

## 2-Lithio-2-trimethylsilyl-1-silacyclopentane as a Synthetic Equivalent of 1-Lithio-1,4-butanediol

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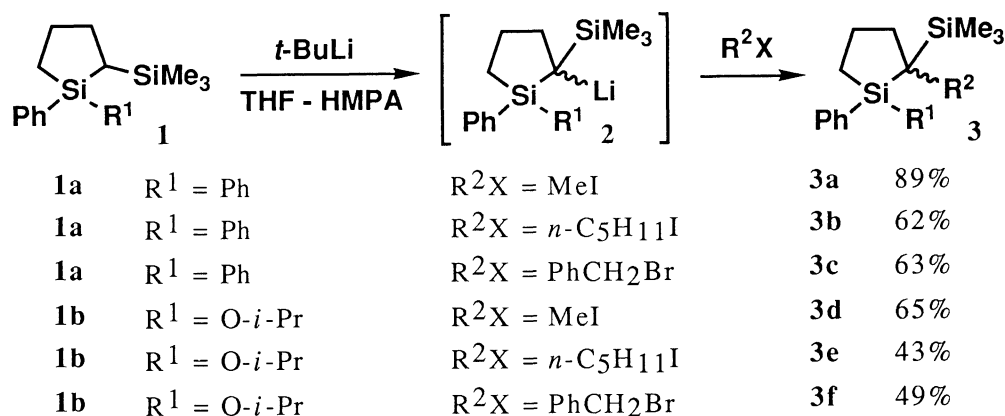
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Addition of alkyl iodide to 2-lithio-2-trimethylsilyl-1-silacyclopentane provided 2-alkyl-2-trimethylsilyl-1-silacyclopentane which was converted into 1-alkyl-1,4-butanediol by treatment with  $\text{H}_2\text{O}_2$ -KF and  $n\text{-Bu}_4\text{NF}$ .

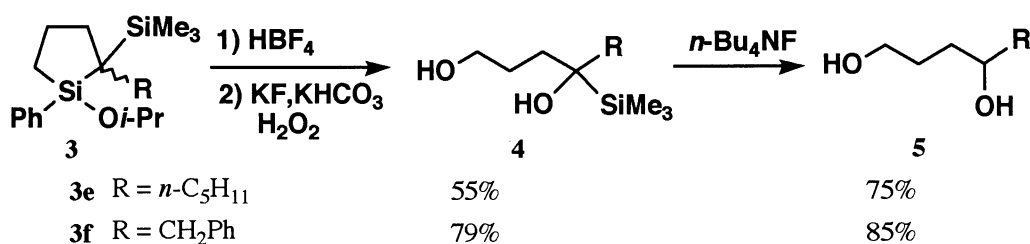
$\alpha$ -Metallated organosilanes play a fundamental role in preparative organosilicon chemistry.<sup>1)</sup> A wide range of variously functionalized organosilanes has been deprotonated  $\alpha$  to silicon. Among them, bis(trimethylsilyl)methylolithium<sup>2)</sup> has been widely used for carbon-carbon bond formation. Here we wish to report a preparation of 2-lithio-2-trimethylsilyl-1-silacyclopentane and its use for organic synthesis.

*tert*-Butyllithium (1.66 mol  $\text{dm}^{-3}$  pentane solution, 0.72 ml, 1.2 mmol) was added to a solution of 1,1-diphenyl-2-trimethylsilyl-1-silacyclopentane **1a** (0.31 g, 1.0 mmol) in THF (3.0 ml) and HMPA (0.4 ml) at  $-78^\circ\text{C}$ . The solution immediately turned dark red. After stirring for 30 min at  $-78^\circ\text{C}$ , iodomethane (0.19 ml, 3.0 mmol) was added and the resulting mixture was warmed to room temperature over 2 h. The mixture was poured into ice-cooled water and extracted with ethyl acetate (10 ml x 3). The combined organic layers were dried over  $\text{Na}_2\text{SO}_4$  and concentrated. The residual oil was submitted to silica-gel column chromatography to give 1,1-diphenyl-2-methyl-2-trimethylsilyl-1-silacyclopentane **3a**<sup>3)</sup> (0.29 g) in 89% yield. The representative results are shown in Scheme 1. 1-Isopropoxy-1-phenyl-2-trimethylsilyl-1-silacyclopentane **1b** also afforded 2-lithio-2-trimethylsilyl-1-silacyclopentane **2b** (orange solution) which provided the corresponding alkylated silacyclopentane **3d**, **3e**, and **3f** as 1:1 stereoisomeric mixtures upon treatment with alkyl halides.

