

## TERT-BUTANOLYSIS OF LICHEN DEPSIDES\*

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**Key Word Index**—Depsides *tert*-butanolysis

**Abstract**—The scope and limit of the *tert*-butanolysis of 12 lichen depsides is described. Neither 2-*O*-methylated nor 2-*O*-acetylated compounds are cleaved by heating with *tert*-butanol.

### INTRODUCTION

The structural elucidation of a depside includes the cleavage of the ester bond as a main step. In 1979 and 1981 Bachelor *et al* [1] and Meyyappan *et al* [2] described the *tert*-butanolysis of atranorin and lecanoric acid, respectively, and Huneck [3] applied this method successfully to the structural elucidation of the new depsides pseudocypbellarin A and B.

To analyse the scope and limit of the *tert*-butanolysis of lichen depsides the reaction of the following compounds with *tert*-butanol was investigated: barbatolic acid (1), chloroatranorin (2), confluent acid (3), methyl evernate (4), 4-*O*-demethylbarbatolic acid (5), perlatolic acid (6), sphaerophorin (7), methyl tri-*O*-methyllecanorate (8), tri-*O*-methylpseudocypbellarin A (9), tri-*O*-acetyllecanoric acid (10), nephroarctin (11) and barbatolic acid (12).

### RESULTS AND DISCUSSION

The depsides 1–7 yielded the corresponding *tert*-butyl esters and phenolics, namely 1 *tert*-butyl rhizonate (13) and  $\beta$ -orcinolcarboxylic acid (14), 2 *tert*-butyl 5-chlorohaematommate (15) and methyl  $\beta$ -orcinolcarboxylate (16), 3 *tert*-butyl 4-*O*-methylolivetonate (17) and 2-*O*-methyl olivetolcarboxylic acid (18), 4 *tert*-butyl evernate (19) and methyl orsellinate (20), 5 *tert*-butyl  $\beta$ -orcinolcarboxylate (21) and 14, 6 *tert*-butyl 4-*O*-methylolivetonolcarboxylate (22) and olivetolcarboxylic acid (23), 7 19 and sphaerophorolcarboxylic acid (24). All new compounds were identified by their <sup>1</sup>H NMR spectra. Neither the 2-*O*-methylated compounds 8 and 9 nor tri-*O*-acetyllecanoric acid (10) were cleaved by prolonged heating with *tert*-butanol. Although nephroarctin (11) has a free hydroxyl group at C-2, only traces of cleavage products could be detected by TLC after heating with *tert*-butanol for 30 hr. Alectoronic and barbatolic acids are the only known depsides where the S-part is connected to the A-part via a benzylic group. Only 3% of 12 was cleaved after heating with *tert*-butanol for 20 hr.

The following conclusions can be drawn from these results: (1) A free 2-hydroxyl group seems to be essential

for the *tert*-butanolysis of depsides, (2) an aldehyde group at position 5 prevents the cleavage, and (3) the rate of the cleavage of benzylic depsides like barbatolic acid is much slower than that of normal depsides.

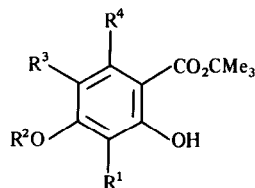
### EXPERIMENTAL

***Tert*-butanolysis** The depside (100–200 mg) was refluxed with *tert*-BuOH (50–100 ml) for 20–40 hr, after removal of solvent the residue was chromatographed on silica gel (with 5% H<sub>2</sub>O) using C<sub>6</sub>H<sub>6</sub> and C<sub>6</sub>H<sub>6</sub>–Et<sub>2</sub>O gradients.

***Tert*-butyl rhizonate (13)** Prismatic plates of mp 68–70° (from *n*-hexane) C<sub>14</sub>H<sub>20</sub>O<sub>4</sub> (252.3) IR  $\nu_{\text{max}}^{\text{KBr}}$  cm<sup>-1</sup> 830, 854, 968, 1000, 1030, 1060, 1130, 1156, 1180, 1228, 1248, 1300, 1370, 1402, 1446, 1460, 1498, 1572, 1616, 1638, 2950, 3300. <sup>1</sup>H NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  1.59 (s, 9H, CMe<sub>3</sub>), 2.04 (s, 3H, Me), 2.48 (s, 3H, Me), 3.80 (s, 3H, OMe), 6.19 (s, 1H, arom H), 11.90 (s, 1H, OH). MS  $m/z$  (rel int.) 252 [M]<sup>+</sup> (34), 197 [M – C(Me)<sub>2</sub> = CH]<sup>+</sup> (83), 178 (100), 150 (91), 135 (18), 122 (21), 107 (27).

