Dioxopyrrolines. XLVII.¹⁾ Thermal Rearrangements of 7-Vinyl Derivatives of 1-Aryl-5-ethoxy-carbonyl-2-azabicyclo[3.2.0]heptane-3,4-diones and Their Imidates

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Thermolysis of the 7-vinyl derivative of 1-aryl-5-ethoxycarbonyl-2-azabicyclo[3.2.0]heptane-3,4-dione caused two different types of skeletal rearrangement depending on the stereochemistry of the 7-vinyl group. The exo-isomer (2) gave the hydroindole (4) by the 1,3-shift of a C_1 - C_7 bond to the vinyl group (path A reaction), while the endo-isomer (3) gave the 7-azabicyclo[4.2.1]nonane (5) by the Cope rearrangement (path B reaction). The product on thermolysis of the corresponding imidate also depends on the stereochemistry of the 7-vinyl group. The endo-isomer (13) gave the Cope product (12), while the exo-isomer (14) gave the dihydropyridone (16), which is produced by the 1,3-shift of the C_1 - C_7 bond to the C_3 position followed by cheletropic loss of CO from the intermediary 2-azanorbornene (path C reaction). The validity of path C was proved by isolation of the 2-azanorbornenes (21 and 22) formed by thermolysis of the 7-vinyl-4-acetoxy imidate (20).

Keywords dioxopyrroline; photocycloaddition; 2-azabicyclo[3.2.0]heptane; vinyl cyclobutane; thermolysis; 1,3-shift; Cope rearrangement; hydroindole; 7-azabicyclo[4.2.1]nonane; 2-azanorbornene

In previous papers^{2,3)} we reported that thermolysis of 7-substituted 5-ethoxycarbonyl-1-phenyl-2-azabicyclo[3.2. 0]heptane-3,4-diones under basic conditions caused 7-epimerization and ring expansion to a dihydroazatropolone, both of the reactions being initiated by C_1-C_5 bond fission. The imidates, 7-substituted 3-ethoxy-5-ethoxycarbonyl-1-phenyl-2-azabicyclo[3.2.0]hept-2-enes underwent 1,3-shift of the C_1-C_7 bond to give 2-azanorborn-2-enes (and dihydropyridones).¹⁾

In this paper we describe details of the thermal reaction of 7-vinyl derivatives of 1-aryl-5-ethoxycarbonyl-2-azabicyclo[3.2.0]heptane-3,4-dione and their imidates, involving three different types of skeletal rearrangement, a 1,3-shift of the C_1 - C_7 bond to the vinyl carbon (path A), a 3,3-sigmatropic rearrangement of the 7-vinyl group to the C_3 -position (path B), and a 1,3-shift of the C_1 - C_7 bond to the C_3 -position (path C).⁴⁾

Thermal Rearrangement of 7-Vinyl Derivatives of 1-Aryl-5-ethoxycarbonyl-2-azabicyclo[3.2.0]heptane-3,4-dione The exo and endo-7-vinyl derivatives of 1-aryl-5-ethoxycarbonyl-2-azabicyclo[3.2.0]heptane-3,4-dione (2 and 3) (vinyl cyclobutanes) were prepared by photocycloaddition of 5-aryl-4-ethoxycarbonyl-1*H*-pyrrole-2,3-dione (dioxopyrroline) (1) to butadiene, whose stereochemistry of the vinyl group was unambigously established.⁵⁾ Heating of 7-

exo-vinyl cyclobutane (2a) in toluene at 140 °C gave the hydroindole (4a) and the 7-azabicyclo[4.2.1]nonane (5a), each in 30% yield. On the other hand, thermolysis of the endo-vinyl isomer (3a), under similar conditions, gave 5a as a sole product in 72% yield. Pyrolysis of the exo-vinyl (2b) and endo-vinyl (3b) 1-methylenedioxyphenyl analogs gave similar results; 2b gave 4b and 5b in 21% and 23% yields, respectively, while 3b gave 5b as a sole product in 40% yield.

The hydroindole (4) was identical with the compound prepared by the Diels-Alder reaction of 1 and butadiene, 6) thus confirming the structure. On the other hand, the structure of 5 was elucidated from the spectral data as follow. The presence of a cis-1,2-disubstituted ethylene moiety was evidenced by the 1 H-nuclear magnetic resonance (1 H-NMR) spectra (δ 5.45 for 5a and 5.50 for 5b). The Ar-C=N- chromophore was indicated by the strong ultraviolet (UV) absorptions (250 nm for 5a, and 274 and 309 nm for 5b) and by the infrared (IR) C=N bands (1570 cm⁻¹ for 5a and 1580 cm⁻¹ for 5b). Furthermore, the IR spectra showed the presence of hydroxy and five-membered keto groups (3300, 1780 cm⁻¹ for 5a, and 3120, 1785 cm⁻¹ for 5b).

Formation of the hydroindole (4) was explained in terms of the 1,3-shift of the C_1 - C_7 bond to the terminal carbon of

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the vinyl group (path A). On the other hand, formation of the product (5) was rationalized in terms of a 3,3-sigmatropic rearrangement (Cope rearrangement) of the imino form (6) derived from the *endo*-isomer (3) (path B). Since it is sterically impossible for the *exo*-isomer (2) to form the Cope product (5), the formation of 5 from the *exo*-isomer requires the prior epimerization of the 7-vinyl group. The above results indicate that the thermolysis products of 7-vinyl cyclobutanes change depending on the stereochemistry of the 7-vinyl group.

Introduction of a methyl group at the vinyl terminus greatly affected these reactions. Thermolysis of the exo-(E)-propenyl derivative (7) in xylene under reflux gave the hydroindole (8) as a sole product, though in a low yield (29%). The other characterizable products including the C_5 -stereoisomer were not isolated from the reaction mixture, suggesting that this thermal 1,3-shift proceeds in a highly stereoselective manner, although the configuration of the C_5 -methyl group in this product was unclear. The endo-(E)-propenyl isomer (9), 5) on similar pyrolysis, gradually decomposed to give no characterizable product. On the other hand, the exo-(Z)-propenyl isomer (10)⁵⁾ showed

Chart 2

different thermal behavior. When heated in xylene for 1 h, it underwent 7-epimerization to give the *endo*-isomer (11)⁵⁾ in 50% yield. The latter compound was thermally stable and remained unchanged on heating for 4 h (recovery 80%).