Oxidation of the silacyclopentane **3a** with  $\text{HBF}_4$  followed by  $\text{H}_2\text{O}_2$ -KF- $\text{KHCO}_3$ <sup>4)</sup> gave the expected diol **4a** only in 20% yield along with unidentified complex mixture. The oxidative cleavage of two carbon-silicon

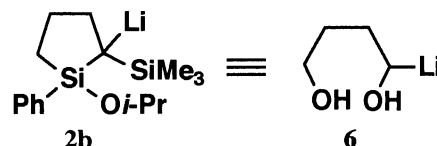


Scheme 1.

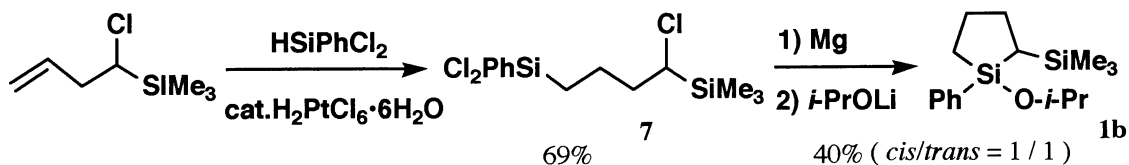


Scheme 2.

bonds of **3e** or **3f**, however, was easily performed by treatment with  $\text{HBF}_4$  followed by  $\text{H}_2\text{O}_2\text{-KF-KHCO}_3$  to provide the corresponding silyl alcohol **4e** or **4f**. Protodesilylation of **4** with  $n\text{-Bu}_4\text{NF}$  in  $\text{DMF}^{5)}$  gave 1,4-butanediol **5** (Scheme 2). Therefore, 1-isopropoxy-2-lithio-1-phenyl-2-trimethylsilyl-1-silacyclopentane **2b** can be regarded as a synthon of 1-lithio-1,4-butanediol **6**.



1,1-Diphenyl-2-trimethylsilyl-1-silacyclopentane **1a** was generated (45%) starting from 1,1-diphenyl-1-silacyclobutane according to the reported procedure ( $\text{ICH}_2\text{SiMe}_3$ , LDA).<sup>6)</sup> Meantime, 1-isopropoxy-1-phenyl-2-trimethylsilyl-1-silacyclopentane **1b** was prepared as follows (Scheme 3). Hydrosilylation of 4-chloro-4-trimethylsilyl-1-butene with  $\text{HSiPhCl}_2$  in the presence of  $\text{H}_2\text{PtCl}_6$  gave **7** which was treated with Mg followed by  $i\text{-PrOLi}$  to afford **1b** as an isomeric mixture ( $\text{cis/trans} = 1:1$ ).<sup>7)</sup>



Scheme 3.

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## References

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- 2) B.-T. Grobel and D. Seebach, *Angew. Chem., Int. Ed. Engl.*, **13**, 83 (1974).
- 3) **3a**: Bp 112-114 °C (0.5 Torr, 1 Torr = 133.322 Pa, bath temp); IR (neat) 3066, 2944, 2856, 1428, 1247, 1109, 857, 833, 735, 699, 658  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  -0.21 (s, 9H), 0.97 (s, 3H), 1.15-1.16 (m, 3H), 1.62-2.25 (m, 3H), 7.25-7.49 (m, 6H), 7.50-7.60 (m, 2H), 7.61-7.75 (m, 2H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  -2.35, 9.71, 14.19, 19.26, 22.29, 38.94, 127.52, 127.72, 128.90, 129.07, 135.03, 135.43, 135.56, 135.75. Found: C, 74.30; H, 8.69%. Calcd for  $\text{C}_{20}\text{H}_{28}\text{Si}_2$ : C, 74.00; H, 8.69%.
- 4) K. Tamao, N. Ishida, and M. Kumada, *J. Org. Chem.*, **48**, 2120 (1983).
- 5) M. Nakada, Y. Urano, S. Kobayashi, and M. Ohno, *J. Am. Chem. Soc.*, **110**, 4826 (1988).
- 6) K. Matsumoto, K. Oshima, and K. Utimoto, *Tetrahedron Lett.*, **31**, 6055 (1990).
- 7) Isomerically pure **1b** (*cis* or *trans*) as well as a 1:1 mixture of **1b** provided the same 1:1 isomeric mixtures **3d**, **3e**, and **3f**. Thus, a *cis/trans* mixture was used without separation.

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