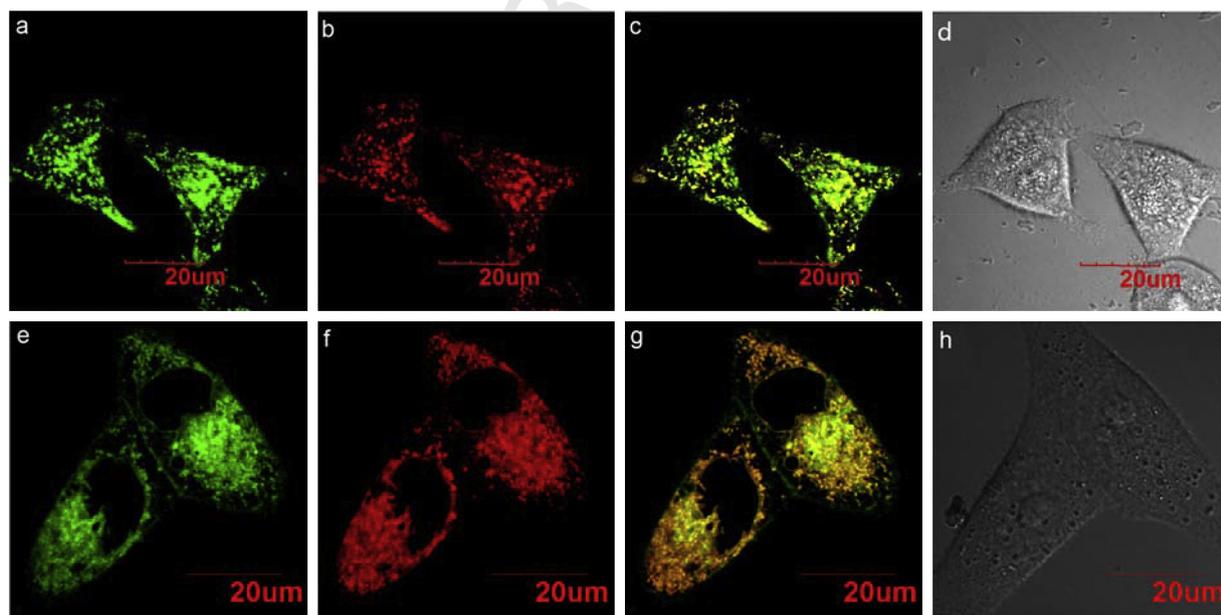
**Fig. 3.** Two photon fluorescence image of MCF-7 cells incubated with PAHPN (1  $\mu\text{mol/L}$ ) for 3 min at 37  $^{\circ}\text{C}$ . (a) Fluorescence image ( $\lambda_{\text{ex}} = 840 \text{ nm}$ ,  $\lambda_{\text{em}} = 520\text{--}560 \text{ nm}$ ). (b) Overlay of fluorescence image and brightfield image.

182 Actually, we have also applied PAHPN in one-photon confocal  
183 imaging (Fig S1 in supporting information) and have found that  
184 there is no difference in imaging quality between two-photon  
185 and one-photon microcopies.

186 Furthermore, PAHPN proves to be a specific mitotracker. A  
187 colocalization imaging experiment has been performed. PAHPN  
188 and two commercial mitrotrackers (tetramethylrhodamine  
189 methyl ester, TMRM and mitotracker deep red) are adopted  
190 to co-stain different cell lines including MCF-7 cells, RAW264.7  
191 cells and COS-7 cells. As TMRM and mitotracker deep red emits  
192 at considerably longer wavelengths than PAHPN, their different  
193 fluorescence signals can be collected in two channels without  
194 crosstalk. As shown in Fig. 4, in MCF-7 cells, the green  
195 fluorescence in the PAHPN channel overlaps well with the red  
196 fluorescence in the TMRM (or mitotracker deep red) channel,  
197 which produces a high colocalization coefficient up to 0.92 (or  
198 0.85 in the case of mitotracker deep red) and indicates that

199 PAHPN and TMRM (or mitotracker deep red) are localized in  
200 similar intracellular areas, namely, mitochondria. Similar  
201 colocalization results are also obtained in RAW264.7 and  
202 COS-7 cells, as shown in the supporting information (Figs. S2  
203 and S3) In other words, PAHPN shows stable localization in  
204 mitochondria of different cell lines. Thus, it can be concluded  
205 that PAHPN is mitochondria-targeted.