Thermal Rearrangements of 7-Vinyl Derivatives of 3-Ethoxy-5-ethoxycarbonyl-2-azabicyclo[3.2.0]hept-2-enes If the argument that the path B product was formed via the Cope rearrangement of the imino form (6) is correct, fixation of the double bond at N-C₃ (for example, formation of the imidate) will facilitate this rearrangement. In fact, the endo-isomer (3a), on treatment with triethyloxonium fluoroborate (Meerwein reagent) at room temperature, afforded the Cope product (12a) in 80% yield. Similar imidations of the endo-(E) and (Z)-propenyl isomers (9 and 11) gave stereospecifically the Cope products (12b and 12c, respectively) in good yields. The spectral characteristics of these products were very similar to those of 5, except that the products possess the OEt group instead of OH and an additional Me group, thus establishing that the compounds are the path B products formed through the intermediary imidates (13). The stereochemistry of the 5-methyl group in 12b and 12c was assigned on the basis of the mechanistic consideration that the configuration should be retained, since this rearrangement proceeds under the control of lowest unoccupied molecular orbital (LUMO)-highest occupied molecular orbital (HOMO) interaction with a boatlike conformation as the preferred transition state.

On the other hand, the exo-isomers (2a, 7, and 10), on treatment with the Meerwein reagent, gave the correspond-

Table I. Thermolysis of 7-exo-Vinyl and 7-exo-Propenyl Derivatives of 3-Ethoxy-5-ethoxycarbonyl-1-phenyl-2-azabicyclo[3.2.0]hept-2-en-4-ones (14)

Substrates 14a	Conditions Temp. (°C) Time (min)		Product (Yield %)		
			16 ^{a)}	12 ^{b)}	15 ^{c)}
	120	30	60	4	10
14b	120	60	51	7	7
14c	120	60	64	8	_

a) Product of path C. b) Product of path B. c) Product of path A.

ing imidates (14). The exo-vinyl (14a) and propenyl imidates (14b and 14c), on heating in toluene under reflux followed by silica gel chromatography of the pyrolysates, afforded three products, the dihydropyridone (16), the hydroindole (15), and the Cope product (12). The results (Table I) showed that the 1,3-shift leading to dihydropyridone (path C) is a major and the 1,3-shift forming hydroindole (path A) is a minor path in the reaction of imidates. In the case of 14c the hydroindole (15c) was not isolated. The structures (16a—c) were elucidated by means of elementary analyses and by comparison of the spectral data with those of the known dihydropyridones. The hydroindoles (15a and 15b) were identical with the compounds obtained by imidation of 4a and 8, respectively.

The path C reaction, as discussed in the preceding paper, 1) involves the 1,3-shift of the C_1 – C_7 bond to C_3 followed by cheletropic loss of CO from the intermediary 2-azanorborn-2-en-7-one (17). Apparently fixation of the double bond at N– C_3 facilitates this rearrangement from the *exo*-isomer. Formation of the Cope product (path B) from this isomer apparently requires the prior epimerization of the 7-vinyl or the propenyl group as described above.

In order to clarify the reaction pathway of path C we carried out thermolyses of 4-acetoxy imidates, where the 2-azanorborn-2-enes are expected to be isolated as discussed in the preceding paper. The substrates (20a—c) were prepared from the lactams (2a, 7, and 10) by reduction with tetra-n-butyl ammonium borohydride, acetylation, and imidation (Chart 4). The stereochemistry of the 4-acetoxy group was elucidated from the observation of γ -effect in the 13 C-NMR spectra as described in the previous paper. 2

Thermolysis of the imidates (20a—c) caused the path C reaction to give exclusively two stereoisomeric 2-azanor-bornenes (21a—c and 22a—c). The structures of 2-azanorborn-2-enes including the stereochemistry of the 6-vinyl or the 6-propenyl groups were deduced by comparisons of the spectral data with those of the 6-phenyl derivatives.¹⁾ In particular, the presence of a strong UV absorption band around 240 nm due to the -N = C - Ph chromophore and the signal patterns of the C_5 and C_6 -ring protons in the 1H -NMR spectra supported this asignment. The 2-azanorborn-2-enes (21 and 22) are the products of C_7 -inversion (si-product) and of C_7 -retention (sr-product), respectively. The product ratios (si/sr=1.2 for 20a, 1.7 for

20b, and 1.0 for **20c**) are very similar to the case of the 7-exo-phenyl analog (si/sr=1.2), indicating that this 1,3-shift proceeds with slight preference of the symmetry-allowed si-process, as discussed in the preceding paper.¹⁾

Finally, we will briefly discuss the mechanisms of path A rearrangement and 7-epimerization. We consider that these two reactions would proceed via a biradical, but probably different species. The 1,3-shift of path A reaction should occur from the biradical (23) formed by fission of the C_1 – C_7 bond, while the 7-epimerization would proceed via the biradical (24) formed by C_1 – C_5 bond fission (Chart 5) as discussed in the previous paper.²⁾

It is noteworthy that the above results suggest that the dioxopyrroline-butadiene photoannulation followed by thermolysis of the resulting vinyl cyclobutanes provides a potential new route to hydroindole, whose skeleton appears in various natural alkaloids.

Experimental

Unless otherwise stated, the following procedures were adopted. Melting points were taken on a Yanagimoto micro hot-stage melting point apparatus, and are uncorrected. IR spectra were taken in Nujol mulls with a Hitachi 260-10 spectrophotometer and are given in cm⁻¹. UV spectra were recorded with a Hitachi 200-10 spectrophotometer and are given in λ_{max} nm (ϵ). ¹H-NMR (100 MHz) and ¹³C-NMR (25.0 MHz) spectra were taken in a CDCl₃ solution with tetramethylsilane (TMS) as an internal standard on a JEOL FX-100 spectrometer. Chemical shifts are reported in δ , and signals are described as s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet), or br (broad). High resolution mass spectra (HRMS) were recorded on a JEOL JMS-D300 mass spectrometer. Chromatography was carried out on silica gel (Wako-gel C-200). Thin layer chromatography (TLC) was performed on Merck precoated Silica gel 60 F₂₅₄ plates. Medium pressure liquid chromatography (MPLC) was performed on Kusano CIG prepacked silica gel columns. The photolysis was done by internal irradiation using a 300 W high pressure mercury lamp (Eikosha

