

A PRACTICAL ENANTIOSELECTIVE SYNTHESIS OF 12-HYDROXYEICOSATETRAENOIC ACIDS

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Abstract

A practical enantioselective synthesis of 12(R)-HETE has been achieved from readily available starting materials. This process also constitutes a formal synthesis of Leukotriene B₄.

12-HETE is generated from arachidonic acid by a 12-lipoxygenase enzyme, found in platelets and skin keratinocytes. The 12(R)-HETE is of particular interest since it has been detected in high concentrations in psoriatic lesions and has been shown to possess potent human neutrophil chemotactic and chemokinetic properties.¹ As part of our program directed towards the synthesis of novel anti-inflammatory agents we required a versatile, enantioselective synthesis of 12(R)-HETE and its 12-substituted analogs for SAR studies.

Starting with the readily available 4-decynoic acid 1², we utilized Evans' innovation for the asymmetric oxidation of chiral imide enolates³ to introduce the critical 12(R) stereochemistry inherent in the natural product. Carboxylic acid, 1 was reduced to the corresponding (Z)-olefin in 95% yield as shown in Scheme I and conversion to the chiral oxazolidinone proceeded uneventfully using Evans' conditions. Metallation and quenching of the enolate derived from 2 with Davis' reagent⁴ afforded the α-hydroxy acyloxazolidinone 3, with a diastereoisomeric excess of ≥ 98:2 and chemical yield of 83%. tert-Butyldiphenylsilylation and subsequent reductive cleavage of the chiral auxiliary with lithium borohydride in the presence of excess cyclohexene gave enantiomerically pure alcohol which was

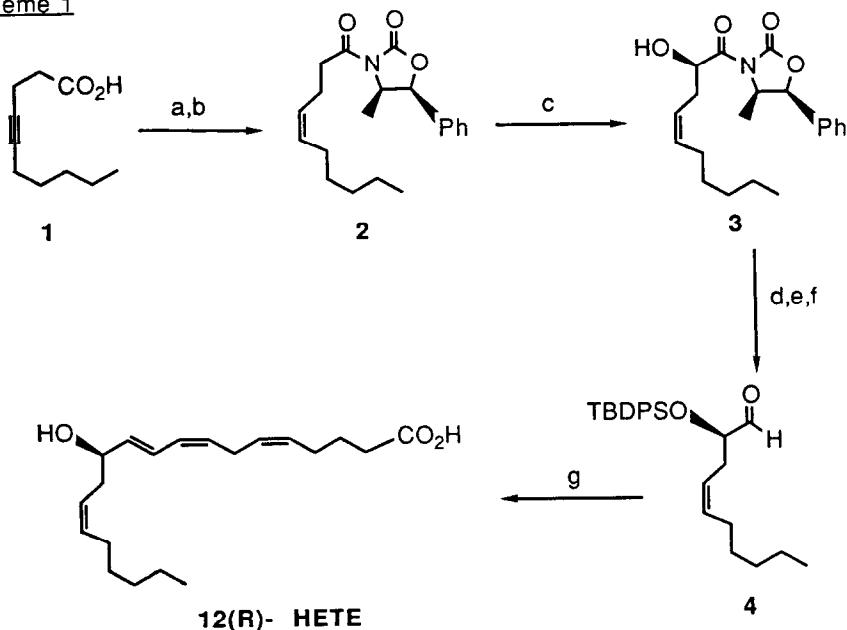
oxidized to the desired aldehyde 4⁵ using the Swern procedure. Significantly, the yield of the borohydride reduction was found to be nearly 20% lower if cyclohexene was omitted from the reaction mixture while lithium aluminum hydride reduction afforded only complex mixtures of products. The aldehyde was converted into 12(R)-HETE using conditions previously reported by the Merck-Frosst group.⁶ As the aldehyde 4 has also been converted into Leukotriene B₄⁷ the approach outlined in this Letter also constitutes a formal total synthesis of this natural product.

Summarily, we have completed a concise, total synthesis of 12(R)-HETE in 11 steps from readily available starting materials in 29% overall yield. The construction of the key intermediate 4 for 12-hydroxy-eicosatetraenoic acid synthesis proceeded in 7 steps and in 63% overall yield from 4-decynoic acid. The advantages of this procedure over previous syntheses of 12(R)-HETE⁸ are a) conciseness and high yield, b) versatility - this procedure provides for a general entry into the 12-HETE family and allows for the synthesis of previously unavailable analogs at the critical C-12 binding site with a high degree of stereogenic control.

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Scheme 1Reagents

a) nickel boride, H_2 , 98%, b) $(COCl)_2$, then $n-BuLi$ /
 (4R,5S)-(+)-4-Methyl-5-phenyl-2-oxazolidinone, THF, -78° to $-10^\circ C$, 99%,
 c) $NaHMDS$, Davis' reagent, THF, -78° to $0^\circ C$, 83%, d) t-butyldiphenylsilyl
 chloride, imidazole, DMF, 94%, e) $LiBH_4$, cyclohexene, Et_2O , $0^\circ C$, 87%,
 f) $(COCl)_2$, DMSO, Et_3N , 98%, g) see Ref. 6.

References

1. P.M. Dowd, A.K. Black, P.M. Woollard, R.D.R. Camp and M.W. Greaves, *J. Invest. Dermatol.*, 84 537 (1985) and T. Ruzika and G. Burg, *J. Invest. Dermatol.*, 88 120 (1987).
2. N.W. Gilman and B.C. Holland, *Chem. Phys. Lipids*, 13 239 (1974).
3. D.A. Evans, M.M. Morrissey and R.L. Dorow., *J. Amer. Chem. Soc.*, 107 4346 (1985).
4. F.A. Davis and O.D. Stringer, *J. Org. Chem.*, 47, 1774 (1982).
5. Aldehyde 4 had $[\alpha]_D^{25} = -15.6^\circ$ (c2.6, CHCl₃) Literature value⁶ $[\alpha]_D^{25} = -16.5$ (c3, CHCl₃)
6. Y. Leblanc, B.J. Fitzsimmons, J. Adams, F. Perea and J. Rokach, *J. Org. Chem.*, 31 789 (1986).
7. C-Q Han, D. Ditullio, Y-F Wang and C.J. Sih, *J. Org. Chem.*, 51 1253 (1980), C. Fuganti, S. Servi and C. Zirotti, *Tetrahedron Letters* 5285 (1983). Also, see K.C. Nicolaou, R. E. Zipkin, R.E. Dolle and B.D. Harris, *J. Amer. Chem. Soc.*, 106 3548 (1984).
8. P. Yadagiri, S. Lumin, P. Mosset, J. Capdevila and J.R. Falck, *Tetrahedron Letters* 6039 (1986) I. M. Taffer and R. E. Zipkin, *Tetrahedron Letters* 6543 (1987) and E. J. Corey, K. Kyler and N. Raju, *Tetrahedron Letters* 5115 (1984). The latter synthesis afforded (+)12-HETE.

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