Stereoselective Synthesis of *trans-2*,5-Disubstituted Tetrahydrofurans *via* the Lewis Acid Mediated Reduction of Cyclic Hemiketals with Triphenylsilane

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An efficient method for the synthesis of *trans*-2,5-disubstituted tetrahydrofuran derivatives has been developed. Dithioacetal-functionalized cyclic hemiketals are reduced by triphenylsilane (Ph₃SiH) in the presence of TiCl₄ to afford the corresponding *trans*-2,5-disubstituted tetrahydrofuran derivatives in good yield and high stereoselectivity. Asymmetrical synthesis of (2S,5S) disubstituted tetrahydrofuran was also achieved by using this reduction method.

There is an increasing interest in the stereocontrolled construction of 2,5-disubstituted tetrahydrofurans¹ in connection with the synthesis of polyethers antibiotics and natural products containing oxacyclic units.² In a previous papers on the synthesis of polytetrahydofuran compounds,³ we have reported a novel and efficient construction of *trans*-2,5-disubstituted tetrahydrofurans *via* stereocontrolled hydride reduction of cyclic hemiketals (1) into 1,- diols followed by their stereospecific one-step cyclodehydration.^{3a}

In the course of our study on the synthesis of chiral polytetrahydofuran compounds, we found a high stereoselective synthetic method of *trans*-2,5-disubstituted tetrahydrofurans (2) *via* reduction of dithioacetal-functionalized cyclic hemiketals (1) with Ph₃SiH (3) in the presence of TiCl₄ (Eq. 1).^{4,6}

Dihydro-2,5-Diphenyl-3,3-trimethylenedithiofuran-2-ol(1a) was chosen as a model substrate and allowed to react with organosilanes in the presence of Lewis acid. Table 1 shows the results of the reduction of 1a under various reaction conditions. When 1 . was treated with one equivalent of 3 and TiCl₄ in CH₂Cl₂ solution at 0°C, trans-2,5-diphenyl-3,3-trimethylendithiotetrahydrofuran (2a) was formed in 76 % yield and moderate trans selectivity (trans:cis = 15:1) (Run 1). However, good yield of 2a was satisfactorily achieved by the use of small excess amount of TiCl₄ (1.2 equiv.) (Run 2). Lowering the reaction temperature (-78 °C) led to increase in the stereoselectivity of 2a (Run 3). Other Lewis acids such as BF₃•Et₂O, Et₂AlCl, EtAlCl₂, AlCl₃, and TMSOTf gave a considerable yield of 2a, but the stereoselectivity is a lower than that of TiCl₄ (Runs 3-8). In the case of organoaluminum compounds, Et₂AlCl, EtAlCl₂, and AlCl₃, increase of the number of the alkyl group on the aluminium compounds led to increase of the stereoselectivity (Run 5-7). In addition, it is noted that the kind of alkyl group on the organosilanes is also contributed to the stereoselectivity on 2a (Runs 3 and 9-11). Thus, when less bulky organosilanes e. g. Ph₂MeSiH, PhMe₂SiH, and Et₃SiH, than 3, were used as

Table 1. Selectivity on the Reduction of Hemiketal with R₃SiH in the Presence of Lewis acid^a

s >		s
/ -\ -s	R₃SiH	, \ s
Ph—(O)—Ph	Lewis acid CH ₂ Cl ₂ (3 mL), 5 min.	Ph O Ph
(1a)	Or 12012 (3 ITIL), 3 ITIIII.	(2a)

(1 <i>a)</i>			(£G)		
Run	R ₃ SiH	Lewis acid	Yield, % ^b (trans / cis) ^c		
1 ^d	Ph ₃ SiH	TiCl ₄	76 (15 / 1)		
2 ^e	Ph₃SiH	TiCl₄	87 (16 / 1)		
3	Ph₃SiH	TiCl ₄	88 (82 / 1)		
4	Ph ₃ SiH	BF ₃ ·OEt₂	86 (10 /1)		
5 ^f	Ph₃SiH	Et ₂ AICI	82 (28 /1)		
6	Ph₃SiH	EtAICI ₂	73 (18 / 1)		
7	Ph₃SiH	AICI ₃	76 (11 /1)		
8	Ph ₃ SiH	TMSOTf	88 (17 / 1)		
9	Ph ₂ MeSiH	TiCl ₄	70 (34 /1)		
10	PhMe ₂ SiH	TiCl ₄	79 (30 / 1)		
11	Et ₃ SiH	TiCl ₄	75 (7 / 1)		

^a Reaction conditions: **1a** (0.15 mmol), R_3SiH (0.15 mmol), Lewis acid (0.18 mmol), CH_2CI_2 (3 mL), -78°C, 5 min.

organosilanes, stereoselectivity of **2a** decreased (Runs 3 and 9-11)