Halos PIH 300) with a Pyrex filter. The preparation of the vinyl cyclobutanes 2a, 3a, 7, 9, 10, and 11 has been reported.⁵⁾

Photocycloaddition of 1b to Butadiene A solution of 1b (1.0 g) and butadiene (10 ml) in dimethoxyethane (150 ml) was irradiated at 0 °C for 1 h. After evaporation of the solvent, the residue was chromatographed. Elution with CH_2Cl_2 gave a mixture of 2b and 3b (1:1) (580 mg, 49%) as colorless crystals. This was separated by fractional crystallizations. The Et_2O insoluble fractions gave dl-(1R*,4S*,7S*)-5-ethoxycarbonyl-1-(3',4'-methylenedioxyphenyl)-7-vinyl-2-azabicyclo[3.2.0]-heptane-3,4-dione (2b) as colorless prisms from MeOH, mp 175—177°C. IR: 1775, 1770, 1740, 1700. ¹H-NMR: 0.82 (3H, t, J=7 Hz, $COOCH_2CH_3$), 2.44 (1H, dd, J=9, 12 Hz, C_6 -H), 3.09 (1H, t, J=12 Hz, C_6 -H), 3.3 (1H, m, C_7 -H), 3.98 (2H, m, $COOCH_2CH_3$), 5.13 (1H, d, J=18 Hz, olefinic H), 5.18 (1H, d, J=10 Hz, olefinic H), 5.60—5.83 (1H, m, olefinic H), 5.97 (1H, s, OCH_2O), 6.83 (3H, brs, ArH). Anal. Calcd for $C_{18}H_{17}NO_6$: C, 62.97; H, 4.99; N, 4.08. Found: C, 62.98; H, 5.15; N, 3.73.

The Et₂O soluble fractions gave dl- $(1R^*,5S^*,7R^*)$ -5-ethoxycarbonyl-(3',4'-methylenedioxyphenyl)-7-vinyl-2-azabicyclo[3.2.0]heptane-3,4-dione (3b) as colorless prisms from Et₂O-hexane, mp 172—177 °C. IR: 1775, 1745, 1730. ¹H-NMR: 0.89 (3H, t, J=7 Hz, COOCH₂CH₃), 2.17 (1H, dd, J=9, 13 Hz, C₆-H), 3.16 (1H, dd, J=9, 13 Hz, C₆-H), 3.90 (2H, q, J=7 Hz, COOCH₂CH₃), 4.08 (1H, m, C₇-H), 5.19 (1H, d, J=17 Hz, olefinic H), 5.32 (1H, d, J=9 Hz, olefinic H), 5.90 (1H, m, olefinic H), 5.98 (2H, s, OCH₂O), 6.82 (3H, m, ArH). Anal. Calcd for C₁₈H₁₇NO₆: C, 62.97; H, 4.99; N, 4.08. Found: C, 63.06; H, 5.02; N, 3.79.

Pyrolysis of 7-Vinyl (2 and 3) and 7-Propenyl Cyclobutanes (7 and 10) (General Procedure) A solution of 2, 3, 7 or 10 (100—200 mg) in toluene (10 ml) was heated in a sealed tube. After removal of the solvent *in vacuo*, the residue was chromatographed and eluted with benzene—CH₂Cl₂ (1:1) to give the products. Each product was crystallized from CH₂Cl₂-Et₂O, unless otherwise stated.

- 1) Pyrolysis of **2a** (200 mg) at 140 °C for 4h gave **4a** (60 mg, 30%) and **5a** (60 mg, 30%). dl-(1R*,6S*)-1-Ethoxycarbonyl-6-phenyl-7-azabicyclo[4.3.0]non-3-ene-8,9-dione (**4a**), colorless prisms, mp 228—230 °C. Compound **4a** was identical with an authentic sample. dl-(1R*,6R*)-1-Ethoxycarbonyl-6-hydroxy-8-phenyl-7-azabicyclo-[4.2.1]nona-3,7-dien-9-one (**5a**). Colorless prisms, mp 150—155 °C. IR: 3100, 1780, 1735, 1600. UV: 250 (14800), 330 (3200). dl-1-NMR: 0.98 (3H, t, J=7 Hz, COOCH₂CH₃), 2.7—3.0 (4H, m, C₂-H and C₅-H), 4.10 (2H, q, J=7 Hz, COOCH₂CH₃), 5.15 (1H, br s, OH), 5.45 (2H, m, olefinic H), 7.4 (3H, m, ArH), 7.7 (2H, m, ArH). HRMS m/z: dl-1 Calcd for dl-1 Calcd for dl-1 Royal 187.
 - 2) Pyrolysis of 3a (180 mg) at 160 °C for 5h gave 5a (130 mg, 72%).
- 3) Pyrolysis of **2b** (100 mg) at 140 °C for 4h gave **4b** (25 mg, 25%) and **5b** (23 mg, 23%). dl-(1R*,6S*)-1-Ethoxycarbonyl-6-(3',4'-methylene-dioxyphenyl)-7-azabicyclo[4.3.0]non-3-ene-8,9-dione (**4b**), colorless prisms, mp 222—224 °C. Compound **4b** was identical with an authentic sample. dl-(1R*,6R*)-1-Ethoxycarbonyl-6-hydroxy-8-(3',4'-methylenedioxyphenyl)-7-azabicyclo[4.2.1]nona-3,7-dien-9-one (**5b**). Colorless prisms, mp 178—183 °C. IR: 3120, 1785, 1740, 1580. UV: 228 (12900), 274 (7600), 309 (8500). dl-1H-NMR: 1.07 (3H, t, dl-7 Hz, COOCH₂CH₃), 2.50—2.95 (4H, m, dl-2-H and dl-3-H), 4.15 (2H, q; dl-7 Hz, COOCH₂CH₃), 5.49 (2H, m, olefinic H), 6.02 (2H, s, OCH₂O), 6.80 (1H, d, dl-8 Hz, ArH), 7.17 (1H, dd, dl-2, 8 Hz, ArH), 7.49 (1H, d, dl-2 Hz, ArH). Anal. Calcd for dl-18 dl-17 C, 6.2.97; H, 4.99; N, 4.08. Found: 62.98; H, 5.17; N, 3.82.
- 4) Pyrolysis of **3b** (200 mg) at 140 °C for 1.5 h gave **5b** (120 mg, 60%). 5) Pyrolysis of 7 (105 mg) at 140 °C for 5 h gave dl-(1R*,6S*)-1-ethoxycarbonyl-5-methyl-6-phenyl-7-azabicyclo[4.3.0]non-3-ene-8,9-dione (8) (30 mg, 29%) as colorless prisms, mp 284—286 °C. IR: 3170, 1780, 1722, 1700. ¹H-NMR: 0.71 (3H, t, J=7Hz, COOCH₂CH₃), 0.95 (3H, d, J=7Hz, CH₃), 2.96 (3H, m, C₂-H and C₃-H), 3.50 (2H, m, COOCH₂CH₃), 5.50—6.20 (2H, m, olefinic H), 7.40 (5H, br s, ArH). *Anal.* Calcd for C₁₈H₁₉NO₄: C, 68.99; H, 6.11; N, 4.47. Found: C, 68.60; H, 6.18: N, 4.20.
- 6) Pyrolysis of 10 (208 mg) at 140 °C for 1 h gave 11 (104 mg, 50%) as colorless prisms from acetone-hexane, mp 152-154 °C.⁵⁾

