Note

Synthesis of the 2- and 3-methyl ethers of L-rhamnose and methyl α -L-rhamnopyranoside

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A recent synthesis¹ of methyl 2-O- (1) and 3-O-methyl- α -L-rhamnopyranoside (2) involved metal ion-catalysed monomethylation of methyl α -L-rhamnopyranoside with diazomethane followed by chromatography. We were not able to repeat the chromatographic separation of 1 and 2, and we now report an alternative synthesis of these compounds, which are important reference compounds in the structural elucidation of microbial polysaccharides² and other natural products¹.

1
$$R^1 = Me, R^2 = R^3 = H$$

2
$$R^2 = Me_1R^1 = R^3 = H$$

5 $R^1 = Me_1R^2 = H_1R^3 = BzI$

$$6 R^1 = H, R^2 = Me, R^3 = BzI$$

$$7 R^1 = R^2 = H_1 R^3 = BzI$$

$$8 R^1 = R^3 = BzI, R^2 = Me$$

$$3 R^1 = Me, R^2 = H$$

 $4 R^1 = H, R^2 = Me$

10
$$R^1 = Me_1R^2 = H$$

11 $R^1 = BzI_1R^2 = Me$

The 4-benzyl ethers 5 and 6, obtained³ by treatment of methyl 4-O-benzyl- α -L-rhamnopyranoside (7) with methyl iodide in the presence of a tetrabutylammonium salt, can be separated³ easily by chromatography on Kieselgel and are therefore readily accessible precursors for 1 and 2. The minor product 6 can be prepared in a good yield by treatment⁴ of the 2,3-O-dibutylstannylene derivative of 7 with methyl iodide. Hydrogenolysis of 5 and 6 gave 1 and 2, respectively.

Partially protected derivatives of methyl α -L-rhamnopyranoside are available³, having HO-2 or HO-3 unsubstituted. Hence, 1 can be prepared also from methyl 3,4-di-O-benzyl- α -L-rhamnopyranoside by methylation followed by hydrogenolysis,

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and 2, likewise, from methyl 2,4-di-O-benzyl-3-O-methyl-α-L-rhamnopyranoside. 2-O-Methyl-L-rhamnose (3) can be prepared by hydrogenolysis of benzyl 4-O-benzyl-2-O-methyl-α-L-rhamnopyranoside³ (10), and the 3-methyl ether (4) by hydrogenolysis of benzyl 2,4-di-O-benzyl-3-O-methyl-α-L-rhamnopyranoside (11) obtained by methylation of benzyl 2,4-di-O-benzyl-α-L-rhamnopyranoside³.

EXPERIMENTAL

General methods. — Melting points (uncorrected) were determined on a Kosler hot-stage. T.l.c. was performed on Kieselgel (Merck, 5562) with A, light petroleum-ethyl acetate (3:2); B, benzene-methanol (3:1); C, benzene-methanol (100:3); and D, chloroform-methanol (9:1); and detection with u.v. light or by charring with sulphuric acid. Optical rotations were measured with a Perkin-Elmer 241 automatic polarimeter. P.m.r. spectra (100 MHz) were recorded for solutions in CDCl₃ with a JEOL MH-100 spectrometer.

Methyl 4-O-benzyl-3-O-methyl- α -L-rhamnopyranoside³ (6). — A solution of methyl 4-O-benzyl- α -L-rhamnopyranoside (265 mg) in dry methanol (10 ml) was stirred with dibutyltin oxide (270 mg) and boiled under reflux for 1 h, and then concentrated in a high vacuum. A solution of the resulting, slightly yellow syrup in N,N-dimethylformamide (5 ml) was stirred with methyl iodide (2 ml) at 45° overnight*. The mixture was concentrated and the residue was subjected to chromatography (solvents A or B), to give 6 (170 mg, 64%), $[\alpha]_D - 80^\circ$ (c 1.1, chloroform); lit.³ $[\alpha]_D - 83^\circ$.

