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Synthesis of 2-sulphur-substituted 1,4-anthraquinones. Application to the Synthesis of Anthracyclinone Precursors

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Abstract 2-sulphur-substituted 1.4-anthraquinones 8a,b and 10 have been prepared by annelation reactions of the anions of furanones 5a,b with naphthoquinone monoketals of type 6 and 9. Diels-Alder reaction of the 1.4-anthraquinones 11a,b with (E)-1.3-bis(trimethylsilyloxy)buta-1.3-diene (12) affords ABCD tetracyclic systems related to those existing in anthracyclinones.

The efficacy of anthracycline antibiotics such as daunomycin (1) and adriamycin (2) in the treatment of a variety of human cancers has stimulated a continued interest in the synthesis of this class of antitumor agents¹ However, these compounds display various side effects, the most serious being a cumulative dosedependent cardiotoxicity, so that considerable efforts have been devoted to develop new structurally modified anthracyclines with an improved antineoplastic activity and a low cardiotoxicity. It is also well known that the synthetic 4-demethoxy derivatives of daunomycin and adriamycin (3, 4) are more effective antitumor agents than the parent compounds²



As a part of a general development of new approaches to compounds related to anthracyclinones, we have been exploring the synthesis of BCD synthons that allows the construction of the tetracyclic system of type **1** and **3** in a subsequent Diels-Alder reaction. In preceding papers³ we have reported the first total synthesis of 5-iminodaunomycinone and its 4-demethoxy derivative. In our strategy, the key step is a regiocontrolled Diels-Alder reaction of an adequately 1,3-disubstituted buta-1,3-diene with fixed derivatives of the 1.4-anthraquinonoid tautomer of 1.4-dihydroxy-9,10-anthraquinone imines, as suitable BCD-ring synthons.

It was therefore of interest to study the synthesis of 1,4-anthraquinones of type 8 and 10, in which the introduction of a sulphur bearing group at 2-position, could be used to control the regiochemistry of the Diels-Alder cycloaddition.

On the other hand, our previous results⁴ show that annelation reactions of anions generated from furan-2(5*H*)-ones substituted at 5-position by sulphur bearing groups such as SEt, SPh, SO₂Et and SO₂Ph, with several napthoquinone monoketals led exclusively to the Michael adduct, without subsequent ring closure to the anthraquinones. However, we have now found that the required 1,4-anthraquinones can be obtained, by annelation reactions of anions of 4-halo-5-sulphur-furan-2(5*H*)-ones of type **5** with naphthoquinone monoketals.

This paper report a novel route to the synthesis of differently substituted 1,4-anthraquinones, their Diels-Alder reaction and investigations about the subsequent conversion of cycloadducts into compounds related to anthracyclinones.

Reaction of the anion generated from 4-bromo-5-ethylthiofuran-2(5H)-one $(5a)^5$ with lithium diisopropylamide, at -78°C in tetrahydrofuran, with naphthoquinone monoketal 6,⁶ after 7 days, at -7°C, afforded a 45:55 mixture of diastereoisomeric Michael adducts $7a^7$ and the 1.4-anthraquinone $8a^8$ in a 15% and 45% yield respectively (Scheme 1). It is noteworthy that simultaneously to the annuelation reaction takes place the substitution of a bromine atom by the ethylthio group.



6.4.1

Under similar reaction conditions, the anion of 4-bromo-5-phenylthiofuran-2(5H)-one (**5b**)⁹ reacted with monoketal **6** to afford, after 13 days, a 60:40 mixture of diastereoisomeric Michael adducts **7b** (10%) and the 1,4-anthraquinone **8b**¹⁰ (50%). However, the annelation reaction between the anion of **5a** with monoketal **9** led exclusively to the 1,4-anthraquinone **10**¹¹ in 40% yield (Scheme 1).

Methylation of the 1,4-anthraquinones 8a and 8b with methyl iodide, in the presence of silver oxide, gives the corresponding dimethyl ethers 11a and 11b in 95% yield.

Our first attempts to reaction of Diels-Alder between the 1.4-anthraquinones **8a,b** and an appropriate 1.3disubstituted diene such as (E)-1.3-bis(trimethylsilyloxy)buta-1.3-diene (**12**), in benzene under reflux, over a period of 15 days, failed to give the expected adducts

However, Diels-Alder reaction of the anthraquinone **11a** with diene **12** (benzene, 60° C, 12 days) occurs regio- and stereo-specifically to afford the adduct **13a**¹² (87%), which without purification was subject to a mild acidic treatment (HCl, THF, 0°C, 5 min) to give the tetracyclic ketone **14a** (95%) by selective hydrolysis of the silyl enol ether (Scheme 2)





Then, it was necessary to protect the keto group as the ethylene dioxyderivative 15a, which was prepared from 14a by treatment with ethylene glycol in presence of *p*-toluenesulphonic acid (benzene, 80°C, 12 h) in 97% yield. Subsequent treatment of the ketal 15a with tartaric acid in THF at room temperature yielded 16a in 95%. The following steps of the synthesis were directed to obtain the appropriate A-ring functionality. Thus, oxidative demethylation of 16a with silver(II) oxide, yielded the previously reported¹³ quinone 17 (85%). It is noteworthy, that under the reaction conditions takes places the removal of EtSOH.

For the synthesis of the quinone 17, an comparable sequence is followed from 4-bromo-5phenylthiofuran-2(5*H*)-one (5b). The sequence $5b\rightarrow 8b\rightarrow 11b\rightarrow 13b\rightarrow 14b\rightarrow 16b\rightarrow 17$ afforded the tetracyclic system 17 (25% overal yield), in only six laboratory operation from 5b (Scheme 2).

In summary, we have develop a novel procedure for the synthesis of 1.4-anthraquinones, starting from furanones and naphthoquinone monoketals. The synthetic methodology described herein, that uses the anthraquinones **11a,b** as suitables BCD ring synthon, may also be relevant for the construction of novel unnatural anthracyclinones.

Acknowledgment

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References and notes

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- 7. All new compounds gave satisfactory spectral data (IR, NMR and MS). Satisfactory combustion analyses were obtained for compounds **7a**, **8a,b**, **10**, **11a,b**, **14b** and **16b**.
- Compound 8a: ¹H NMR δ 14.85 (s, 1H), 8.50-8.48 (m, 1H), 8.35-8.31 (m, 1H), 7.78-7.73 (m, 2H), 6.63 (s, 1H), 4.03 (s, 3H), 2.87 (g, 2H), 1.45 (t, 3H).
- 9. Alguacil, R.; Fariña, F.; Martín, M.V. unpublished results.
- 10. Compound **8b**: ¹H NMR δ 14.71 (s, 1H), 8.43-8.40 (m, 1H), 8.38-8.30 (m, 1H), 7.72-7.66 (m, 2H), 7 52-7 43 (m, 5H), 6.08 (s, 1H), 4.0 (s, 3H).
- Compound 10: ¹H NMR δ 16 04 (s, 1H), 7.94 (d, 1H), 7.68 (t, 1H), 7.14 (d, 1H), 6.62 (s, 1H), 4.06 (s, 3H), 3.98 (s, 3H), 2.86 (q, 2H), 1.44 (t, 3H).
- 12. The structure was conclusively established on the basis of its ¹H NMR spectrum, in which the methine proton on the carbon bearing the OTMS (C-7) appears as a doublet (δ 4.62, J=6.0 Hz) coupled with the olefinic proton
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