In order to know the scope and limitation of the stereoselective synthesis of *trans*-2,5-disubstituted tetrahydrofurans derivatives, the reduction of various cyclic hemiketals with Ph₃SiH in the presence of TiCl₄ was examined under the same reaction conditions as shown in run 3 in Table 1 (Table 2).⁸ The transformation of 5-methyl-2-phenyl, 5-ethyl-2-phenyl, and 5-isopropyl-2-phenyl substituted hemiketals to the corresponding *trans* 2,5-disubstituted tetrahydrofuran derivatives proceeded very effectively (81-83%) and high stereoselectivity (trans:cis = 13:1 - 83:1) (Runs 1-3). 2-Alkyl substituted derivative, dihydro-2-methyl-5-diphenyl-3,3-trimethylene-dithiofuran-2-ol, was also reduced by 3 in the presence of TiCl₄ to afford trans-2-methyl-5-phenyl tetrahydrofuran derivatives in substantial yield and high stereoselectivity (Run 4)

Next, we examined the application of this method on the synthesis of the asymmetrical 2,5-disubstituted tetrahydrofuran. (2SR, 5S)-5-Methyl-2-phenyl-3,3-trimethylenedithiofuran-2-ol

b Isolated yield. C Determined by 400 Mz H NMR.

^d One equivalent of Lewis acid was used. ^e0°C. ^f 35 min.

Table 2. Reaction of Various Hemiketals with Ph₃SiH in the Presence of TiCl₄

Run	R ₁	R ₂	Yield / % ^a (trans / cis) ^b
1	СН₃	Ph	83 (13 / 1)
2	C_2H_5	Ph	81 (16 / 1)
3	CH(CH ₃) ₂	Ph	82 (9 / 1)
4	Ph	CH ₃	72 (63 / 1)
5	CH ₃	CH ₃	70 (10 / 1)
6	CH ₃	C_2H_5	63 (7 / 1)

a Isolated yield.

(4) was treated with Ph₃SiH in the presence of TiCl₄ to afford (2S, 5S)-dihydro-5-methyl-2-phenyl-3,3-trimethylenedithiofuran in 84 % yield and high stereoselectivity [(2S,5S):(2R,5S)=12.7:1.0] (Eq. 2). We have already shown the stereoselective

synthesis of (2R,5R)-5 via one-step cyclodehydration of anti-diol, generated by stereoselective reduction of 4 with LiAlH₄.^{3a} Both results show that synthesis of respective both enatiomers of trans disubstituted tetrahydrofurans from one chiral starting materials was successfully achieved by using these two methods.

Although we can not explain the reason of the high stereoselectivity in detail yet, we suggest that the organosilane attacks an oxonium ion intermediate (6), which was generated in situ from 1 and Lewis acid, from the perisheral side owing to the steric hindrance of 3,3-trimethylenedithio group and/or Lewis acid

Scheme 1

coodinated on oxygen atom of oxonium ion intermediate leading to trans isomer (Scheme 1). 10

Since 3,3-trimethylenedithio group can easily converted into the methylene by Raney nickel, the present method is a simple one for the synthesis of trans-2,5-disubstituted tetrahydrofuran derivatives. Further investigation is now under way on the mechanism and the synthetic application of the process.

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References and Notes

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- 4 Some workers reported that ketals and acetals were reduced by organosilane in the presence of Lewis acid to afford the corresponding ethers in moderate to good yields.⁵ However, to the best of our knowledge, there is no reports on the reduction of hemiketals and hemiacetals with organosilane.
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- 7 Configuration of each isomers was assigned by the NOE analysis for the ring protons. In the case of miner product (cis isomer), when irradiated at the frequency of C2 proton signal, NOE was observed of C5 proton signal. On the other hand, irradiation of C2 proton of major product (rans isomer) did not show an NOE fifference spectroscopy.
- 8 To a dichloromethane (3 mL) of cyclic hemiketal (0.15 mmol) was added TiCl₄ (0.18 mmol) under nitrogen atmosphere at -78°C, and the mixture was stirred at that temperature for 15 min. Triphenylsilane (0.15 mmol) was added to the resulting solution at -78°C, and the solution was stirred at that temperature for 5 min. The resulting mixture was quenched with aq. NaHCO₃, and extracted with diethyl ether. The combined extracts was dried over MgSO₄. Removal of the solvent under the reduced pressure afforded yellow liquid, which was purified by column chromatography on silica gel (n-hexane:AcOEt =10:1 v/v as eluant) to give the corresponding 2,5-disubstituted tetrahydrofuran derivatives.
- 9 The compounds was identified by comparision of its ¹H NMR with those of authentic samples, ^{3a)}
- 10 On the stereoselective synthesis of disubstituted cyclic ethers by the reaction of lactols with organometals in the presence of Lewis acid, high trans selectivity was exaplained by considering the attack of the nucleophiles on the oxonium ion intermediate from the opposite face to the substituent, see: Ref 6.

^b Determined by 400 Mz ¹H NMR.