206 Finally, PAHPN is used in fluorescence life time imaging (FLIM)  
207 to reflect polarity in mitochondria. Fig. 5 shows heterogeneous  
208 polarity distribution in mitochondria in a single cell or in different  
209 cells, which indicates that polarity of mitochondria under different  
210 physiology status could be different. Thus, PAHPN can be used as  
211 an indicator for polarity in mitochondria. In future, the combina-  
212 tion of PAHPN's polarity-sensitivity and FLIM technique will be  
213 further applied in order to monitor the mitochondrial micro-  
214 environmental variations under different physiological, pharma-  
215 cological, and toxicological conditions.



**Fig. 4.** Colocalization imaging of MCF-7 cells costained by PAHPN and commercial mitotracker TMRM (a-d) mitotracker deep red (e-h) for 3 min at 37  $^{\circ}\text{C}$ . (a and e) Fluorescence image of PAHPN ( $\lambda_{\text{ex}} = 405 \text{ nm}$ ,  $\lambda_{\text{em}} = 470\text{--}530 \text{ nm}$ ). (b) Fluorescence image of TMRM,  $\lambda_{\text{ex}} = 559 \text{ nm}$ ,  $\lambda_{\text{em}} = 600\text{--}630 \text{ nm}$ . (c) Overlay image of a and b, colocalization coefficient 0.92. (f) Fluorescence image of mitotracker deep red,  $\lambda_{\text{ex}} = 635 \text{ nm}$ ,  $\lambda_{\text{em}} = 655\text{--}755 \text{ nm}$ . (g) Overlay image of e and f, colocalization coefficient 0.85. (d and h) Brightfield image.

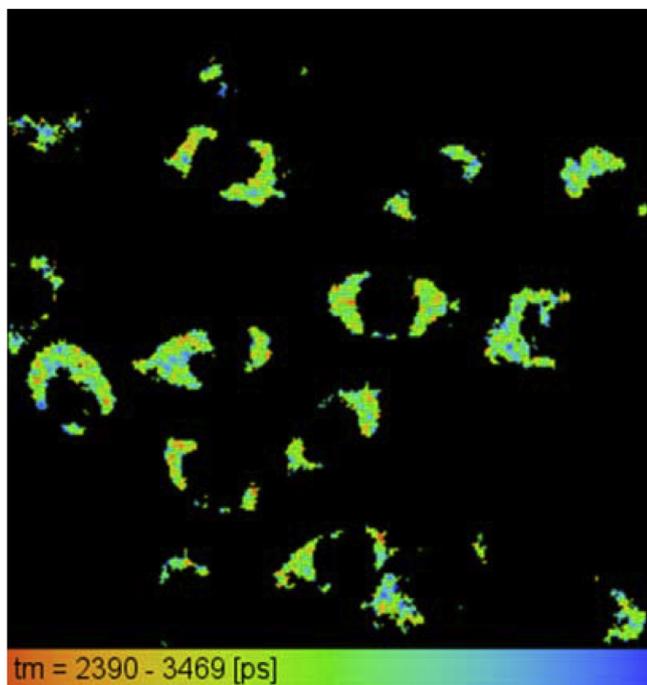


Fig. 5. Fluorescence life time imaging of MCF-7 cells (1 mol/L PAHPN for 3 min at 37 °C). The excited light is 405 nm, observing emission wavelength at 535 ± 15 nm.

#### 4. Conclusion

In conclusion, PAHPN, a new two-photon mitotracker has been efficiently synthesized through the attachment of a triphenylphosphonium moiety to the naphthalimide fluorophore. PAHPN exhibits strong fluorescence in nonpolar solvents but extremely weak fluorescence in water, which favors the reduction of background fluorescence from the aqueous environment outside the mitochondrial membrane. This new molecular probe shows reasonable two-photon excitation emission activity and presents high-quality fluorescence images under two-photon microscopy. PAHPN's specificity toward mitochondria has been confirmed by the colocalization imaging studies on three different cell lines with two commercial mitotrackers TMRM and mitotracker red as the references. FLIM imaging using PAHPN has mapped the different polarity of mitochondria. The applications of PAHPN in the

investigations on cellular processes related to mitochondrial dynamics are now being studied.

#### Acknowledgments

We thank National Natural Science Foundation of China (Nos. 21174022, 21376038), National Basic Research Program of China (No. 2013CB733702), Key Project of the education department of Sichuan Province (No. 12ZA087), and Specialized Research Fund for the Doctoral Program of Higher Education (No. 20110041110009).

#### Appendix A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at <http://dx.doi.org/10.1016/j.ccllet.2014.05.020>.

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