***Tert*-butyl 5-chlorohaematommate (15)** Needles of mp 108–109° (from *n*-hexane) C<sub>12</sub>H<sub>15</sub>ClO<sub>5</sub> (274.7) IR  $\nu_{\text{max}}^{\text{KBr}}$  cm<sup>-1</sup> 716, 850, 1048, 1164, 1208, 1262, 1340, 1390, 1410, 1440, 1590, 1648, 2800, 3030, 3500. <sup>1</sup>H NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  1.60 (s, 9H, CMe<sub>3</sub>), 2.63 (s, 3H, Me), 10.25 (s, 1H, CHO), 12.5–13.1 (br s, 2H, 2 × OH).

***Tert*-butyl 4-*O*-methylolivetonate (17)** Oil C<sub>19</sub>H<sub>28</sub>O<sub>5</sub> (336.4) IR  $\nu_{\text{max}}^{\text{film}}$  cm<sup>-1</sup> 754, 820, 850, 960, 1046, 1114, 1154, 1198, 1266,



	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	R <sup>4</sup>
<b>13</b>	Me	Me	H	Me
<b>15</b>	CHO	H	Cl	Me
<b>17</b>	H	Me	H	CH <sub>2</sub> –CO– <i>n</i> –C <sub>5</sub> H <sub>11</sub>
<b>19</b>	H	Me	H	Me
<b>21</b>	Me	H	H	Me
<b>22</b>	H	Me	H	<i>n</i> –C <sub>5</sub> H <sub>11</sub>

\*Part 142 in the series "Lichen Substances". For part 141 see Connolly, J. D., Freer, A. A., Kalb, K. and Huneck, S. (1984) *Phytochemistry* 23, 857.

1304, 1334, 1374, 1428, 1464, 1572, 1610, 1640, 1708, 2990, 3400  $^1\text{H}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  0.80 (t, 3H,  $\text{CH}_2\text{-Me}$ ), 1.20 (m, 6H,  $(\text{CH}_2)_3\text{-Me}$ ), 1.50 (s, 9H,  $\text{CMe}_3$ ), 2.34 (t, 2H,  $\text{CO-CH}_2\text{-CH}_2$ ), 3.72 (s, 3H, OMe), 3.93 (s, 2H, benzyl  $\text{CH}_2$ ), 6.13, 6.32 (2  $\times$  d, 2H, 3-H, 5-H), 11.64 (s, 1H, OH)

**Tert-butyl everninate (19)** Crystals, mp  $28^\circ$  (from *n*-pentane)  $\text{C}_{13}\text{H}_{18}\text{O}_4$  (238.3) IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$  700, 758, 818, 850, 952, 992, 1040, 1062, 1118, 1160, 1200, 1262, 1300, 1330, 1370, 1420, 1450, 1576, 1610, 1640, 3000, 3450  $^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3$ )  $\delta$  2.80 (s, 9H,  $\text{CMe}_3$ ), 3.45 (s, 3H, Me), 4.45 (s, 3H, OMe), 6.39, 6.45 (2  $\times$  d, 2H, 3-H, 5-H), 10.80 (s, 1H, OH)

**Tert-butyl  $\beta$ -orcinolcarboxylate (21)** Prisms, mp  $128\text{--}130^\circ$  (from  $\text{Et}_2\text{O-n-hexane}$ )  $\text{C}_{14}\text{H}_{18}\text{O}_4$  (250.3) IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$  730, 842, 966, 1024, 1058, 1100, 1140, 1158, 1248, 1300, 1368, 1394, 1430, 1450, 1590, 1620, 1640, 3000, 3480  $^1\text{H}$  NMR (100 MHz,  $\text{CDCl}_3$ ) 1.55 (s, 9H,  $\text{CMe}_3$ ), 2.04 (s, 3H, 3-Me), 2.36 (s, 3H, 6-Me),