Methyl 3-O-methyl-α-L-rhamnopyranoside (2). — Methylation (MeI and Ag₂O in N,N-dimethylformamide) of methyl 2,4-di-O-benzyl-α-L-rhamnopyranoside³ and chromatography of the product (solvent A) gave the 3-O-methyl derivative 8, $[\alpha]_D$ —29° (c 0.9, chloroform). P.m.r. data: δ 1.30 (d, 3 H, $J_{5,6}$ 6 Hz, Me), 3.28 (s, 3 H, OMe), 3.76 (m, 1 H, H-2), 4.72 (b, 1 H, H-1), 4.52–4.96 (m, 4 H, 2 CH₂-Ph), and 7.34 (m, 10 H, 2 Ph). Hydrogenolysis (10% Pd/C) of 6 or 8 in ethanol, with chromatography of the product on Kieselgel H (solvent D), gave 2, $[\alpha]_D$ —60° (c 1.2, chloroform); lit. $[\alpha]_D$ —61°.

Methyl 2-O-methyl- α -L-rhamnopyranoside (1). — Methylation of methyl 3,4-di-O-benzyl- α -L-rhamnopyranoside³, as described above, with chromatography of the product (solvent A or C), gave the syrupy 2-O-methyl derivative 9, $[\alpha]_D - 38^\circ$ (c 1.8, chloroform). P.m.r. data: δ 1.33 (d, 3 H, $J_{5,6}$ 6 Hz, Me), 3.32 (s, 3 H, OMe), 3.50 (s, 3 H, OMe), 3.86 (dd, 1 H, $J_{2,3} \sim 3$, $J_{3,4} \sim 9$ Hz, H-3), 4.70 (b, 1 H, H-1), 4.78 (q, 4 H, 2 C H_2 -Ph), and 7.36 (m, 10 H, 2 Ph). Hydrogenolysis of 5 or 9 and chromatography of the product, as described above, gave 1, $[\alpha]_D - 37^\circ$ (c 1.3, chloroform); lit. $[\alpha]_D - 49^\circ$.

2-O-Methyl-L-rhamnose (3). - Hydrogenolysis of benzyl 4-O-benzyl-2-O-

^{*}Note added in proof. Methylation can also be effected at 35-40° in the absence of N,N-dimethyl-formamide, to give 6 in 70% yield.

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methyl- α -L-rhamnopyranoside³ (10) and chromatography of the product (solvent B) gave syrupy 3, $[\alpha]_D + 25^\circ$ (c 0.3, water); lit.⁵ m.p. 113–114°, $[\alpha]_D + 31^\circ$; lit.⁶ $[\alpha]_D + 24^\circ$.

3-O-Methyl-L-rhamnose (4). — Methylation of benzyl 2,4-di-O-benzyl-α-L-rhamnopyranoside³, as described above for 8, gave the 3-O-methyl derivative (11) as a syrup, $[\alpha]_D - 51^\circ$ (c 1.3, chloroform). P.m.r. data: δ 1.30 (d, 3 H, $J_{5,6}$ 6 Hz, Me), 3.40 (s, 3 H, OMe), 3.42–3.90 (m, 4 H, H-2,3,4,5), 4.32–4.96 (m, 7 H, 3 CH₂Ph, H-1), and 7.25–7.75 (m, 15 H, 3 Ph). Hydrogenolysis of 11 and chromatography of the product (solvent B) gave 4, m.p. 109–111°, $[\alpha]_D + 30^\circ$ (c 0.6, water); lit. m.p. 115°, $[\alpha]_D + 39.1^\circ$; lit. m.p. 110–112.5°, $[\alpha]_D + 32^\circ$.

REFERENCES

- 1 R. TOMAN, S. KARÁCSONYI, AND R. PALOVCIK, Carbohydr, Res., 56 (1977) 191-194.
- 2 M. HEIDELBERGER, in J. B. G. KWAPINSKI (Ed.), Research in Immunochemistry and Immunobiology, Vol. 3, University Park Press, Baltimore, 1973, pp. 1-40.
- 3 V. Pozsgay, Carbohydr. Res., 69 (1979) 284-286.
- 4 M. A. NASHED, Carbohydr, Res., 60 (1978) 200-205.
- 5 F. G. YOUNG AND R. C. ELDERFIELD, J. Org. Chem., 7 (1942) 241-249.
- 6 P. Andrews, L. Hough, and J. K. N. Jones, J. Am. Chem. Soc., 77 (1955) 125-130.
- 7 E. L. HIRST AND S. DUNSTAN, J. Chem. Soc., (1953) 2332-2337.
- 8 J. R. TURVEY AND L. M. GRIFFITHS, Phytochemistry, 12 (1973) 2901-2907.