Treatment of the endo-Vinyl (3a) and Propenyl Cyclobutanes (9 and 11) with Triethyloxonium Fluoroborate (General Procedure) A solution of 3a, 9, or 11 (100 mg) and an excess of Et₃OBF₄ in CH₂Cl₂ (10 ml) was stirred at room temperature overnight. The raction mixture was washed with 5% NaHCO₃ and water, dried over MgSO₄, and concentrated. Chromatography of the residue and elution with benzene gave the Cope rearrangement product (12), which was crystallized from Et₂O-hexane.

 $d\bar{l}$ -(1 R^* ,6 R^*)-6-Ethoxyl-1-ethoxycarbonyl-8-phenyl-7-azabicyclo[4.2.1]-nona-3,7-dien-9-one (12a): 76.5 mg, 70%. Colorless prisms, mp 95—97

°C. IR: 1780, 1730, 1600. UV: 250 (12200). ¹H-NMR: 1.05 (3H, t, J=7 Hz, COOCH₂CH₃), 1.28 (3H, t, J=7 Hz, OCH₂CH₃), 2.4—2.9 (4H, m, C₂-H and C₅-H), 3.56 (2H, m, COOCH₂CH₃), 4.18 (2H, q, J=7 Hz, OCH₂CH₃), 5.31 (2H, m, olefinic H), 7.5 (3H, m, ArH), 7.8 (2H, m, ArH). HRMS m/z: M⁺ Calcd for C₁₉H₂₁NO₄ 327.1471. Found: 327.1503.

dl-(1R*,5S*,6R*)-6-Ethoxy-1-ethoxycarbonyl-5-methyl-8-phenyl-7-azabicyclo[4.2.1]nona-3,7-dien-9-one (12b): 82 mg, 75%. Colorless prisms, mp 86 °C. IR: 1780, 1740, 1735, 1600. UV: 250 (14700). ¹H-NMR: 1.07 (3H, t, J=7 Hz, COOCH₂CH₃), 1.27 (3H, t, J=7 Hz, OCH₂CH₃), 1.37 (3H, d, J=8 Hz, CH₃), 2.6—3.0 (3H, m, C₂-H and C₅-H), 3.53 (2H, quint, J=7 Hz, COOCH₂CH₃), 4.18 (2H, q, J=7 Hz, OCH₂CH₃), 5.31 (2H, m, olefinic H), 7.5 (3H, m, ArH), 7.8 (2H, m, ArH). HRMS m/z: M⁺ Calcd for C₂₀H₂₃NO₄ 341.1626. Found: 341.1611.

dl-(1R*,5R*,6R*)-6-Ethoxy-1-ethoxycarbonyl-5-methyl-8-phenyl-7-azabicyclo[4.2.1]nona-3,7-dien-9-one (12c): 78 mg, 72%. Colorless prisms, mp 90—91 °C. IR: 1780, 1745, 1600. UV: 250 (15000). ¹H-NMR: 1.05 (3H, t, J=7 Hz, COOCH₂CH₃), 1.13 (3H, d, J=7 Hz, CH₃), 1.28 (3H, t, J=7 Hz, OCH₂CH₃), 2.9 (3H, m, C₂-H and C₅-H), 3.60 (2H, quint, J=7 Hz, COOCH₂CH₃), 4.20 (2H, q, J=7 Hz, OCH₂CH₃), 5.45 (2H, m, olefinic H), 7.5 (3H, m, ArH), 7.8 (2H, m, ArH). HRMS m/z: M⁺ Calcd for C₂₀H₂₃NO₄ 341.1625. Found: 341.1625.

Imidation of the exe-Vinyl (2a) and Propenyl Cyclobutanes (7 and 10) with Triethyloxonium Fluoroborate (General Procedure) A solution of 2a, 7, or 10 (200 mg) and an excess of Et₃OBF₄ in CH₂Cl₂ (30 ml) was allowed to stand at room temperature overnight. The reaction mixture was diluted with CH₂Cl₂, washed with 5% NaHCO₃ and water, dried over MgSO₄, and concentrated. The product was crystallized from Et₂O-hexane to give the imidate (14).

dl-(1R*,5S*,7S*)-3-Ethoxy-5-ethoxycarbonyl-1-phenyl-7-vinyl-2-azabicyclo[3.2.0]hept-2-en-4-one (14a): 180 mg, 82%. Colorless prisms, mp 96—97 °C. IR: 1760, 1730, 1630. ¹H-NMR: 0.67 (3H, t, J=7 Hz, COOCH₂CH₃), 1.52 (3H, t, J=7 Hz, OCH₂CH₃), 2.15—2.45 (1H, m, C₆·H), 2.7—3.0 (2H, m, C₆ and C₇-H), 3.79 (2H, q, J=7 Hz, COOCH₂CH₃), 4.59 (2H, q, J=7 Hz, OCH₂CH₃), 4.8—5.2 (2H, m, olefinic H), 5.45—5.85 (1H, m, olefinic H), 7.3 (5H, m, ArH). HRMS m/z: M⁺ Calcd for C₁₉H₂₁NO₄ 327.1470. Found: 327.1490.