5.50 (br s, 1H, 4-OH), 12.18 (s, 1H, 2-OH)

**Tert-butyl 4-O-methylolivetolcarboxylate (22)** Oil  $\text{C}_{17}\text{H}_{26}\text{O}_4$  (294.4) IR  $\nu_{\text{max}}^{\text{film}}$   $\text{cm}^{-1}$  710, 754, 780, 820, 832, 850, 960, 1042, 1110, 1154, 1194, 1260, 1300, 1330, 1370, 1422, 1462, 1570, 1606, 1636, 2970, 3400  $^1\text{H}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  0.83 (t, 3H,  $\text{CH}_2\text{-Me}$ ), 1.28 (m,  $(\text{CH}_2)_3\text{-Me}$ ), 1.56 (s, 9H,  $\text{CMe}_3$ ), 2.80 (t, 2H, benzyl  $\text{CH}_2$ ), 6.17, 6.23 (2  $\times$  d, 2H, 3-H, 5-H), 11.84 (s, 1H, OH)

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# FLAVONOIDS FROM *ACHYROCLINE FLACCIDA*

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**Key Word Index**—*Achyrocline flaccida*, Compositae, Inuleae, aerial parts, prenylated flavonoids, flavonoids, caffeic acid derivatives

**Abstract**—Three new flavonoids 5-hydroxy-7-(3-methyl-2,3-epoxybutoxy)flavanone, 5-hydroxy-3,8-dimethoxy 7-(3-methyl-2,3-epoxybutoxy)flavone and 4'-hydroxy-5-methoxy-7-(3-methyl-2,3-epoxybutoxy)flavone were isolated and identified from the aerial parts of *Achyrocline flaccida*. Tamarixetin, gnaphalin, isognaphalin, 5,7,8-trihydroxy-3-methoxyflavone, chrysoeriol, galangin 3-methyl ether, naringenin 5-methyl ether, caffeic acid, chlorogenic acid and isochlorogenic acid were also isolated

## INTRODUCTION

In continuation of our chemosystematic search of the tribe Inuleae (Compositae), we have now investigated *Achyrocline flaccida* (Weinm.) DC, a shrub, widely distributed in the North of Argentina and the South of Brazil. In a previous paper we reported the identification of galangin, galangin 3-methyl ether, quercetin 3-methyl ether and two esters of calleryanin (3,4-dihydroxybenzyl alcohol 4-glucoside) with caffeic acid and protocatechuic acid from *Achyrocline sativoides* [4]. Investigation of the acetone extract of *A. flaccida* resulted in the isolation and determination of the structure of 7,4'-dihydroxy-5-methoxyflavanone and the corresponding 4,2',4'-trihydroxy-6-methoxychalcone [5].

The most characteristic features distinguishing members of the Inuleae from those of other Compositae tribes is the presence of flavonols lacking B ring hydroxylation, 6 and/or 8 hydroxyflavonols and their methyl ethers [6]. In the present report we describe the

occurrence of such typical flavonoids, together with the identification of three new prenylated flavonoids

## RESULTS AND DISCUSSION

The hexane extract of the aerial parts of *A. flaccida* was subjected to silica gel CC affording three new flavonoids. The first of these, compound 1 showed a brown colour in UV (365 nm) and a yellow-green colour with methanolic ferric chloride. Its UV spectrum exhibited maxima at 272 and 280 (sh) nm characteristic of a flavanone. The shifts induced in the UV spectra by aluminium chloride, sodium acetate and sodium methoxide led us to conclude that there is only one free hydroxyl attached to C-5. The  $^1\text{H}$  NMR spectrum (in  $\text{CDCl}_3$ ) showed a multiplet at  $\delta$  7.6 characteristic of an unsubstituted aromatic ring (B ring),  $\delta$  6.2 and 5.8 signals from protons H-6 and H-8,  $\delta$  5.3 corresponding to H-2 and  $\delta$  2.6 (multiplet) to H-3 *trans* and H-3 *cis*. The aliphatic chain showed the gem-dimethyl