dl-(1R*,5S*,7S*)-3-Ethoxy-5-ethoxycarbonyl-1-phenyl-7-(E)-propenyl-2-azabicyclo[3.2.0]hept-2-en-4-one (14b): 157 mg, 72%. Colorless prisms, mp 104—106 °C. IR: 1760, 1730, 1630. ¹H-NMR: 0.67 (3H, t, J = 7 Hz, COOCH₂CH₃), 1.48 (3H, t, J = 7 Hz, OCH₂CH₃), 1.53 (3H, d, J = 5 Hz, CH₃), 2.1 (1H, m, C₆-H), 2.6—3.2 (2H, m, C₆ and C₇-H), 3.75 (2H, q, J = 7 Hz, COOCH₂CH₃), 4.53 (2H, q, J = 7 Hz, OCH₂CH₃), 5.3 (2H, m, olefinic H), 7.2 (5H, m, ArH). Anal. Calcd for C₂₀H₂₃NO₄: C, 70.36; H, 6.79; N, 4.10. Found: C, 70.33; H, 6.87; N, 3.99.

dl-(1 R*,5S*,7S*)-3-Ethoxy-5-ethoxycarbonyl-1-phenyl-7-(Z)-propenyl-2-azabicyclo[3.2.0]hept-2-en-4-one (14c): 211 mg, 97%. Colorless prisms, mp 108—111 °C. IR: 1750, 1720, 1630. ¹H-NMR: 0.67 (3H, t, J= 7 Hz, COOCH₂CH₃), 1.43 (3H, d, J= 3 Hz, CH₃), 1.50 (3H, t, J= 7 Hz, OCH₂CH₃), 2.25 (1H, m, C₆-H), 2.7—3.28 (2H, m, C₆ and C₇-H), 3.83 (2H, q, J= 7 Hz, COOCH₂CH₃), 3.97 (2H, q, J= 7 Hz, OCH₂CH₃), 5.4 (2H, m, olefinic H), 7.3 (5H, m, ArH). Anal. Calcd for C₂₀H₂₃NO₄: C, 70.36, H, 6.79; N, 4.10. Found: C, 70.30; H, 6.83; N, 4.01.

Pyrolysis of the 4-Keto Imidates (14) (General Procedure) A solution of 15 (100 mg) in toluene (5 ml) was heated in a sealed tube. Chromatography of the products and elution with benzene gave a mixture of 12 and 14 which was separated by preparative TLC (developed with CH_2Cl_2). Further elution with CH_2Cl_2 gave 16. The products (12a—c and 16a) were identical with the compounds obtaind from the *endo*-derivatives and the reported dihydropyridone, ⁵⁾ respectively.

- 1) Pyrolysis of 14a at 120 °C for 30 min gave 12a (4 mg, 4%), 15a (10 mg, 10%), and 16a (50 mg, 60%). dl-(1R*,6S*)-8-Ethoxy-1-ethoxy-carbonyl-6-phenyl-7-azabicyclo[4.3.0]nona-3,7-dien-9-one (15a): Colorless prisms, mp 94—96 °C. IR: 1750, 1730, 1630. ¹H-NMR: 0.57 (3H, t, J=7 Hz, COOCH₂CH₃), 1.42 (3H, t, J=7 Hz, OCH₂CH₃), 2.60—2.78 (4H, m, C₂-H and C₅-H), 3.42 (2H, quint, J=7 Hz, COOCH₂CH₃), 4.45 (2H, q, J=7 Hz, OCH₂CH₃), 5.63—5.84 (2H, m, olefinic H), 7.12—7.50 (5H, m, ArH). Anal. Calcd for C₁₉H₂₁NO₄: C, 69.70; H, 6.47; N, 4.28. Found: C, 69.64; H, 6.47; N, 4.34.
- 2) Pyrolysis of 14b at 120 °C for 1 h gave 12b (7 mg, 7%), 15b (7 mg, 7%), and 16b (50 mg, 51%). dl-(1R*,6S*)-8-Ethoxy-1-ethoxycarbonyl-5-methyl-6-phenyl-7-azabicyclo[4.3.0]nona-3,7-dien-9-one (15b): Colorless prisms, mp 112—114 °C. IR: 1755, 1725, 1640. ¹H-NMR: 0.60 (3H, t, J=7 Hz, COOCH₂CH₃), 0.98 (3H, d, J=7 Hz, CH₃), 1.47 (3H, t, J=7 Hz. OCH₂CH₃), 2.9 (3H, m, C₂-H and C₅-H), 3.39 (2H, m, COOCH₂CH₃), 4.55 (2H, q, J=7 Hz, OCH₂CH₃), 5.5 (1H, m,

olefinic H), 5.8 (1H, m, olefinic H), 7.3 (5H, m, ArH). HRMS m/z: M⁺ Calcd for $C_{20}H_{23}NO_4$ 341.1626. Found: 341.1615. 5-Ethoxycarbonyl-6-phenyl-3-(E)-propenyl-3,4-dihydropyridin-2(1H)-one (16b): Colorless needles, mp. 111—112 °C. IR: 3350, 3300, 1710, 1660, 1640, 1600. UV: 227 (9200), 285 (9750). ¹H-NMR: 0.93 (3H, t, J=7Hz, COOCH₂CH₃), 1.72 (3H, d, J=5Hz, CH₃), 2.67—2.90 (2H, m, C₄-H), 3.2 (1H, m, C₃-H), 3.93 (2H, q, J=7Hz, COOCH₂CH₃), 5.67 (2H, m, olefinic H), 7.35 (5H, m, ArH). HRMS m/z: M⁺ Calcd for $C_{17}H_{19}NO_3$ 285.1365. Found: 275.1362.

3) Pyrolysis of 14c at 120 °C for 1 h gave 12c (8 mg, 8%) and 16c (53 mg, 64%). 5-Ethoxycarbonyl-6-phenyl-3-(Z)-propenyl-3,4-dihydropyridin-2(1H)-one (16c): Colorless needles, mp 121—122 °C. IR: 3200, 3100, 1695, 1680, 1635, 1600. UV: 225 (9800), 284 (9800). 1 H-NMR: 0.90 (3H, t, J=7 Hz, COOCH₂CH₃), 1.68 (3H, d, J=7 Hz, CH₃), 2.53 (1H, dd, J=12, 16 Hz, C₄-H), 2.93 (1H, dd, J=7, 16 Hz, C₄-H), 3.30—3.73 (1H, m, C₃-H), 3.90 (2H, q, J=7 Hz, COOCH₂CH₃), 5.23—6.00 (2H, m, olefinic H), 7.2 (5H, m, ArH). HRMS m/z: M+ Calcd for C₁₇H₁₉NO₃ 285.1365. Found: 285.1410.

Imidation of the Hydroindole (4a and 8) with Triethyloxonium Fluoroborate A solution of 4a and 8 (20 mg) in ${\rm CH_2Cl_2}$ (5 ml) was treated with an excess of ${\rm Et_3OBF_4}$ at room temperature for 6 h, and worked up as usual to give the products 15a (18 mg, 82%) and 15b (17 mg, 80%), which were identical with the hydroindoles (15a and 15b) described above, respectively.

Preparation of 4-Acetoxy Lactams (19) (General Procedure) A solution of **2a**, **7**, or **10** (100 mg) in CH_2Cl_2 (30 ml) was treated with n- $(C_4H_9)_4NBH_4$ (0.5 mol eq) at 0 °C for 30 min. The mixture was diluted with CH_2Cl_2 , washed with water, dried over $MgSO_4$, and evaporated. The residue was acetylated with Ac_2O (1 ml) and pyridine (2 ml) at room temperature overnight. Chromatography of the product, elution with benzene, and recrystallization from CH_2Cl_2 - Et_2O gave the acetate (19).

dl-(1R*,4S*,5S*,7S*)-4-Acetoxy-5-ethoxycarbonyl-1-phenyl-7-vinyl-2-azabicyclo[3.2.0]heptan-3-one (19a): 65 mg, 57%. Colorless prisms, mp 179—182 °C. IR: 1755, 1720. UV: 258sh (1700). ¹H-NMR: 0.83 (3H, t, J= 7 Hz, COOCH₂CH₃), 2.13 (3H, s, OAc), 2.5—2.9 (2H, m, C₆-H), 3.2—3.5 (1H, m, C₇-H), 3.82 (2H, q, J=7 Hz, COOCH₂CH₃), 5.05 (1H, d, J= 10 Hz, olefinic H), 5.07 (1H, d, J=17 Hz, olefinic H), 5.73 (1H, ddd, J=6.5, 10, 17 Hz, olefinic H), 5.80 (1H, s, C₄-H), 7.1—7.3 (5H, m, ArH), 7.81 (1H, br s, NH). ¹³C-NMR: 13.6 (q, COOCH₂CH₃), 20.4 (q, OCOCH₃), 31.1 (t, C₆), 50.5 (d, C₇), 56.1 (s, C₅), 60.6 (t, COOCH₂CH₃), 70.3 (s, C₁), 76.4 (d, C₄), 116.5 (t, CH=CH₂, 125.9 (d, 2C, CH=CH₂), 127.6 (d, Ph), 128.0 (d, 2C, Ph), 136.3 (d, Ph), 136.3 (s, Ph), 168.4 (s, C₃), 169.4 (s, COOCH₂CH₃), 172.6 (s, OCOCH₃). Anal. Calcd for C₁₉H₂₁NO₅: C, 66.46; H, 6.16; N, 4.08. HRMS m/z: 343.1419. Found: C, 66.34; H, 6.16; N, 3.84. HRMS m/z: 343.1419.

dl-(1 R*,4S*,5S*,7S*)-4-Acetoxy-5-ethoxycarbonyl-1-phenyl-7-(E)-propenyl-2-azabicyclo[3.2.0]heptan-3-one (19b): 60 mg, 52%. Colorless prisms, mp 193—194 °C. IR: 1750, 1725, 1720sh, 1710. UV: 260sh (300). ¹H-NMR: 0.83 (3H, t, J=7 Hz, COOCH₂CH₃), 1.58 (3H, d, J=5 Hz, CH₃), 2.12 (3H, s, OAc), 2.66 (1H, d, J=10 Hz, C₆-H), 2.70 (1H, d, J=10 Hz, C₆-H), 3.16—3.3 (1H, m, C₇-H), 3.81 (2H, q, J=7 Hz, COOCH₂CH₃), 5.2—5.6 (2H, m, olefnic H), 5.79 (1H, s, C₄-H), 7.24—7.33 (5H, m, Ar-H), 7.64 (1H, s, NH). ¹³C-NMR: 13.6 (q, COOCH₂CH₃), 17.8 (q, CH₃), 20.5 (q, OCOCH₃), 31.6 (t, C₆), 49.8 (d, C₇), 56.0 (s, C₅), 60.6 (t, COOCH₂CH₃), 70.7 (s, C₁), 76.6 (d, C₄), 125.9 (d, 2C, Ph), 127.5 (d, CH=CH), 128.0 (d, 2C, Ph), 128.9 (d, Ph), 136.4 (s, Ph), 168.6 (s, C₃), 169.6 (s, COOCH₂CH₃), 172.8 (s, OCOCH₃). HRMS m/z: M * Calcd for C₂₀H₂₃NO₅ 357.1575. Found: 357.1560.

dl-(1 R*,4S*,5S*7S*)-4-Acetoxy-5-ethoxycarbonyl-1-phenyl-7-(Z)-propenyl-2-azabicyclo[3.2.0]heptan-3-one (19c): 74 mg, 65%. Colorless prisms, mp. 166—170 °C. IR: 1760, 1730, 1710. UV: 260sh (200). ¹H-NMR: 0.84 (3H, t, J=7 Hz, COOCH₂CH₃), 1.53 (3H, d, J=5 Hz, CH₃), 2.13 (3H, s, OAc), 2.69 (1H, d, J=10 Hz, C₆-H), 2.70 (1H, d, J=10 Hz, C₆-H), 3.4—3.6 (1H, m, C₇-H), 3.84 (2H, q, J=7 Hz, COOCH₂CH₃), 5.35—5.45 (2H, m, olefinic H), 5.84 (1H, s, C₄-H), 7.2—7.4 (5H, m, Ar-H), 7.51 (1H, s, NH). ¹³C-NMR: 13.6 (q, COOCH₂CH₃), 13.8 (q, CH₃), 20.5 (q, OCOCH₃), 33.0 (t, C₆), 44.8 (d, C₇), 56.3 (s, C₅), 60.7 (t, COOCH₂CH₃), 71.4 (s, C₁), 76.6 (d, C₄), 125.8 (d, 2C, Ph), 126.9 (d, CH=CH), 127.5 (d, CH=CH), 128.0 (d, 2C, Ph), 128.8 (d, Ph), 136.4 (s, Ph), 168.6 (s, C₃), 169.6 (s, COOCH₂CH₃), 172.8 (s, OCOCH₃). Anal. Calcd for C₂₀H₂₃NO₅: C, 67.21; H, 6.49; N, 3.92. HRMS m/z: 357.1575. Found: C, 67.00; H, 6.49; N, 3.72. HRMS m/z: 357.1560.

Preparation of 4-Acetoxy Imidates (20) (General Procedure) A solution of 4-acetoxy lactam (19) (100 mg) in CH₂Cl₂ (5 ml) was treated with an excess of Et₃OBF₄ at room temperature for 6 h. The mixture was diluted

with CH₂Cl₂ (30 ml), washed with 5% NaHCO₃ and water, dried over MgSO₄, and concentrated. The residue in benzene was passed through a short column of SiO₂ to give the 4-acetoxy imidate (20).

dl-(1R*,4S*,5S*,7S*)-4-Acetoxy-3-ethoxy-5-ethoxycarbonyl-1-phenyl-

7-vinyl-2-azabicyclo[3.2.0]hept-2-ene (20a): 94 mg, 87%. Colorless needles, mp 82—86 °C. IR: 1755, 1720, 1645, 1630. 1 H-NMR: 0.84 (3H, t, J=7 Hz, $COOCH_2CH_3$), 1.39 (3H, t, J=7 Hz, OCH_2CH_3), 2.07 (3H, s, OAc), 2.2– 2.5 (1H, m, C₆-H), 2.6—2.9 (1H, m, C₆-H), 2.9—3.2 (1H, m, C₇-H), 3.81 (2H, q, J=7 Hz, COOCH₂CH₃), 4.41 (2H, q, J=7 Hz, OCH₂CH₃), 5.01(1H, d, J = 11 Hz, olefinic H), 5.03 (1H, d, J = 17 Hz, olefinic H), 5.83 (1H, ddd, J=6, 11, 17 Hz, olefinic H), 5.88 (1H, s, C_4 -H), 7.1—7.4 (5H, m, ArH). HRMS m/z: M⁺ Calcd for C₂₁H₂₅NO₅ 371.1733. Found: 371.1734. dl-(1R*,4S*,5S*,7S*)-4-Acetoxy-3-ethoxy-5-ethoxycarbonyl-1-phenyl-7-(E)-propenyl-2-azabicyclo[3.2.0]hept-2-ene (20b): 101 mg, 90%. Colorless gum. IR (CH₂Cl₂): 1740, 1720, 1640. H-NMR: 0.83 (3H, t, J=7 Hz, COOCH₂CH₃), 1.39 (3H, t, J=7 Hz, OCH₂CH₃), 1.57 (3H, d, J=7 Hz4 Hz, CH_3), 2.08 (3H, s, OAc), 2.36 (1H, dd, J=9, 12 Hz, C_6 -H), 2.73 (1H, dd, J=9, 12 Hz, C₆-H), 2.85—3.0 (1H, m, C₇-H), 3.73 (2H, q, J=7 Hz, $COOCH_2CH_3$), 4.40 (2H, q, J=7 Hz, OCH_2CH_3), 5.4—5.5 (2H, m, olefinic H), 5.86 (1H, s, C₄-H), 7.15—7.45 (5H, m, ArH). HRMS m/z: M⁺ Calcd for C22H22NO5 385.1890. Found: 385.1902.

dl-(1 R^* ,4 S^* ,5 S^* 7 S^*)-4-Acetoxy-3-ethoxy-5-ethoxycarbonyl-1-phenyl-7-(Z)-propenyl-2-azabicyclo[3.2.0]hept-2-ene (20c): 104 mg, 93%. Colorless gum. IR (CH₂Cl₂): 1750, 1720, 1650. ¹H-NMR: 0.85 (3H, t, J=7 Hz, COOCH₂CH₃), 1.39 (3H, t, J=7 Hz, OCH₂CH₃), 1.51 (3H, d, J=5 Hz, CH₃), 2.09 (3H, s, OAc), 2.43 (1H, dd, J=9.5, 13 Hz, C₆-H), 2.65 (1H, dd, J=8, 13 Hz, C₆-H), 3.1—3.0 (1H, m, C₇-H), 3.83 (1H, q, J=7 Hz, COOCH₂CH₃), 4.42 (2H, q, J=7 Hz, OCH₂CH₃), 5.3—5.6 (2H, m, olefinic H), 5.90 (1H, s, C₄-H), 7.15—7.4 (5H, m, ArH). HRMS m/z: M⁺ Calcd for C₂₂H₂₇NO₅ 385.1889. Found: 385.1920.

Pyrolysis of 4-Acetoxy Imidates (20) (General Procedure) A solution of 20 in toluene (5 ml) was heated in a sealed tube. After evaporation of the solvent, the residue was chromatographed with benzene as an eluent to give a mixture of 21 and 22, which was separated by preparative MPLC eluting with AcOEt-hexane (1:3).

1) Pyrolysis of **20a** (50 mg) at 180 °C for 2 h gave **21a** (21 mg, 42%) and 22a (18 mg, 36%). dl-(1R*,4R*,6S*,7R*)-7-Acetoxy-1-ethoxy-4ethoxycarbonyl-3-phenyl-6-vinyl-2-azabicyclo[2.2.1]hept-2-ene (21a): Colorless gum. IR (CH₂Cl₂): 1745, 1725, 1650. UV: 249 (10300). ¹H-NMR: 1.07 (3H, t, J=7 Hz, COOCH₂CH₃), 1.29 (3H, t, J=7 Hz, OCH_2CH_3), 2.01 (1H, dd, J=10, 13Hz, C_5 endo-H), 2.03 (3H, s, OAc), 2.29 (1H, dd, J=5, 13 Hz, $C_5 exo-H$), 2.5—2.7 (1H, m, $C_6 endo-H$), 3.55—4.46 (2H, m, COOC \underline{H}_2 C \underline{H}_3), 4.07 (2H, q, J=7Hz, OC \underline{H}_2 C \underline{H}_3), 5.18—5.35 (1H, m, olefinic H), 5.26 (1H, s, C_7 -H), 5.85—6.20 (2H, m, olefinic H), 7.2—7.7 (5H, m, ArH). HRMS m/z: M^+ Calcd for $C_{21}H_{25}NO_5$ 371.1731. Found: 371.1723. $dl-(1R^*,4R^*,6R^*,7R^*)-7$ -Acetoxy-1-ethoxy-4-ethoxycarbonyl-3-phenyl-6-vinyl-2-azabicyclo[2.2.1]hept-2-ene (22a): Colorless gum. IR (CH₂Cl₂): 1745, 1730. UV: 249 (12000). ¹H-NMR: 1.09 (3H, t, J=7 Hz, COOCH₂CH₃), 1.29 (3H, t, J=7 Hz, OCH_2CH_3), 1.56 (1H, dd, J=4, 13 Hz, C_5 endo-H), 2.04 (3H, s, OAc), 2.74 (1H, dd, J = 10, 13 Hz, $C_5 exo-H$), 2.9—3.2 (1H, m, $C_6 exo-H$), 4.06 (2H, q, J=7 Hz, COOC \underline{H}_2 CH₃), 4.09 (2H, q, J=7 Hz, OC \underline{H}_2 CH₃), 4.96—5.16 (1H, m, olefinic H), 5.22 (1H, s, C₇-H), 5.4—6.0 (2H, m, olefinic H), 7.2— 7.8 (5H, m, ArH). HRMS m/z: M⁺ Calcd for $C_{21}H_{25}NO_5$ 371.1733. Found: 371.1745.

2) Pyrolysis of 20b (50 mg) at 180 °C for 1 h gave 21b (20 mg, 40%) and 22b (11.5 mg, 23%). dl-(1R*,4R*,6S*,7R*)-7-Acetoxy-1-ethoxy-4ethoxycarbonyl-3-phenyl-6-(E)-propenyl-2-azabicyclo[2.2.1]hept-2-ene (21b): Colorless gum. IR (CH₂Cl₂): 1750, 1730. UV: 248 (13700). ¹H-NMR: 1.07 (3H, t, J=7 Hz, COOCH₂CH₃), 1.29 (3H, t, J=7 Hz, OCH_2CH_3), 1.73 (3H, d, J=5 Hz, CH_3), 2.04 (3H, s, OAc), 2.05 (1H, dd, J=9, 13 Hz, C₅endo-H), 2.24 (1H, dd, J=5, 13 Hz, C₅exo-H), 2.4—2.6 (1H, m, C_6 endo-H), 4.07 (4H, q, J=7 Hz, $COOC_{\frac{1}{2}}CH_3$, OCH₂CH₃), 5.26 (1H, s, C₇-H), 5.42—5.62 (2H, m, olefinic H), 7.22—7.61 (5H, m, ArH). HRMS m/z: M⁺ Calcd for C₂₂H₂₇NO₅ 385.1888. Found: 385.1908. dl- $(1R^*,4R^*,6R^*,7R^*)$ -7-Acetoxy-1-ethoxy-4-ethoxycarbonyl-3-phenyl-6-(E)-propenyl-2-azabicyclo[2.2.1]hept-2-ene (22b): Colorless gum. IR (CH₂Cl₂): 1750, 1730. UV: 248 (11400). ¹H-NMR: 1.08 (3H, t, J=7 Hz, COOCH₂CH₃), 1.28 (3H, t, J=7 Hz, OCH₂CH₃), 1.48 (1H, dd, J=4, 13 Hz, C₃endo-H), 1.60 (3H, d, J=7 Hz, CH₃), 2.03 (3H, s, OAc), 2.73 (1H, dd, J=10, 13 Hz, C_5exo-H), 2.9—3.2 (1H, m, C_6exo-H), 3.99 (2H, q, J=7 Hz, COOC \underline{H}_2 CH₃), 4.08 (2H, q, J=7 Hz, OC \underline{H}_2 CH₃), 5.19 (1H, s, C₇-H), 5.2—5.6 (2H, m, olefinic H), 7.2—7.7 (5H, m, ArH). HRMS m/z: M⁺ Calcd for C₂₂H₂₇NO₅ 385.1888. Found: 385.1888.

3) Pyrolysis of 20c (50 mg) at 180 °C for 2 h gave 21c and 22c (38

mg, 78%) as a 1:1 mixture. An attempt to separate the mixture by preparative TLC failed. dl-(1R*,4R*,6S*,7R*)-7-Acetoxy-1-ethoxy-4-ethoxycarbonyl-3-phenyl-6-(Z)-propenyl-2-azabicyclo[2.2.1]hept-2-ene (21c) and dl-(1R*,4R*,6R*,7R*)-7-Acetoxy-1-ethoxy-4-ethoxycarbonyl-3-phenyl-6-(Z)-propenyl-2-azabicyclo[2.2.1]hept-2-ene (22c): The spectral data of the mixture: IR (CH₂Cl₂): 1735, 1720. UV: 248 (11500). ¹H-NMR: 1.07 (3H, t, J=7 Hz, COOCH₂CH₃), 1.09 (3H, t, J=7 Hz, COOCH₂CH₃), 1.27 (3H, t, J=7 Hz, OCH₂CH₃), 1.28 (3H, t, J=7 Hz, OCH₂CH₃), 1.45 (1H, m, C₆-H), 1.63 (3H, d, J=5 Hz, CH₃), 1.68 (3H, d, J=7 Hz, CH₃), 2.03 (3H, s, OAc), 2.04 (3H, s, OAc), 2.11 (1H, m, C₆-H), 2.6—3.0 (3H, m, C₆ and C₇-H), 3.4 (1H, m, C₇-H), 4.0 (4H, m, 2 × OCH₂CH₃), 4.07 (2H, q, J=7 Hz, COOCH₂CH₃), 4.08 (2H, q, J=7 Hz, COOCH₂CH₃), 5.1—5.8 (4H, m, olefinic H), 5.24 (1H, s, C₄-H), 5.30 (1H, s, C₄-H), 7.26—7.80 (10H, m, 2 × Ar).

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